# **ADF Modeling Suite**

# Powerful Computational Chemistry



# Li-Ion Diffusion Coefficients in cathode materials with ADF/ReaxFF



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# Contents

This ReaxFF recipe will demonstrate the following techniques:

- 1. Importing a CIF file from an external database and equilibrating the structure
- 2. Manipulating the structure, e.g. inserting particles
- 3. Creating an amorphous structure by running an annealing program
- 4. Calculate Diffusion Coefficients directly from trajectories
- 5. Calculate low temperature Diffusion Coefficients indirectly by means of extrapolation

# The System

This tutorial will be using two representative Li x S systems. The systems and workflows presented here are originally described in the publication "ReaxFF molecular dynamics simulations on lithiated sulfur cathode materials" by van Duin and coworkers[1] (see <u>here</u>).

To speed up the calculations, we will use downsized  $Li_xS$  systems consisting of only 177 and 281 atoms respectively.



Starting from an experimental structure of Sulfur( $\alpha$ ), Li-atoms were added at random positions followed by an annealing run to create the above amorphous structures.

# 1. Importing the Sulfur( $\alpha$ ) crystal structure

The crystal structure can be directly imported from a CIF file. There are several resources for crystollgraphic data available online and you can choose according to your liking. Here we used the structure of  $S_8$  alpha Sulfur from the <u>American Mineralogist Crystal Structure Database</u>.

The downloaded CIF can now directly be imported into ADFinput:

• File → Import Coordinates

Next, map the coordinates into the ReaxFF unit cell and display the lattice vectors:

- Edit  $\rightarrow$  Crystal  $\rightarrow$  Map atoms to (0..1)
- View → Periodic → Show lattice vectors

Before adding any Li-ions to the system, we need to relax the structure using a full geometry optimization, i.e. including the optimization of lattice vectors.

**Tip:** Most of the time we find that running a couple of hundred steps NPT dynamics at low temperature and 0.0 pressure using a small timestep is sufficient as a relaxation or at least speeds up a subsequent geometry optimization significantly. Try the following settings:

- + × ADFinput 2017.108				
SCM File Edit Select Atoms Bonds View Help				
× 20-66 4	ReaxFF Main Model	Details	Q	
	Task:	Molecular Dynamics 🔹	•	
	Periodicity:	Bulk	٠	
	Force field:	LiS.ff	0	
	Number of iterations:	2500		
	Start with:	0 non-reactive iterations		
	Time step:	0.05 fs		
	Method:	NPT Berendsen		
	Temperature:	5.0 K		
	Damping constant:	100.0 fs		
	Pressure:	0.0 MPa		
	Damping constant:	500.0 fs		
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• Save and Run this calculation

You can inspect the trajectory with ADFmovie and look at the total Energy and pressure, both will drop dramatically already during the very first MD steps.

- Click "yes" when asked if the structure in ADFinput should be updated
- Edit  $\rightarrow$  Crystal  $\rightarrow$  Map atoms to (0..1)

## 2.1 Generating the $\rm Li_{0.4}S$ system

#### Preparation

Open up a new ADFinput draw a single Li-atom and export the geometry to a file called Li.xyz

• File → Export Coordinates

#### Adding atoms

We use the builder functionality of ADFinput to randomly insert 51 Li-atoms into our equibrated Sulfur system:



- Import your previously saved Li-atom by clicking on the folder icon
- Set the Distance to 1.0

Relax this structure by running a geometry optimization or low-T NPT dynamics as with the plain sulfur system.

# 2.2 Generating the $Li_{1.2}S$ system

We'll follow the same strategy for the  $Li_{1,2}S$  system but since now we'll have to add 154 Liatoms we need a bigger unit cell to be able to fit all atoms.

We are creating an amorphous system later on, so we can just enlarge the unit cell in a rather crude way by increasing the z-lattice vector and manually moving part of the sulfur atoms up:

- Edit → Builder
- Set the z-vector to 30.0 Angstrom
- Reduce the Distance to 1.0 Angstrom

<ul> <li>ADFinput 2017.110: sulfur</li> </ul>	-npt.adf		- + ×
SCM File Edit Select Atoms Bonds View Help			
SCM File Edit Select Atoms Bonds View Help	Builder Lattice vectors (Angstrom 10.26158452 0.0 0.0 + Molecules: - Current v - Fill box with: v Random start: Yes Distance: 1.0 Angst	n): 0.0 12.51587278 0.0 154 copies of: f/Old rom	0.0 0.0 30.00 Move to center of box e/LiDiffusion/Li
	Current density: 1.764 g/ New density will be arou	/mL und: 2.229 g/mL	
		Close	Generate Moleculos
► C O N H X, O ↔ □ ★ 🛛 🕄		close	Generate Molecules

• Manually drag the sulfur atoms further apart. Insert the Li-atoms and relax the system as before.



## 3. Creating the amorphous systems

Amorphous systems can be created by slowly annealing up to certain temperature followed by a rapid cooling down.

