



# **COSMO-RS**

## **fluid thermodynamics**

Erik van Lenthe

# Outline

Methods to describe liquids

COSMO-RS

- ADF COSMO result files

- $\sigma$ -profiles

- interaction energy

- chemical potential

COSMO-SAC 2013-ADF

Binary mixture examples

Properties

# Methods to describe liquids

Molecular dynamics, Monte Carlo simulations

MM or DFT potential energy surface

expensive, often not feasible

COSMO-RS, COSMO-SAC

molecule is represented by charge density on DFT COSMO surface

statistically thermodynamics of (fitted) interacting surfaces

general applicable

UNIFAC, UNIQUAC

molecule is sum of functional groups

properties functional group fitted to experiment

can fail for complex molecules

# COSMO-RS

## COnductor-like Screening MOdel for Realistic Solvents

A. Klamt J. Phys. Chem 99 (1995) 2244, A. Klamt et al. J. Phys. Chem. A 102 (1998) 5074

C.C. Pye, T. Ziegler, E. van Lenthe, J.N. Louwen, Can. J. Chem. 87 (2009), 790

thermodynamic properties, equilibrium

calculates chemical potential

$\sigma$ -profile: charge density on COSMO surface

electrostatic interactions + hydrogen bond

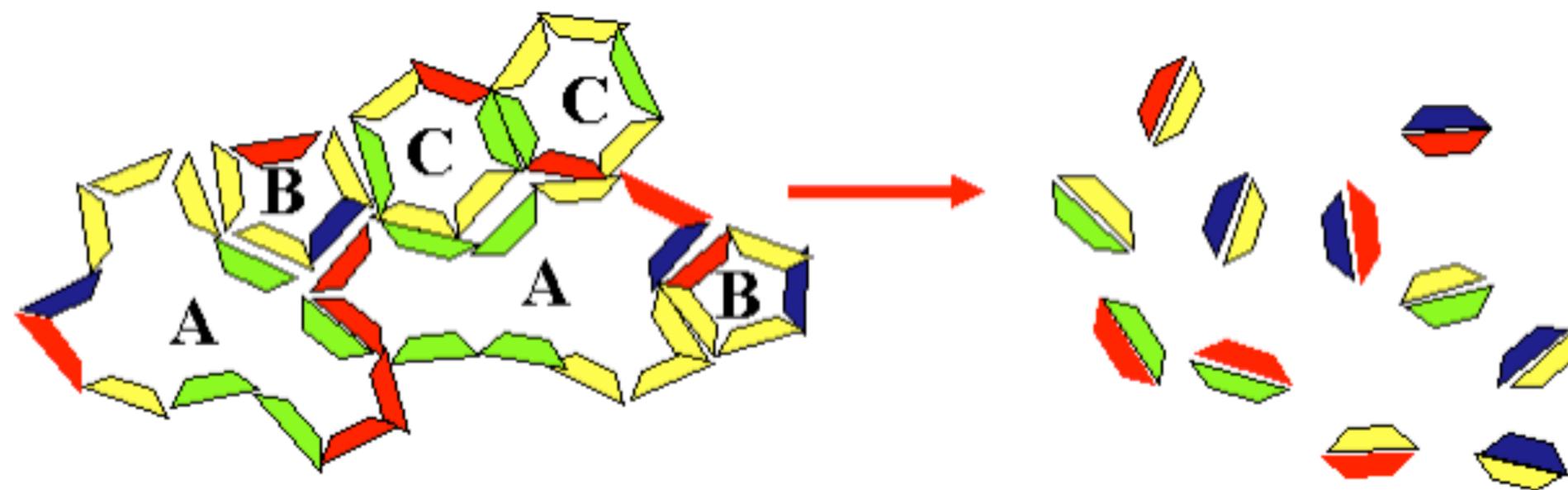
effective contact area

random mixture approximation

Boltzmann distribution

COSMO-RS parameters

fitted to experiment



# COSMO

## **C**Onductor-like **S**creening **M**odel

solute embedded in molecule-shape cavity

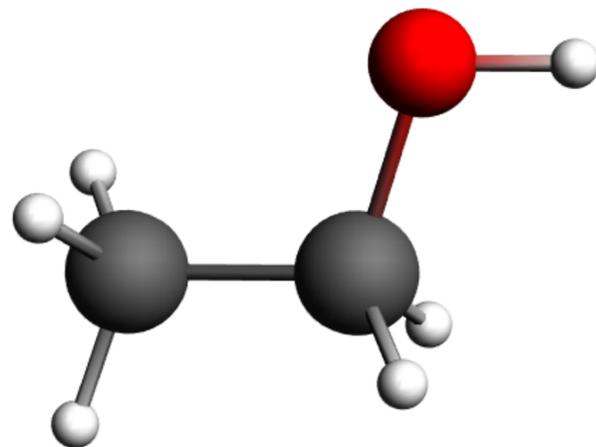
element specific COSMO-radii

dielectric continuum outside cavity

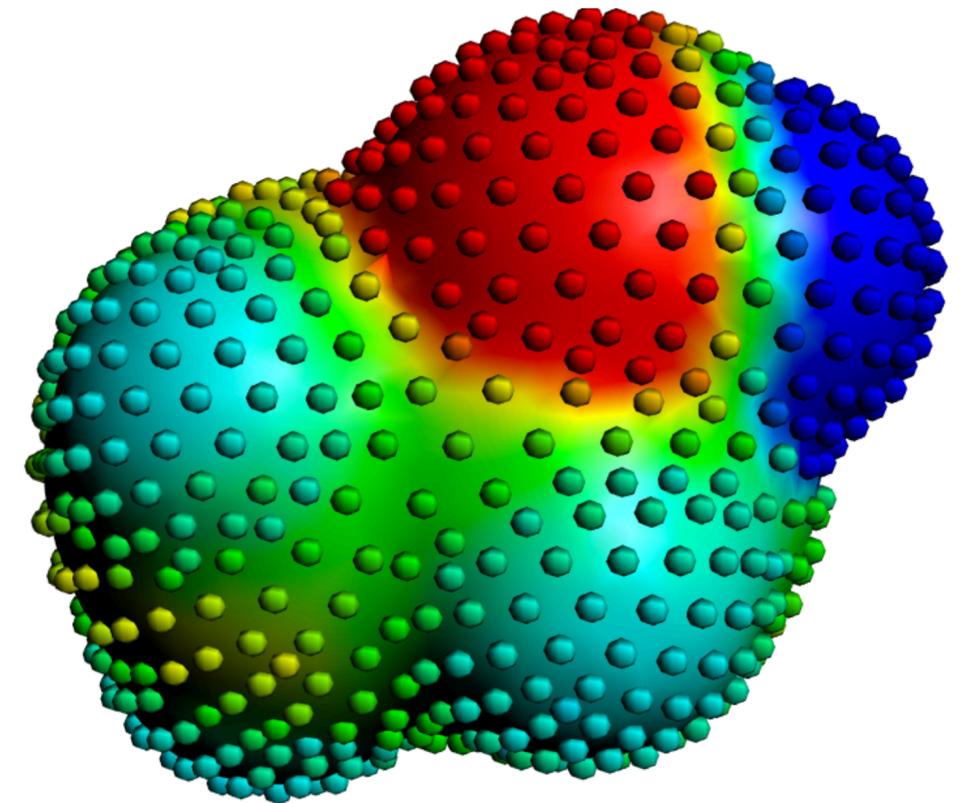
for COSMO-RS: ideal conductor  $\epsilon = \infty$

