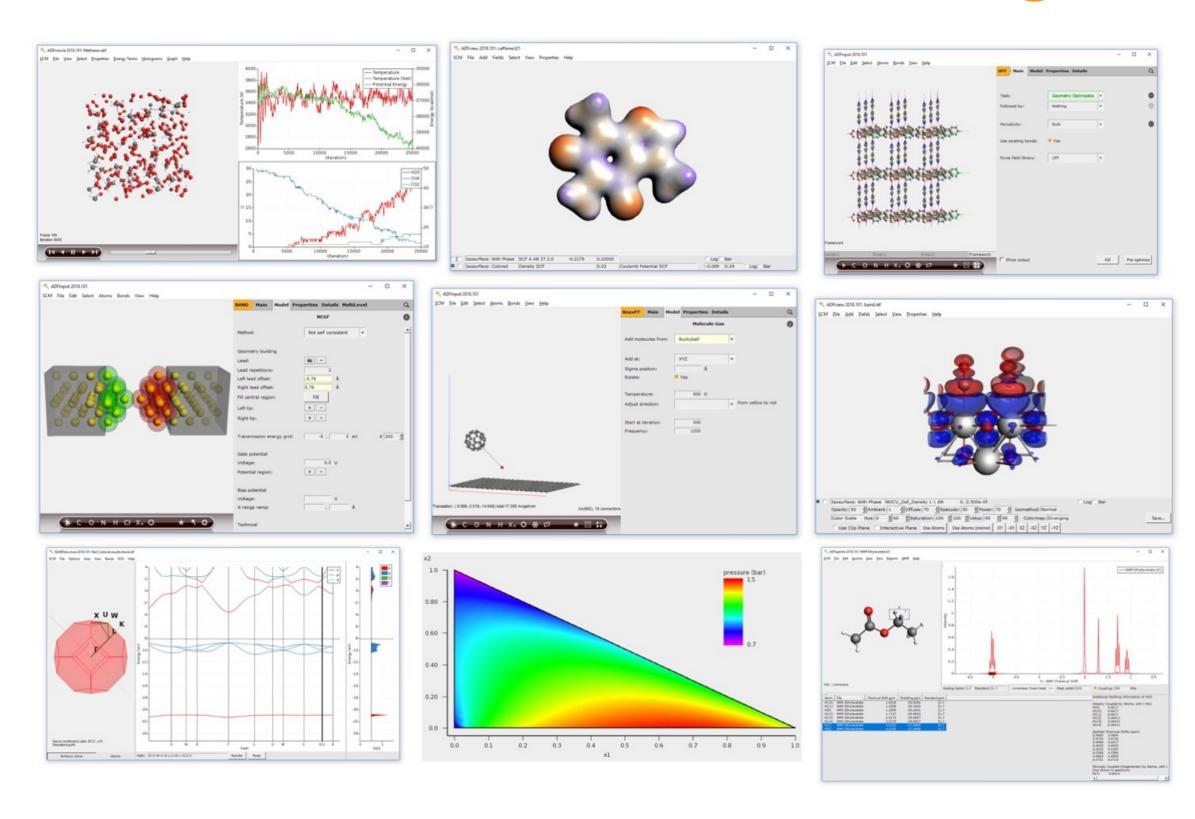
Hands-on workshop: Chemistry & Materials with the Amsterdam Modeling Suite





School of Chemistry, Glasgow University, hands-on workshop, 19 Jun 2023 Maria Jose Aliaga, aliaga@scm.com

SCM support: support@scm.com

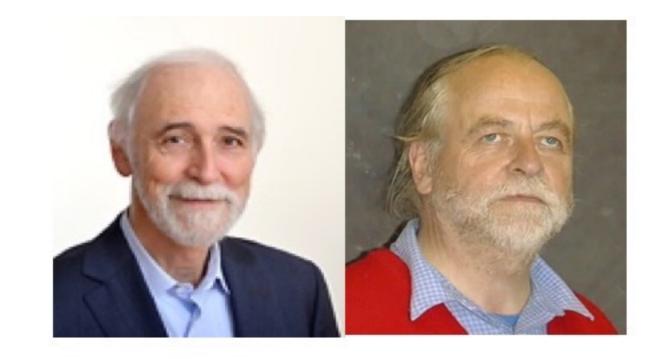
Program

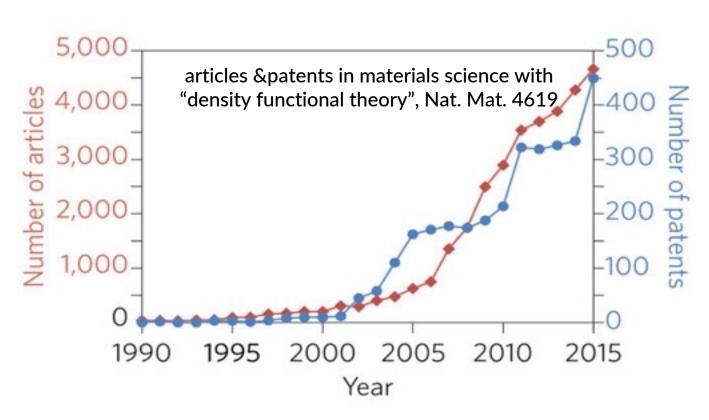
- Brief introduction Amsterdam Modeling Suite
- Hands on:
 - https://www.scm.com/about-us/news-agenda/adf-hands-on-workshops/
- Q&A
- NOTE: we'll be using AMS2023 (released in April)
- www.scm.com/doc2023
- www.scm.com/bin2023
- Slides: www.scm.com/news/19-june-2023-1-day-ams-workshop-in-glasgow/

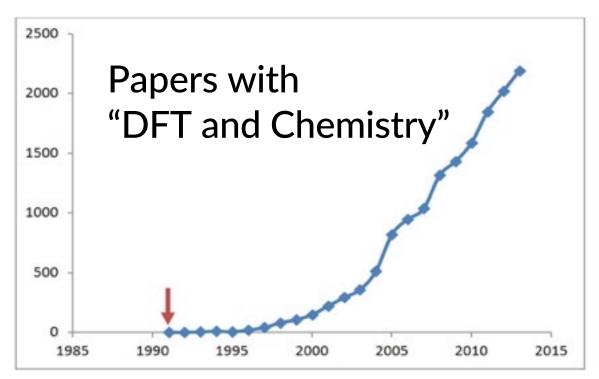


Background: SCM, ADF & AMS

- ADF = first DFT code for chemistry (1970s)
 Baerends@VU (>'73), Ziegler@Calgary⁽⁺⁾ (>'75)
 - o 80s: support industrial users Mitsui, Shell, Akzo, Unilever
- SCM: Spin-off company 1995
- 24 people (17 senior PhD's) + 3 EU fellows
- Many academic collaborators / EU networks
 - ~ 160 authors
 - New functionality
 - Academia, government & industry users worldwide
- SCM: development, debug, port, optimize, docs & <u>support</u>







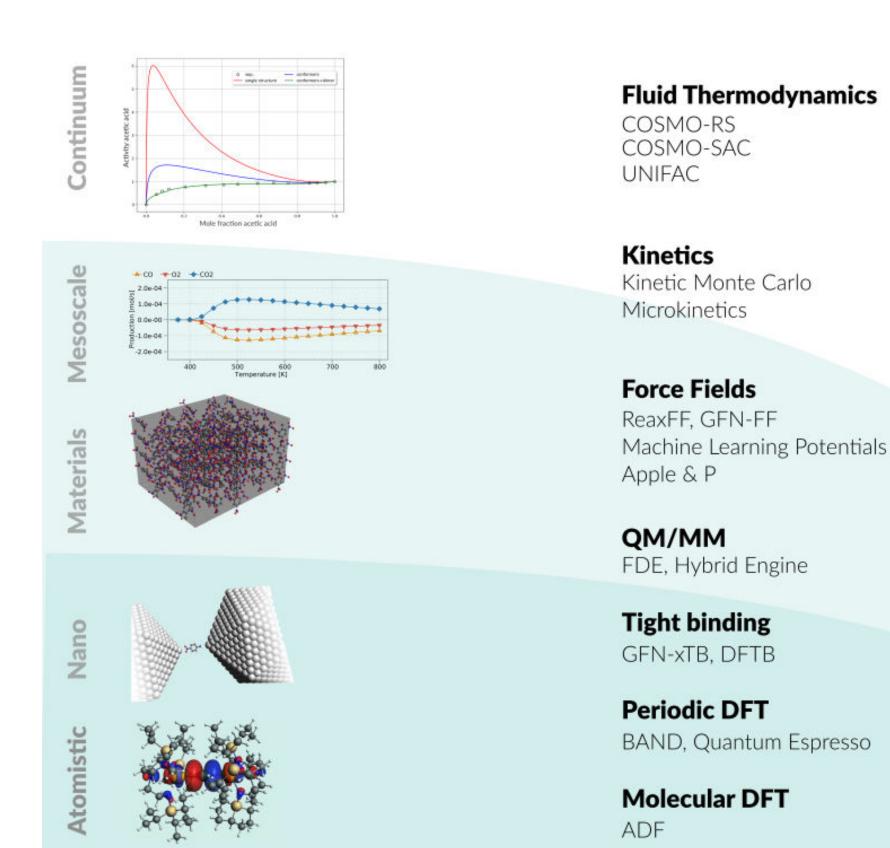




- 17 developers
- 4 support scientists
- 3 office / business
- 3 PhD students (EU)

