



# Day 1: Introduction to AMS2018 Getting started

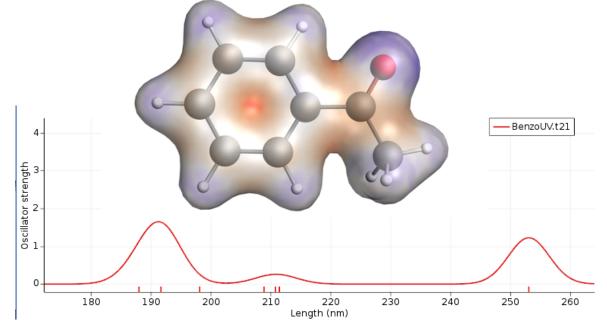
Hands-on workshop Chemistry & Materials with the Amsterdam Modeling Suite

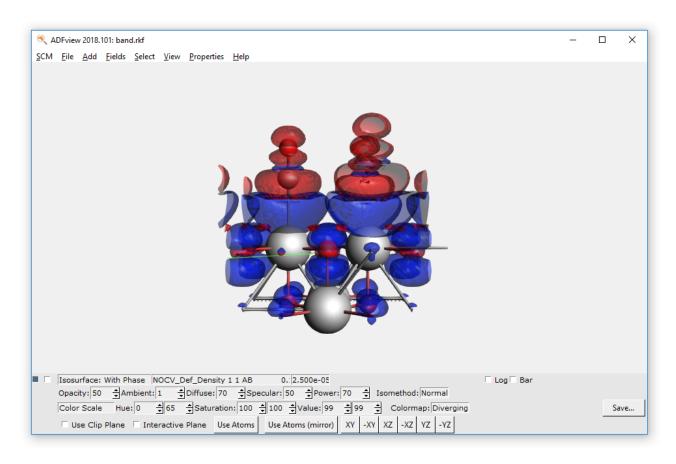
2-day workshops, October 2018, China Fedor Goumans, goumans@scm.com SCM support: support@scm.com FermiTech support: wiki, support@fermitech.com.cn

Making Computational Chemistry Work for You

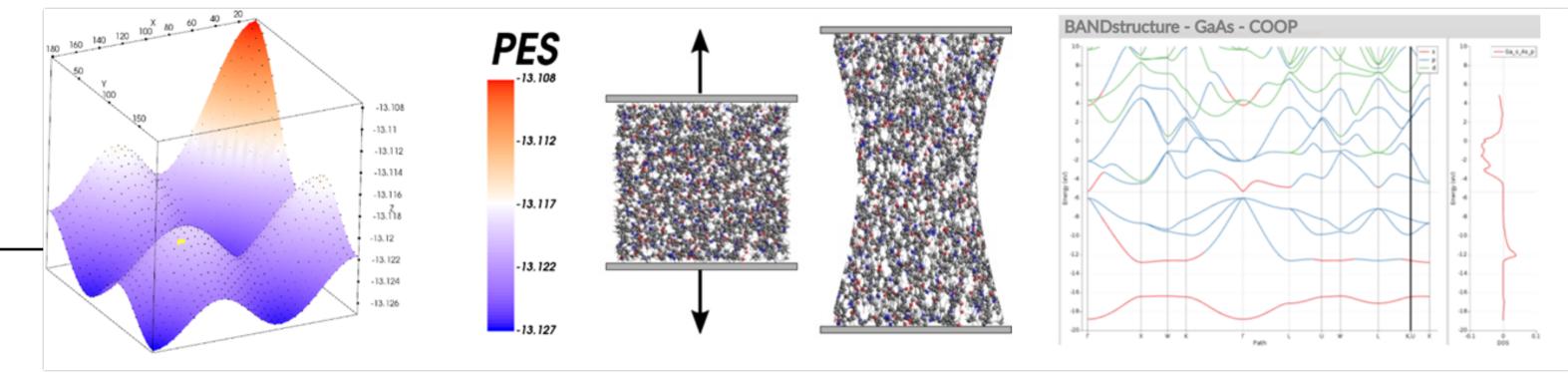
## Program

- Introduction
  - SCM & AMS
  - Computational chemistry& materials science
- Molecules: builder, database, import SMILES, xyz
  - Spectroscopy
  - Advanced: bonding analysis, charge transfer, SOCME
- Periodic structures: database, cif, surfaces
  - Advanced: bonding analyses (PEDA-NOCV, COOP)
- COSMO-RS: properties, optimizing mixtures
- ReaxFF: acceleration, fitting parameters (tomorrow)





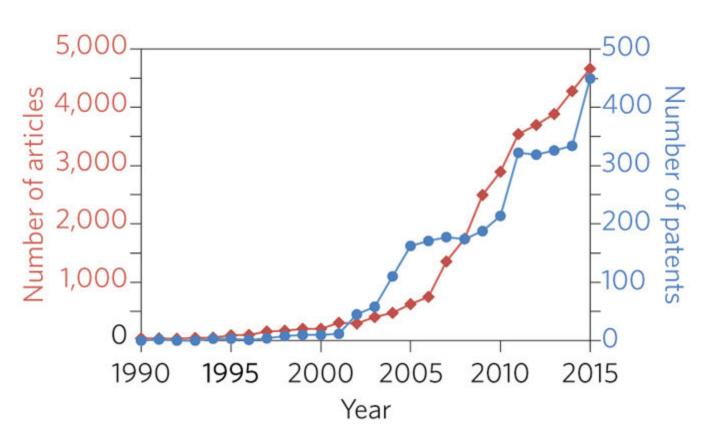




## Background: SCM & ADF

- ADF = first DFT code for chemistry (1970s)
   Baerends@VU (>'73), Ziegler@Calgary † (>'75)
- SCM: Spin-off company 1995
- 15 people (10 senior PhD's) + 5 EU fellows
- Many academic collaborators / EU networks
  - ~120 authors
  - New functionality
- SCM: development, debug, port, optimize, docs & <u>support</u>





articles &patents in materials science with "density functional theory", Nat. Mat. 4619



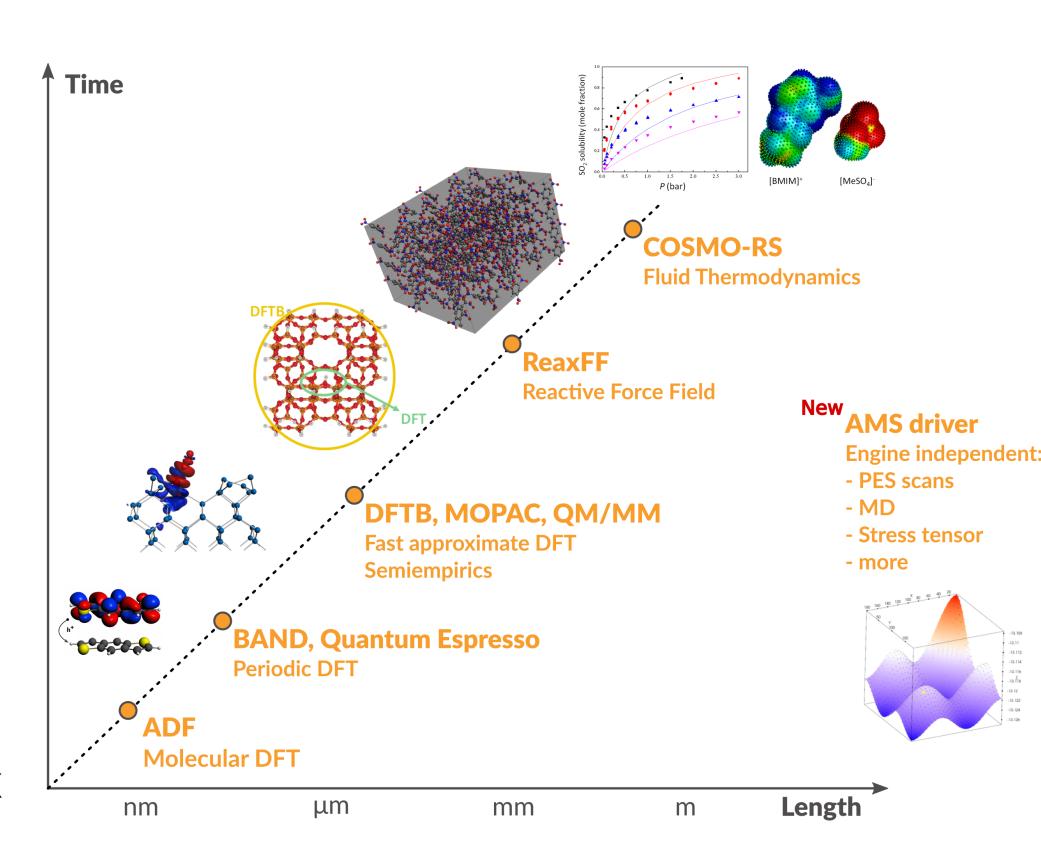
### The SCM team





## Amsterdam Modeling Suite

- ADF: powerful molecular DFT
  - Spectroscopy: NMR, EPR, VCD, UV, XAS
  - Advanced solvation / environments
  - Bonding & density analysis
- BAND: periodic DFT
  - (2D) Materials
- DFTB: fast approximate DFT
- ReaxFF: Reactive MD
  - Dynamics of large complicated systems
- COSMO-RS: fluid thermodynamics
  - VLE, LLE, logP, solubility
- Integrated GUI use out of the box
- Scripting: workflows & automation

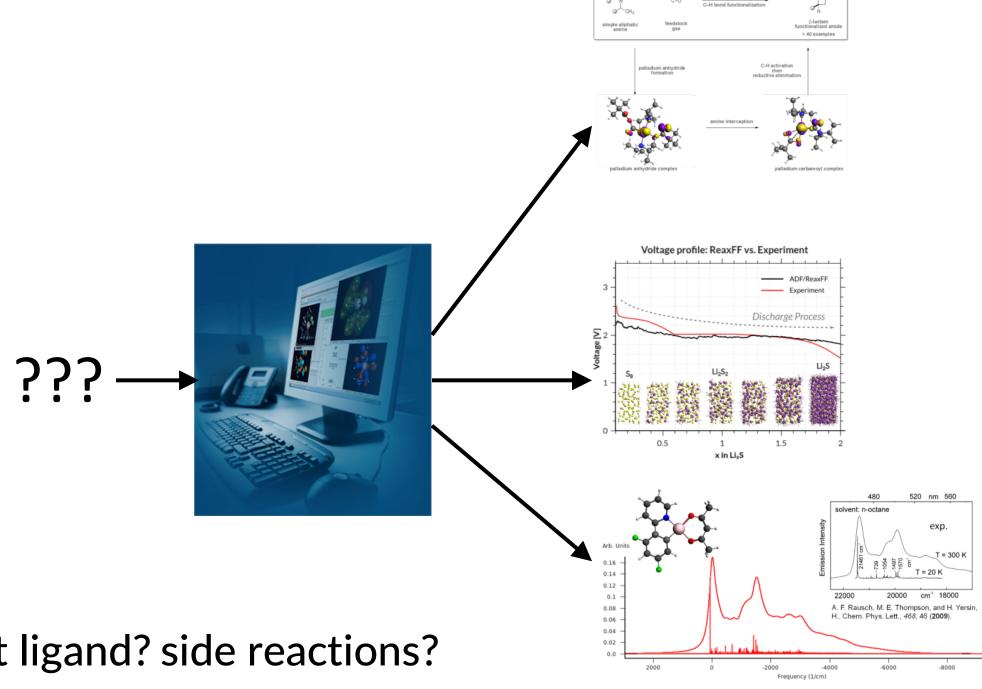




## Why bother with calculations?

#### Computational chemistry & materials modeling

- Accelerate research, reduce costs & environmental impact
  - Reduce experimental search space
  - Analyze structure-property-reactivity
- Models: physics & empiricism
  - Accuracy?
- Synergy experiment-calculations
  - Ask relevant questions
  - Limitations model
  - Constraints experiments



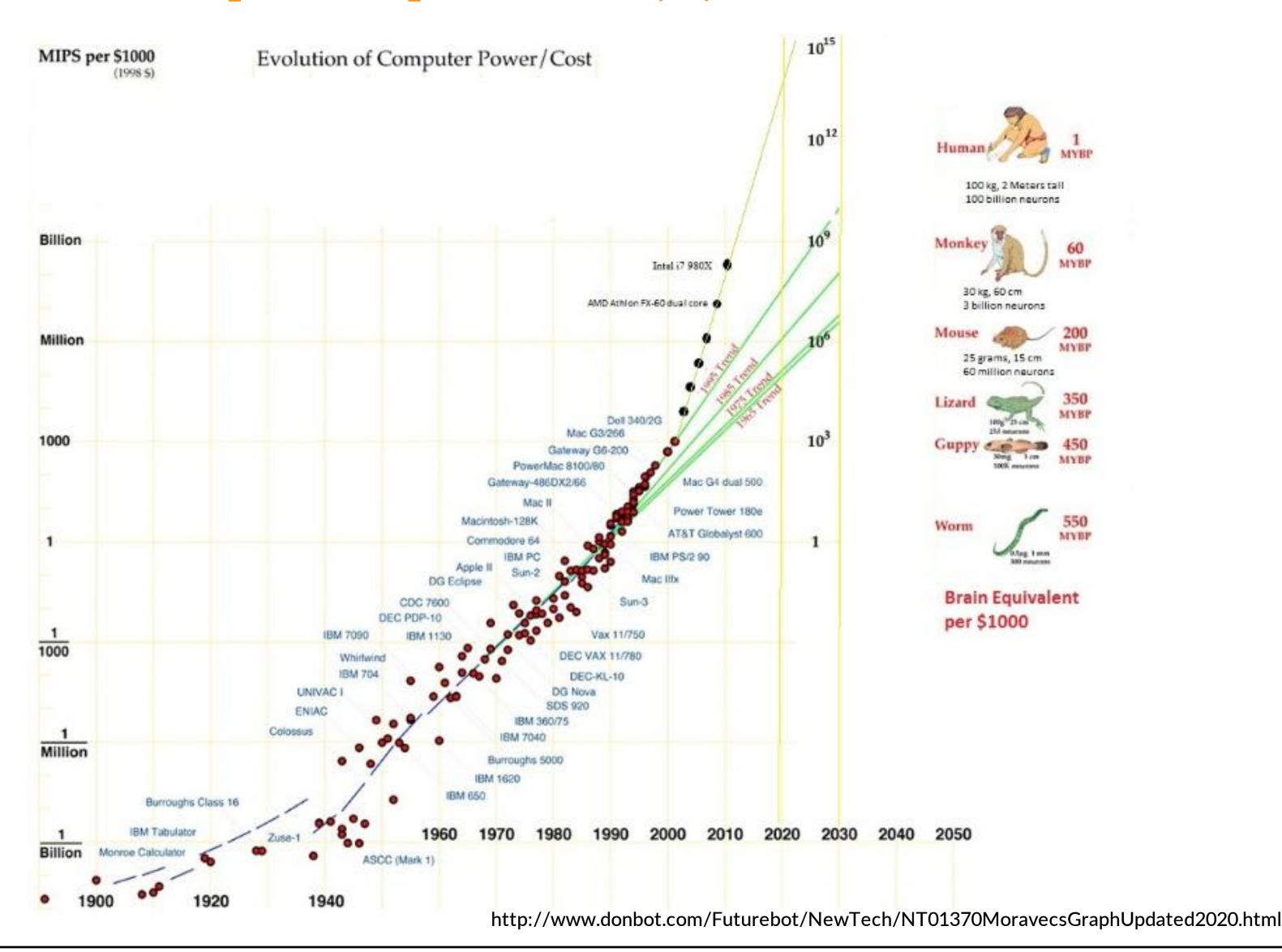
Best catalyst? => mechanism? lowest  $E_a$ ? best ligand? side reactions?

Best battery? => discharge? voltage? interaction with electrolyte?

Best OLED? => charge & exciton mobility? emission speed & color?

