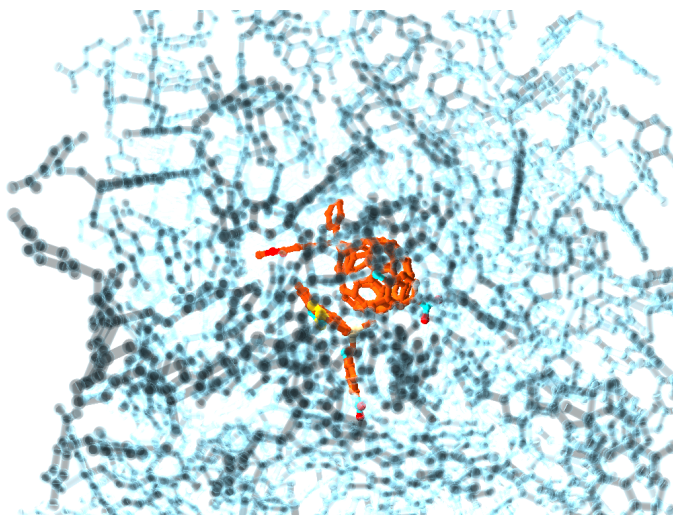


OPV research with ADF

Remco W. A. Havenith



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Theoretical Chemistry

Theoretical Chemistry Group

Mission:

Provide independent knowledge on structures, properties and their relation, with interpretation in terms of chemical concepts

- Emphasis on
 - Molecular properties
 - Photo-physical properties
 - Spectra and excited states
 - Magnetic and electric properties
 - Non-orthogonal methods
- Application driven method development (GAMESS-UK/ADF)
- Computational experiments



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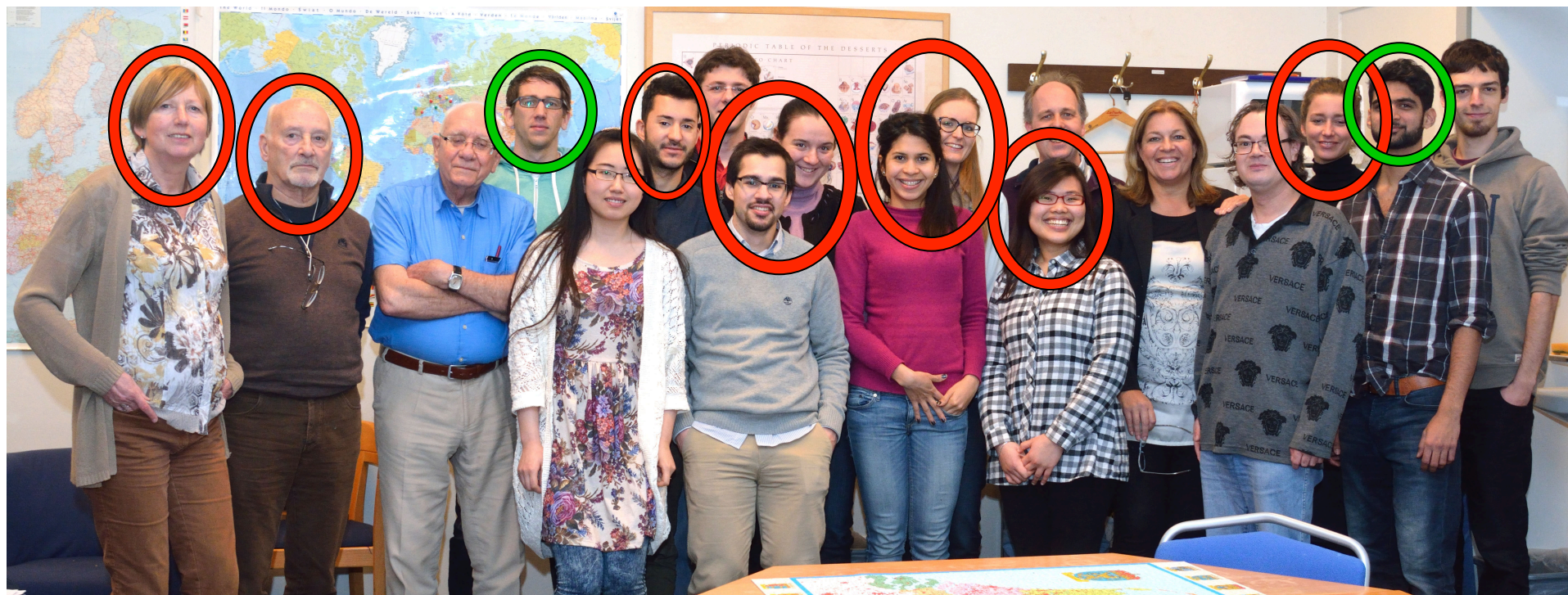
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Theoretical Chemistry