Amsterdam Modeling Suite

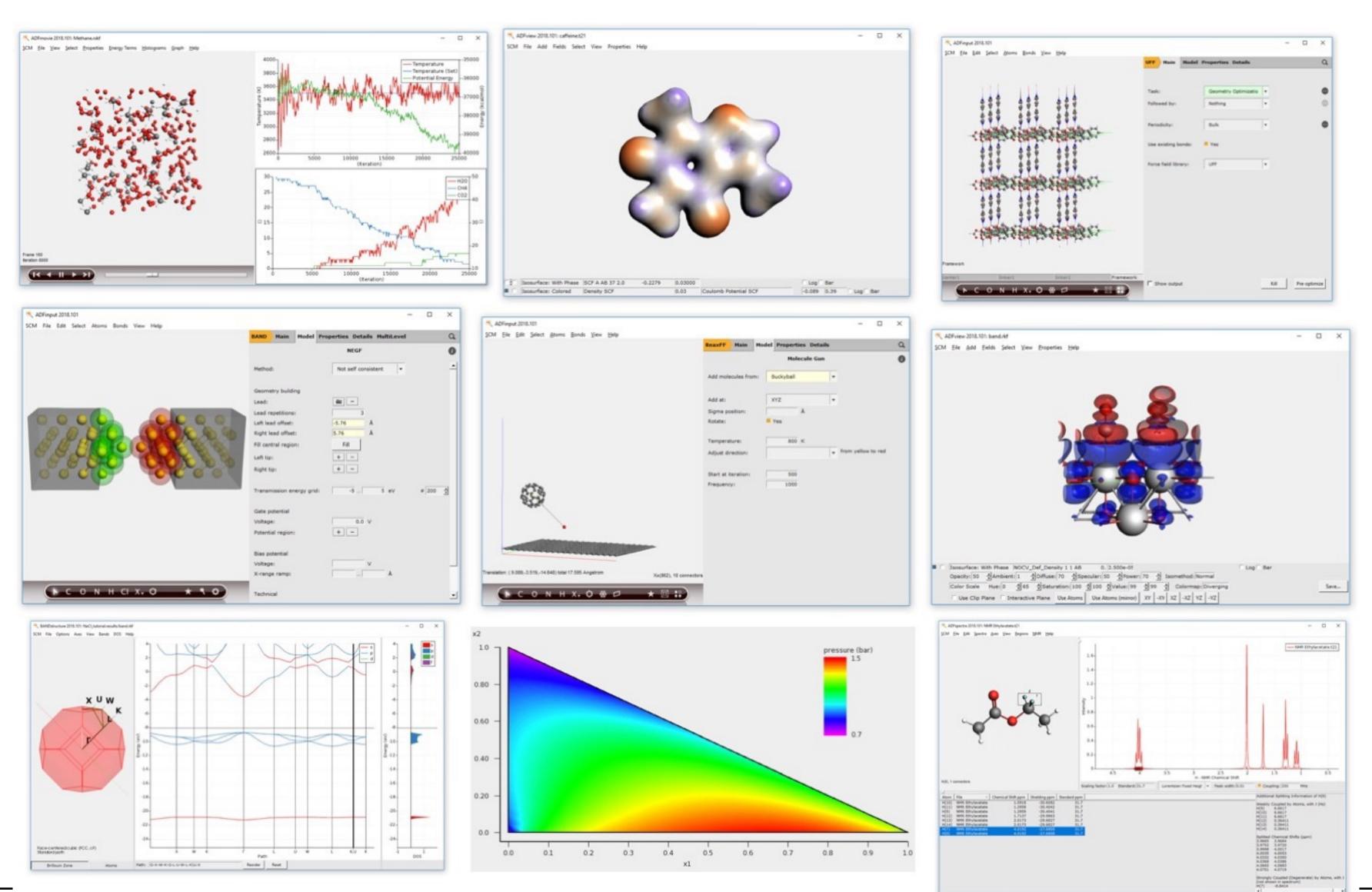
- ADF: powerful molecular DFT
 - Reactivity, spectroscopy
 - Spectroscopy: NMR, EPR, VCD, UV, XAS
- BAND: periodic DFT
 - (2D) Materials, spectroscopy, analysis
 - Interface with QE, VASP
- DFTB & MOPAC: fast electronic structure
- ReaxFF: Reactive MD
 - Dynamics of large complicated systems
- MLPotential
 - Several backends, ANI-2x, M3GNet
- COSMO-RS: fluid thermodynamics
 - VLE, LLE, logP, solubility
- AMSdriver: PES exploration, MD, MC
 - Hybrid: multi-layer, QM/MM, QM/QM'