COSMO volume predicts molar density

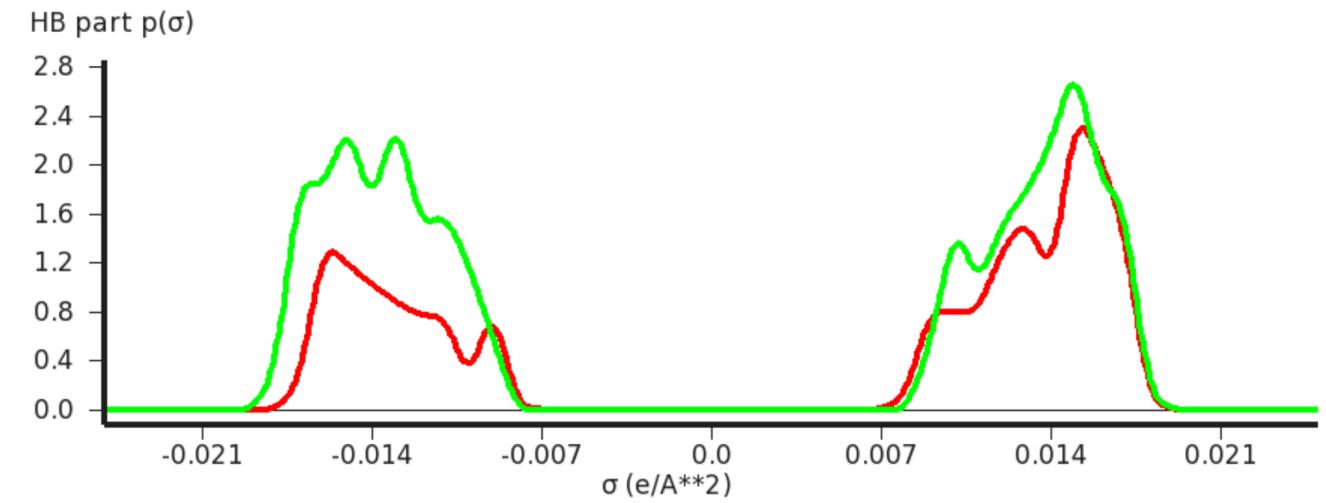
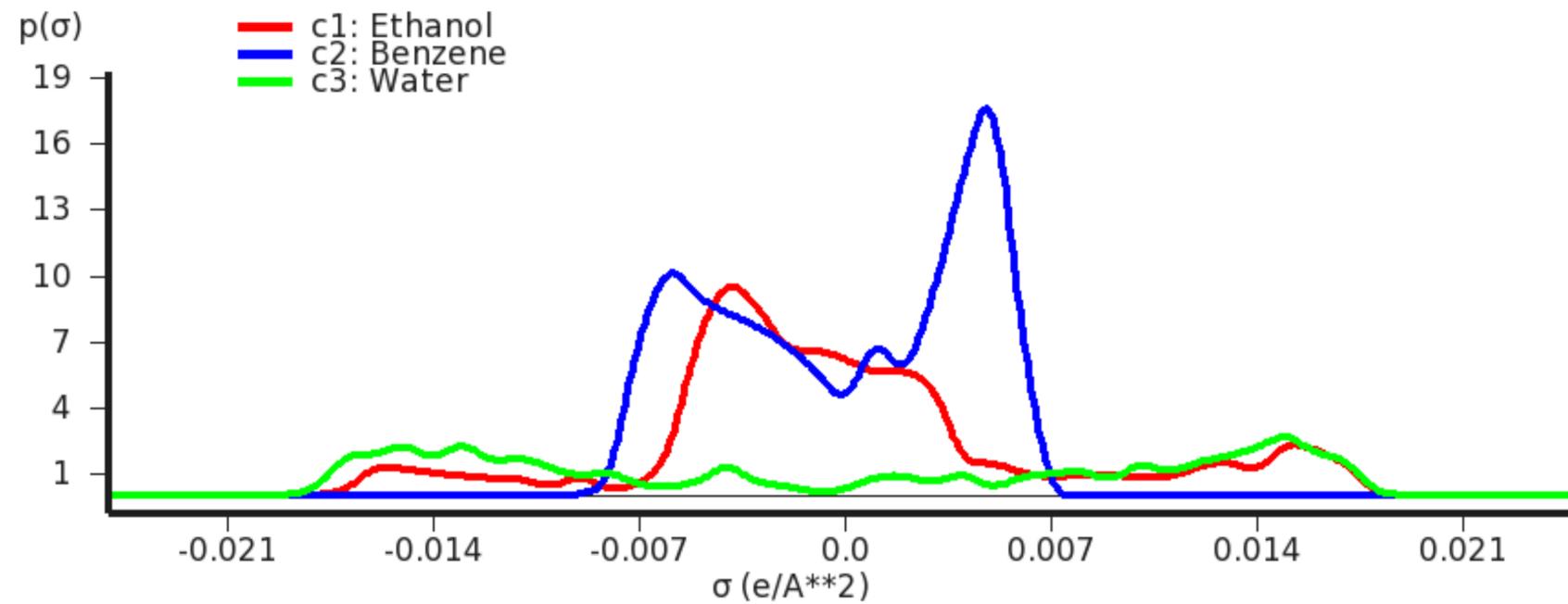
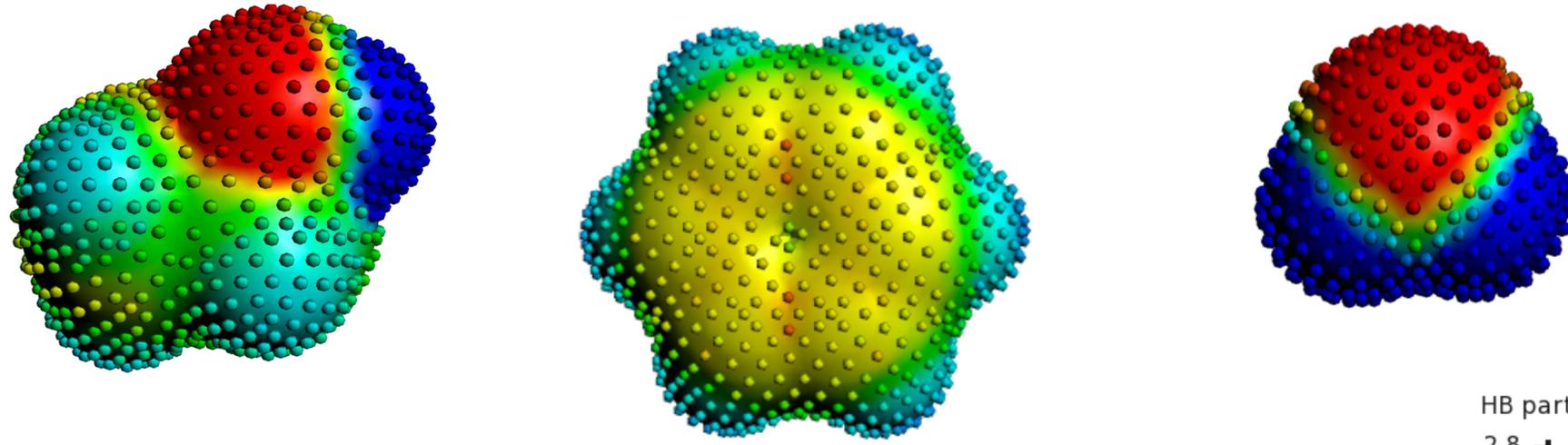
ADF DFT molecular calculation (BP86 TZP)