The OPV gang

Theoretical Chemistry Group



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Theoretical Chemistry

FOM Focus Group Groningen 'Next Generation Organic Photovoltaics'

- Aim:
 - Deliver the science for highly efficient, long-lived, and low-cost organic photovoltaic devices
- Challenge:
 - Charge separation at the donor/acceptor interface
- Approach:
 - Multi-disciplinary:
 - Material development
 - Physical characterisation (OPV device physics)
 - *Theoretical modelling*



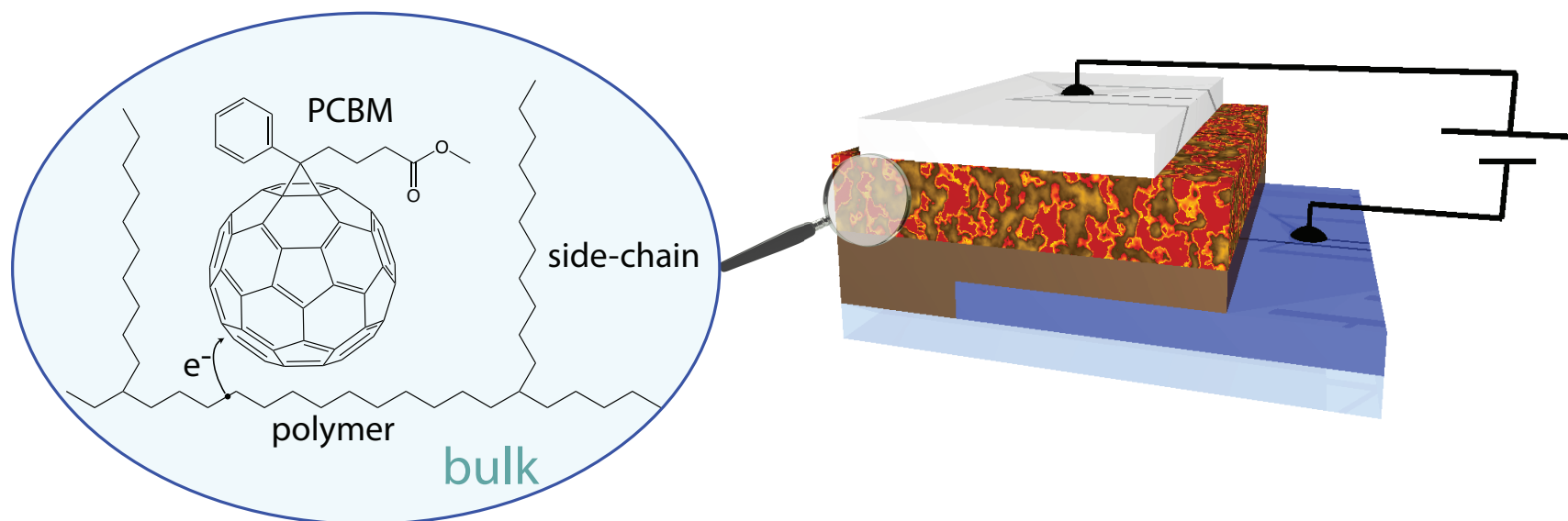
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Theoretical Chemistry

Organic photovoltaics



Theoretical Chemistry gives access to the fundamental processes that lead to photocurrent generation



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G. Yu *et al.*, *Science* **270** (1995), 1789

Theoretical Chemistry

What happens in an organic solar cell?

- Organic solar cell:
 - Nowadays bulkheterojunction
 - Blend of (DA) polymer and PCBM derivatives
- Light is absorbed
- Exciton diffusion to the interface
- Charge transfer from donor to acceptor
- Electron/hole diffusion to the electrodes
- While exciton/electron/hole diffuses through the material anything can happen