Integrated GUI, python scripting (workflows), ParAMS: parametrize ReaxFF & xTB

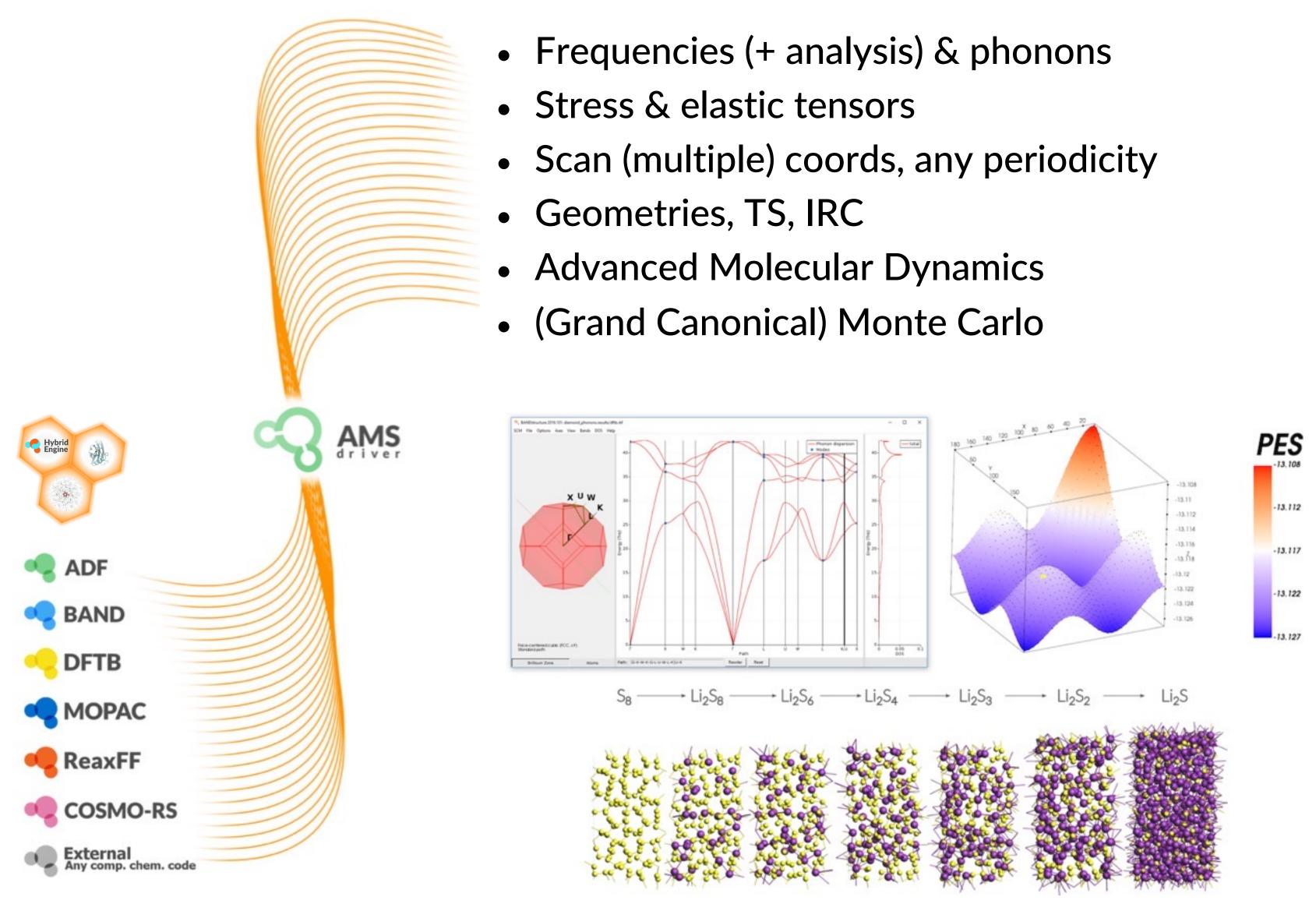


1 GUI: build, run & analyze



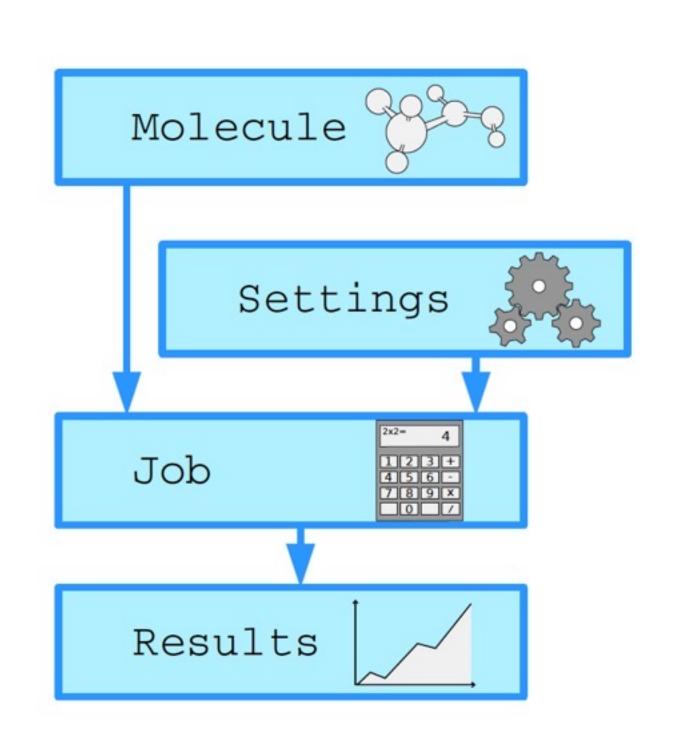


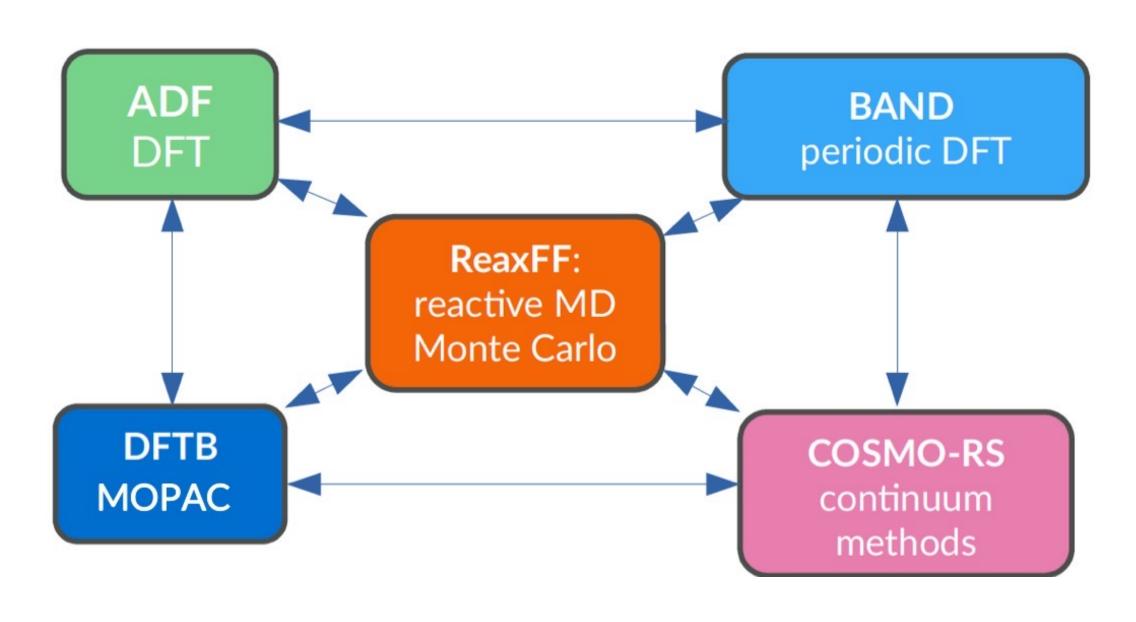
The AMS driver: decouple from Engine





PLAMS: python scripting







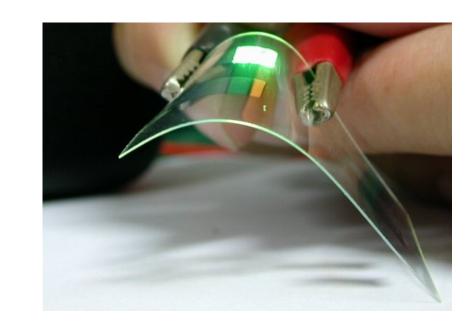
Links all modules + various tools

- → workflows & screening
- → (custom) post-processing
- → rapid prototyping

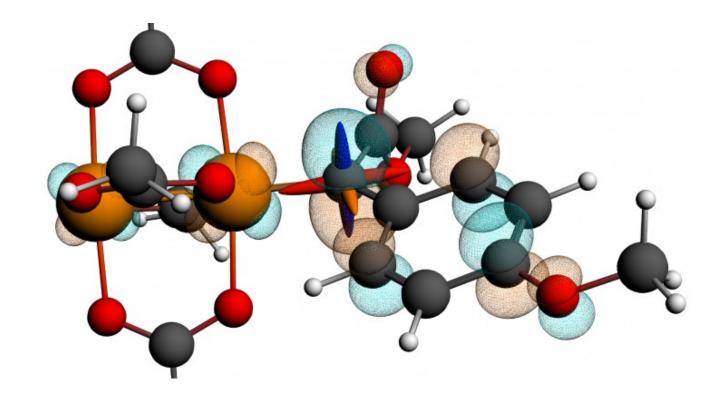


ADF: Molecular DFT





Organic electronics



NMR calculations locate ¹³C di-Rh carbene catalyst intermediate, <u>Science</u>, **342**, 351 (2013)

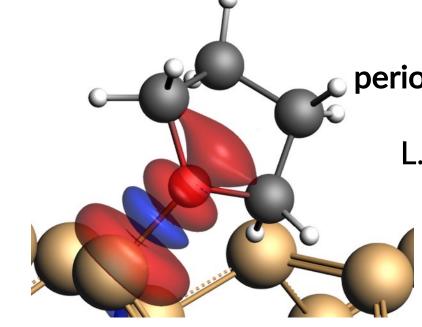
Strong & unique points

- All-electron Slaters, H-Og
- Relativity: ZORA (SR, SOC)
- Spectroscopy
 - EPR, NMR, IR (VCD), UVVIS, XAS
 - qsGW+BSE
 - Phosphorescence
- Bonding analysis:
 - Fragment-based approach
 - ETS-NOCV, QTAIM, MO diagrams, NCI,
 - Transfer integrals (charge mobility)
- Environments
 - Subsystem DFT (FDE), DIM/QM, QM/MM, QM/FQ, 3D-RISM, COSMO, SM12

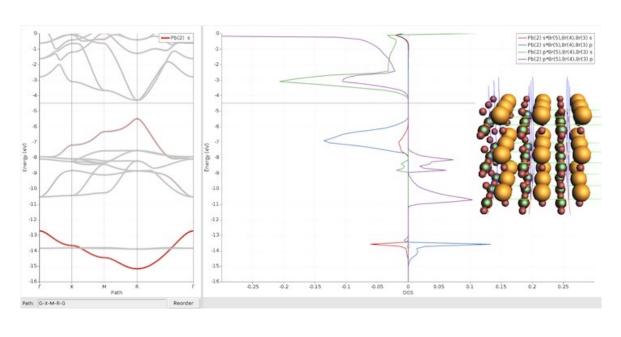


Periodic DFT: BAND vs Plane Waves

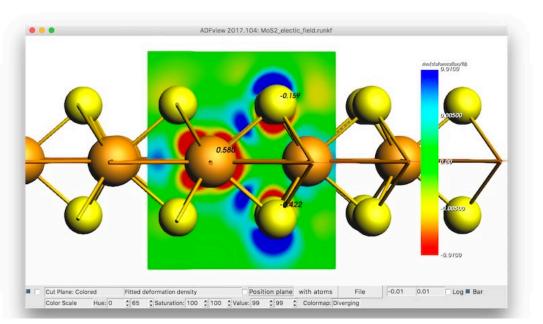
- Atom centered basis functions, STO or NAO
 - Compare cluster with periodic
 - No pseudopotentials, all elements
 - Core spectroscopy (core holes)
 - Easy orbital analysis: pDOS, COOP, EDA
 - Fast for empty (1D, 2D, porous)
 - o xc: r2SCAN, MN15-L, HSE06, GLLB-sc, D3(BJ), D4, DFT-1/2
 - Self-consistent NEGF
 - Gate & bias potential
 - Spin transport
- True 2D surfaces, 1D polymers
 - Het. catalysis: polarization, COSMO, SM12
 - 2D electronics (homogeneous E field)
 - Nanotubes
 - QM/MM and QM/QM' for 2D
- Integrated Graphical Interface:
 - Easy set up & analysis
 - Switch: ADF, BAND & Quantum Espresso, VASP



periodic energy decomposition analysis (tutorial) L. Pecher and R. Tonner WIREs CMS, (2018)



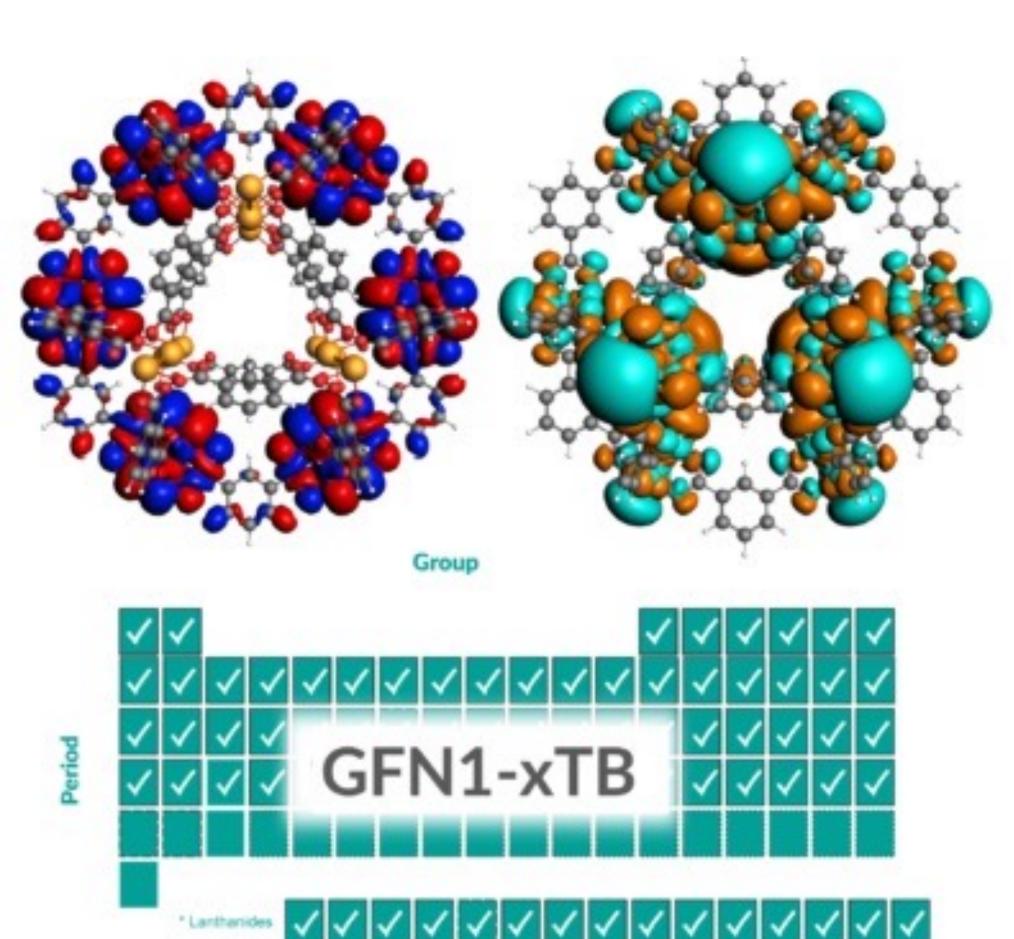
COOP in perovskites (tutorial)
Goesten & Hoffmann
JACS (2018)



Polarizing 2D semiconductor (tutorial)
N. Zibouche et al.
PCCP (2014)



DFTB: 'fast DFT' for molecules & periodic



Approximated DFT

- Nearest neighbor & minimal basis
- Tabulated elec & rep. parameters:
 - Grimme GFN-xTB (Z = 1-86)
 - QuasiNaNo & DFTB.org

Capabilities & Features

- UV/VIS (fast!)
- MOs, band structures, DOS

Through AMS

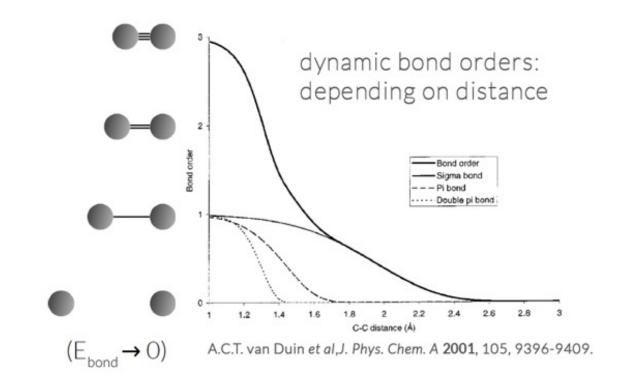
- Geometries, frequencies, phonons
- Stress tensors (optimize under p)
- Advanced MD, PES scans
- GCMC, molecule gun
- Multi-layer, QM/MM, QM/QM'
- Reparametrize xTB