As in the publication [1] we will anneal up to 1600 K followed by a rapid cool down to room temperature. In order to speed up the calculation, only 30 000 steps are calculated here using the following temperature program:

- 1. From start until step 5000: T = const. = 300 K
- 2. From step 5000 25 000: T = T + dT, dT = 0.065 K/step → Heating up.
- 3. From step  $25\,000 30\,000$ : T = T + dT, dT = -0.26 K/step  $\rightarrow$  Rapid cooling.

•	ADFinput 2017.110: LiS-Annealing.adf					-	+ ×
SCM File Edit Select Atoms Bonds View	Help						
	ReaxFF Main	Model	Details	5			Q
	+ Regime	Te	mperat	ture Reg	ime		0
	Iteration			Т (К)	Damp (	fs) dT (K)	+
		All	-	300.0	100.0	0.0	-
	- 5000						+
	- 25000	AII	•	300.0	100.0	0.065	-
		All	•	1600.0	100.0	-0.26	
		1					

Use the following settings for the MD run:

• NVT Berendsen, 300K, 30 000 steps

After the annealing run has finished:

- Accept the new structures
- Equibrate with the following settings NPT (Berendsen), 300 K, 0.1 MPa, 40 000 steps



You can check the temperature during the annealing run with ADFmovie

The density of the system should at least be somewhat equilibrated for the following calculations of the diffusion coefficients:



# 4.1 Calculating the diffusion coefficients - ReaxFF

In order to calculate a diffusion coefficient with the TRAVIS program[2] an additional keyword needs to be set in the ReaxFF control file. This keyword will cause ReaxFF to calculate a non-backtranslated trajectory from which the mean square displacements are calculated.

We will use a minimal proof-of-principle setup of 10 000 equilibration steps followed by only 100 000 production steps. The settings are:

- NVT, 1600K, 110 000 steps, 10 000 non-reactive steps, timestep 0.25 fs
- The framerate can be set to 50 fs:

Details  $\rightarrow$  Molecular Dynamics  $\rightarrow$  KF result file: 200

Finally, edit the run file that is about to be created from the GUI.

• Details → Run Script

Use the scroll bar on the right to scroll past the definition of the geometry to the MD settings found at the end of the file. In the section starting with #MD-Parameters add the following (new!) line:

- 1 iout3 Create moldyn.xxx files (0: yes, 1: no)
- 1 itrout Calculate a not-back translated trajectory
- 0 ivels 0: Use velocities from vels restart-file;

- ADFinput 2017.110: LiS16	i00.adf		×
SCM File Edit Select Atoms Bonds View Help			
	ReaxFF Main Mode	l Details	Q
- WX II		Run Script	0
	Run script: Auto up 3 IXMU U XMU 1 ichupd Cha 4 icharg alv 600.00 mdtemp MU 100.0 tdamp1 1st 0 icheck 0: 10000 nrstep MD-parameters 1 imdmet MD- 0.250 tstep MD- 0.100000 mdpres 500.0 pdamp1 Ber 0 inpt 0: 10000 nmdit Num 00000 iout1 Out 1 iout3 Cre 1 itrout Cal 0 ivels 0: 2000 iout6 Fre 200 iout7 Fre 200 iou	date Front 0: Xy2 onty, 1: Xy2 + Vets + m arge update frequency ways 4: Full ystem EEM D-temperature (K), unless tregime fi t Berendsen/Anderson temperature dam Normal run 1:Check first derivative -method. 1:Velocity Verlet+Berendsen time step (fs) MD-pressure (MPa) rendsen pressure damping constant (f Change all cell parameters in NPT-r mber of MD-iterations tput frequency to unit 71 and unit 7 ve coordinates (xmolout, moldyn.vel, sate moldyn.xxx files (0: yes, 1: no lculate non-backtranslated trajector Use velocities from vels restart-fi equency of reavout.kf writing equency of removal of rotational and . of iterations in previous runs nge for back-translation of atoms ou d point criterium for MM energy mini	su 3)))/lc
	1 imaxmo 0: 10000 imaxit Max	conjugate gradient, -1: temperature kimum number of iterations	1
►СОNНX,0%□ ★ 🛛 🕄	1 iout4 Fre	equency of structure output during m	i

Save and run the calculation.

#### 4.2 Calculate the Diffusion Coefficients - TRAVIS

Open a terminal and navigate to the folder you ran your calculations in. Just use your default Linux shell or – when on Windows – follow <u>these steps</u> to get into a Unix-like shell shipped with ADF.

First convert the binary rxkf file into a non-backtranslated xyz trajectory for usage with TRAVIS. Assuming your calculation was called LiS-1600, run the following command *on a single line*:

\$ADFBIN/startpython
\$ADFHOME/scripting/trajectory\_analysis/kf\_to\_xyz.py
LiS-1600.rxkf -itrout

This will create a file named LiS-1600\_travis.xyz which can be processed by TRAVIS.