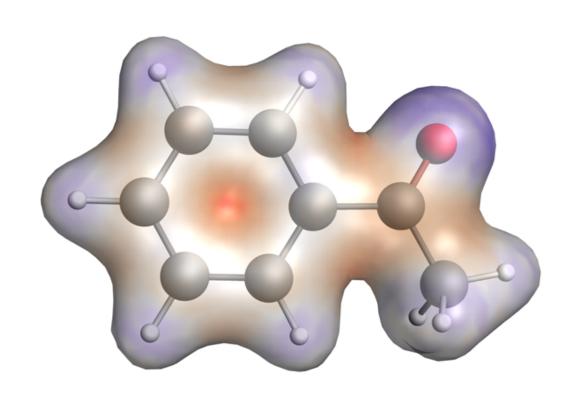


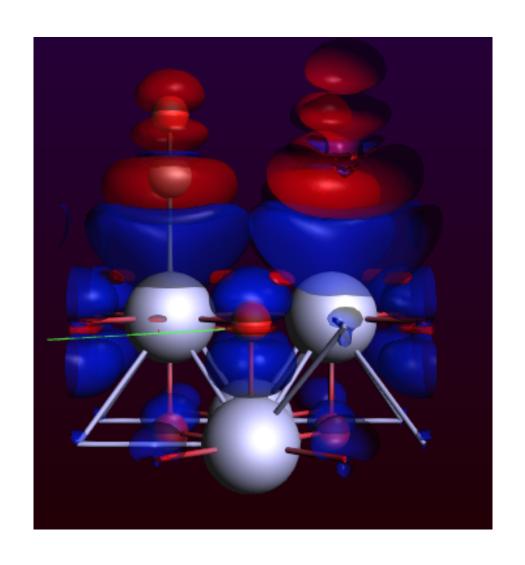
### Compute power (r)evolution





## Computational Chemistry & Materials





Electronic structure methods: Schrödinger equation

Electrons in molecules & materials

Expand  $\psi$ : atomic orbitals / plane waves

Solve self-consistently

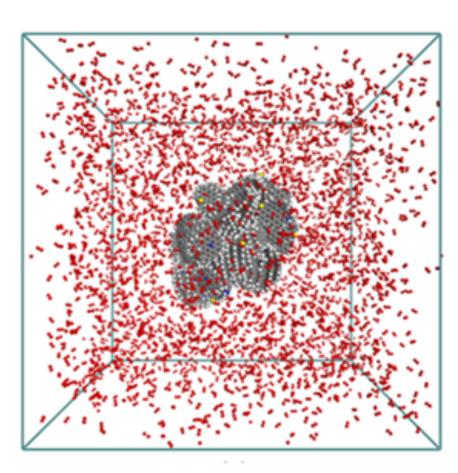
Pragmatic: DFT

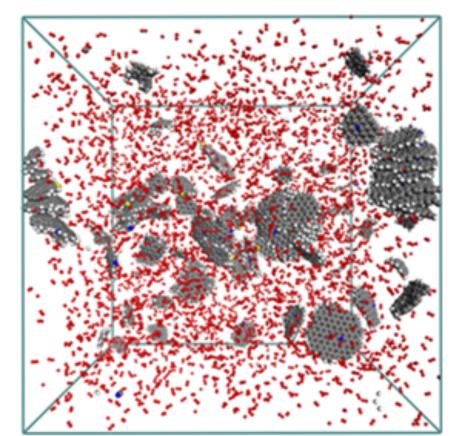
**Properties**: energies (gradients), MOs, densities & related, spectroscopy (EPR, NMR, IR, UV/VIS, ....)



## Computational Chemistry & Materials

$$d^2x/dt^2 = F(x)$$
$$F(x) = -dV(x)/dx$$





Molecular dynamics: Newton's equations of motion

Movement of atoms: solve numerically + propagate,

Properties: reaction rates, diffusion coefficients, stress-strain, ....

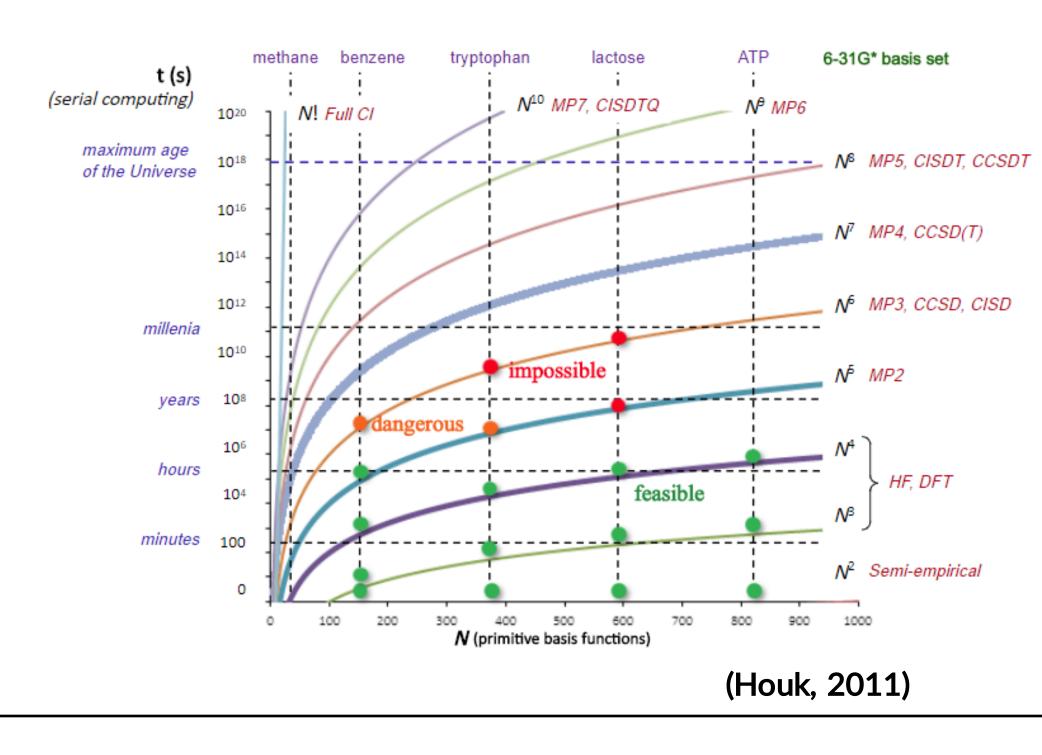


### **Electronic Structure methods**

• ab initio (basis set dependencies!)

 $H\psi = E\psi$ 

- Hartree-Fock (HF): mean field (no explicit e-e interaction)
- MP2: perturbation theory (if HF = good guess)
- CC: coupled cluster
- CI: configuration interaction (full CI = max. Accuracy)
- Multi-reference / active space
- Density Functional Theory (DFT)
  - 'first principle' functionals (physics)
  - empirical functionals (fit to data)
- DFT-based tight binding (DFTB)
  - Fit to DFT data
  - Nearest neighbor, minimal basis
- Semi-empirical (MOPAC: PM7)
  - Fit to exp. Data
  - Nearest neighbor, minimal basis





### Computational Methods

Relative costs, scaling & accuracy for computational methods (\* depending also strongly on the system & property!)

Method	~ max atoms	~ relative cost	scaling	Typical Accuracy*
Classical force field (UFF, Amber,)	1,000,000	0.0005	N <sup>1</sup>	<20 kcal/mol
Reactive force field (ReaxFF)	500,000	0.001	$N^1$	<15 kcal/mol
Semi-empirical methods (e.g. AM1, PM7)	5,000	1	N <sup>1~2</sup>	<10 kcal/mol
DFTB	5,000	1	N <sup>1~2</sup>	<10 kcal/mol
DFT	500	500	N <sup>3~4</sup>	<5 kcal/mol
MP2	100	2000	<b>N</b> <sup>5</sup>	<5 kcal/mol
CCSD(T)/cc-pVTZ	30	100000	N <sup>7</sup>	~1 kcal/mol



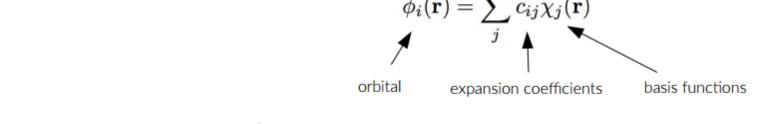
## **Density Functional Theory**

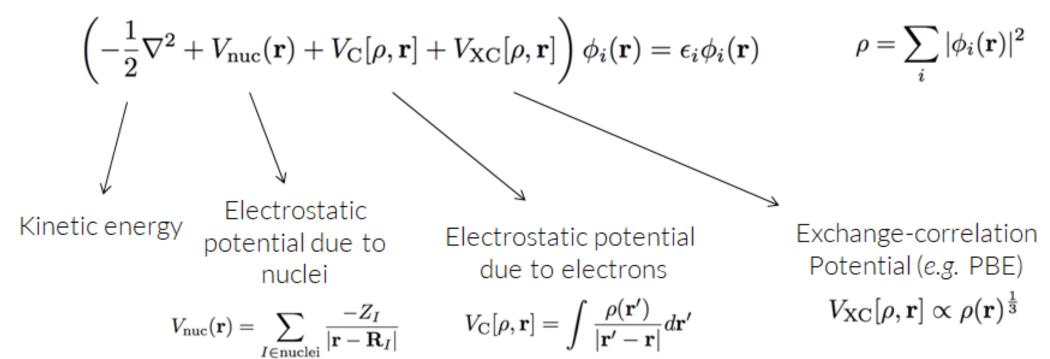
- Density  $\rho$  = central quantity
- Density functional  $\rho$  gives E  $E[\rho] = T[\rho] + E_{ee}[\rho] + E_{ne}[\rho]$
- Usually expanded in orbitals
  - Linear combination of atomic orbitals
    - Basis set
  - Kohn-Sham DFT
  - 'Non-interacting' reference T<sub>s</sub>

$$E[\rho] = T_s[\rho] + E_{ee}[\rho] + J[\rho] + E_{xc}[\rho]$$

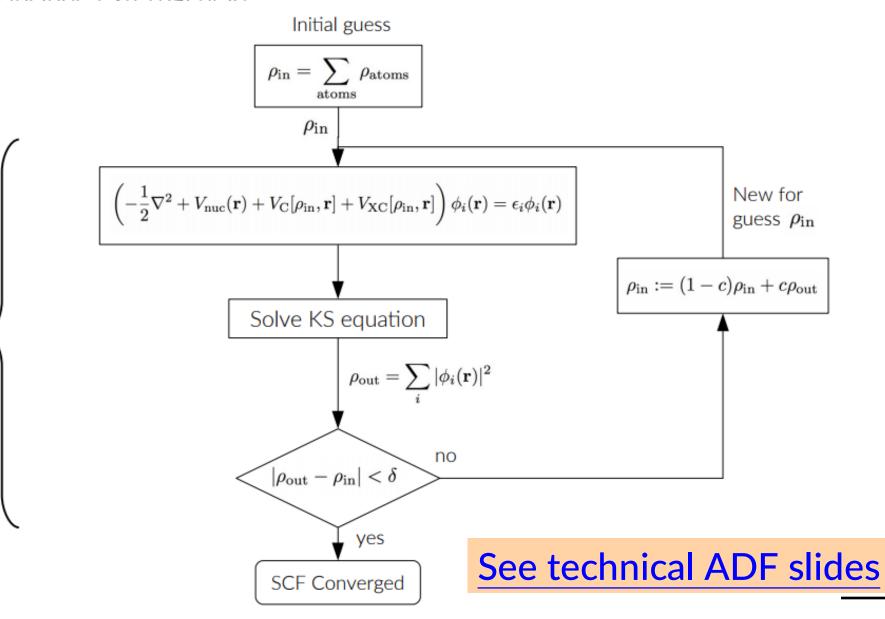
- Solve self-consistently
- $v_{xc}[\rho] = E_{xc}[\rho]/d\rho = 'functional'$ 
  - approximate: LDA, BP86, PBE, M15L, ...
- Which basis & functional?

Check literature or benchmark!





More general:  $v_{\text{ext}}$ 



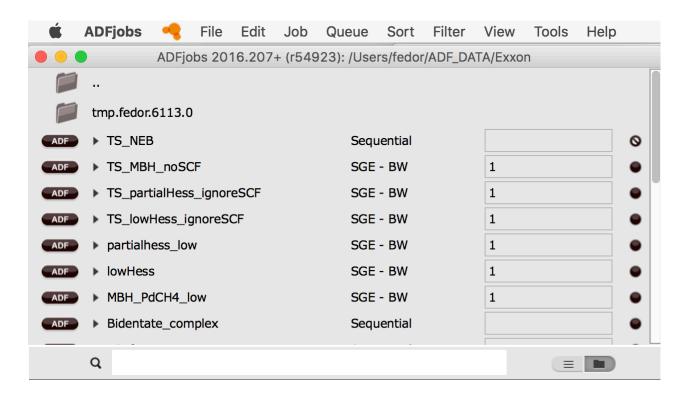


SCF

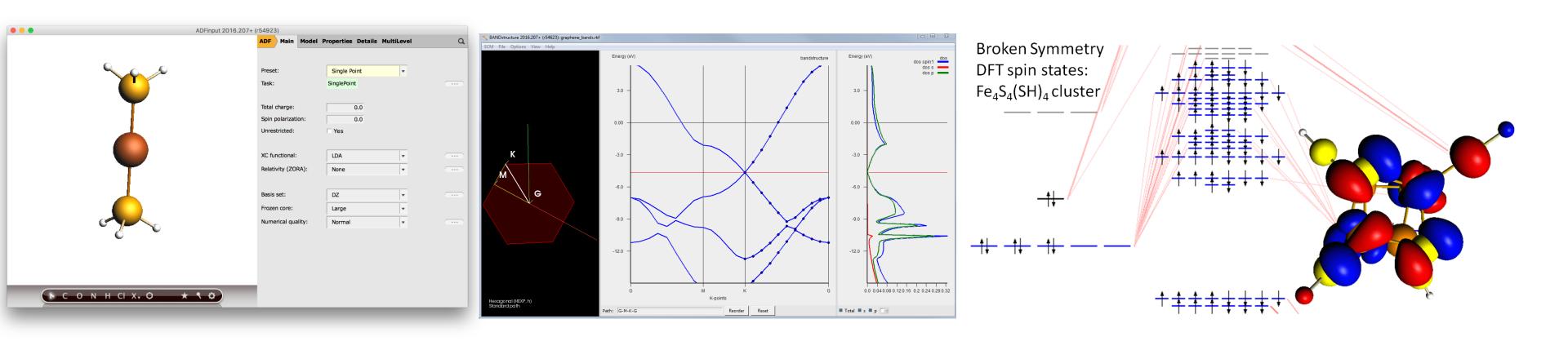
## Getting started with the GUI

Starting ADFjobs: job bookkeeping tool

- Win: dbl-click desktop item
- Mac: open Application
- Linux: run \$ADFBIN/adfjobs

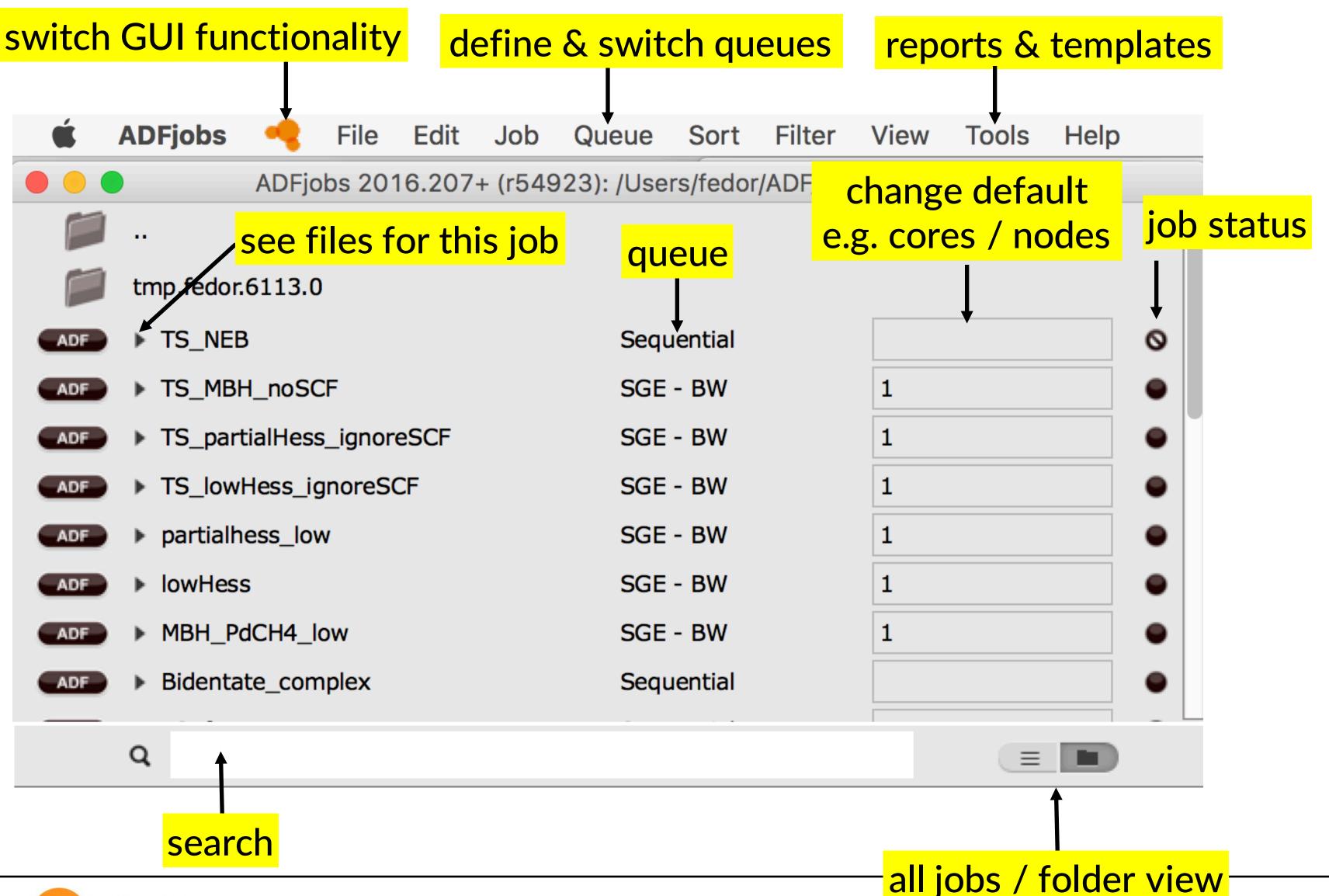


- Other GUI modules: (Input, View, Levels, Movie, Spectra, Band Structure, COSMO-RS, ...)
  - o Can be opened by dbl-clicking '.exe' (Win) or opening e.g. '\$ADFBIN/adfinput'