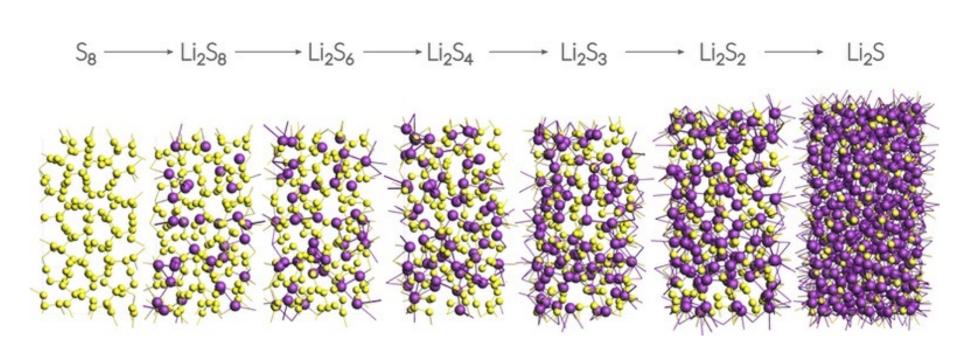


ReaxFF - reactive molecular dynamics

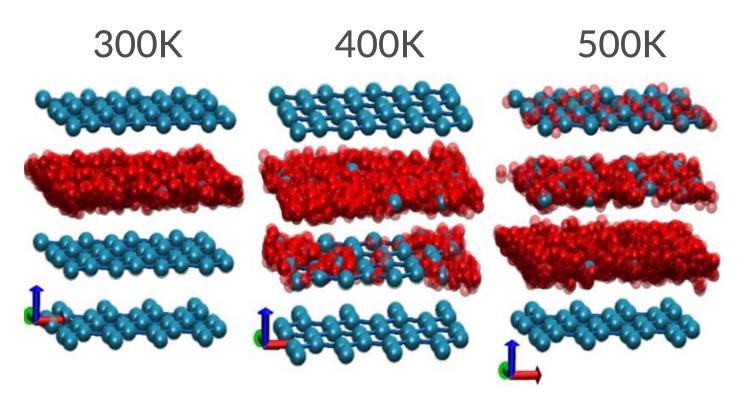
- No discontinuities in energy or forces
- No pre-defined reaction sites or types
 - Dynamic bond orders, charge equilibration
 - Only 1 atom type per element

$$\begin{split} E_{\text{system}} &= E_{\text{bond}} + E_{\text{lp}} + E_{\text{over}} + E_{\text{under}} + E_{\text{val}} + E_{\text{pen}} + \\ E_{\text{coa}} &+ E_{\text{C}_2} + E_{\text{tors}} + E_{\text{conj}} + E_{\text{H-bond}} + E_{\text{vdWaals}} + E_{\text{Coulomb}} \end{split}$$





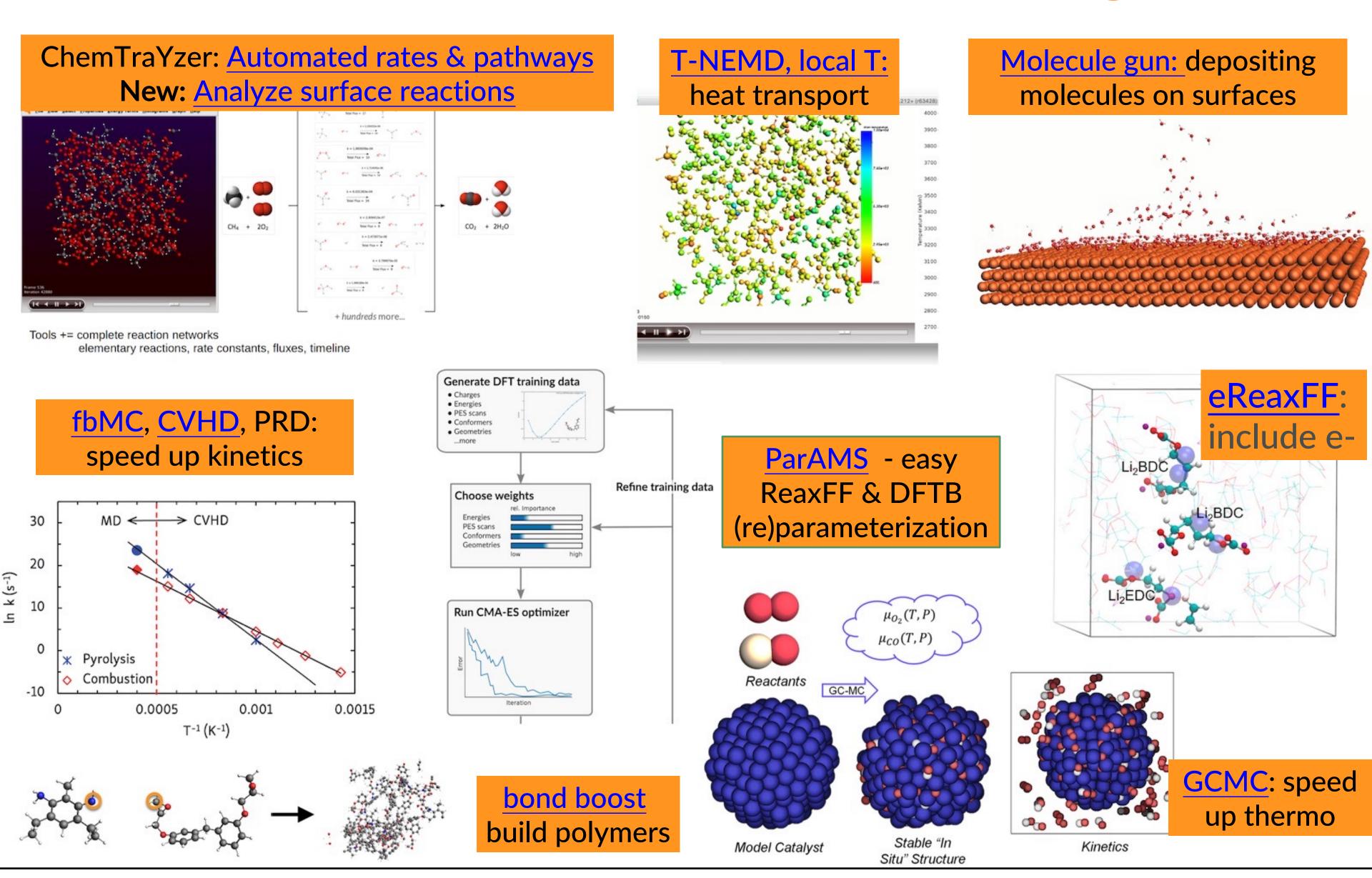
Li battery discharge: J. Electrochem. Soc. **161**, E3009 (2014); PCCP, **17**, 3383 (2015), tutorial



eReaxFF to study electron mobility & Li ion reduction, explicit electrons & electric field, J. Electrochem. Soc. 169,110540 (2022)



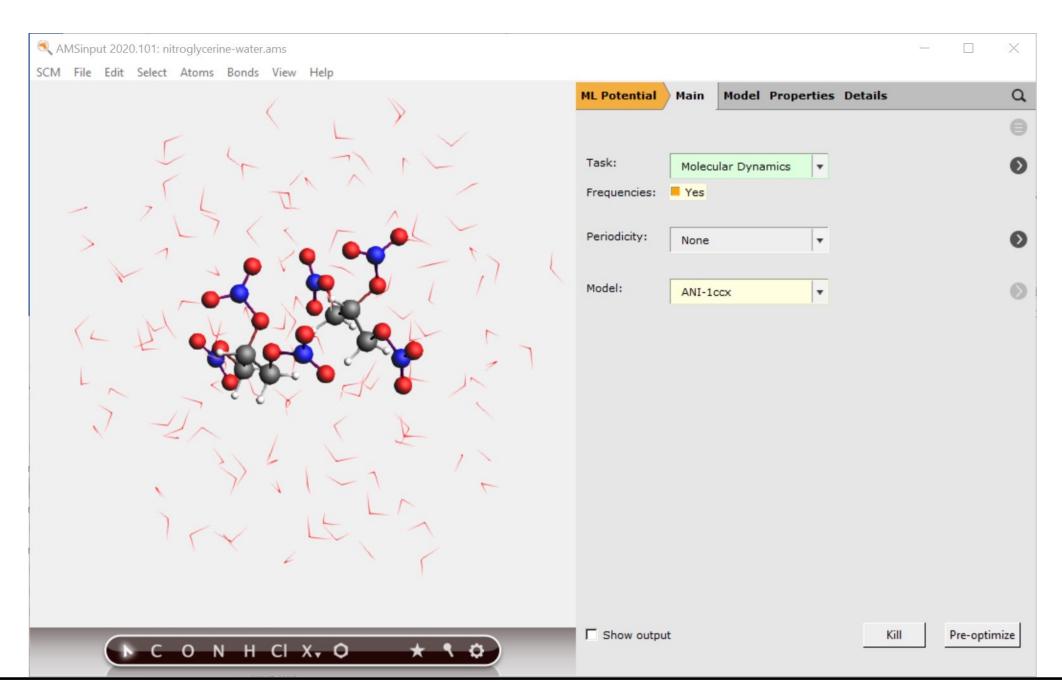
Reactive MD tools Amsterdam Modeling Suite





Machine Learned Potentials

- Use machine learning potentials with AMS driver
 - PES scans, conformers, MD, reaction energies, ...
 - Also with Hybrid engine (multi-layer)
- Automatically install popular ML Backends
 - SchNetPack, sGDML, PiNN, TorchANI
 - Pre-parametrized neural network potentials ANI-1ccx, ANI-2x, M3GNet-UP(2022)
 - CUDA-enabled PyTorch and Tensorflow can be used