Open the interactive settings dialogue of TRAVIS by running the following command:

#### \$ADFBIN/travis.exe -p LiS-1600\_travis.xyz

Enter the following settings in the TRAVIS dialogue:

- Use the advanced mode  $[...] \rightarrow$  yes
- Use periodic boundary conditions? → no
- Execute molecule recognition for which time step (-1 = disable)?  $[0] \rightarrow -1$
- Accept the defaults until you reach the "List of functions"



- Select the Mean Square Displacement / Diffusion Coefficients → msd
- Use advanced mode for the main part?  $\rightarrow$  no
- Enter the length of one trajectory time step in fs?  $\rightarrow$  50
- Which molecule should be observed?  $\rightarrow$  Choose Li
- Which atom(s) to take from OM Li (e.g. "C1,C3-5,H", "\*"=all)? [#2] → ENTER
- Enter the resolution (=depth) for this MSD (in time steps):  $[601] \rightarrow 60$
- Accept defaults until asked at which timestep to start the analysis  $\rightarrow$  50
- Accept all defaults until the calculation starts.

TRAVIS will write a logfile (travis.log) it's settings (input.txt) and resultsfiles called msd\_Li\_#2.csv the latter one contains three columns:

(1) Time[ps] (2)  $MSD[pm^2]$  (3) Derivative of the MSD

Plotting either column 1 against column 2/3 will give the graphs shown below.

The Diffusion coefficients are calculated from a linear regression and are printed on the screen as well as in the logfile travis.log. With the described settings the Diffusion Coefficient for the Li<sub>1.2</sub>S system at 1600 K is calculated as:

Diffusion coefficient D =  $50680.522077 \text{ pm}^2/\text{ps}$ =  $5.06805\text{E}-08 \text{ m}^2/\text{s}$ .

If you want to re-run the exact same analysis, e.g. on a trajectory at a different temperature, you can ask TRAVIS to re-use the settings from a previous inputfile as follows:

\$ADFBIN/travis.exe -p LiS-1600\_travis.xyz -i input.txt

The MSD for Li in  $Li_{1,2}S$  system at 1600 K looks as follows:



The same analysis can also be applied to the sulfur atoms, yielding a Diffusion Coefficient of D =  $3.16567E-08 \text{ m}^2/\text{s}$ .

#### 6. Extrapolate to Diffusion Coefficients at room temperature

Since the Diffusion Coefficients a very small for the Li<sub>x</sub>S systems calculating the Diffusion Coefficient at 300K would require a very, very long trajectory. However, it is possible to provide an upper bound to the Li diffusion by means of extrapolation from elevated temperatures using the following Arrhenius equation<sup>[1]</sup>.

$$D(T) = D_0 \exp(\frac{-E_a}{T})$$

where  $D_0$  is the pre-exponential factor,  $E_a$  is the activation energy, k is the Boltzmann constant, and T is the temperature. The activation energy and pre-exponential factors can then be obtained from an Arrhenius plot of D(T) against 1/T.

In order to extrapolate the Diffusion Coefficients for Li<sub>0.4</sub>S and Li<sub>1.2</sub>S we calculate trajectories for at least four different temperatures (600K, 800K, 1200K, 1600K) for each system summing to eight trajectories in total using the exact same settings discussed above:

x in Li <sub>x</sub> S	D(T = 600K)	D(T = 800K)	D(T = 1200K)	D(T = 1600K)
0.4	2.5684e-09	7.27756e-09	2.25917e-08	5.06805e-08
1.2	8.1066e-11	2.28662e-10	4.68517e-09	1.45497e-08



We used a Levenberg-Marquardt algorithm w. statistical weighting from QtiPlot to fit the Arrhenius equation and obtained the following Diffusion Coefficients:

 $Li_{0.4}S: D(T = 300K) = 3.31x10^{-7}exp(-3081/300) = 1.15x10^{-11} m^2 s^{-1}$ 

#### $Li_{1.2}S: D(T = 300K) = 6.41x10-07exp(-6072/300) = 1.04x10^{-15} m^2 s^{-1}$

When comparing these values with the Diffusion Coefficients presented in publication by A. van Duin and coworkers [1] it becomes clear that the Diffusion Coefficient of the dense Li<sub>1.2</sub>S system is one or two orders of magnitude too high. Clearly this is due to the minimal setup (small systems and short trajectories). Nonetheless, the Li<sub>0.4</sub>S system as well as the overall trend are already nicely reproduced.

#### References

**1.** M. M. Islam, A. Ostadhossein, O. Borodin, A. T. Yeates, W. W. Tipton, R. G. Hennig, N. Kumar, and A. C. T. van Duin, *ReaxFF molecular dynamics simulations on lithiated sulfur cathode materials*, Phys. Chem. Chem. Phys. **17**, 3383-3393 (2015).

**2.** TRAVIS is commandline tool for analyzing and visualizing trajectories from all kinds of Molecular Dynamics or Monte Carlo simulations initiated by Kirchner, written and developed by Brehm and Thomas (<u>J. Chem. Inf. Model. 51(8) (2011</u>).

TRAVIS is included as a commandline tool from on ADF2017.

Questions and comments: support@scm.com