Theoretical challenges

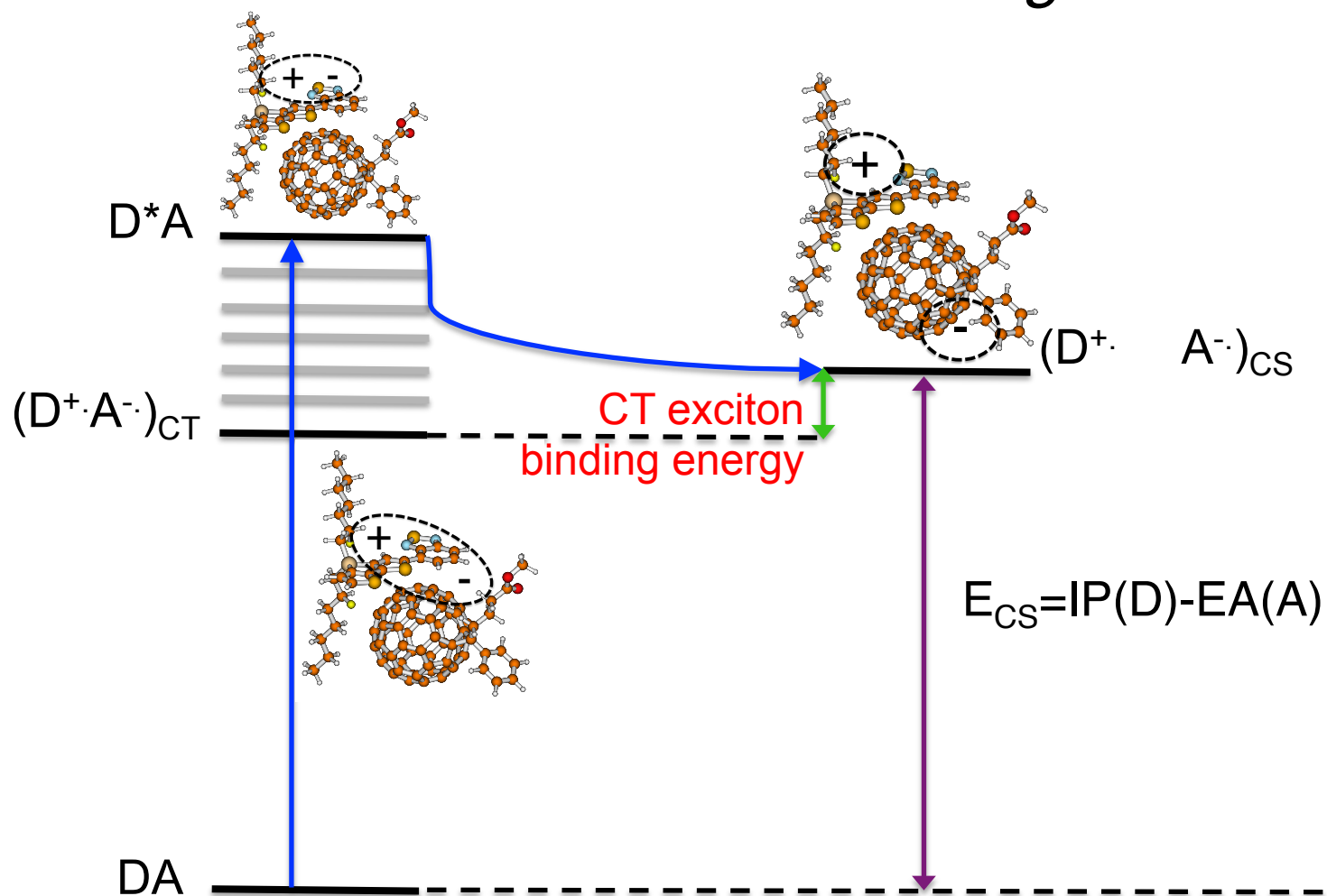
- Predict molecular properties that determine the dielectric properties of the interface
 - Dipole moments
 - Polarisability
- Modelling of the donor/acceptor interface
 - Molecular Dynamics simulations
 - Time scales of molecular motion
- Calculation of the excited states
 - Theoretical methods
 - Influence of molecular structure
 - Influence of the embedding using multiscale modelling
- Approximation of the electron transfer rates



- Required:
 - Quantum chemical calculations
 - HOMO/LUMO levels
 - Polarisability
 - Dipole moment
 - Excited states
 - Molecular Dynamics calculations
 - Representative structures
 - Morphologies
 - Mobility of molecules in the blend
 - Combined QM and MD
 - Excited states and excited state dynamics



Electronic state diagram



Brédas J.-L., *et al.*, *Acc. Chem. Res.* **42**, 1691-1699 (2009)

Bakulin A.A., *et al.*, *Science* **335**, 1340-1344 (2012)

Clarke T.M., Durrant J.R., *Chem. Rev.* **110**, 6736-6767 (2010)



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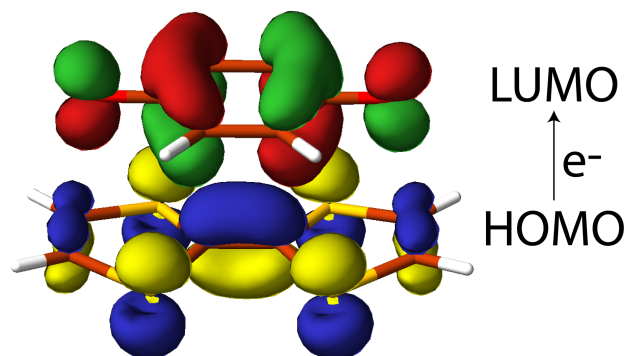
Theoretical Chemistry

Which QM method and package to use?