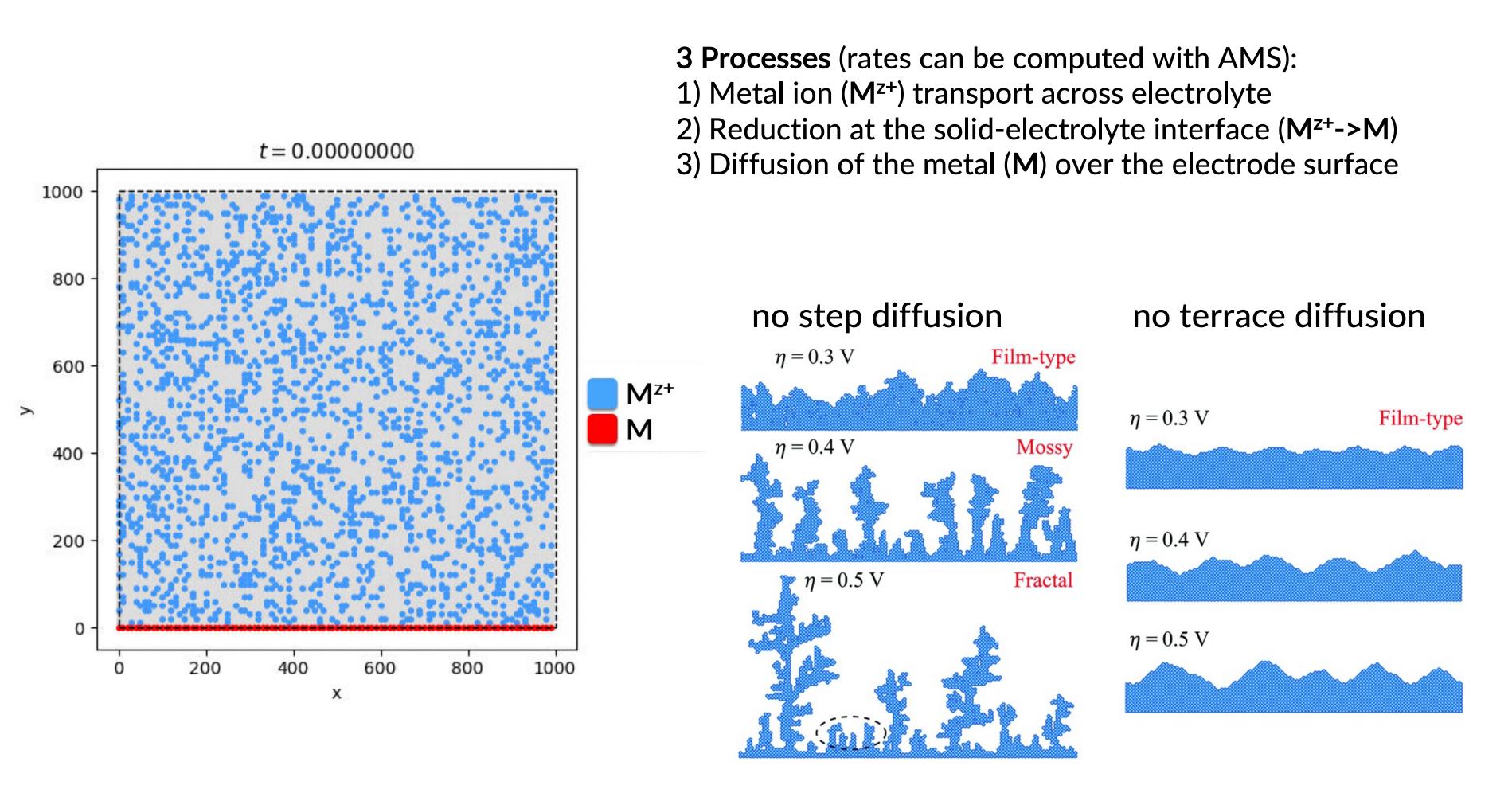


Demo video



Kinetic Monte Carlo: SEI dendrite formation

Modified (py)Zacros kMC to study dendrite formation



Inspired by Surface diffusion manifestation in electrodeposition of metal anodes, PCCP 2020 (22), 11286

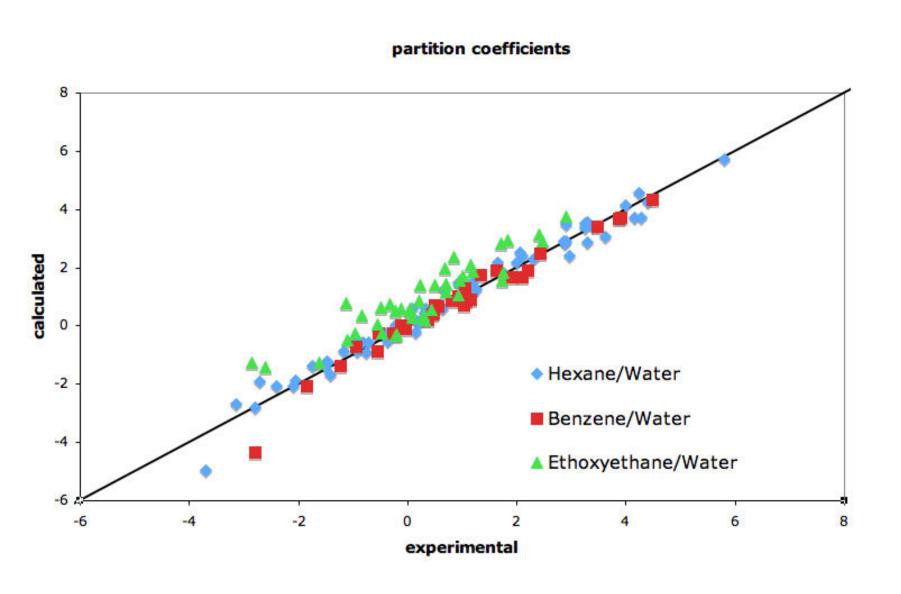


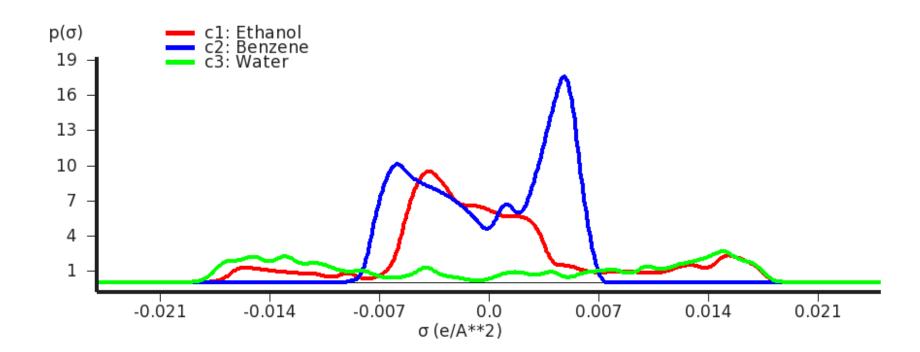
COSMO-RS/SAC: thermodynamic properties of fluids

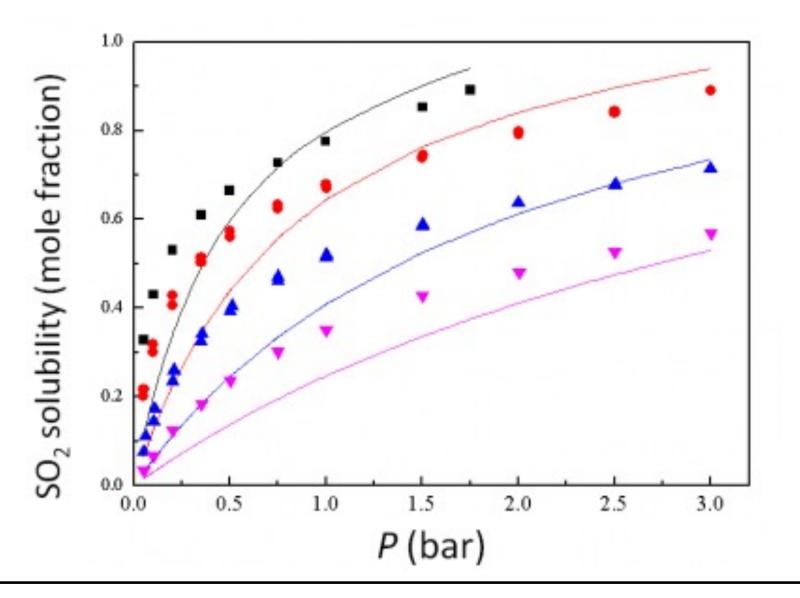
Quantum Chemistry & QSPR for quick property predictions

COntinuum Solvation MOdel + RS (Klamt), SAC (Sandler) chemical potential => activity coefficients => instantaneous properties

- Solvation & excess energies, pKa
- Solubilities, LLE, VLE, boiling points
- Optimize mixtures: solubility, LLE
- Polymers: Flory-Huggins X