Ethanol



# $\sigma$ -profiles



# Interaction energy

pair-wise interactions, random mixture approximation

misfit energy repulsive

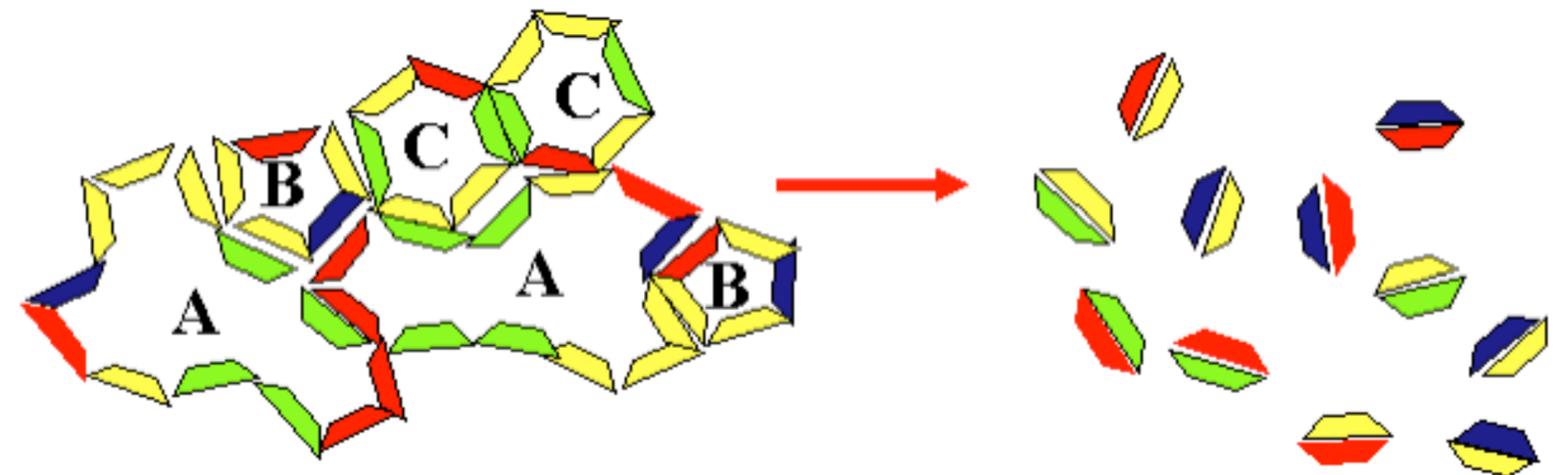
$$E(\sigma, \sigma') = c (\sigma + \sigma')^2$$

hydrogen bonding attractive

$$E_{\text{HB}}(\sigma_{\text{HB}}, \sigma_{\text{HB}}')$$

$\sigma$ -potential

$$\mu_s(\sigma) = -kT \ln \sum_{\sigma'} p_s(\sigma') \exp\{[-E_{\text{int}}(\sigma, \sigma') + \mu_s(\sigma')]/kT\}$$



# Chemical potential

liquid chemical potential

$$\mu^i = \sum_{\sigma} p^i(\sigma) \mu_s(\sigma) + \text{combinatorial terms}$$

gas chemical potential

$$\mu_{\text{gas}}^i = E_{\text{gas}}^i - E_{\text{cosmo}}^i + \text{dispersion terms}$$