- We want to describe:
 - Charge-transfer states
 - The effect of the medium
 - Electron/hole diffusion
- CT states
 - DFT?
 - HF?
 - Correlated methods?
- Medium effects
 - Explicit molecules?
 - Continuum?



Methods: Charge-transfer states with (PCM)-DFT



Method/ Functional	E_{exc} (vac) (eV)	E_{exc} (H ₂ O) (eV)	ΔE_{exc} (eV)
HF	3.53	3.43	0.09
B3LYP	0.91	0.77	0.14
<i>BHandH</i>	<i>1.70</i>	<i>1.57</i>	<i>0.13</i>
CAMB3LYP	1.66	1.55	0.11
HCTH407	0.47	0.33	0.13
PBE0	1.04	0.90	0.14
CC2	1.87	--	--

Improved methods/functionals required



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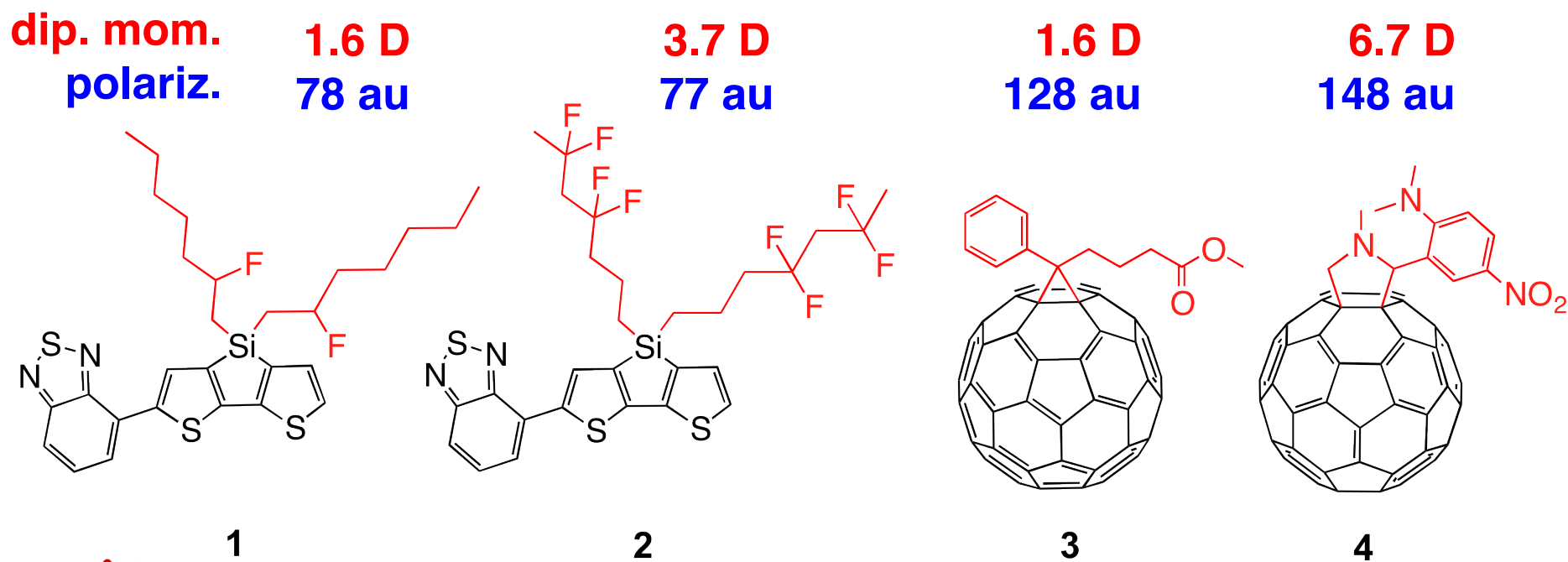
Screening effect using a continuum model

- Calculate IP(D) and EA(A)
- Determine the CS state energy in the medium: $E(\text{CS}) = \text{IP}(\text{D}) - \text{EA}(\text{A})$

Systems under study:

D: donor-acceptor “co-polymer” with two different side-chains (**1** and **2**)

A: PCBM (**3**) and PCBM derivative (**4**)