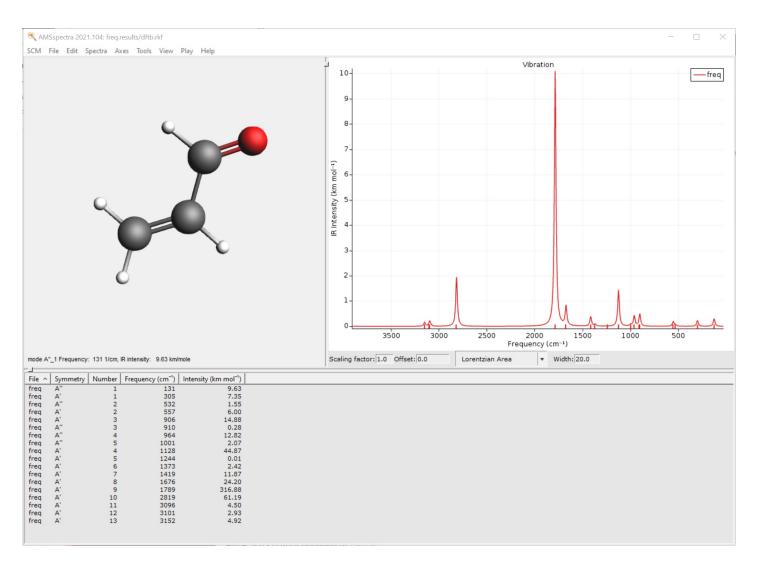


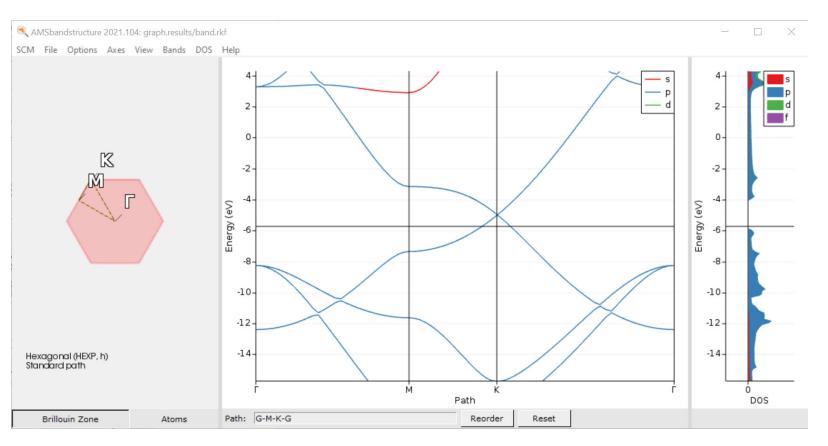


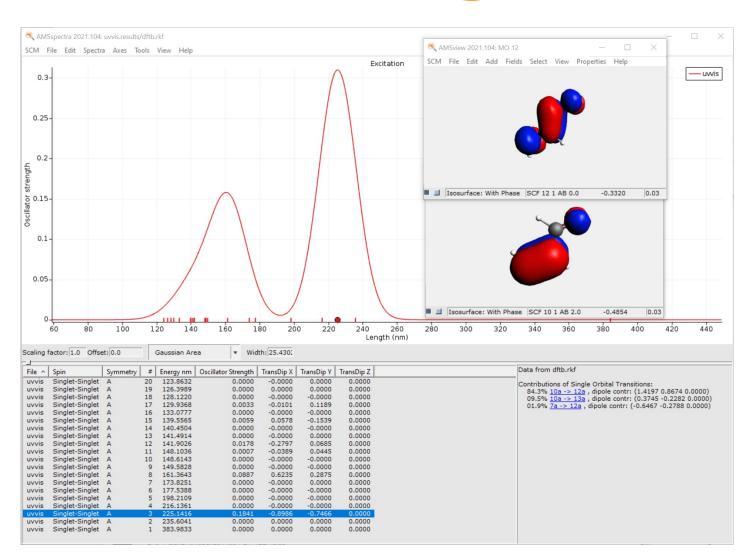


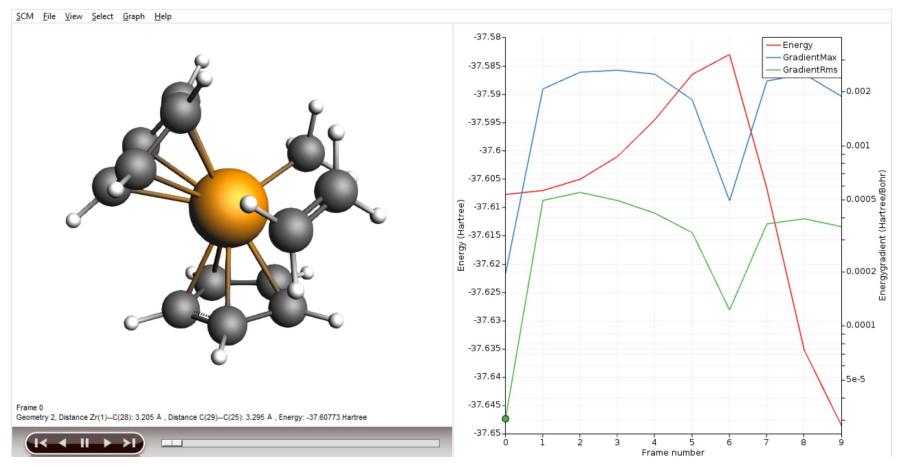


Hands on + self-learning









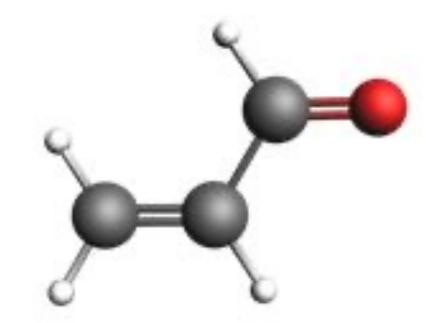


Building molecules, spectra

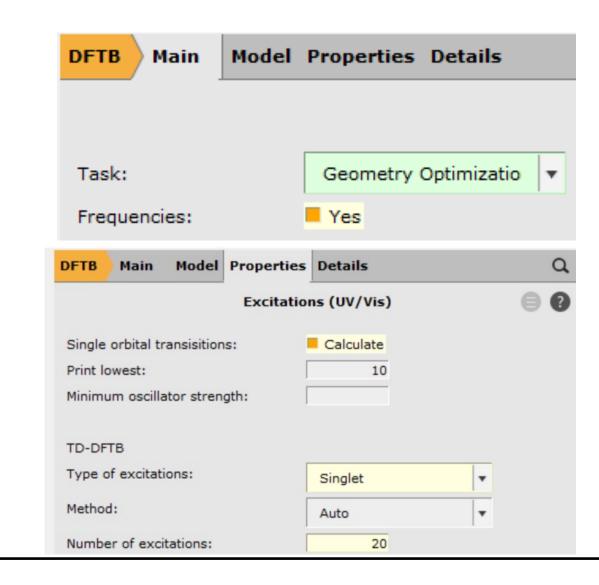
www.scm.com/doc/Tutorials/GUI_overview/Building_Molecules.html

- Search molecules
- Import: InChI, SMILES, xyz, cif, pdb, ...
- Included library + building





- IR frequencies (<u>tutorial</u>)
 - Geometry needs to be optimized at same level as frequency calc.
 - Try ADF, DFTB, MOPAC, MLPot, ...
 - SCM -> Spectra
- UV/VIS FAQ for tips + links to tutorials
 - ADF: (s)TDDFT, (s)TDA, TD-DFT+TB
 - Visualize MOs Single Orbital Transition (ADF: also NTOs)
 - Compare to NIST (IR), NIST (UV/VIS)





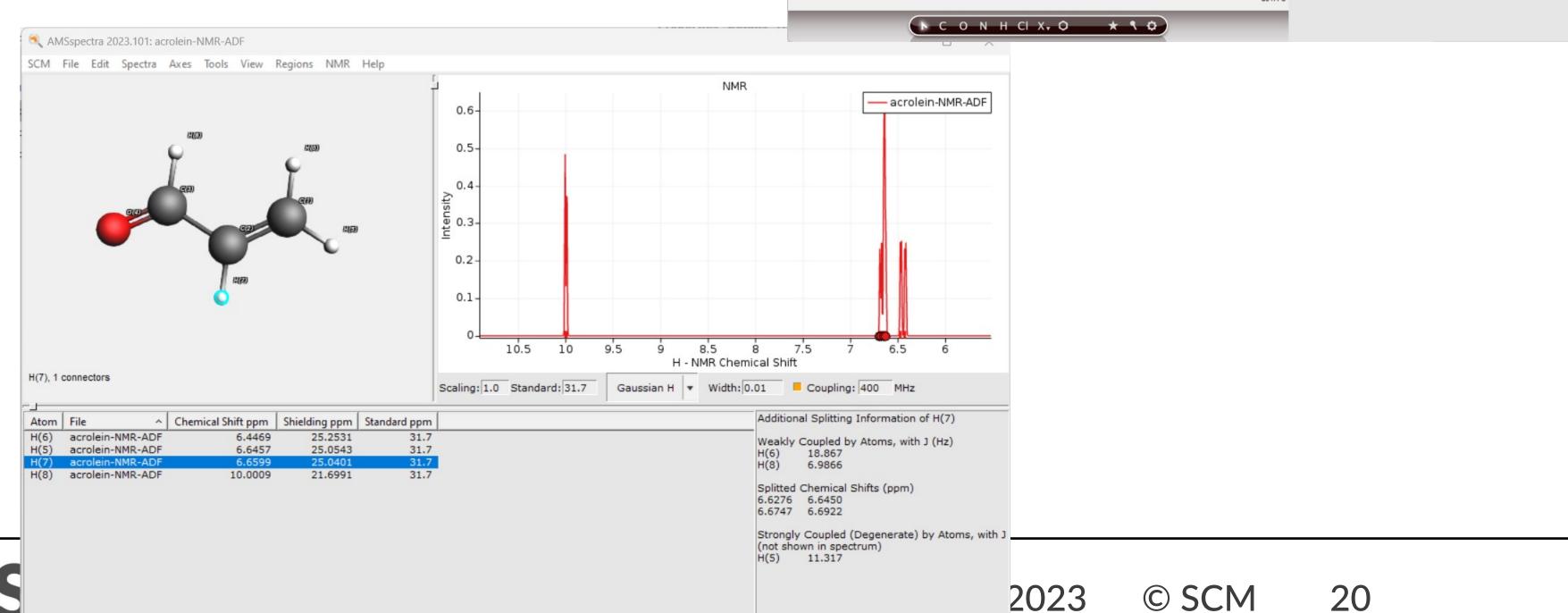
Spectra: NMR

AMSinput 2023.101: acrolein-NMR-ADF.ams

SCM File Edit Select Atoms Bonds View Help

• Exercise:

- See also NMR FAQ for tips + advanced tutorials
- Use PBEO + TZP, Scalar, no core
- Select Properties -> NMR
 - select shielding & coupling for all H Set 400 MHz and tick coupling, compare
- Note the internal ref. uses different settings!
 - Recalculate TMS with same set up





+ -+ -

Quick thermodynamic properties

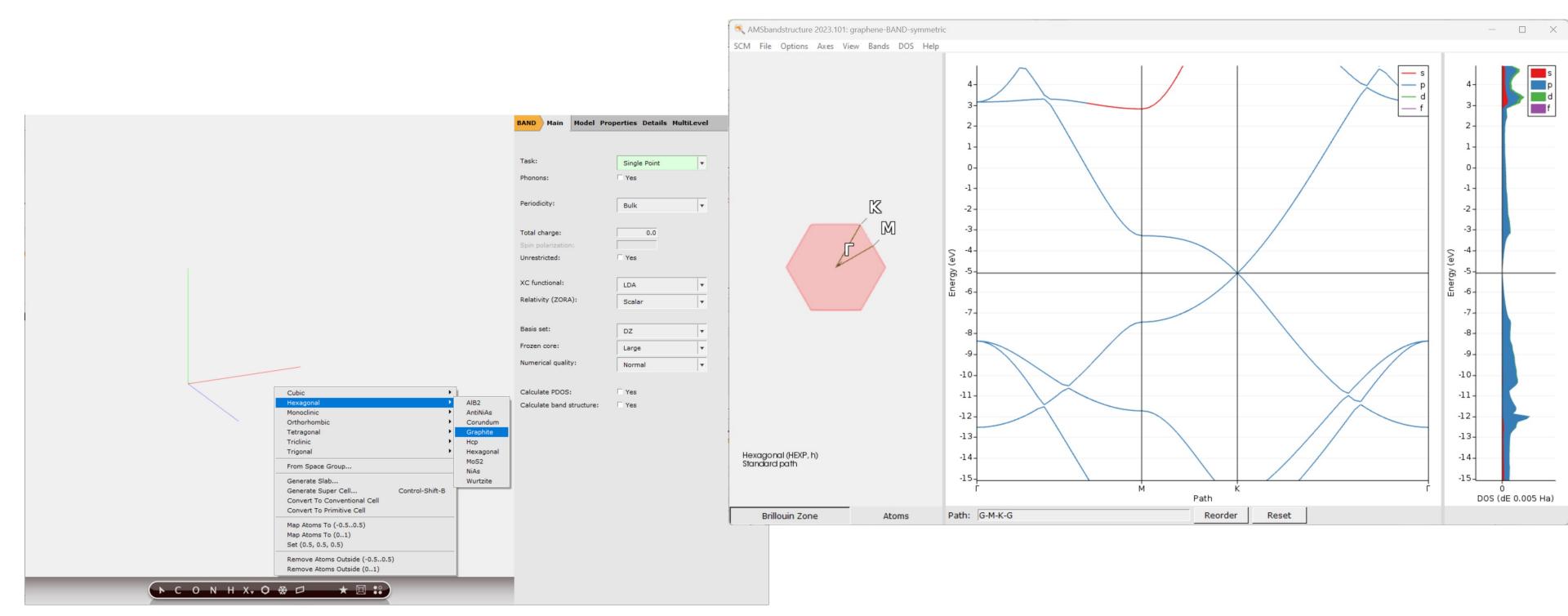
- Open SCM -> COSMO-RS (no need to install database)
- In the SMILES input, put C=CC=O and Add
- Properties -> Pure compound
- Compare some properties (density, boiling point, flash point) (e.g. Wikipedia)
- Try COSMO-RS Properties -> Boiling point pure compound