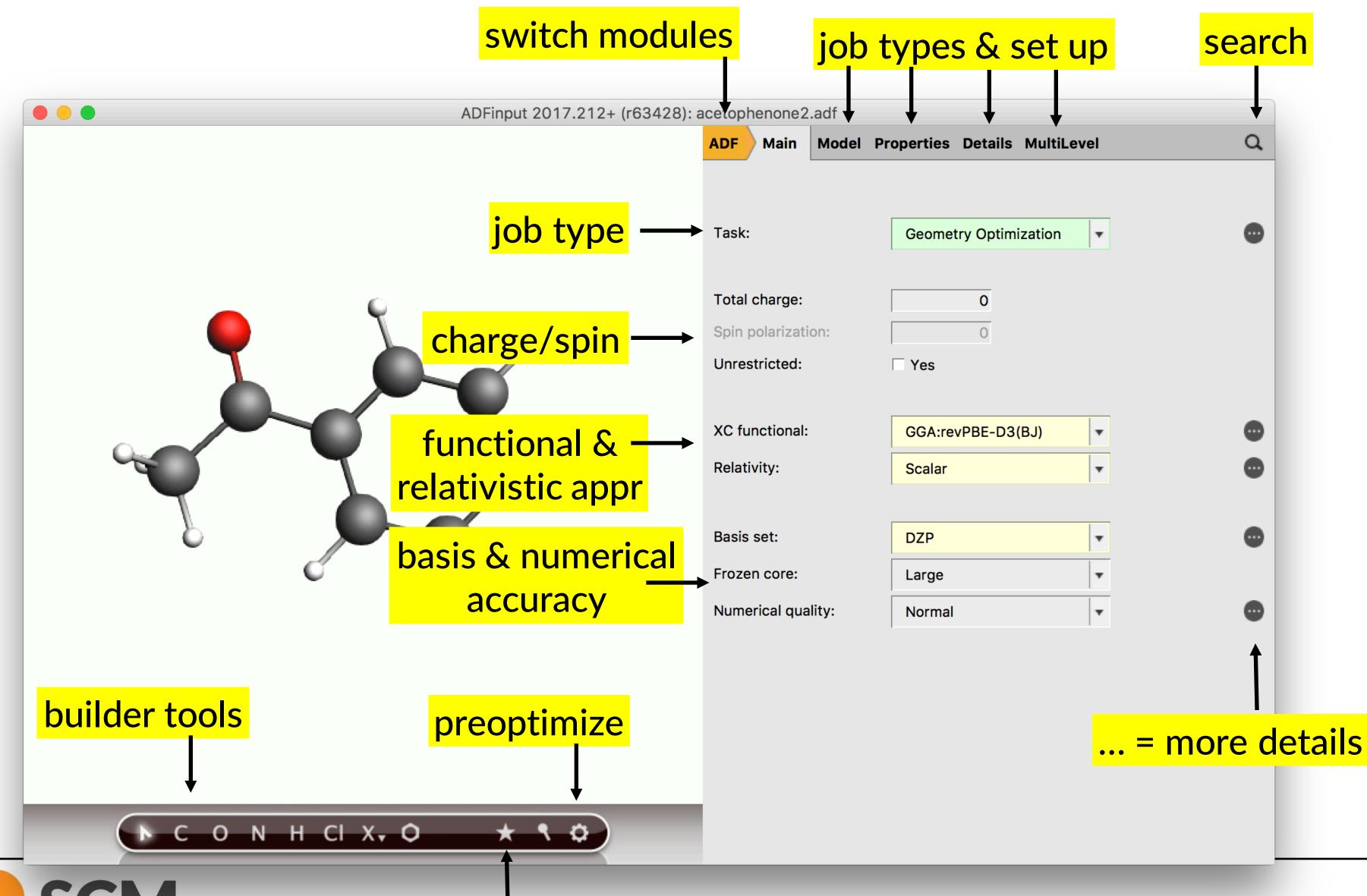




## ADFjobs: job bookkeeping



## Basic calculations & settings



AM

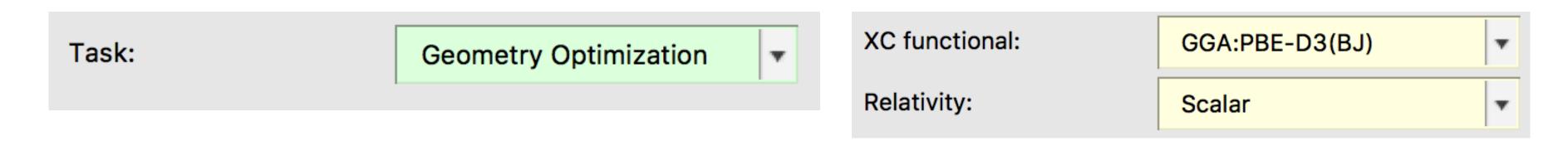
## Building molecules

#### www.scm.com/doc/Tutorials/GUI\_overview/Building\_Molecules.html

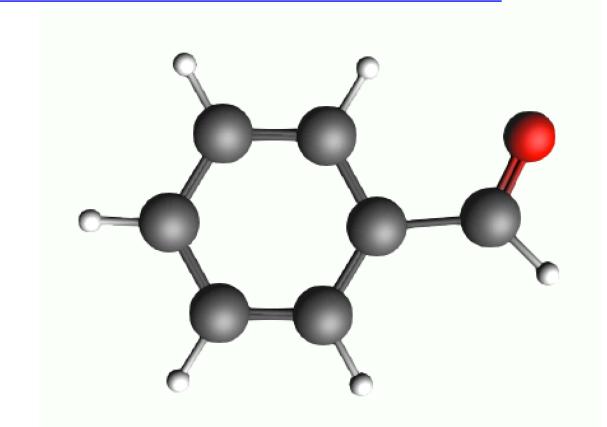
- NB: tutorials also offline with your ADF!
- Import: SMILES, xyz, cif, pdb, ...
- Included library + building



- By searching for it in the GUI
- By starting from the benzene template (press 2 for double bond, Ctrl+E to add Hs)
- By importing smiles CC(=O)c1ccccc1 (e.g. from Wikipedia or Chemspider)
- Exercise 2: Symmetrize, pre-opt & optimize: SR-ZORA-PBE-D3(BJ)/DZP



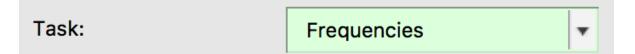




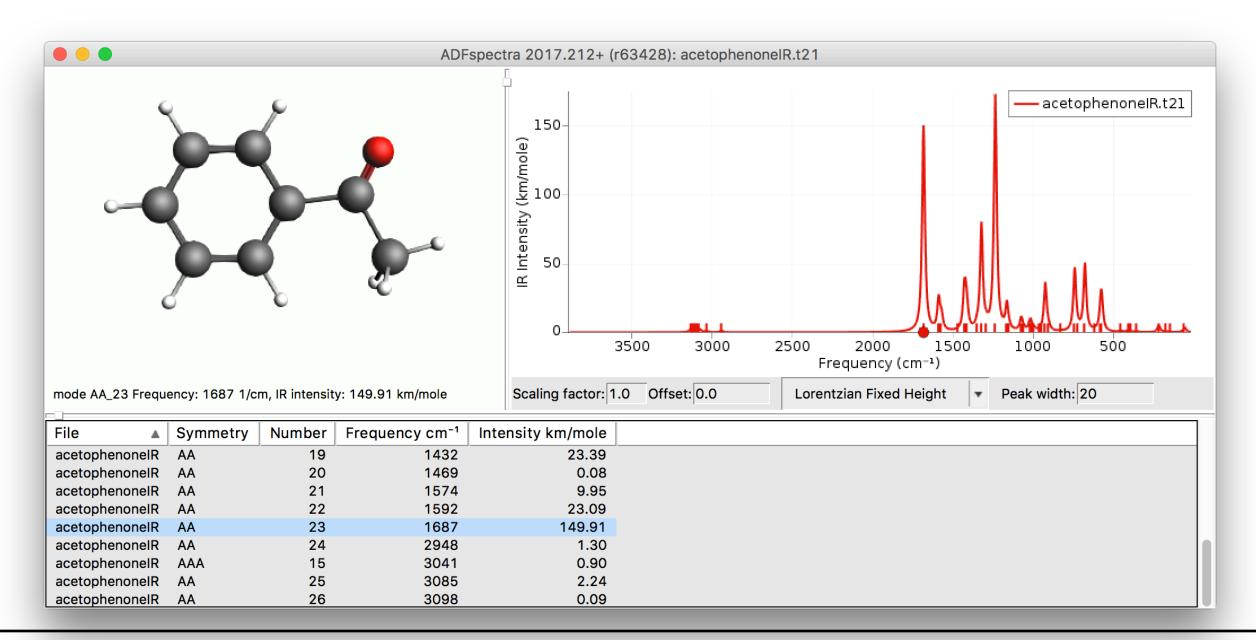
## Spectra: IR

#### www.scm.com/doc/Tutorials/ADF/ADF-GUI\_tutorials.html#spectroscopy

#### • Excercise 3: Calculate & visualize frequencies



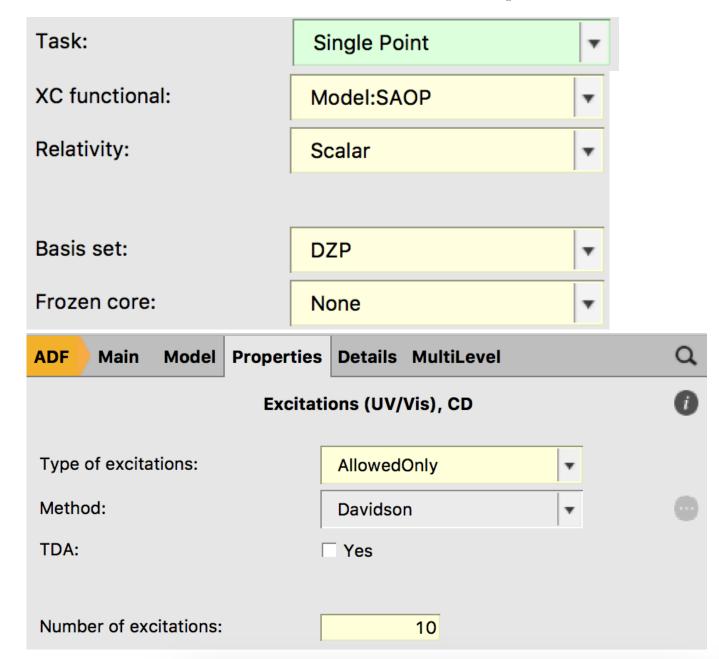
- NB analytical frequencies available for most GGAs, not for hybrids
- Go to spectra, visualize the CO stretch at ~1690cm<sup>-1</sup>
- Increase the line width to ~20
- Compare to NIST data

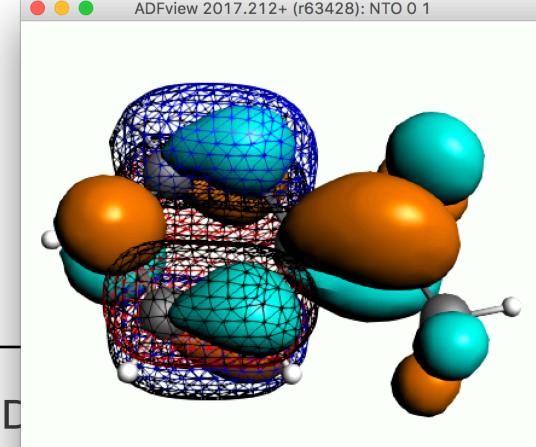




## Spectra: UV/VIS

- Exercise 4: Calculate 10 allowed excitations with the SAOP model potential
  - See also UV/VIS FAQ for more tips
  - Go to spectra, change x-axis to nm
  - Increase the line width to ~10
  - Visualize the pi-pi\* NTOs at ~250 & 285nm
  - Compare to NIST data
  - Now rerun with method 'sTDA' and tick TDA
  - Compare timings & spectra (File -> add spectra)
  - AMS2018: optimize excited states also with
    - Range-separated hybrids
    - COSMO solvation







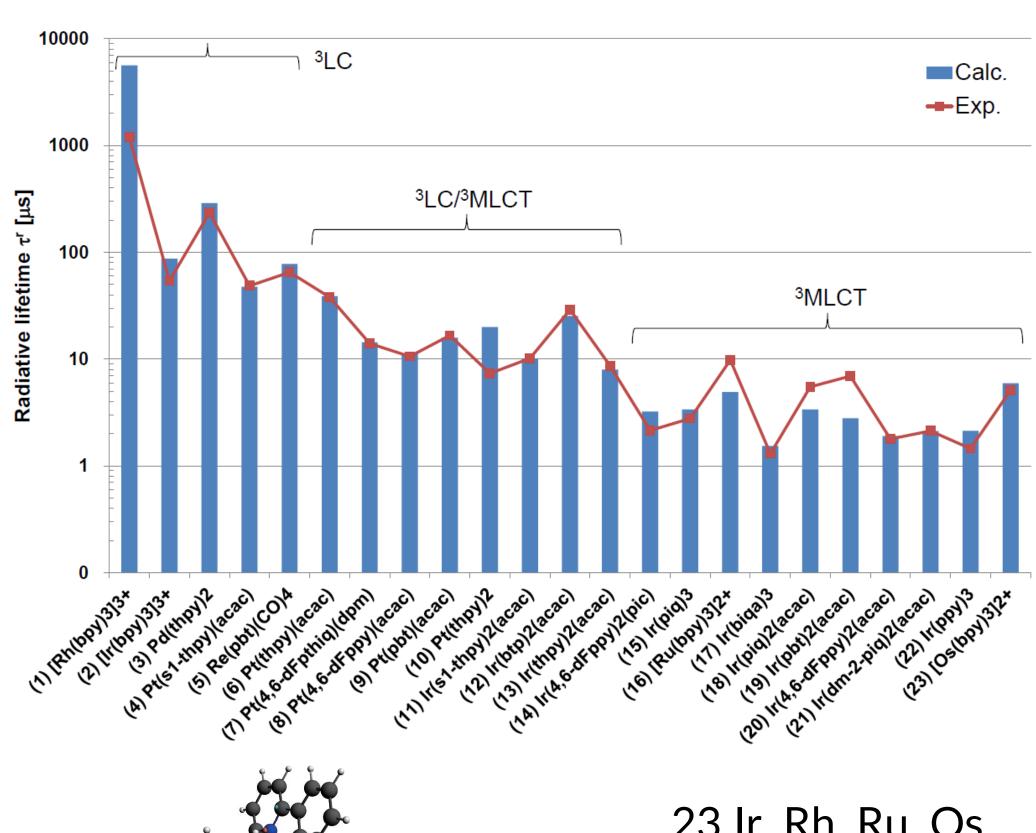
## Phosphorescent OLED emitters

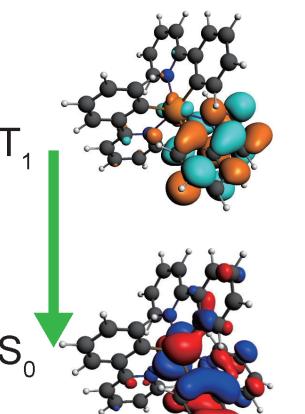
#### Optimize OLED performance

- $_{\circ}$  Phosphorescence T<sub>1</sub>→ S<sub>0</sub>
- o TADF: min. S-T, max. SOCME
- ∘ Fast ISC:  $S_n \rightarrow T_1$
- High charge mobility

#### ADF features

- SOC-TDDFT: k<sub>phos</sub>
- SOCME: k<sub>ISC</sub>
- Transfer integrals (mobilities)
- Other couplings (FDE)
- Vibrationally resolved abs/em