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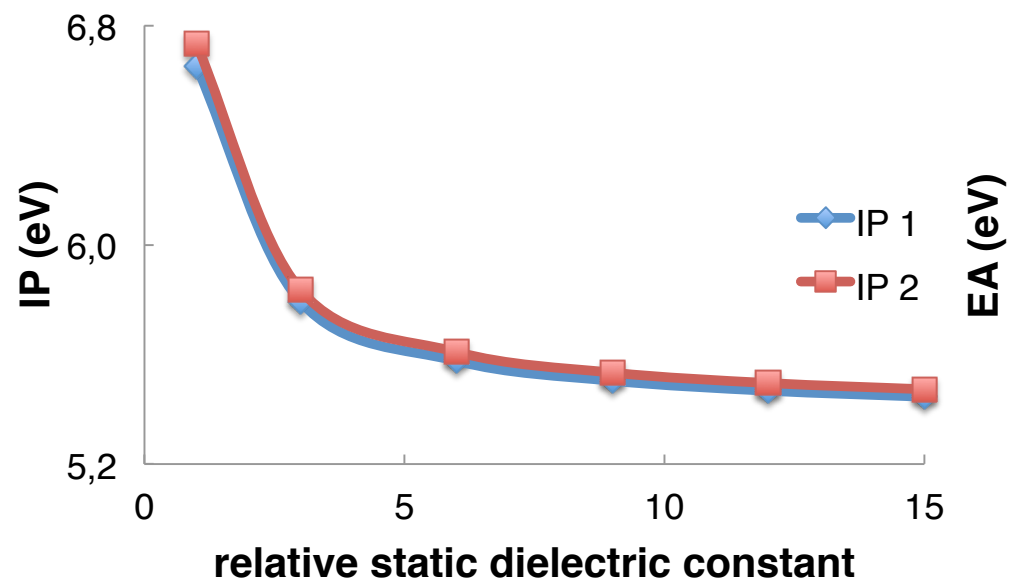
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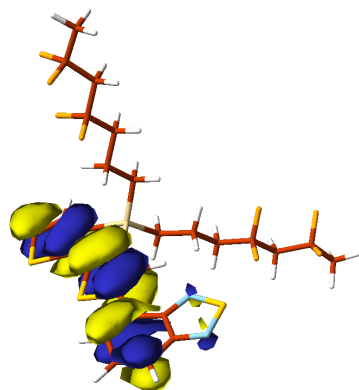
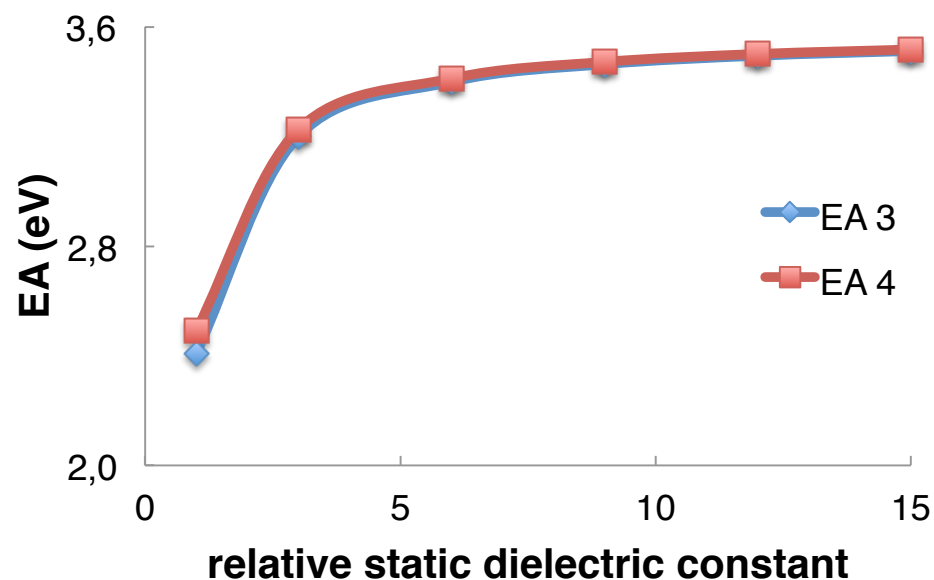
Theoretical Chemistry

Results

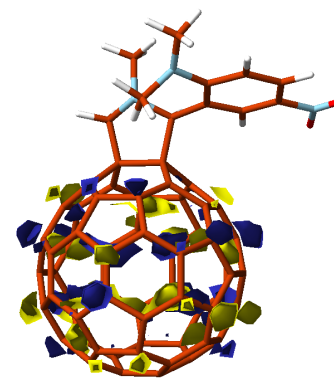
IP (1 and 2)



EA (3 and 4)