Property		Unit
Boiling point	346.746	K
Critical pressure	51.21	bar
Critical temperature	524.603	K
Critical volume	0.195	L/mol
Liquid density	0.816	kg/L
Dielectric constant	10.985	
Absolute entropy (ideal gas)	309.848	J/(mol K)
Flash point	270.523	K
Gibbs energy of formation (ideal gas)	-111.468	kJ/mol
Net enthalpy of combustion	-1652.692	kJ/mol
Std. state enthalpy of formation	-188.906	kJ/mol
Enthalpy of fusion	17.129	kJ/mol
Enthalpy of formation (ideal gas)	-157.71	kJ/mol
Enthalpy of sublimation	49.534	kJ/mol
Melting point	213.82	K
Liquid molar volume	0.069	L/mol
Parachor	161.085	
Solubility parameter	10.098	√(MPa)
Triple point temperature	213.778	K
Van der Waals area	91.576	Ų
Van der Waals volume	63.229	\mathring{A}^3



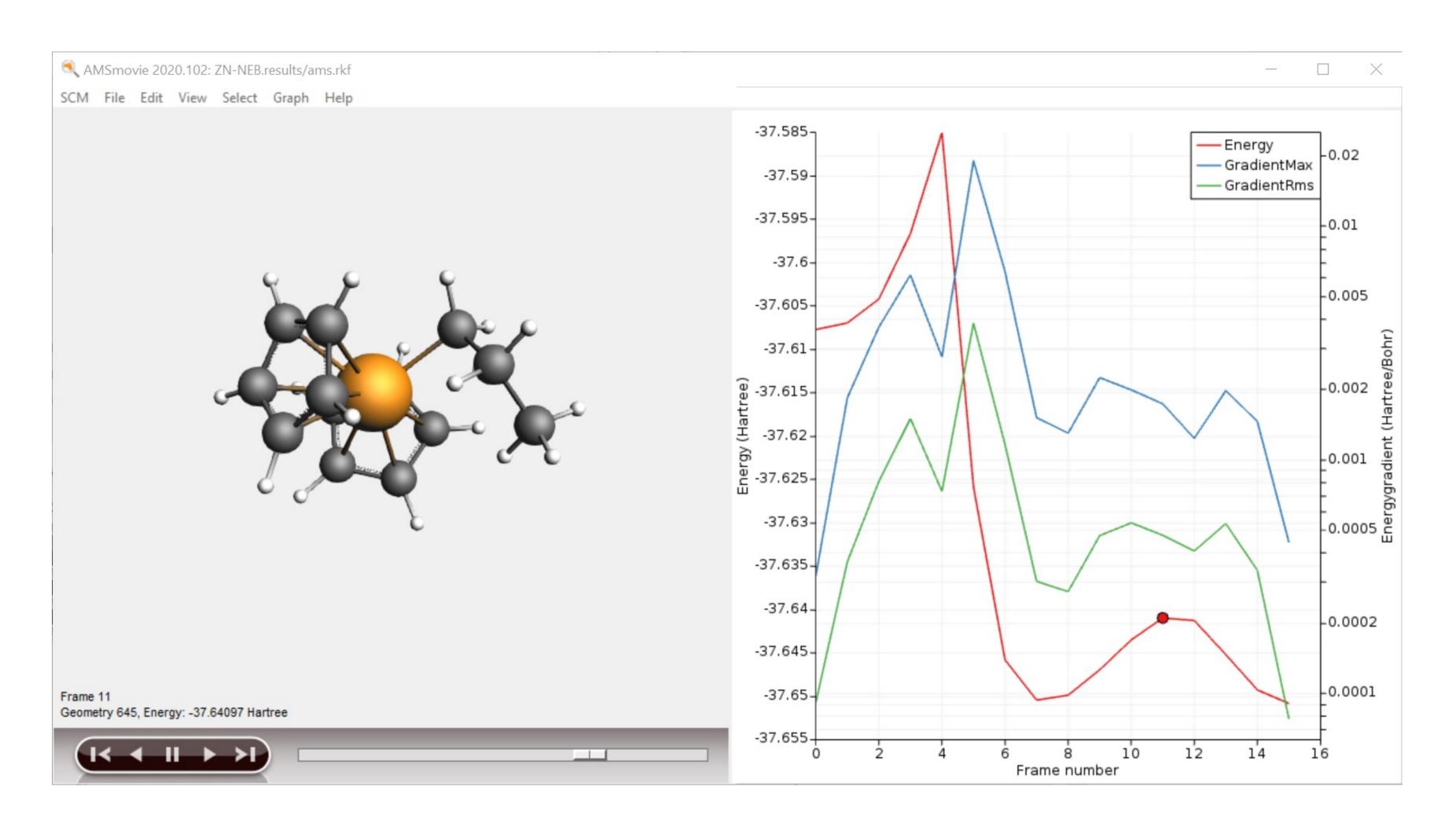
Periodic DFT(B) calculations

- Import cif, periodic pdb/xyz, use structure builder
- Run graphene 2D with BAND and DFTB
 - To get the Dirac point, right use a symmetric k-grid in BAND
 - 3D with QE (think about z lattice vector, and reorder the visualization path!)





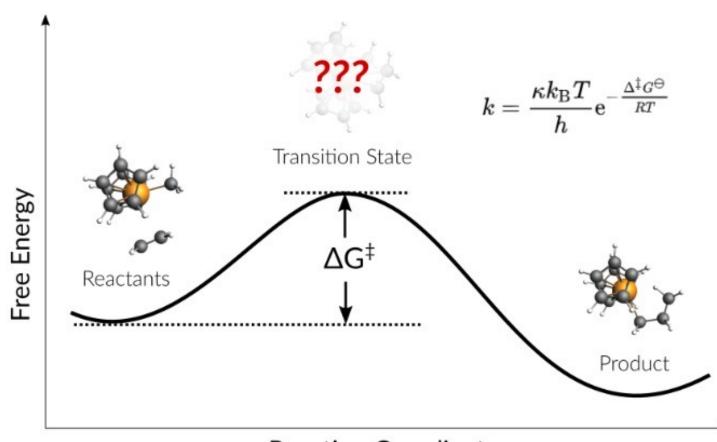
Transition States: Tips & Tricks





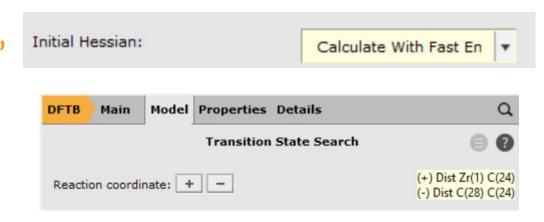
Pointers & ideas for TS search

- 1. Good starting point (geometry close to TS)
 - First explore reactant and product
 - (Initially) use low convergence criteria
 - Scan the PES with MOPAC or GFNx-TB
 - > Sometimes scanning from product side is easier
 - > Can use complex composite scan coordinates
 - Use CI-NEB (<u>tutorial</u>)
 - > Always a good idea to explore Reactants
 - Pass on coordinates (+Hessian) to ADF
 - Previous TS 'similar' system
 - Consider first doing constrained/partial optimization



Reaction Coordinate

- 2. Good description of the curvature: lowest Hessian eigenmode(s)
 - Partial Hessian
 - Low-level Hessian from previous, ADF choose 'Calculate With Fast Engine'
 - TSRC: define (complex) reaction coordinate
 - Consider 'TS point characterization' to check only 1 negative eigenmode
- 3. Automated reaction path search (tutorial)
- 4. Accuracy?
 - Solvent effects: COSMO and/or SM12 (single point only), COSMO-RS
 - metaGGA? TZP/DZP? Good numerical quality?
 - Double hybrids (single point)
 - Low-eigenmode corrections to frequencies
 - Sometime spurious 2nd imaginary modes are hard to get rid of -> ScanFreq