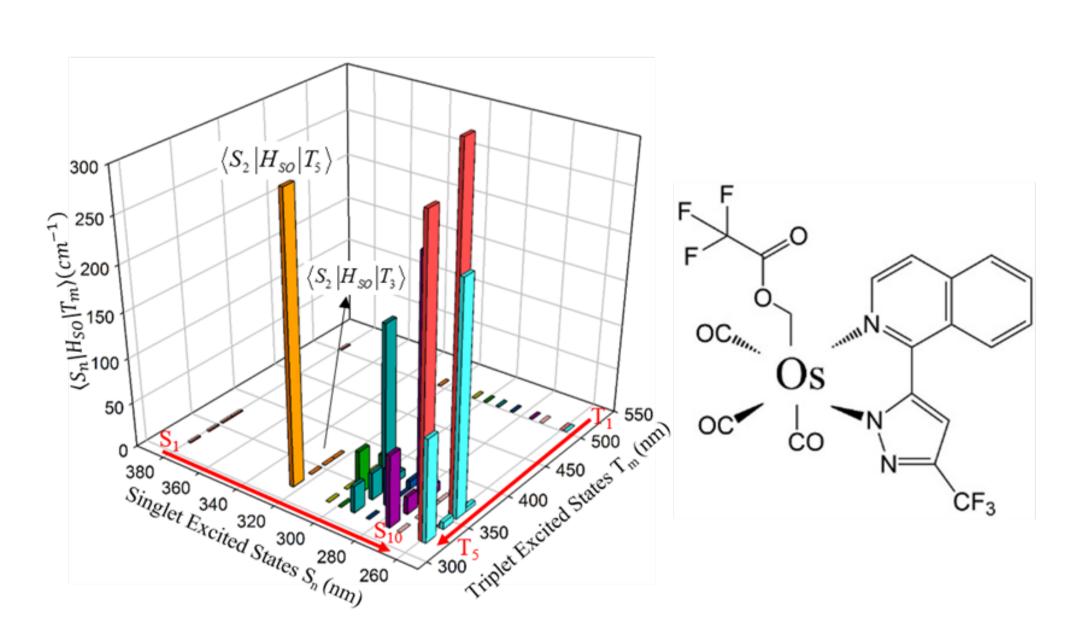
23 Ir, Rh, Ru, Os, Pt, Re complexes

K. Mori, T. P. M. Goumans, E. van Lenthe, F. Wang, PCCP 16, 14523 (2014)

ADF tutorial online



#### Intersystem crossing: spin-orbit coupling



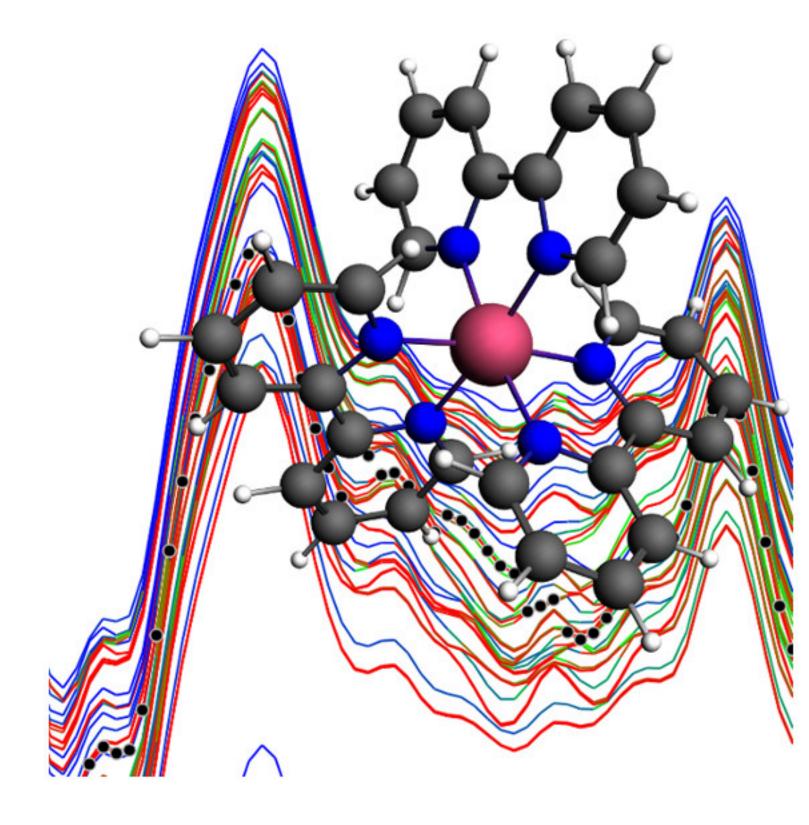
#### **El-Sayed for organometallics:**

SOC is largest when:

- both S ( ${}^{1}d\pi^{*}$ ) and T ( ${}^{3}d'\pi^{*}$ ) are MLCT
- different d-orbitals are involved (d ≠ d').

 $\frac{\lambda_{exc}\text{-dependent quantum yield}}{\text{SOCME negligible for S}_1\text{-T}_n}$ ISC from higher S<sub>n</sub> states

Phys. Chem. Chem. Phys. **16**, 26184-26192 (2014) NB: See also full k<sub>ISC</sub> Paul et al. JPCL, 2017, **8**, 4893



#### Surface-Hopping Dynamics Ru(bpy)<sub>3</sub><sup>2+</sup>

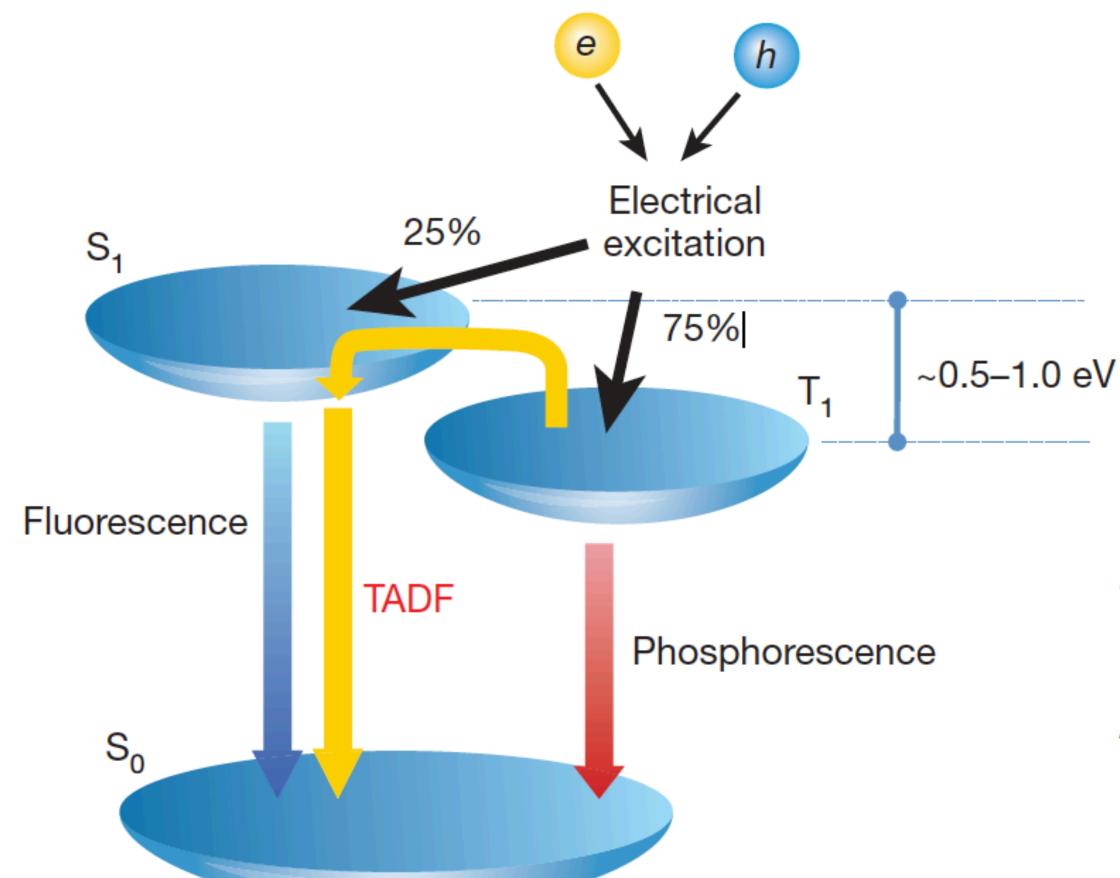
- Interface SHARC-ADF with ISC (SOC)
- ISC from higher S<sub>n</sub> states within 26 ± 3 fs

**Atkins & Gonzalez** 

J. Phys. Chem. Lett., 8, 3840–3845 (2017)



#### Thermally Activated Delayed Fluorescence



Optimize radiative rate

- Minimize S1-T1 gap
- Maximize SOC
- Maximize k<sub>phos</sub> & k<sub>TADF</sub>

$$k_{\text{UISC}} = \frac{2\pi}{\hbar} \rho_{\text{FC}} |\langle S_1 | \hat{H}_{\text{SO}} | T_1 \rangle|^2$$

$$\rho_{FC} = \frac{1}{\sqrt{4\pi\lambda_{M}k_{B}T}} \sum_{n=0}^{\infty} \exp(-S)$$

$$\frac{S^n}{n!} \exp \left[ -\frac{(\Delta E_{\rm ST} + n\hbar\omega_{\rm eff} + \lambda_{\rm M})^2}{4\lambda_{\rm M}k_{\rm B}T} \right]$$

Z.-M. su et al. Dyes & Pigments **145**, 277-2847 (2017) Bredas et al. J. Am. Chem. Soc. **139**, 4042-4051 (2017)

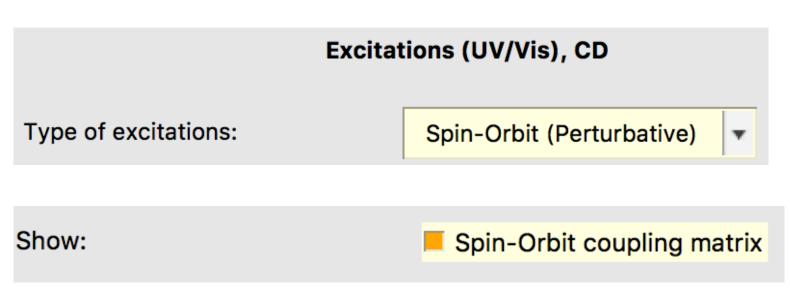
ADF tutorial in progress



## Spin-orbit coupling TDDFT

- Exercise 5: Calculate the SOC spectrum and the SOCME
  - Switch to perturbative Spin-Orbit and tick SOCM
    - ADF can do full and perturbative SOC
    - SOCME useful for (R)ISC, e.g. TADF/OLEDs
  - Visualize the spectrum: any effect?
    - SOC small for purely organic
  - Check SCM -> Output
  - Go to Response Properties -> All Singlet-Singlet Excitations. Compare with All Spin-Orbital Coupling Excitation Energies. Go to -> SO Matrix. This is the full matrix. Further down you will find the averaged spin-orbit coupling matrix elements (SOCMEs)

All Sr	All Spin-Orbital Coupling Excitation Energies					
no.	E/a.u.	E/eV	f	tau/s	Symmetry	
1:	0.12004	3.26655	0.1880E-08	1.149	A'	
2:	0.12004	3.26657	0.4159E-06	0.5193E-02	A'	
3:	0.12004	3.26655	0.2580E-09	8.371	A''	
4:	0.13339	3.62962	0.2890E-04	0.6054E-04	A''	
5:	0.13363	3.63626	0.1693E-11		A'	
6:	0.13363	3.63626	0.4123E-10		A''	
7:	0.13367	3.63732	0.4724E-05	0.3687E-03	A''	
8:	0.14001	3.80992	0.7306E-11		A'	
0 -	0 14001	2 00002	0 710CH 11		2011	



Spin-orbit couplings calculated as root mean squares: square root of (the sum of squares of spin-orbit coupling matrix elements of

21.60

0.01

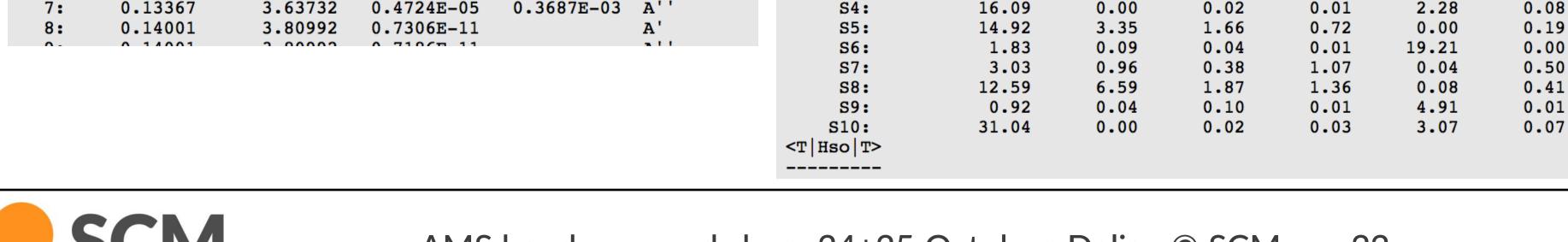
0.60

all sublevels of the uncoupled states) in cm-1

0.38

3.41

0.06





<S | Hso | T>

S1:

S2:

T4

3.87

0.08

5.22

3.45

0.00

1.61

Т6

0.72

0.00

19.20

0.33

5.84

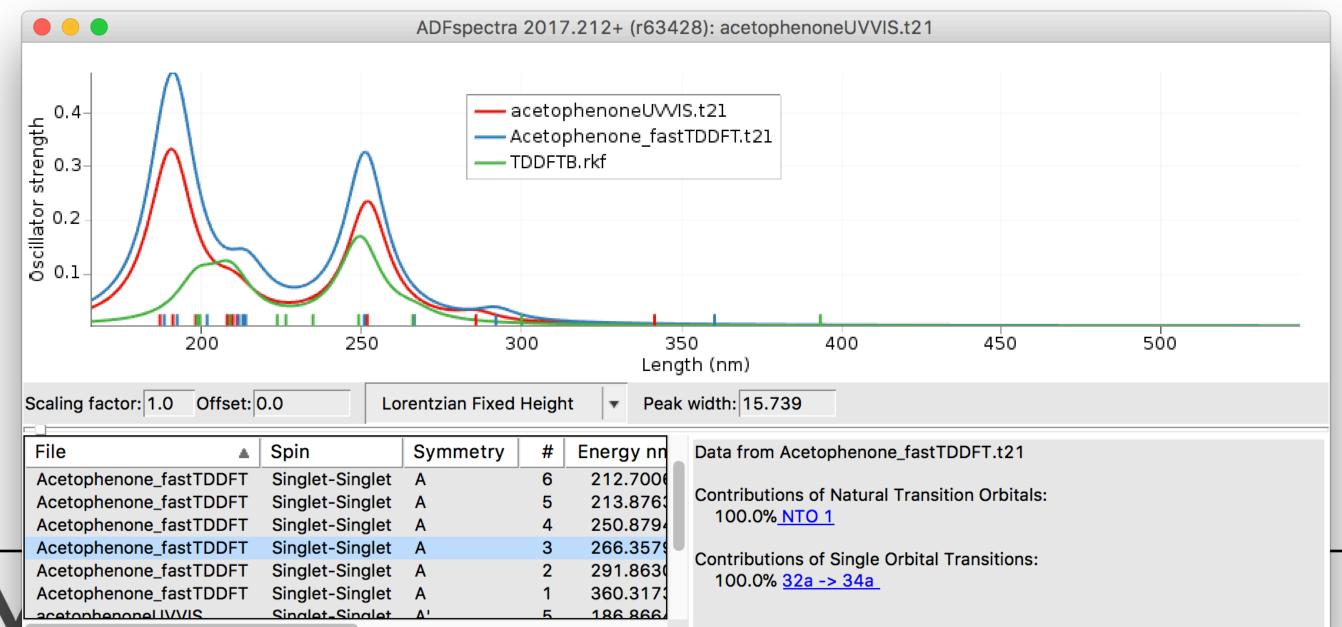
0.09

## Speeding it up: DFTB

Parameter directory:

#### www.scm.com/doc/Tutorials/DFTB/DFTB-GUI\_tutorials.html

- Exercise 6: Switch to DFTB and re-run the spectra
  - The default SCC-DFTB method is fine
  - Chose the 3ob-3-1 parameter set
  - Re-optimize, recalculate frequencies, recalculate UV/VIS spectrum
  - How do the spectra compare: TDDFT, sTDA, TDDFTB? Timings?
  - Also try the IR spectrum with DFTB & MOPAC
- To visualize DFTB MOs: single point SCC-DFTB/QuasiNano15!