solid chemical potential

$$\mu_{\text{solid}}^i - \mu^i = \Delta G_{\text{fus}}$$

activity coefficients shows deviations from ideal

$$\gamma_s^i = \exp \{(\mu_s^i - \mu_{\text{pure}}^i)/kT\}$$

vapor pressure pure compound

$$p = \exp \{-(\mu_{\text{gas}}^x - \mu_{\text{pure}}^x)/kT\}$$

# COSMO-RS in ADF

## COSMO-RS in ADF

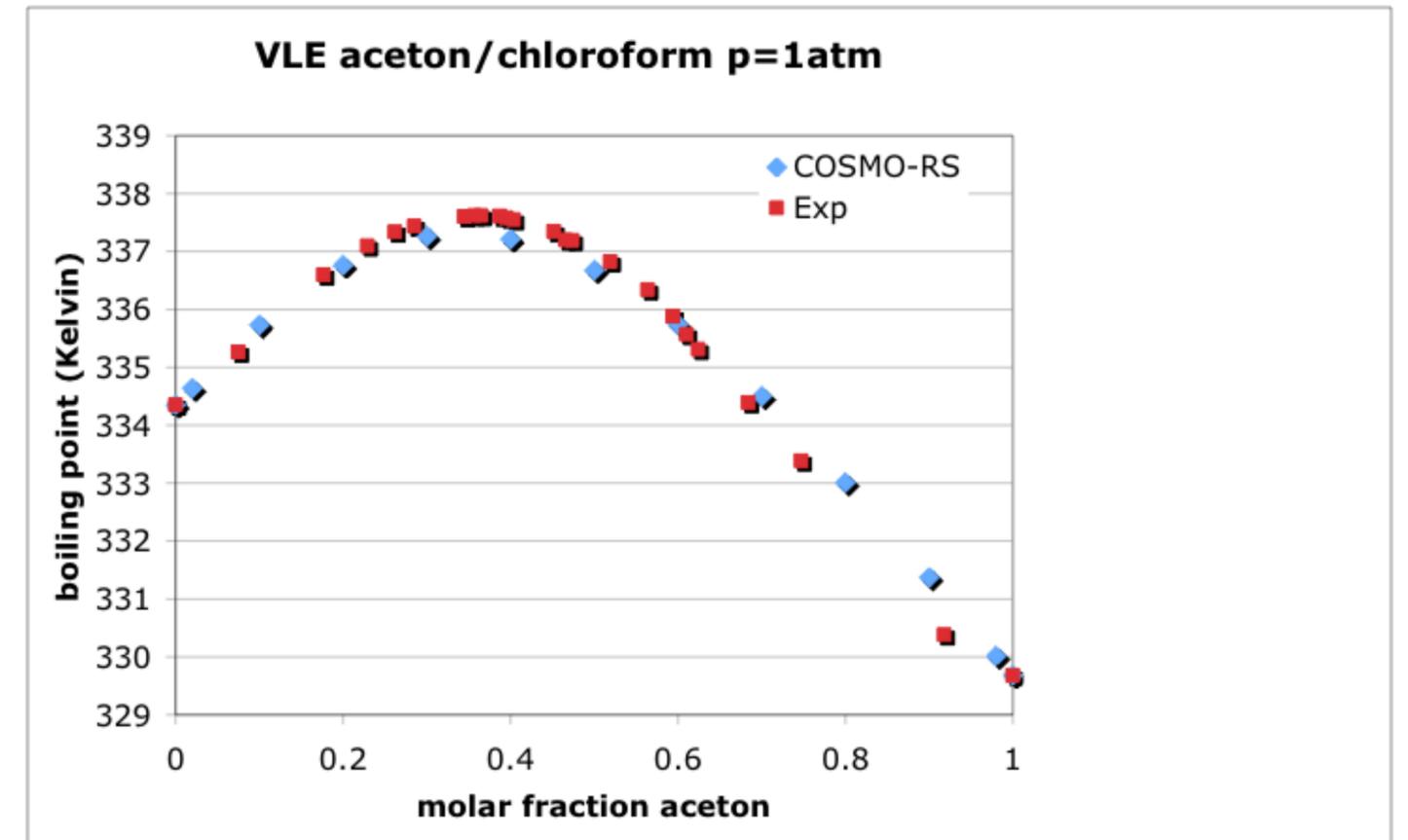
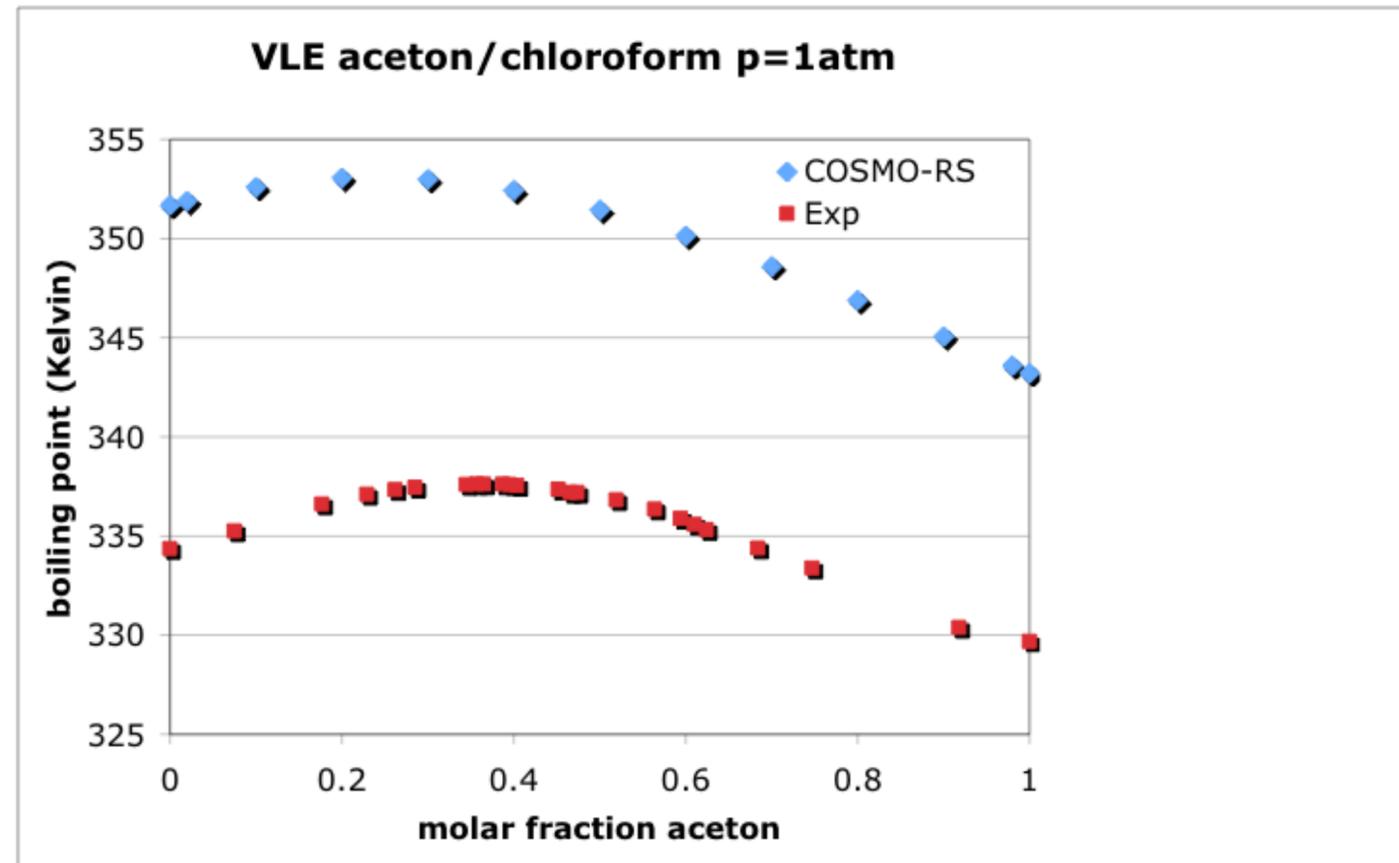
C.C. Pye, T. Ziegler, E. van Lenthe, J.N. Louwen, Can. J. Chem. 87 (2009), 790  
216 molecules, 642 experimental data      accuracy (rms)  
log vapor pressures (bar)                      0.24 log units  
hydration energies                              0.37 kcal/mol  
partition coefficients                          0.35 log units  
(octanol/water, benzene/water, hexane/water, diethyl ether/water)