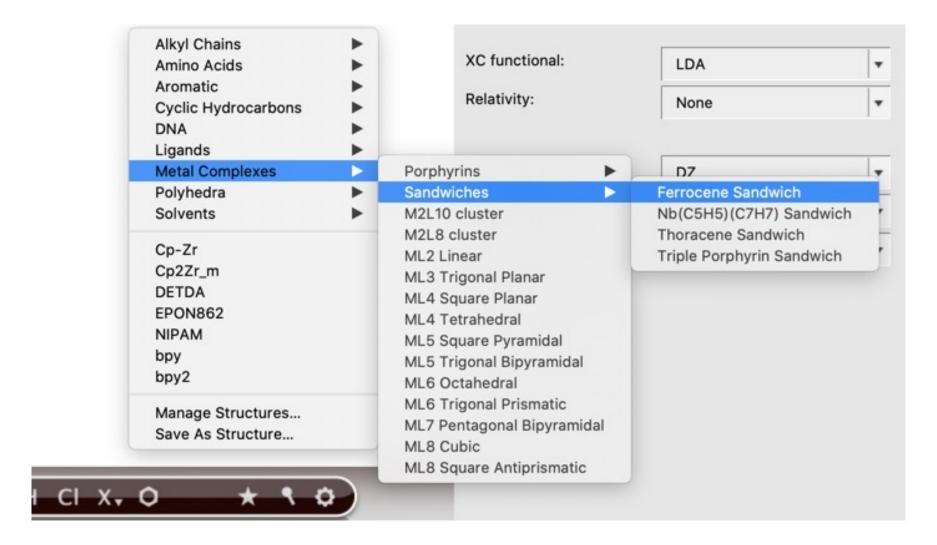


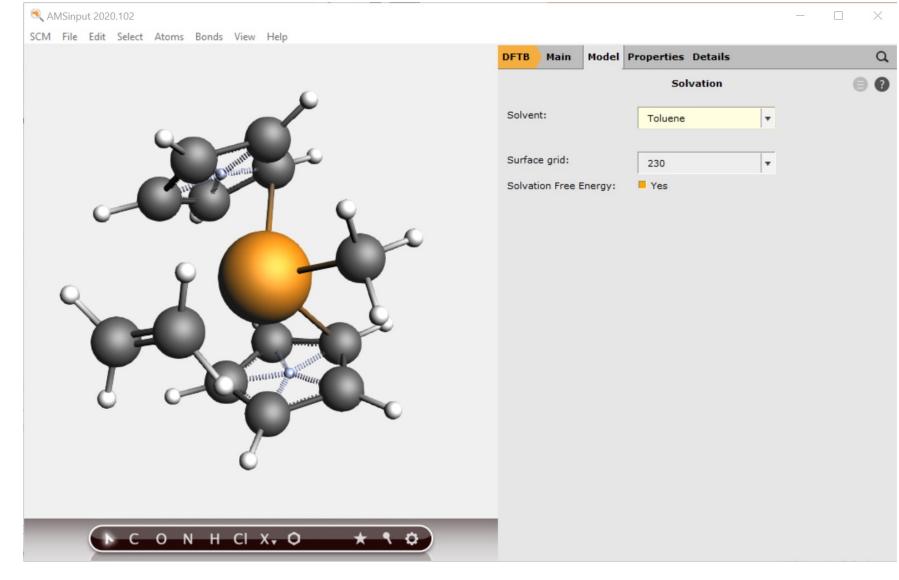




Transition State through PES scan

- Exercise (Ziegler-Natta tutorial):
 - New DFTB input
 - Build Cp₂ZrMe⁺
 - structure tool => ferrocene
 - Right-click Fe -> Element -> Zr
 - Add C and replace by methyl
 - (Use dummies to change CpMCp angle)
 - Model -> Solvent -> Toluene
 - Add charge and optimize
 - \circ Add ethene (use the \mathbb{Q})
 - Manipulate it in position
 - Right-click + drag = translate
 - Left-click + drag = rotate
 - Ctrl+M = select molecule
 - Remove dummies
 - Optimize

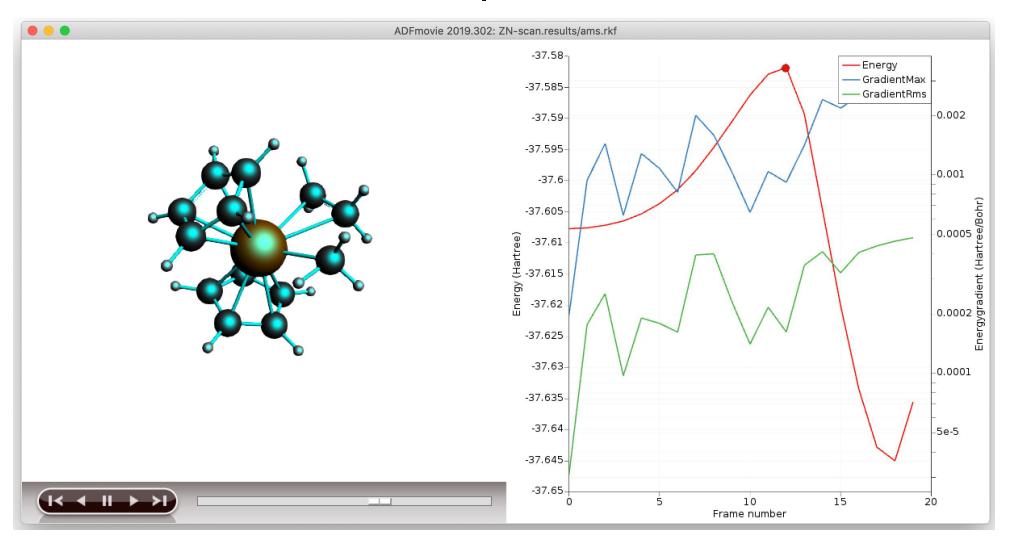


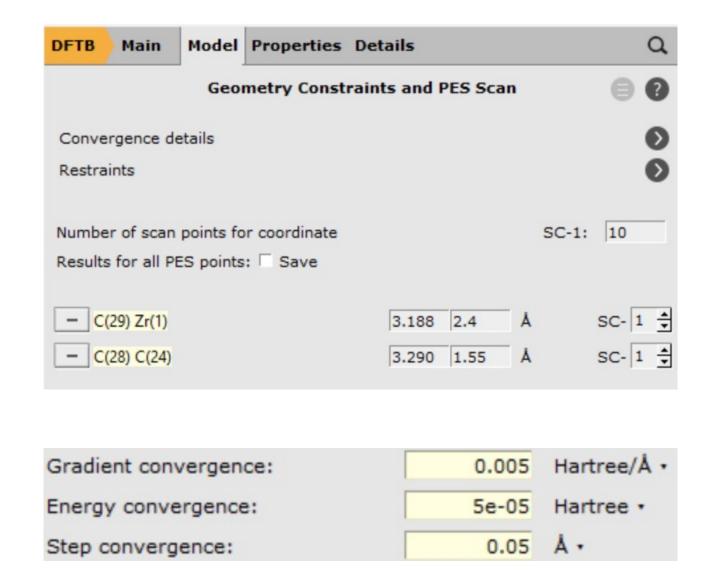




Transition State through PES scan

- Set up a PES scan, using 1 combined scan coordinate
 - Zr-C to 2.4 & C-C to 1.55
- Details -> Geometry Optimization:
 - Loosen the Convergence criteria by a factor of 5
- Run, and when finished, open AMSMovie





- Copy the highest energy structure (e.g. Ctrl+U), or save as structure
- Set up a TS run (ideally make a new input)
 - make sure there are no constraints, and regular convergence
 - use a calculate Hessian as initial guess and calculate frequencies



Transition State, start with NEB (Ziegler-Natta)

• Exercise:

- Optimize reactant & product from the last PES scan points
- Rotate the Zr-C-C-H dihedral to ~0 (metallacycle Zr-C-C-H) & optimize => alpha-agostic
- Set up a climbing nudged elastic band calculation Task: NEB
- Choose this structure as final, the other product as intermediate, and R as initial (import structures from results/ams.rkf) and choose 14 points
- Don't optimize R&P, relax convergence by factor 5
- Open AMSMovie when finished
 - Which is the rate-determining step?

AMSmovie 2020.102: ZN-NEB.results/ams.rki SCM File Edit View Select Graph Help -0.02 GradientMax -37.59 GradientRms -37.595 -37.6 0.005 -37.605 -37.61 0.002 9 -37.615--37.62 -0.001 e -37.625--37.63 -0.0005 -37.645 -0.0001 Geometry 645, Energy: -37.64097 Hartree 12

Tip: use CI-NEB to get close enough to TS, use Hessian-based optimization to find it

Transition States: Further considerations

Reaction rates:

 \circ Calculate $\Delta G^{\#}$: include H_{vib} , S (AMSoutput -> Other Properties -> Statistical Thermal Analysis)

Temp		Transl	Rotat	Vibrat	Total
298.15 Entropy (cal/mol-K):		42.601	31.288	52.371	126.261
Nuclear Internal Energy (kcal/mol):		0.889	0.889	160.907	162.684
Constant Volume Heat Capacity (cal/mol-K):		2.981	2.981	56.260	62.222
(c) Constant Volume Heat	Capacity (cal/mol-K):	2.981	2.981	52.800	58.761
Summary of energy terms					
	hartree	eV		cal/mol	kJ/mol
Energy from Engine:	-37.585024915182686	-1022.7406		3584.96	-98679.47
Nuclear Internal Energy:	0.259253970109237	7.0547		162.68	680.67
Nuclear Internal Energy:	0.257445594179003	7.0055		161.55	675.92
Internal Energy U:	-37.325770945073451	-1015.6859	-2	3422.28	-97998.80
pV/n = RT:	0.000944186013486	0.0257		0.59	2.48
Enthalpy H:	-37.324826759059967	-1015.6602	-2	3421.68	-97996.32
-T*S:	-0.059990510871653	-1.6324		-37.64	-157.51
-T*S:	-0.058176671371776	-1.5831		-36.51	-152.74
Gibbs free energy:	-37.384817269931617	-1017.2926	-2	3459.33	-98153.82
The properties marked with the interpolation corrections'. In but the contribution to the print the following paper:	For those terms all positive partition function of small	re frequencies vibrational f	are used requencie	(also ones es is modifi	below 20 1/cm), ed as described
S. Grimme, 'Supramolecular B: (Chem. Eur. J. 2012, 18, 995;			cted Dens	ity Functio	nal Theory'

- $_{\circ}$ hTST as first approximation: k ~ exp (-G $^{\#}/RT$), Low-level frequencies: Grimme corrections
- Could also try <u>mircokinetics</u>. More underway (<u>ReaxPro</u>)

Suggestions & troubleshooting

- 1. get close to TS: NEB, PES Scan, constrained opt, previous TS + change ligands (PLAMS script!),
- 2. get a good curvature (Hessian): pre-calc with GFN-xTB, MOPAC; partial Hessian
- 3. check final curvature / path: Frequency calculation, PES point characterization, IRC
- 4. Spurious imag. Freq.: displace along that mode (AMSSpectra, Ctrl+8 *5) & retry



TS for periodic systems

- How to get good TS geometry guess?
 - Literature or previous results:
 - similar TS + constrained optimization
 - Quick (?) NEB (you want to find R & P anyway)
 - PES scan (can you use ReaxFF, MOPAC, GFN-xTB? Otherwise: cheaper DFT?) (video)
 - Maybe you can get away with periodic DFT/DFTB (video) or DFT/MM?
- How to get good guess for lowest few Hessian eigenmodes?
 - Could try MOPAC or DFTB Hessian
 - Inspect modes first?
 - Define reaction coordinate (TSRC)
 - Use a partial Hessian ('active' region)