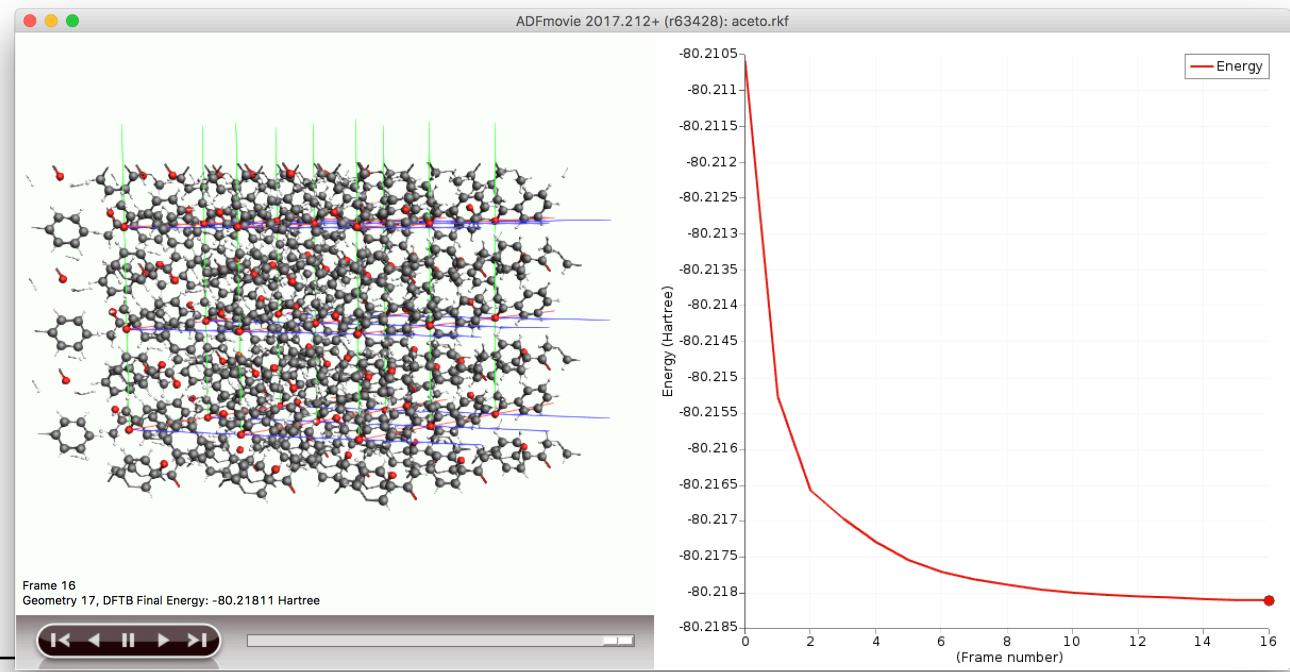




DFTB.org/3ob-3-1

## Going periodic with DFTB

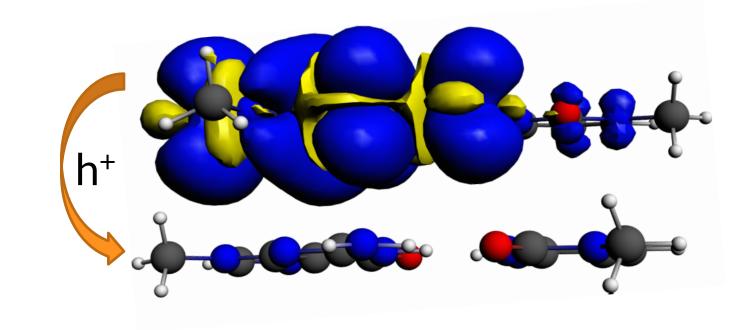
- Exercise 7: Acetophenone crystal
  - Make a new input, search for acetophenone, select the crystal (optimized with MOPAC)
  - Switch from BAND to DFTB and choose D3-BJ dispersion and the 3-ob-3-1 parameter set
  - Ignore the small unit cell warning
    - NB: consider to go to a larger supercell instead of k-point sampling (will use FIRE)
    - You may also consider optimizing the lattice (slow)
  - Check progress with SCM -> Movie





### Methods to calculate charge mobilities

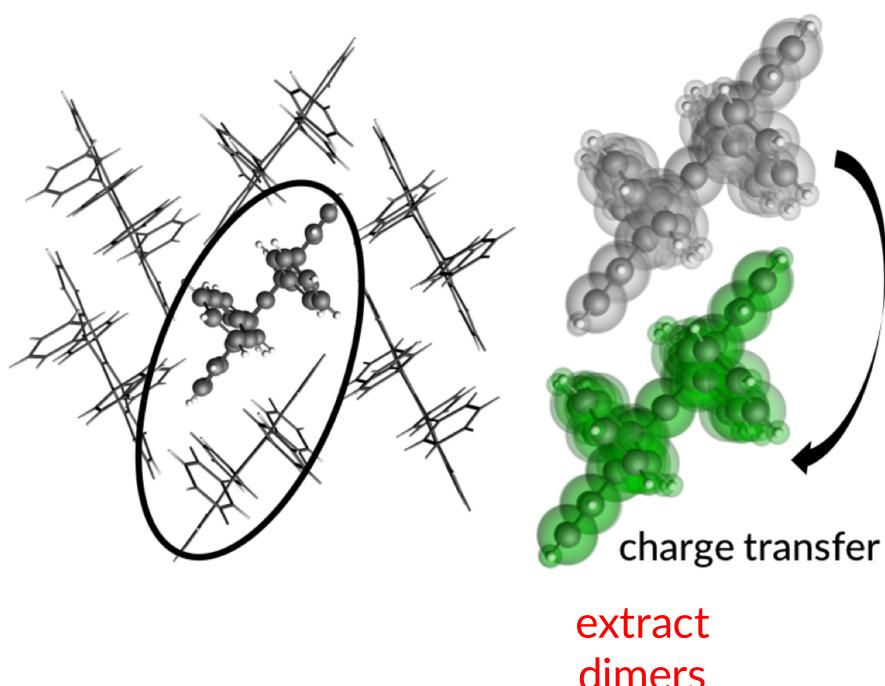
- Hopping transport:
  - Charge transfer integrals (FO) easy
  - Electronic couplings from FDE

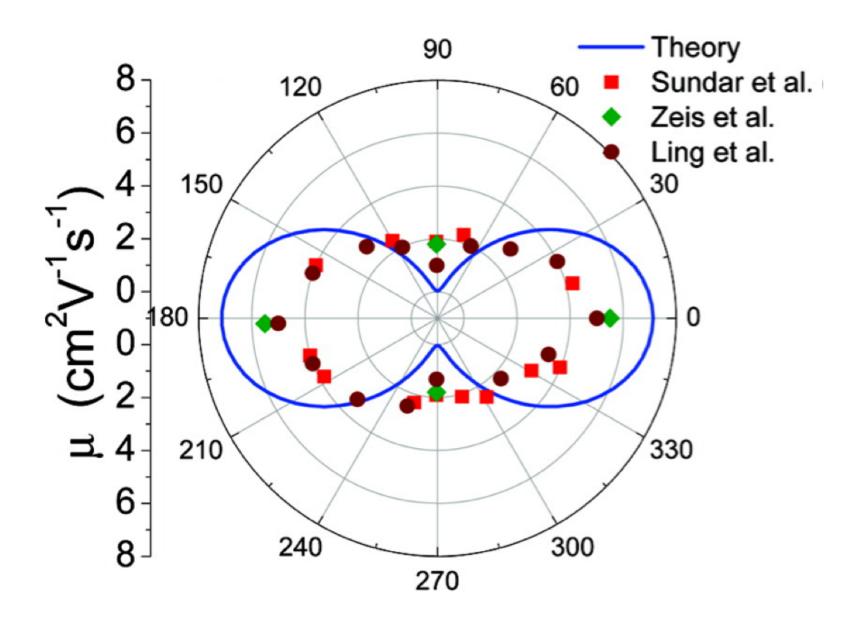


- Band transport: effective mass tensors in BAND
  - AMS2018: for any k-point & also for DFTB
  - Simulating excited state: <u>create a core hole</u>
- Non-equilibrium Green's Function (NEGF)
  - transmission probabilities for single-molecule junctions
  - quick calculation: wide-band limit
  - o also in BAND (periodic structures, fully self-consistent, bias) and in DFTB
    - See online tutorials



#### Effective transfer integral $J_{eff}$ = electronic coupling V





## dimers

#### Anisotropic mobility:

$$V = \frac{J_{\text{RP}} - S_{\text{RP}} (H_{\text{RR}} + H_{\text{PP}})/2}{1 - S_{\text{RP}}^2}$$

$$k = \frac{4\pi^2}{h} \frac{V^2}{\sqrt{4\pi\lambda k_B T}} \exp\left\{-\frac{\lambda}{4k_B T}\right\}$$

$$\mu_{\Phi} = \frac{e}{2k_{\rm B}T} \sum_{i} W_{i} r_{i}^{2} P_{i} \cos^{2} \gamma_{i} \cos^{2}(\theta_{i} - \Phi)$$

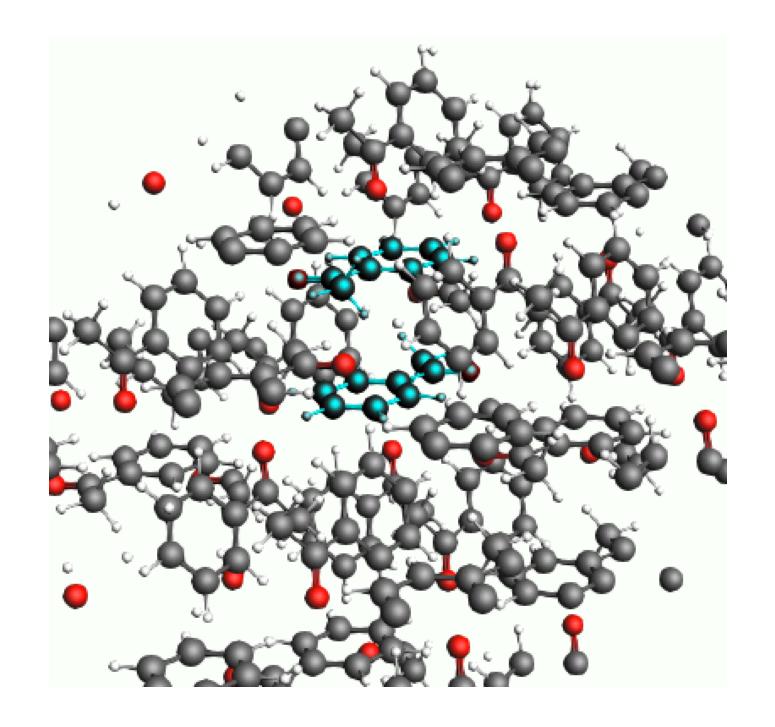
S.-H. Wen et al., J. Phys. Chem. B 113, 8813 (2009)

ADF tutorial online, ADF prints  $V/J_{eff} = >$ use with Marcus theory for hopping rates



### Charge transfer integrals & recombination

- Exercise 8: 1<sup>st</sup> Select the dimer for the charge transfer event
  - Update the geometry from your DFTB run
  - Make a 2x2x2 supercell (Edit-> Crystal -> Generate Supercell)
  - Switch from DFTB to ADF
  - Select 1 atom from the middle 'flat' acetophenone and one from the molecule above
  - Press Ctrl+M to select both molecules





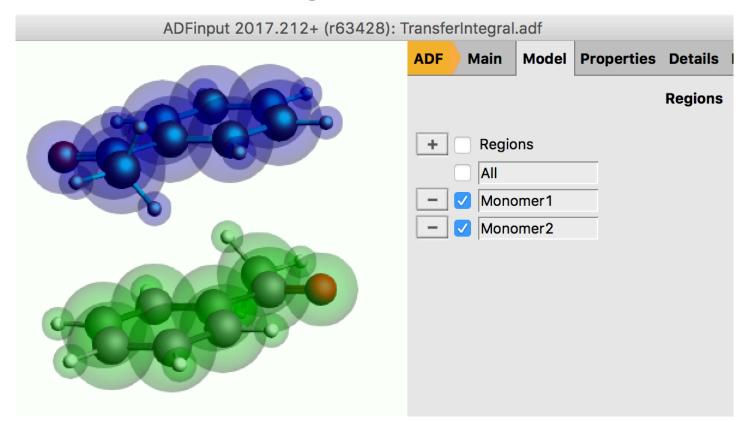
## Charge transfer integrals & recombination

- Exercise 8: Calculate the charge transfer integral
  - Select -> Invert selection. Delete all other atoms
  - o In the Model Region Panel select each molecule as a new region (delete the crystal region):

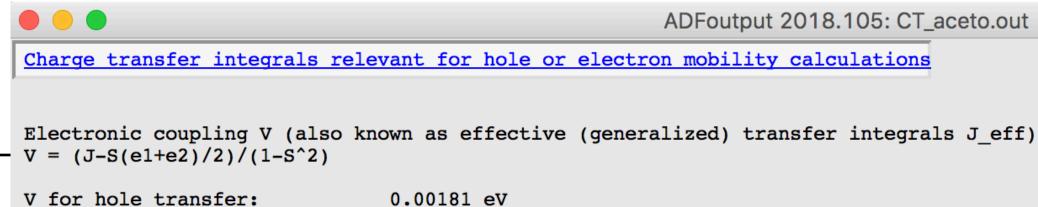
V for electron transfer:

V for charge recombination 1-2:

V for charge recombination 2-1:



- Select PW91 functional (often used) with a DZP basis
  - Frozen cores are incorrect in principle; in practice usually fine
- Find 'transfer integral' in the GUI search, and tick it.
- In Multilevel -> Fragments pane, tick 'Use fragments'
- SCM-> Output; Properties -> Charge transfer integrals
- ADF tutorial: how to get mobilities
- AMS2018: charge recombination



0.00433

-0.00434

eV

eV

-0.09641 eV

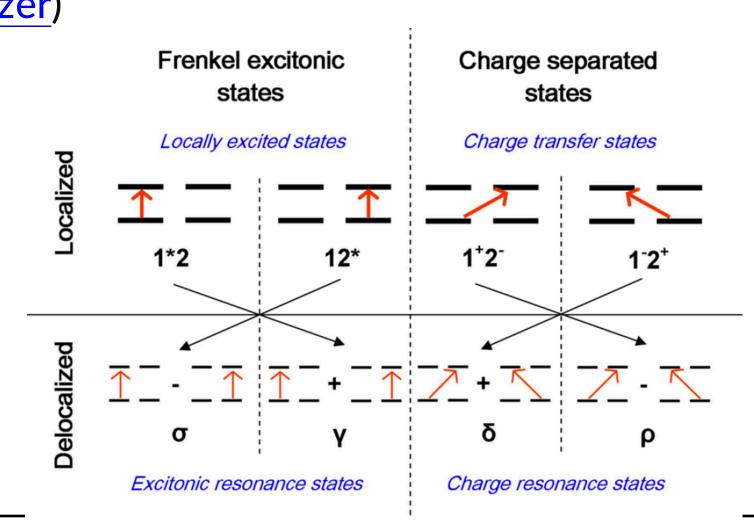


## (Charge transfer) analysis excited state

Charge transfer descriptors

- Exercise 9: Calculate the excitations with SFO analysis + CT descriptors
  - Untick charge transfer integrals
  - Properties -> Excitations: select Allowed only select Allowed
  - Tick SFO Analysis + CT descriptors
  - Save as a different file & run
  - Visualize the spectrum and NTOs of two lowest states (CT)
  - In output search for Plasser (see paper with Lischka)
  - $_{\circ}$  CT = 1 & POS = 1.5 and CTNET = 0 => Charge resonance
  - Very small exciton splitting
  - $_{\circ}$  LAMBDA = 0.4655 => perhaps use hybrid or RSH (Tozer)

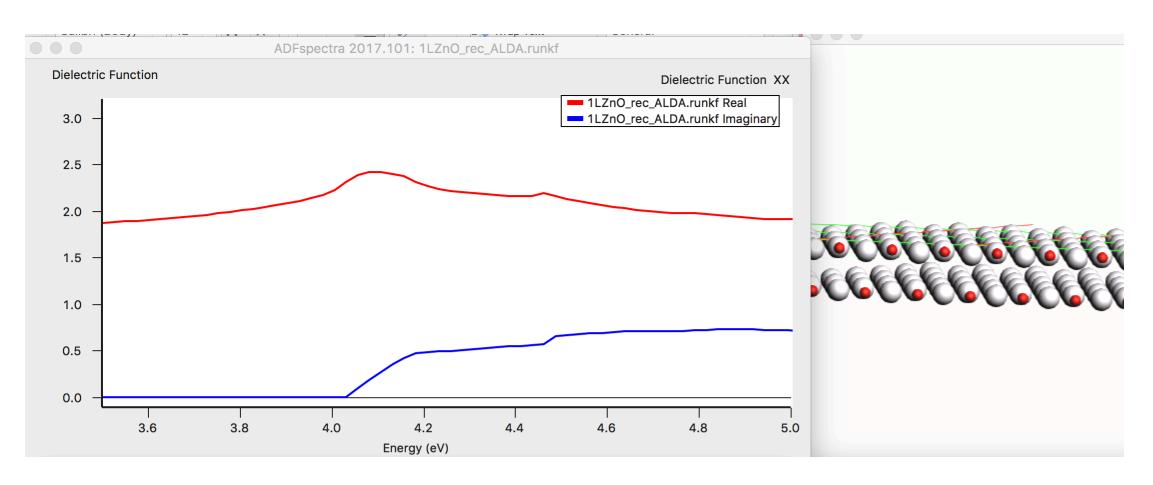
-	•	Plasser, Lisch R HE =	•	.)	
CT =	0.8778	PR = CTNET =	1.9530	СОН =	1.2649
Descriptors	(Peach, Toz	zer, et al.) R HE =			
FRAGMENT ->	FRAGMENT AN	NALYSIS BASED	on SFOs		
Frag ->	Frag	weight n	nu(x) mu	(y) mu(z)	
1 ->	1	0.0698 0.	.0247 -0.00	011 -0.0072	
	2			157 -0.0011	
1 ->		0.5172 -0.	.0082 0.01	182 -0.0066	
2 ->	1	0.3630 -0.	.0089 0.00	099 -0.0036	
711 ->	Δ11	1.0000 0.	0275 0 0	114 _0 0186	