## COSMO-SAC 2013-ADF in ADF

R. Xiong, S.I. Sandler, R.I. Burnett, Ind. Eng. Chem. Res. 53 (2014) 8265  
SAC = segment activity coefficients  
optimized parameters for ADF  
dispersion contribution in the mixture interaction

# Aceton Chloroform VLE

Use of experimental  
pure compound data



# Acetone Chloroform azeotrope

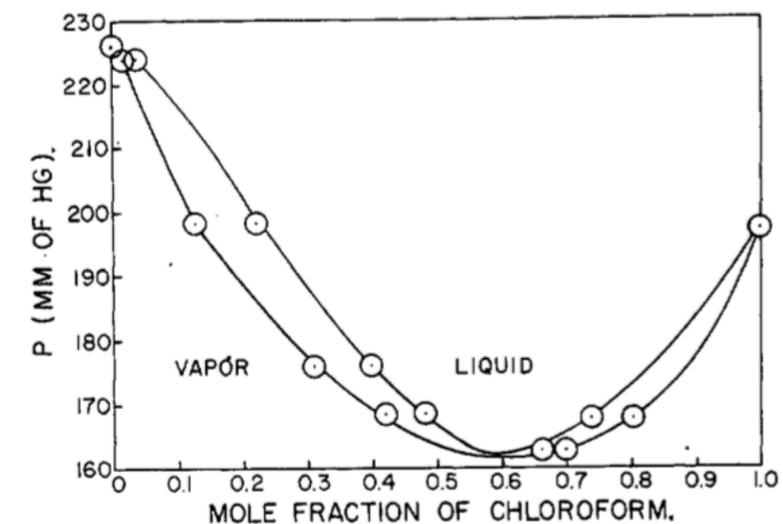
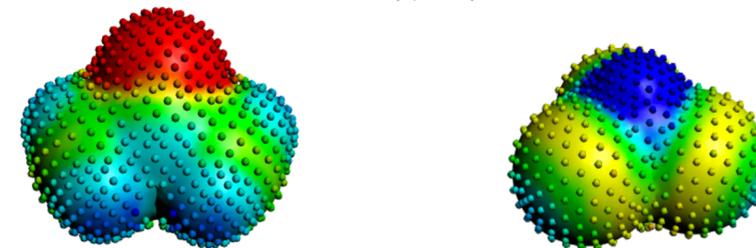
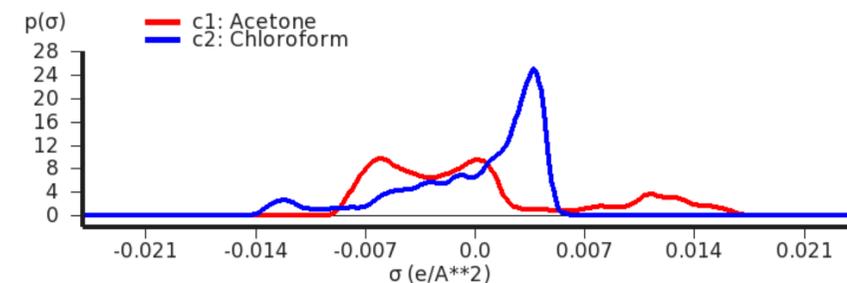