SOMO of
2 and 4



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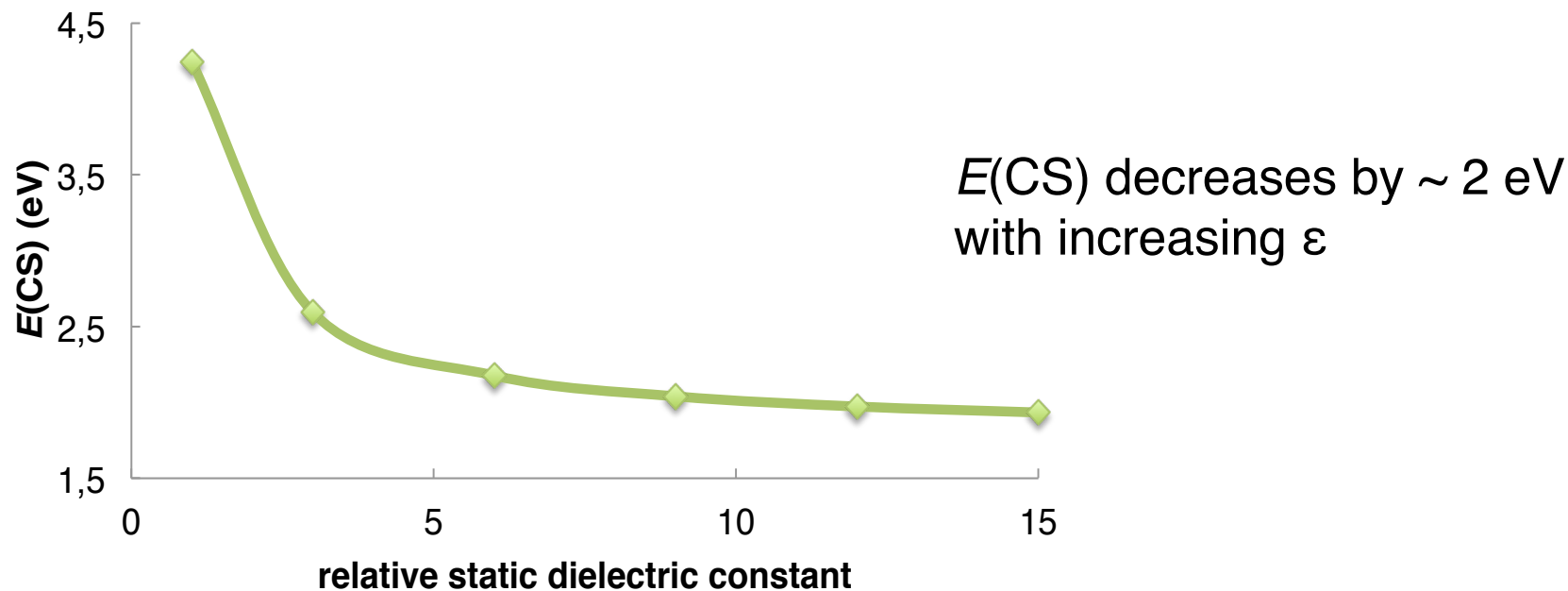
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Theoretical Chemistry

Results

CS state energy (1 and 3)



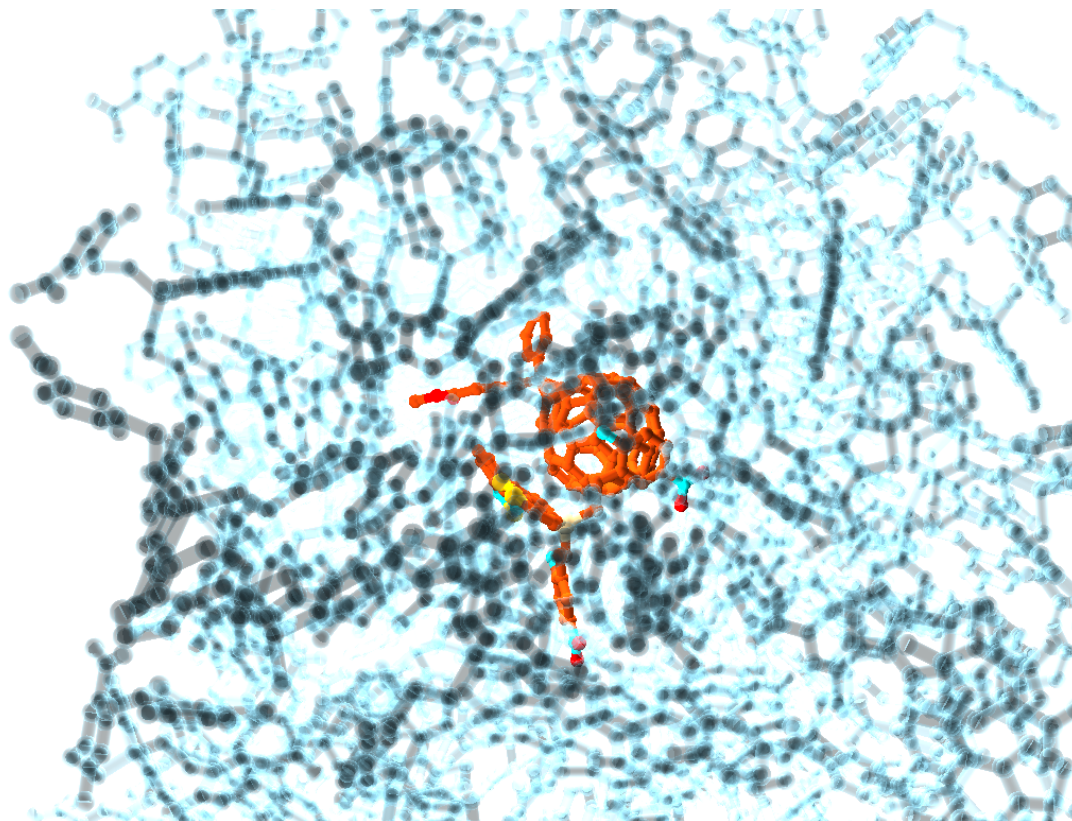
But:

We need to know ϵ

No information on relaxation effects and local field effects



Multiscale modelling: Molecular Dynamics



Embed the molecules in a box filled with environment

Perform (TD-)DFT calculations while treating the embedding with DRF (ADF program)



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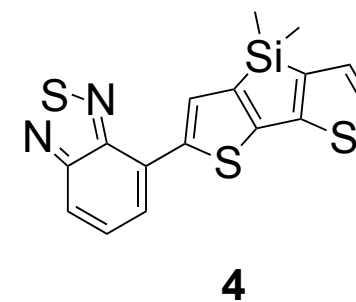
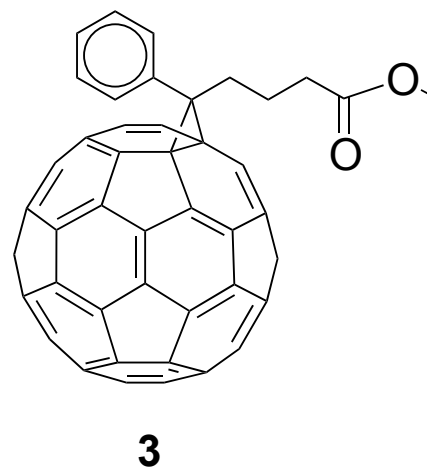
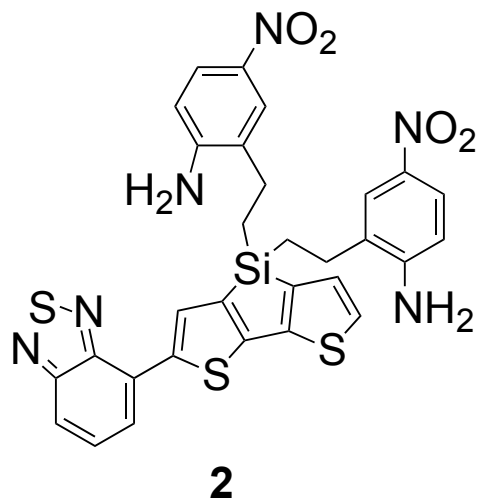
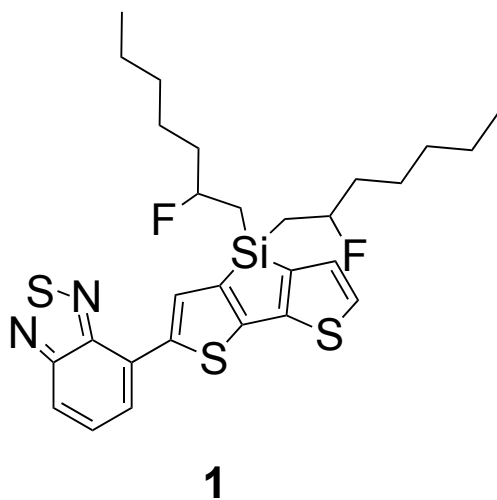
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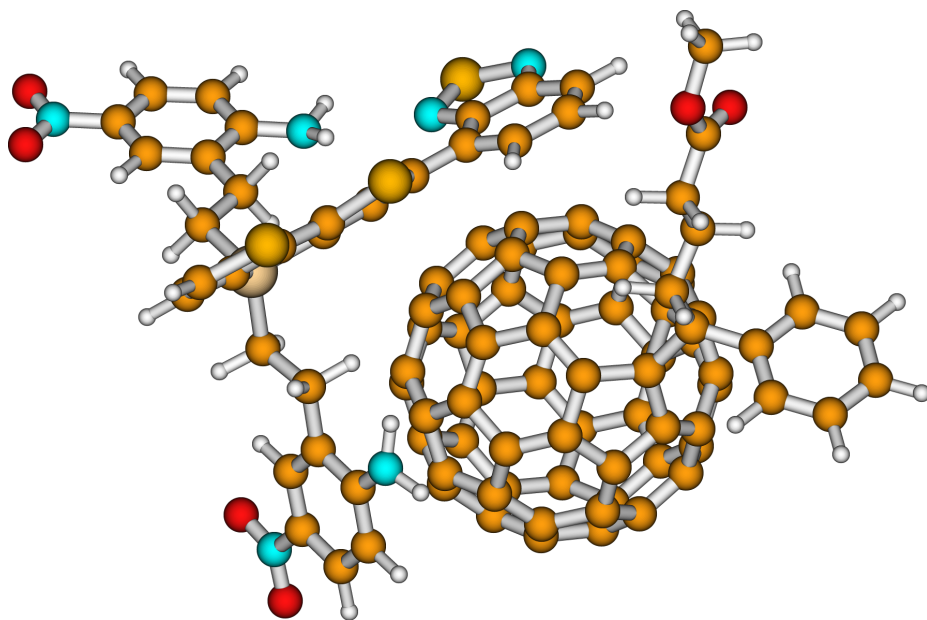
Theoretical Chemistry