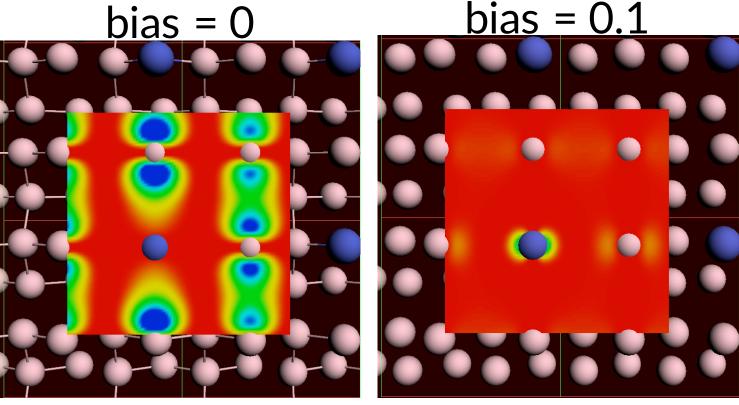




ADFview 2018.105: NTO 1 1

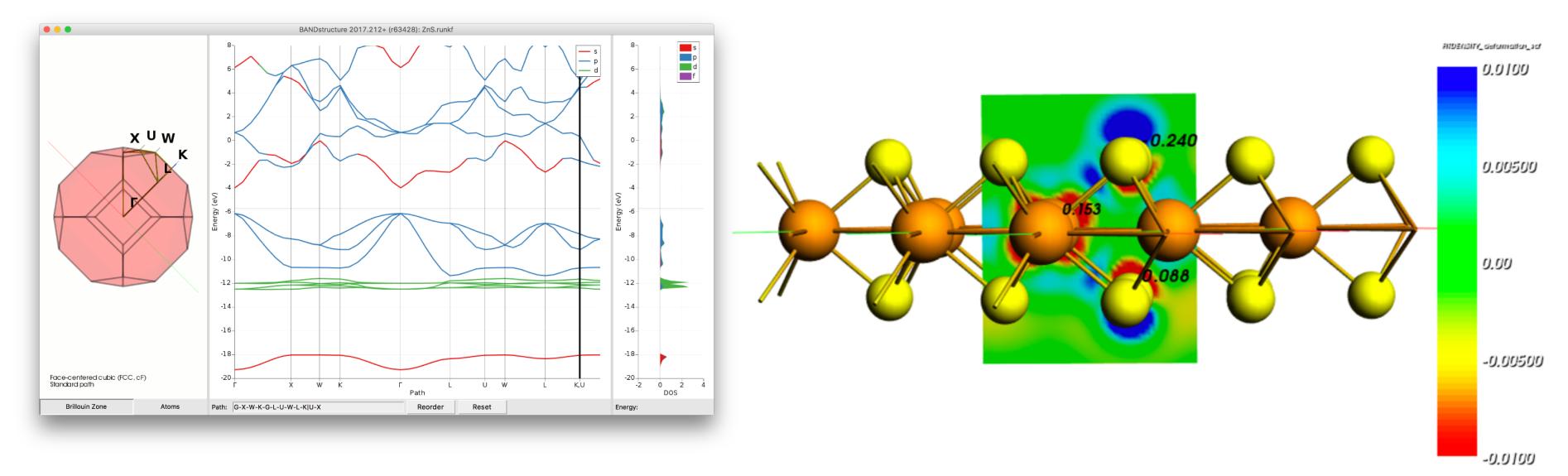
## BAND & QE: Periodic DFT





Dielectric function ML ZnO

STM PtGe(100)



Band structure, pDOS, fat bands ZnS

Polarizing MoS<sub>2</sub> with an electric field



## BAND vs. Plane Wave codes (QE)

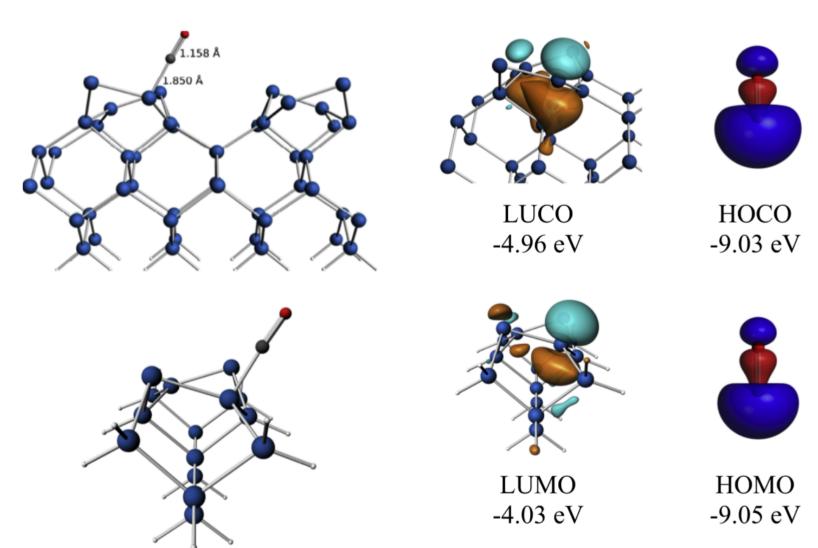
- Atom centered basis functions, STO or NAO
  - Compare cluster with periodic
  - No pseudopotentials, all elements
  - Core spectroscopy (core holes)
  - Easy (orbital, density) analysis with GUI
  - Fast for empty (1D, 2D, porous)
  - o xc: SCAN, MN15-L, HSE06, GLLB-sc, D3(BJ), ...
  - Self-consistent NEGF
    - Gate & bias potential
    - Spin transport



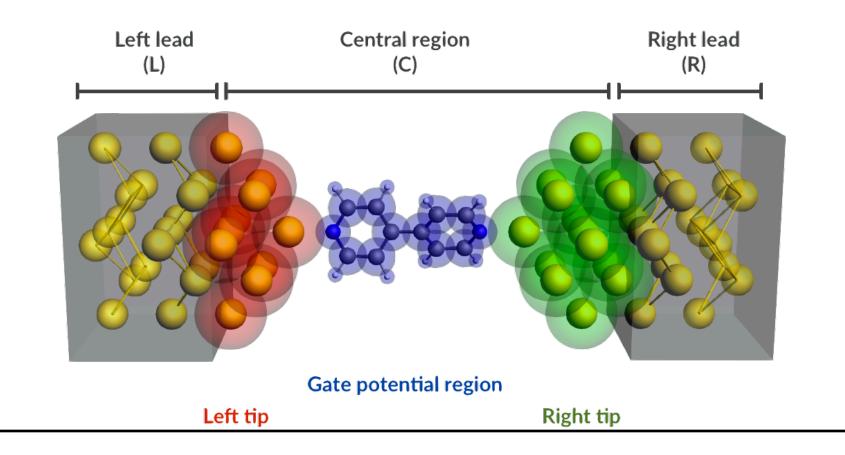
- Het. catalysis: polarization, COSMO
- 2D electronics (homogeneous E field)
- Nanotubes



- Easy set up & analysis
- Switch: ADF, BAND & Quantum Espresso



crystal orbitals, periodic energy decomposition analysis M. Raupach and M. Tonner, J. Chem. Phys. **142**, 194105 (2015)

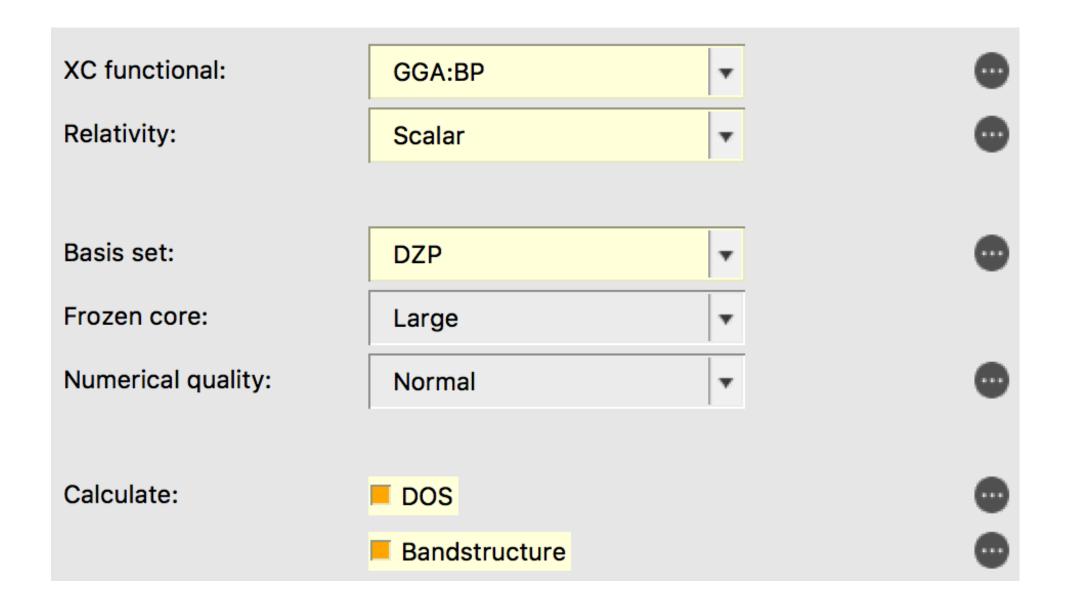




### Band structure, pDOS, fat bands, COOP

#### • Exercise 10: ZnS bulk

- New input, go to BAND
- click on the 'crystal' builder tool in the bottom
- select cubic -> Zincblende and accept the default
- Settings: BP, SR-ZORA, and DZP
- Select DOS and Bandstructure (default interpolation)
- o Run it!

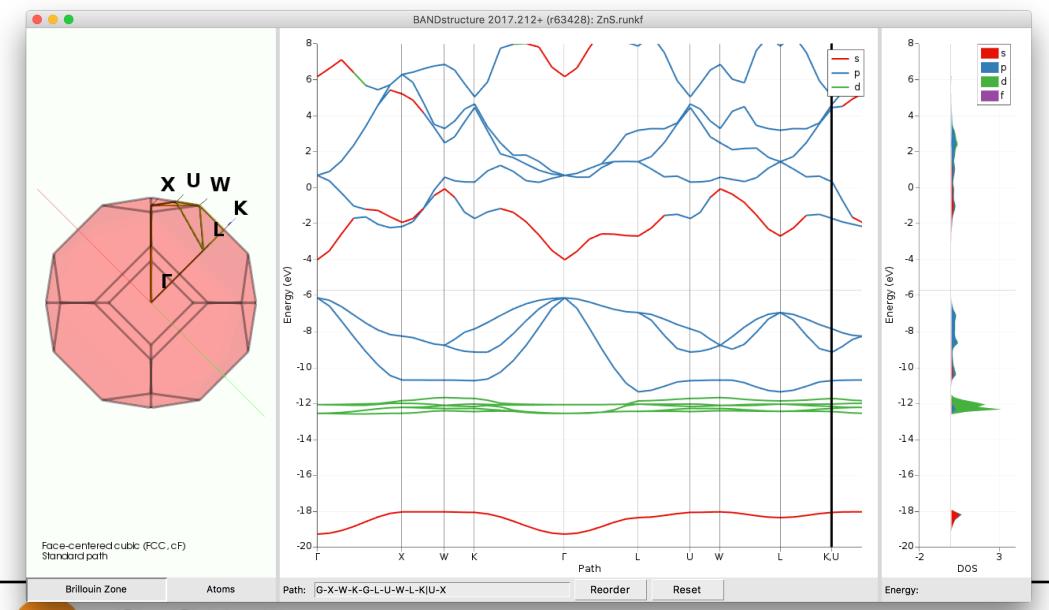




### Band structure, pDOS, fat bands, COOP

#### • Exercise 10: ZnS bulk

- Visualize the band structure (SCM Menu). You will automatically see the pDOS and 'fat bands'
- ZnS is a direct band gap semiconductor (p-s transition)
  - Check the logfile and output for band gap info and kmesh
  - Low band gap: try model potentials (TB-mBJ, GLLB-sc) and HSE06? see benchmark study.
  - Should also be converged wrt kpoints, basis, etc.
- Restart the calculation from SCF and in the DOS details tick 'COOP'
  - Visualize the crystal orbital overlap population between the Zn s and S p orbitals





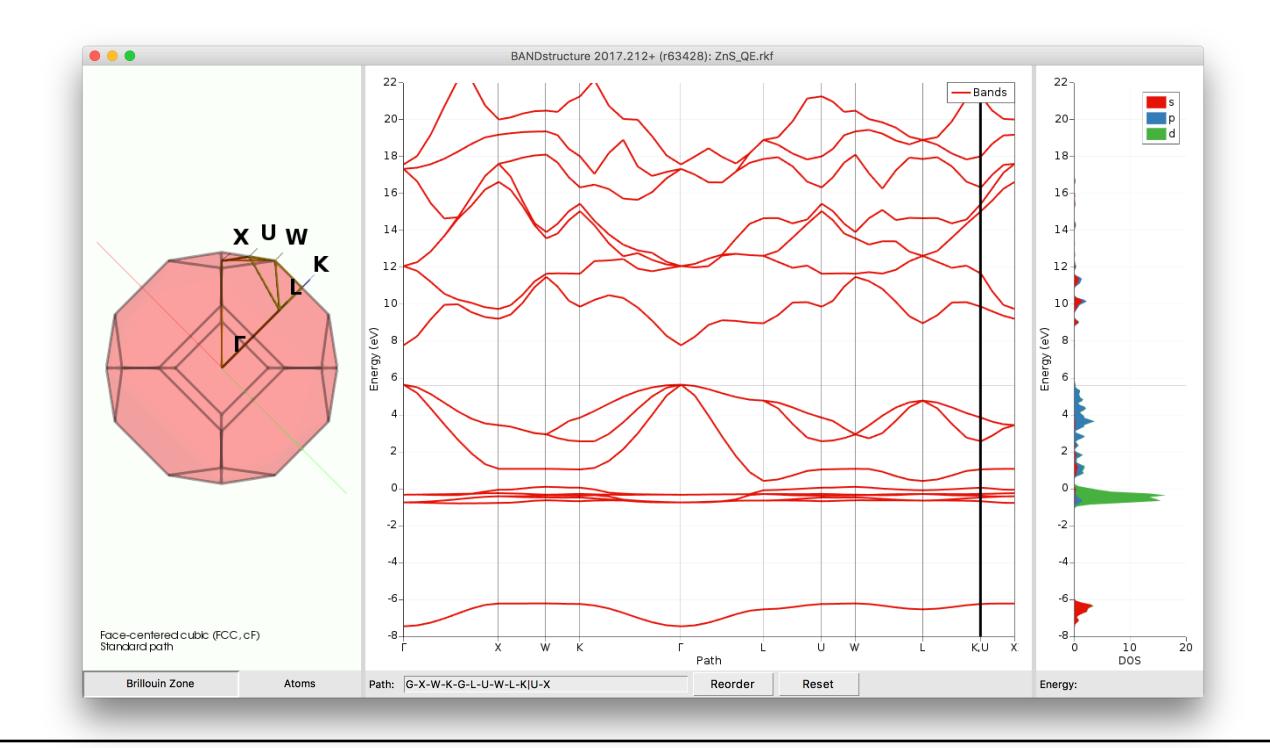
recent JACS study by

Hoffmann: COOP in perovskites

(tutorial)

## Band structure, pDOS with QE

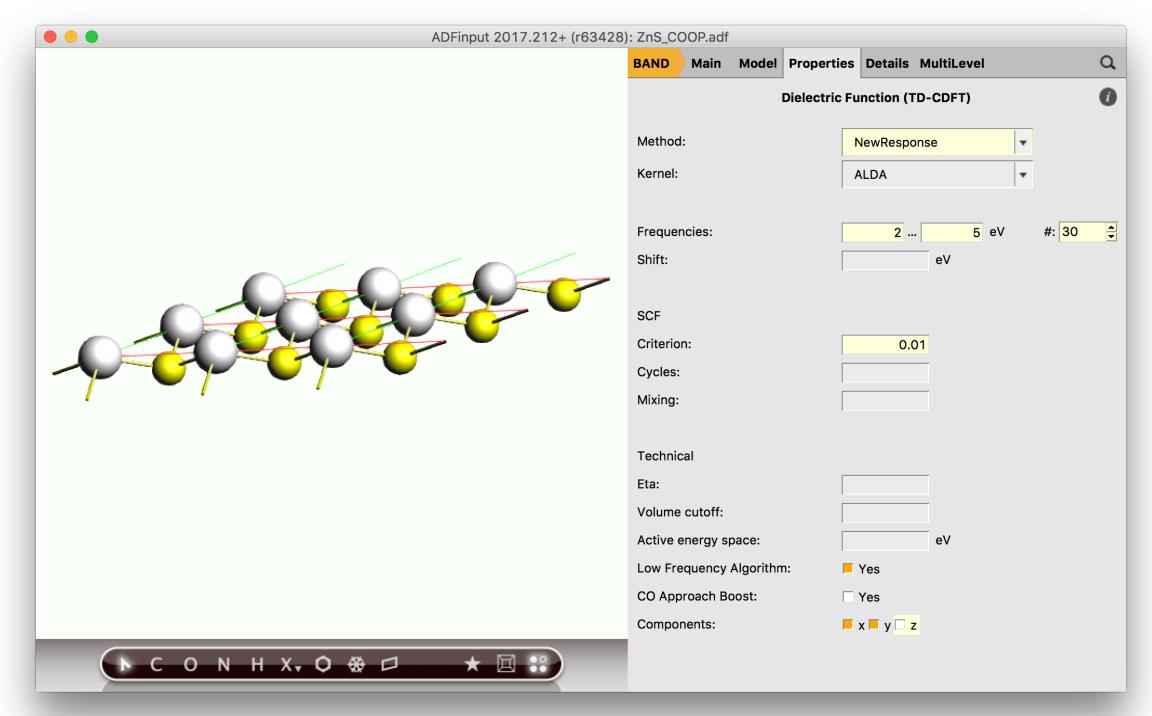
- Exercise 10: ZnS bulk with QE
  - Switch from BAND to Quantum ESPRESSO
  - $\circ$  Choose the same k-mesh (5x5x5), functional and Vanderbilt pseudopotentials
  - You will see a similar band structure, but they aren't colored according to character
    - DOS can be projected by QE