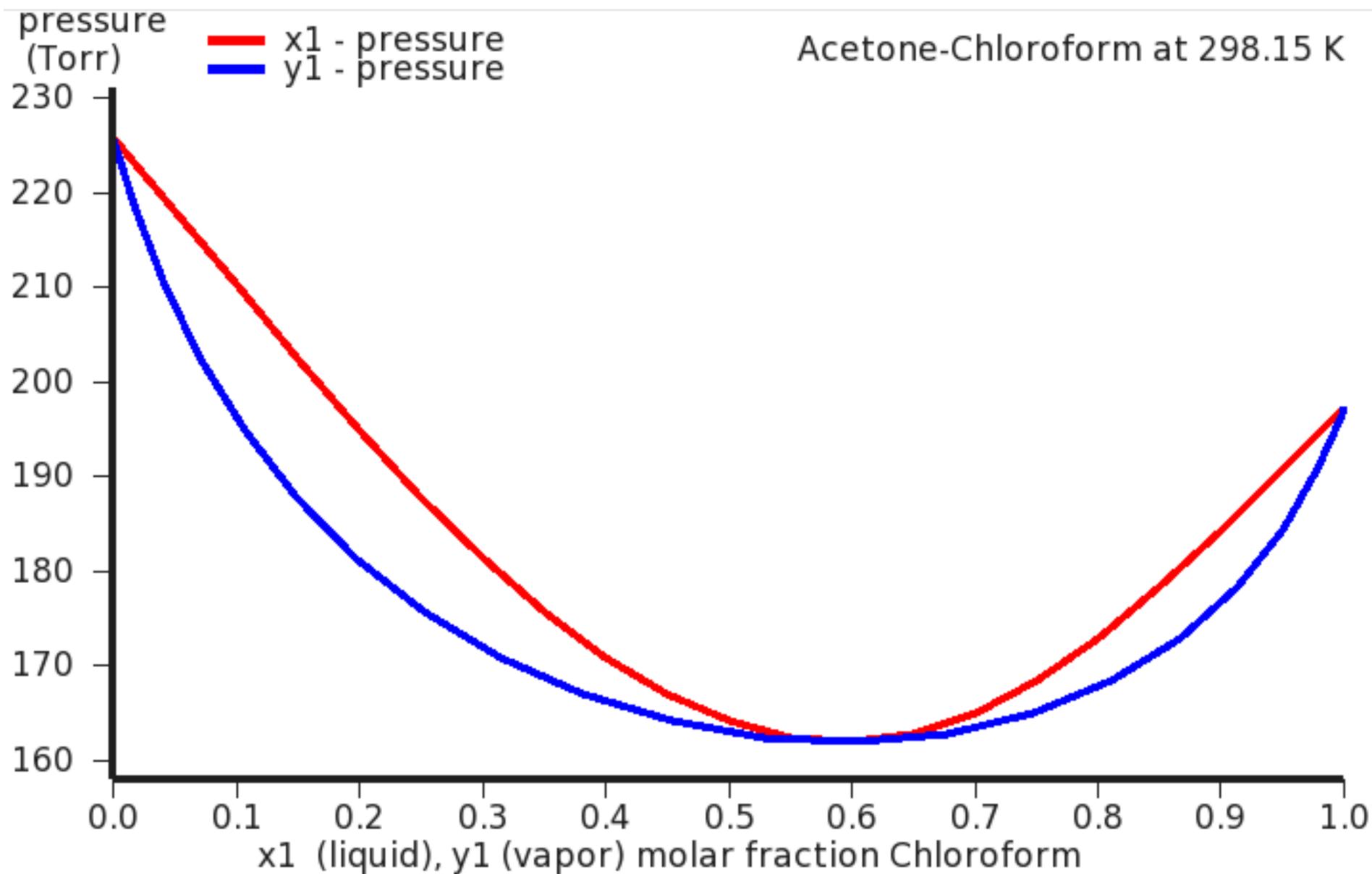
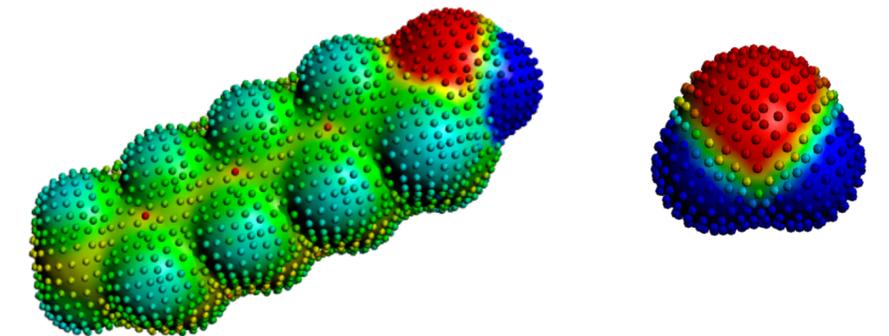
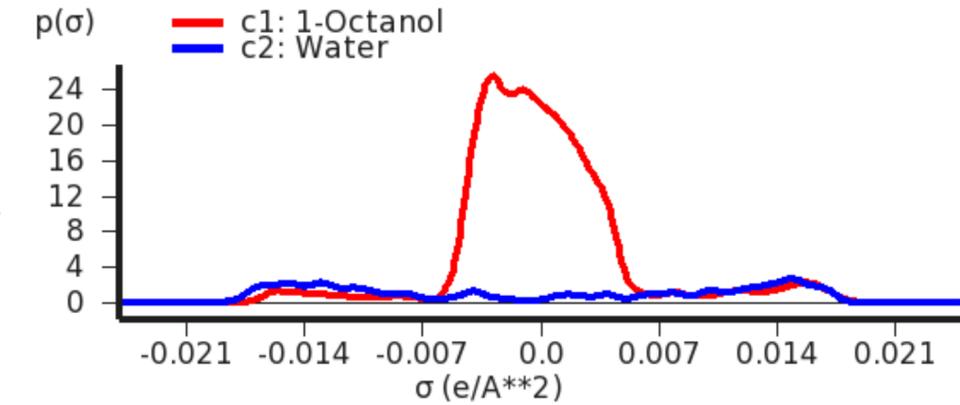
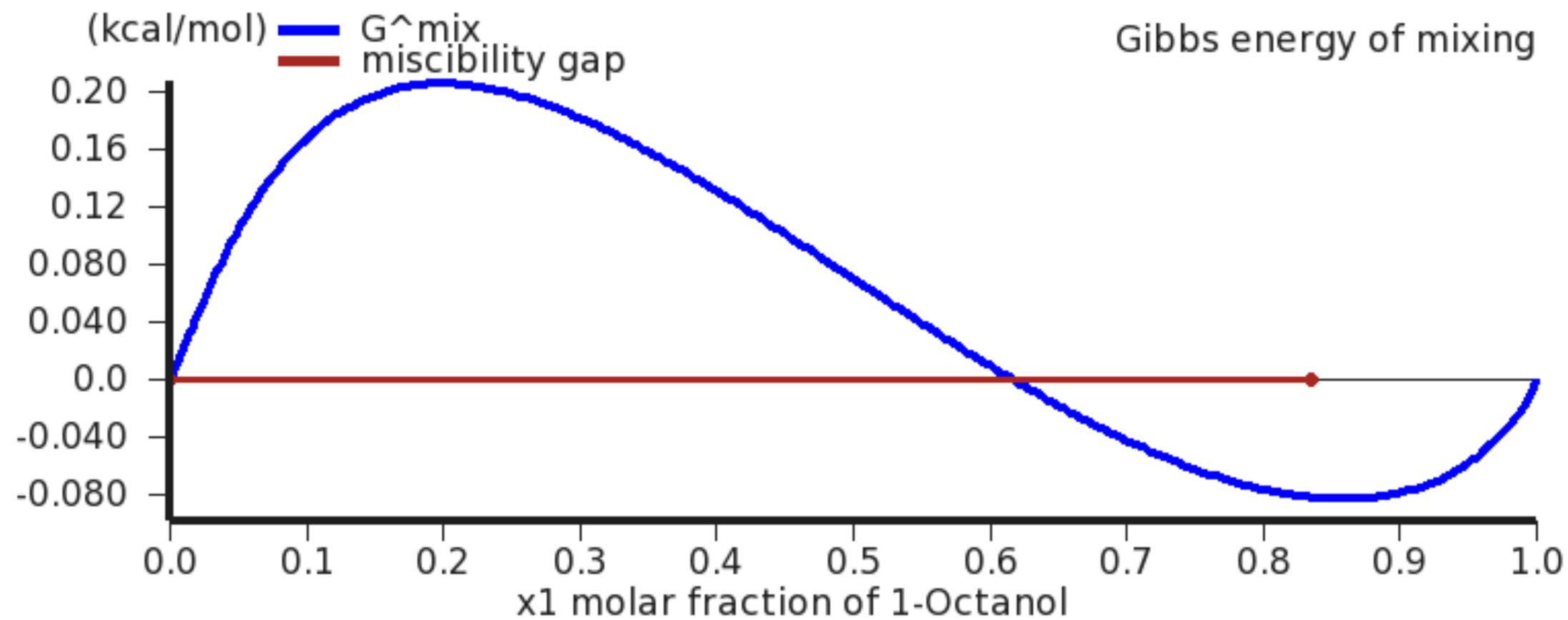


Fig. 4.—Vapor pressure-composition diagram of acetone and chloroform at 25°.

Mueller, Kearns, J. Phys. Chem., 1958, 62 1441

# 1-Octanol Water miscibility gap



# Properties

Solubility gas, liquid, or solid in solvent

for solids experimental melting point +  $\Delta H_{\text{fus}}$  needed

Partition coefficients

$K_{\text{OW}}$  relative solubilities in 1-Octanol and Water

pKa values

activity coefficients, solvation free energies, Henry's law constants

vapor pressures, boiling points

vapor-liquid diagrams binary and ternary mixtures (VLE/LLE)

excess energies, azeotropes, miscibility gaps

composition lines, flash points

# 2016

ionic liquid database

Dr. Zhigang

command-line scripts

pure compound COSMO-SAC ADF-2013