## Surfaces, dielectric function

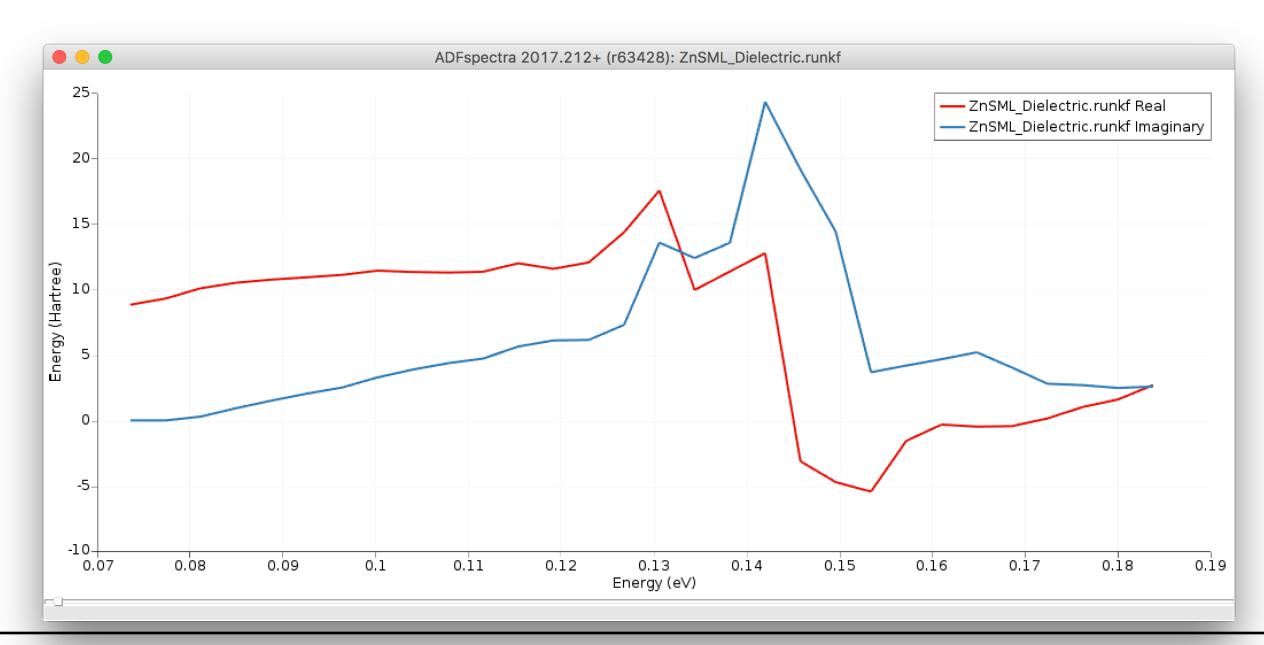
- Exercise 11: ZnS monolayer: 2D-TDCDFT
  - Cut the 111 surface with the slicer tool, and choose 1 layer
  - From properties -> dielectric function choose NewResponse
  - Calculate 30 frequencies between 2-5 eV
  - Set the SCF convergence criterion to 0.01 and switch off the z-component
  - Run it (you will prompted Nosymm is used)





## Surfaces, dielectric function

- Exercise 11: ZnS monolayer: 2D-TDCDFT
  - SCM -> Spectra will show the averaged dielectric function
  - Look at the susceptibility, polarizability and refractive index in Spectra->TDCDFT
  - You could use a 'scissor' shift to upshift the virtuals
  - Converge with respect to k-points!
  - Geometry of the ions should be optimized, this will effect electronic properties
    - For free-standing ML, also optimize lattice ?!



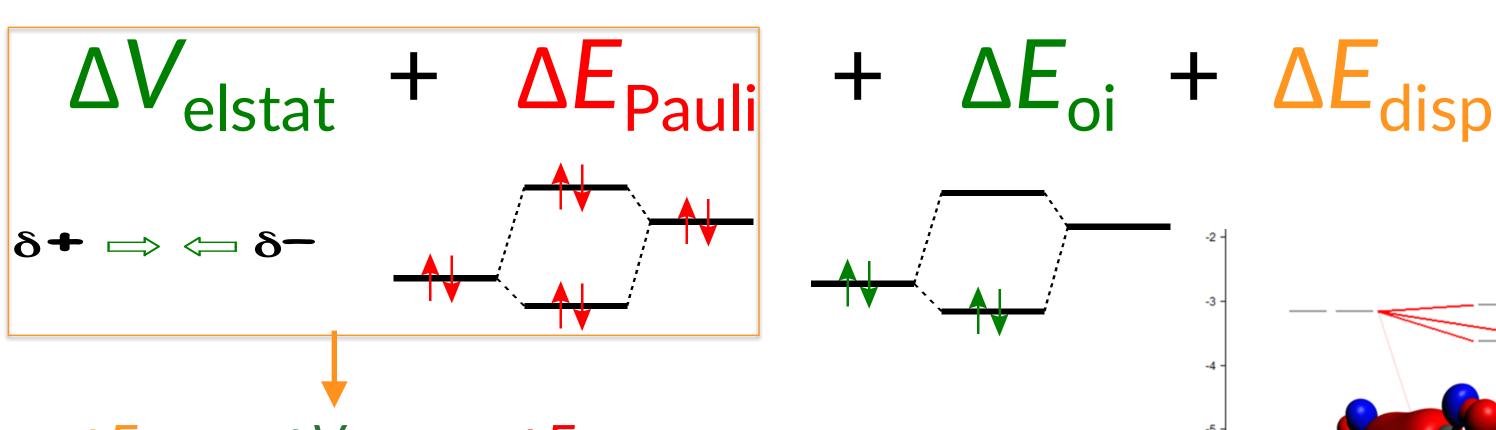


## Energy decomposition analysis

$$\Delta E = \Delta E_{\text{prep}} + \Delta E_{\text{int}}$$

•  $\Delta E_{\text{prep}}$  = geometry 'deformation' energy

Chem. Soc. Rev. 2014, **43**, 4953; WIRES Comput. Mol. Sci. 2015, **5**, 324 Oline tutorial & teaching materials

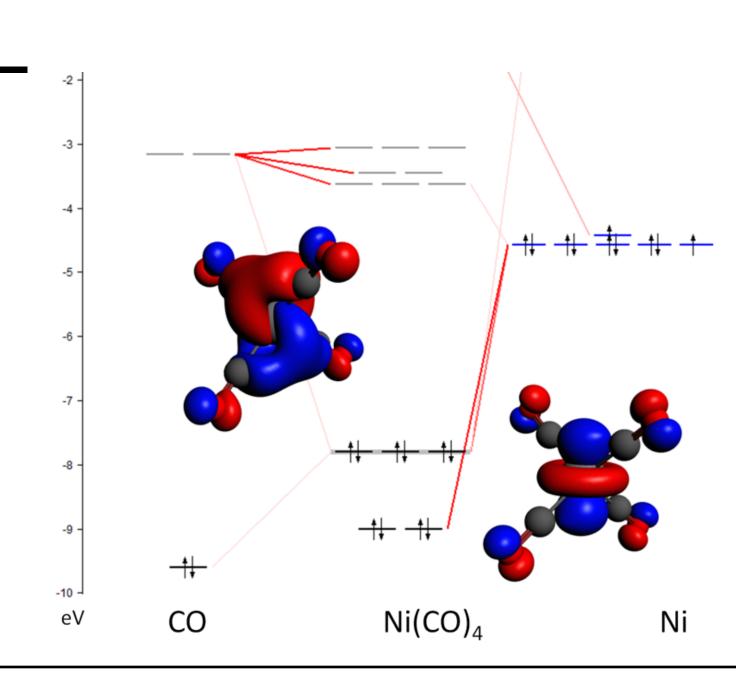




•  $\Delta E_{oi}$  = decomposed in irreps.

#### **Extensions:**

- Periodic EDA: M. Raupach & R. Tonner, J. Chem. Phys. 142, 194105 (2015)): molecule-surface interactions
- Ziegler, Michalak, Mitoraj: ETS-NOCV



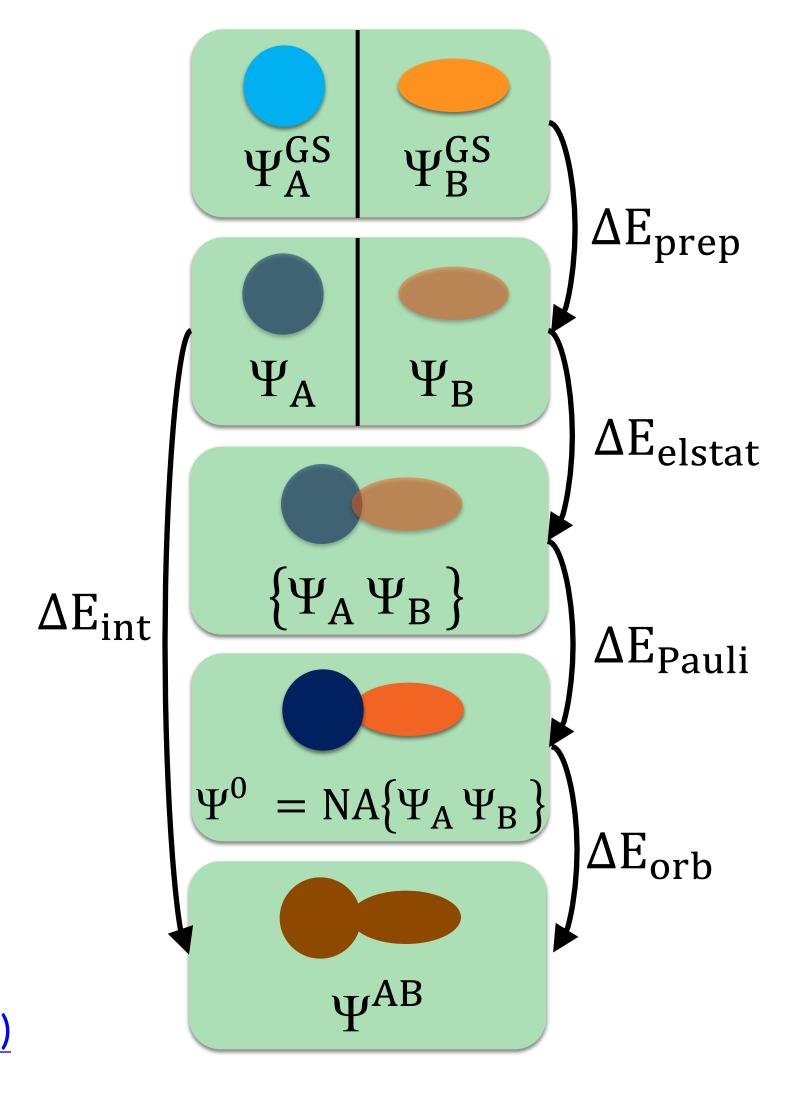


### EDA/ETS-NOCV: bond & density decomposition

- Molecule built from fragments
- Bond analysis with meaningful terms:

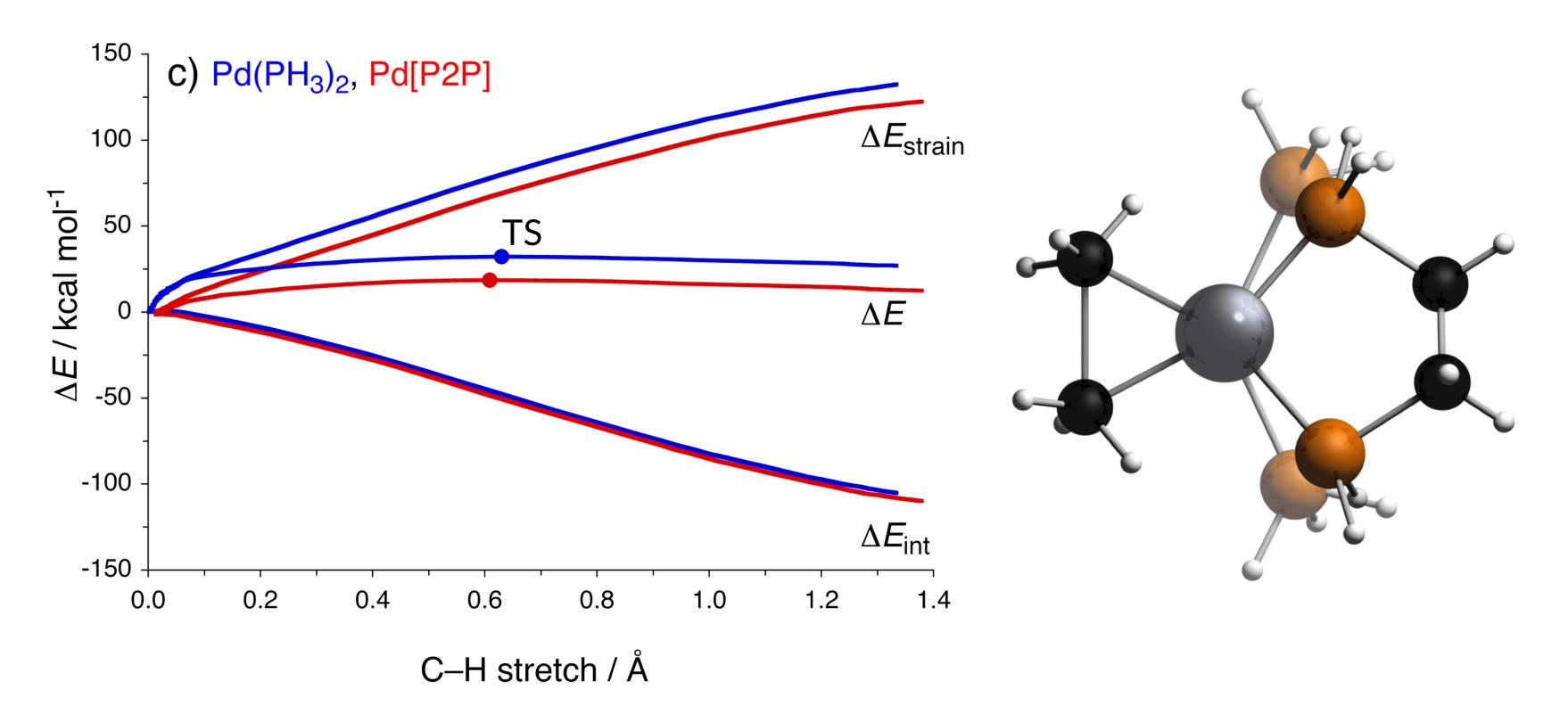
$$\Delta V_{\text{elstat}} + \Delta E_{\text{Pauli}} + \Delta E_{\text{oi}}$$

- Combine with NOCV (webinar)
  - Orbital interactions / charge transfer
- Also periodic 1D, 2D, 3D (tutorials)
  - (Ad)sorption nanotbues, surfaces, MOFs etc.
  - Raupach & Tonner, <u>J. Chem. Phys. 142, 194105 (2015)</u>





#### Catalyst design: activation strain model



Chem. Eur. J. (communication) 2009, 15, 6112 Org. Biomol. Chem. 2010, 8, 3118 Nature Chem. 2010, 2, 417

Latest reviews:

Chem. Soc. Rev. 2014, 43, 4953 WIRES Comput. Mol. Sci. 2015, 5, 324 EDA along the reaction path Bite-Angle Effect: Activation Strain analyses:

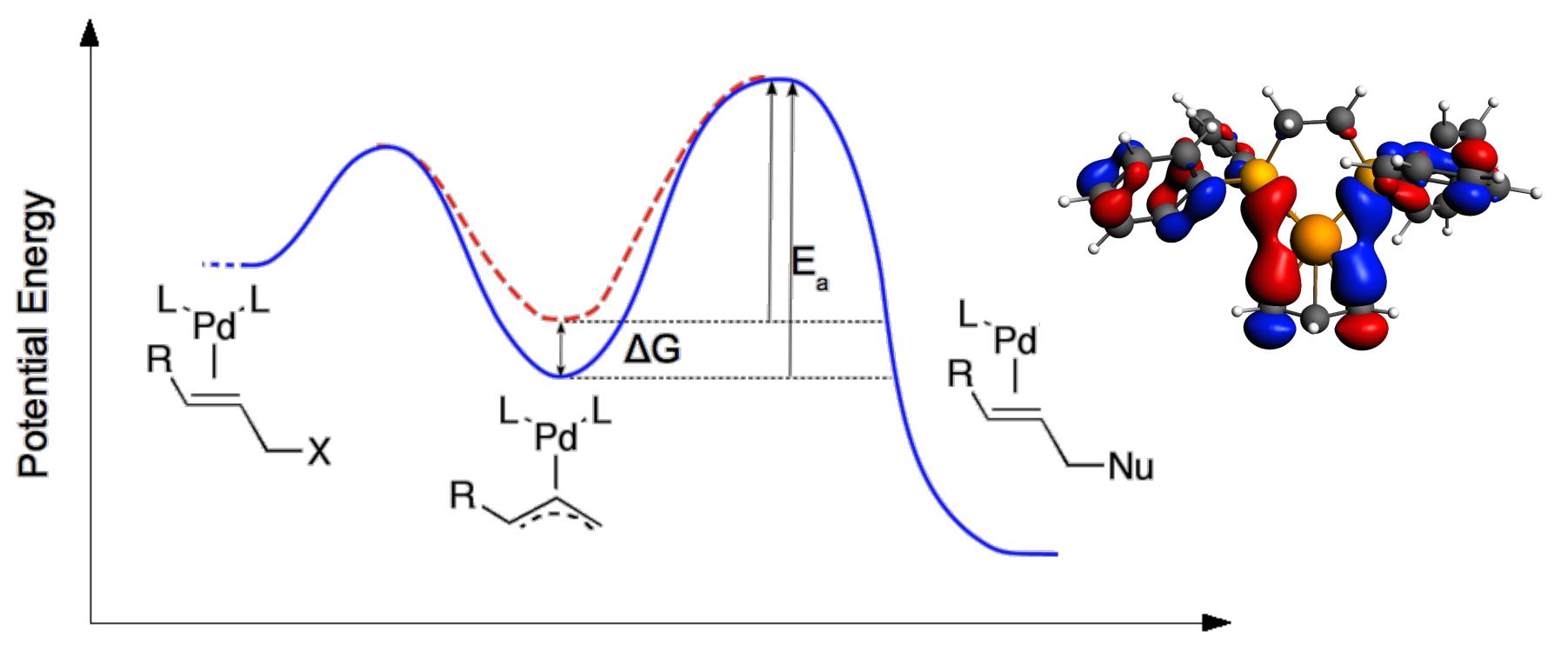
- HOMO-LUMO interaction marginally improved
- But: strain reduced by building it into catalyst



#### Catalyst natural selection: survival of the weakest

Increased bite angle dppe → dppb

- Improved electronic interaction and larger repulsive strain
- Strain outweighs electronic interactions  $\rightarrow$  dppb most destabilized  $\rightarrow$  best catalyst



#### **Reaction Coordinate**

J. Wassenaar, et al. Catalyst selection based on intermediate stability measured by mass spectrometry. Nature Chem. 2, 417 (2010)

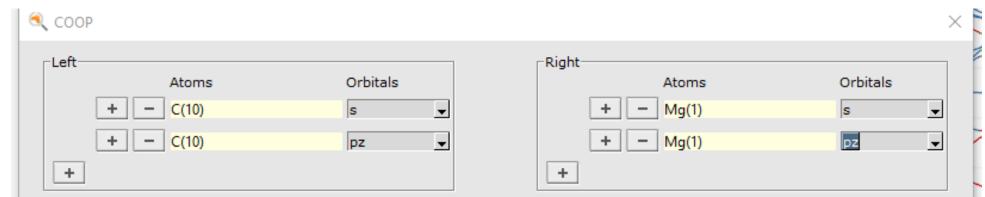


## CO on MgO: COOP

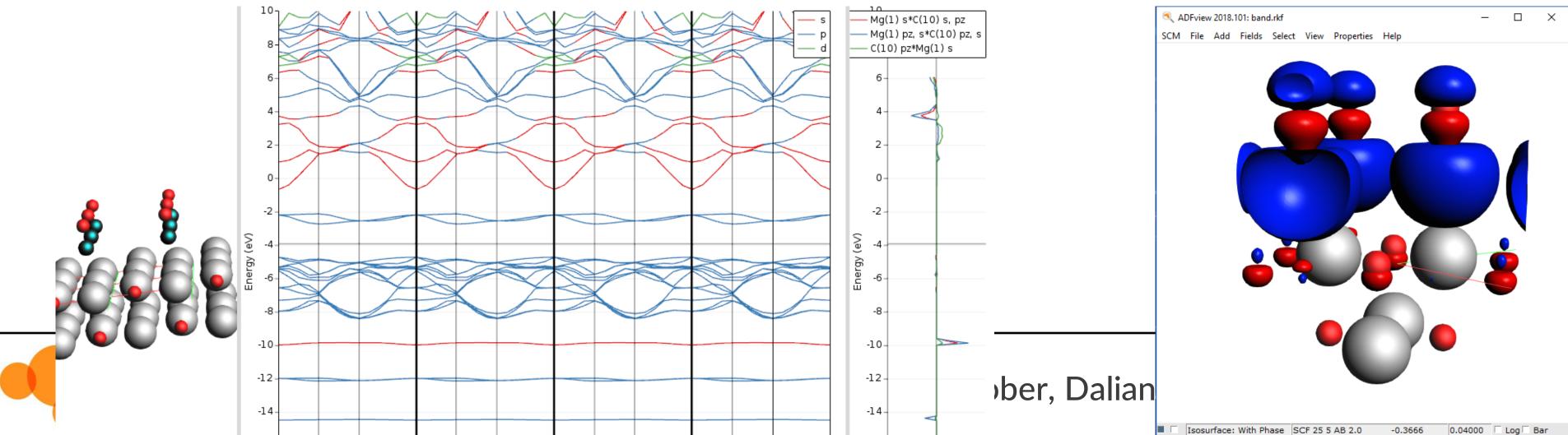
- Exercise 13: First do the PEDA-NOCV tutorial
- New input with same coordinates
- Scalar, PBE, DZP/small, basic, DOS, BandStructure
  - DOS details (...) => COOP, -20 to + 20eV

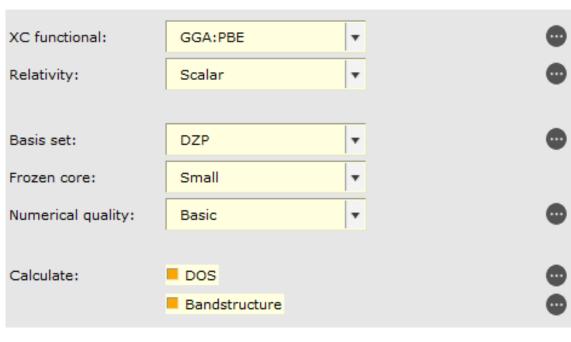


- o DOS-> COOP... Try different Mg (s) + Mg ( $p_z$ ) and C ( $p_z$ ) + C (s) combo's
- $\circ$  What is the biggest contribution to the -10 eV band? Mg(s)+C(s-p<sub>z</sub>); Mg's p<sub>z</sub> helps a bit



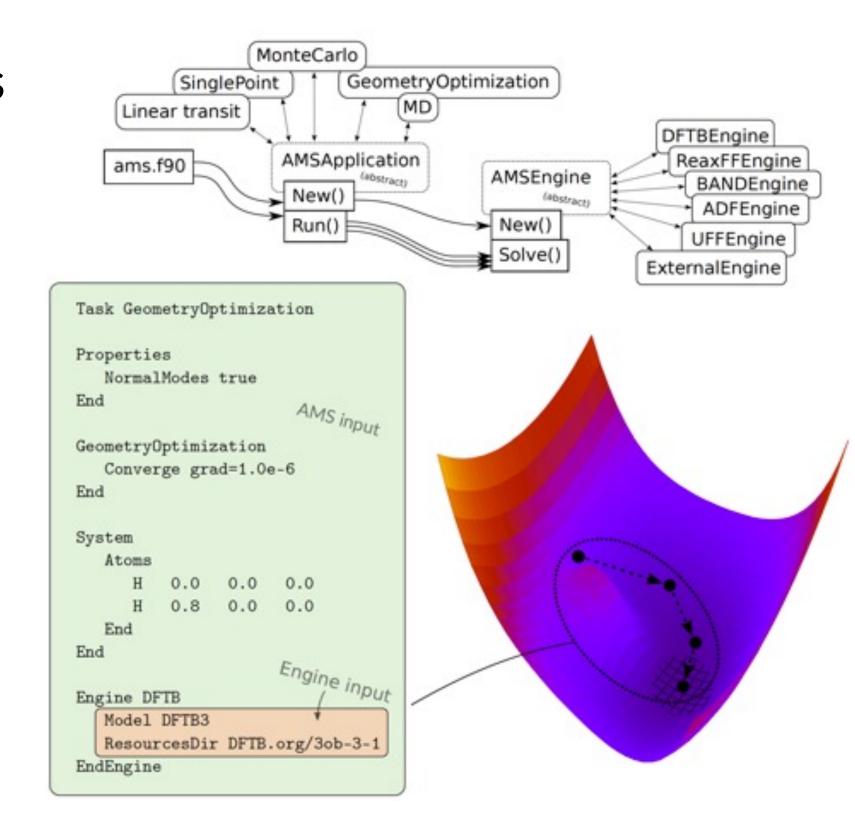
Show MO of that band with ADFview, improve grid & tweak cut-off





## AMS: Powerful driver

- Scan multiple coordinates for any periodicity
  - Works with ADF, BAND, DFTB, MOPAC, UFF, ReaxFF
  - Can be extended/interfaced with other codes
- Advanced MD barostats and thermostats
  - Also with DFT(B): AIMD
  - Other 'ReaxFF' functionality to be ported:
    - fbMC, CVHD, Molecule Gun, GCMC
- Reuse PES information
- Task farming (double parallelization)
  - Elastic tensors, phonons, frequencies



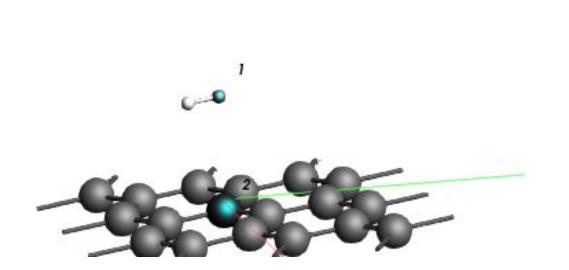


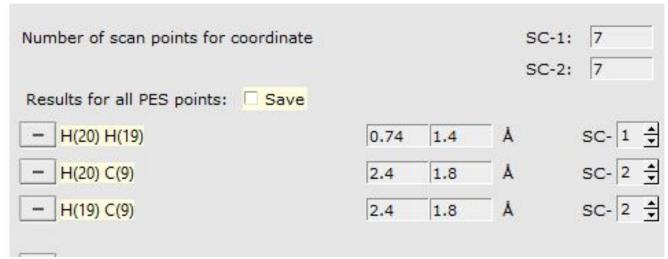
### AMS: PES scan & TS search

- Exercise 14: H<sub>2</sub> on graphene
  - $_{\circ}$  Find graphite (or build it), slice a 1L 001 surface & remove top layer to get graphene
  - Create 3x3 super cell and build H<sub>2</sub> somewhere in the input window
  - Adjust Model -> Coordinates to put it atop, 2.4 above surface

H(19)	0.0	0.37	2.4
H(20)	0.0	-0.37	2.4

- Go to DFTB, choose DFTB3-D3BJ/3ob-3-1 and set Task: PES Scan, click '...'
- In the PES Scan panel set the Scan coordinates by selecting atoms and clicking +:
  - Set SC-1 to H<sub>2</sub> from 0.74-1.4 (7 points)
  - Set SC-2 both H-C coordinates from 2.4-1.8 (7 points)





Lower convergence criteria in Details -> Geometry Optimization

Save & Run

Gradient convergence:

1.0e-2 Hartree/Å

Energy convergence:

1.0e-4 Hartree

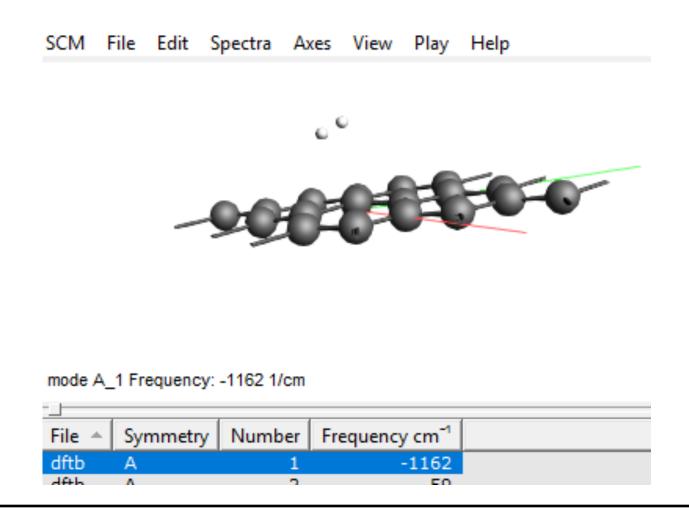
Step convergence:

1.0e-2 Å

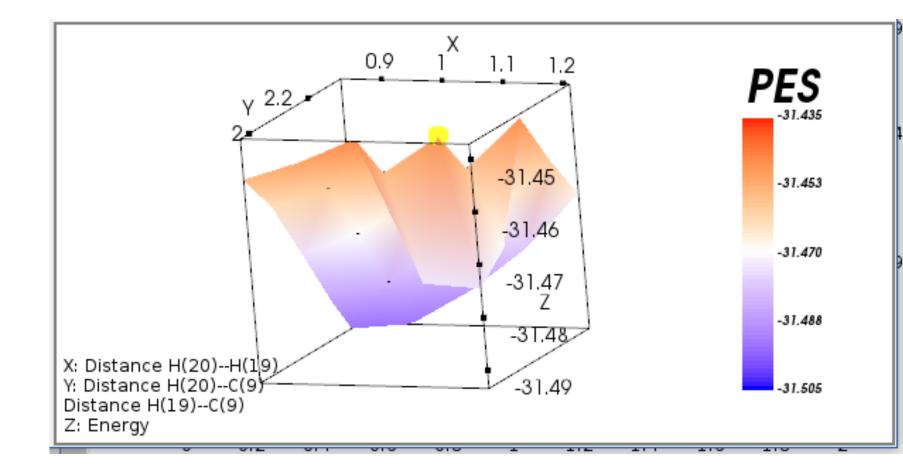


### AMS: PES scan & TS search

- Exercise 14: Explore the 2D PES
  - When Finished: SCM -> Movie
  - $\circ$  Find a plausible TS start point (x=1.18, y = 2.0)
  - File -> Save Geometry
  - Make a New input & import that geometry
  - Now we calculate the Hessian:
  - Go to DFTB, choose DFTB3-D3BJ/3ob-3-1
    - Task: Single point, followed by frequencies
  - Save & Run
  - Visualize the Spectra and check that there is 1 large imaginary mode

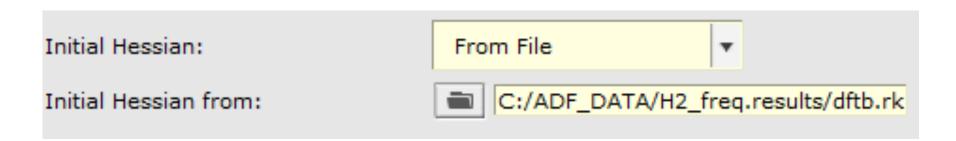




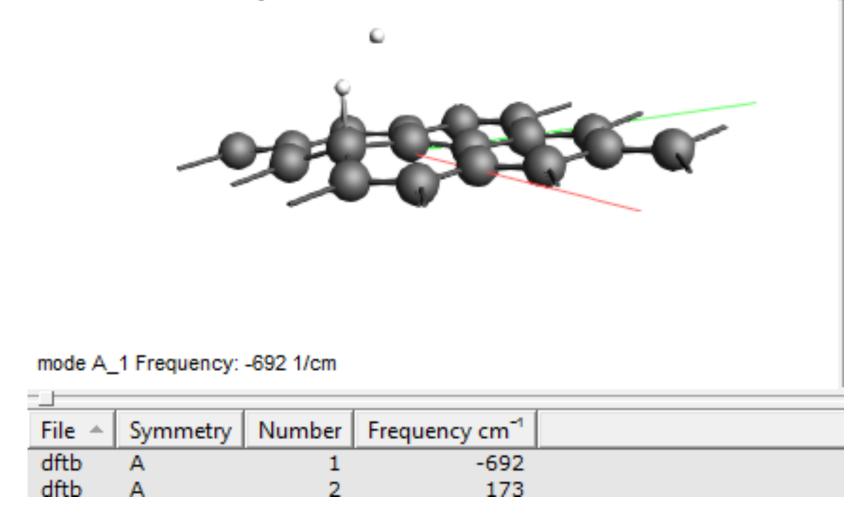


### AMS: PES scan & TS search

- Exercise 14: Find the TS
  - Change Task to Transition State
  - In Details -> Geometry Optimization: Initial Hessian -> From file -> find dftb.rkf



- Save as a different file & Run
- Check the TS geometry and it's normal mode spectrum



 $\circ$  Calculate the activation energy as E(TS) – E (graphene + H<sub>2</sub>) (you can put H<sub>2</sub> at 50A)