Model systems

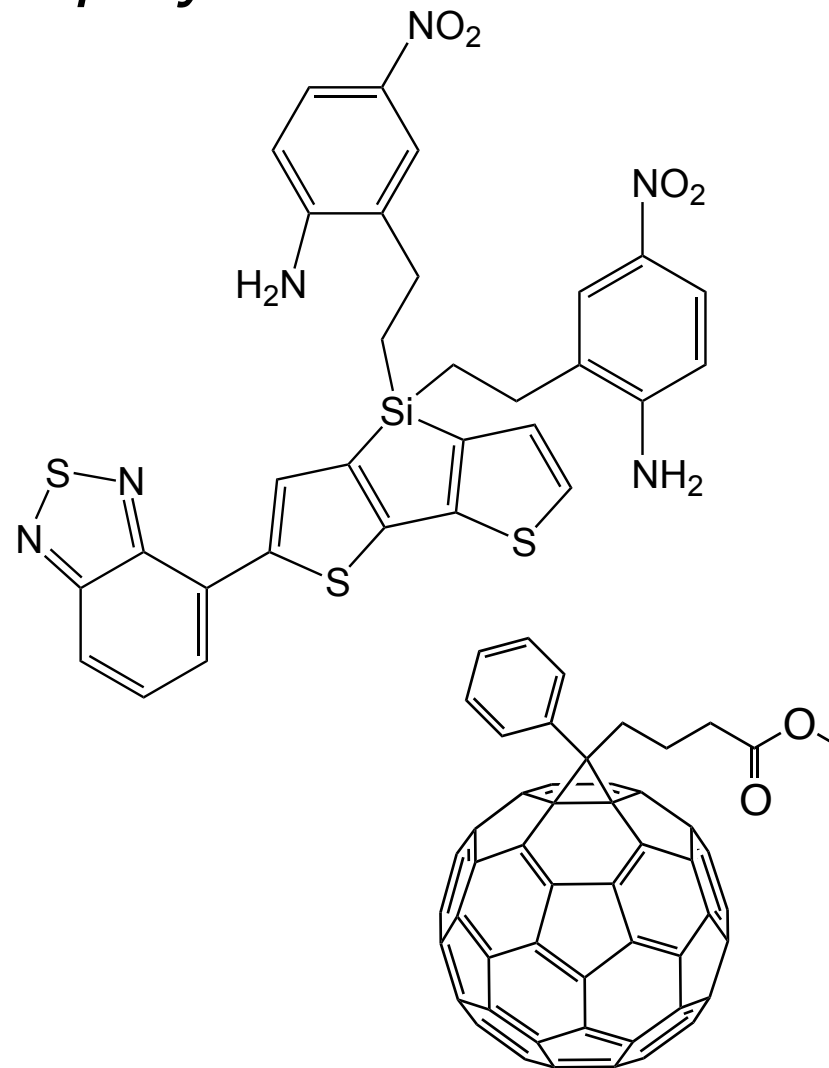
One donor-acceptor co-monomer (**1** or **2**) and one PCBM molecule (**3**),
embedded in only monomers (**1**, **2** or **4**)



Embedding study: effect of polymer side-chains



DFT geometry optimisation



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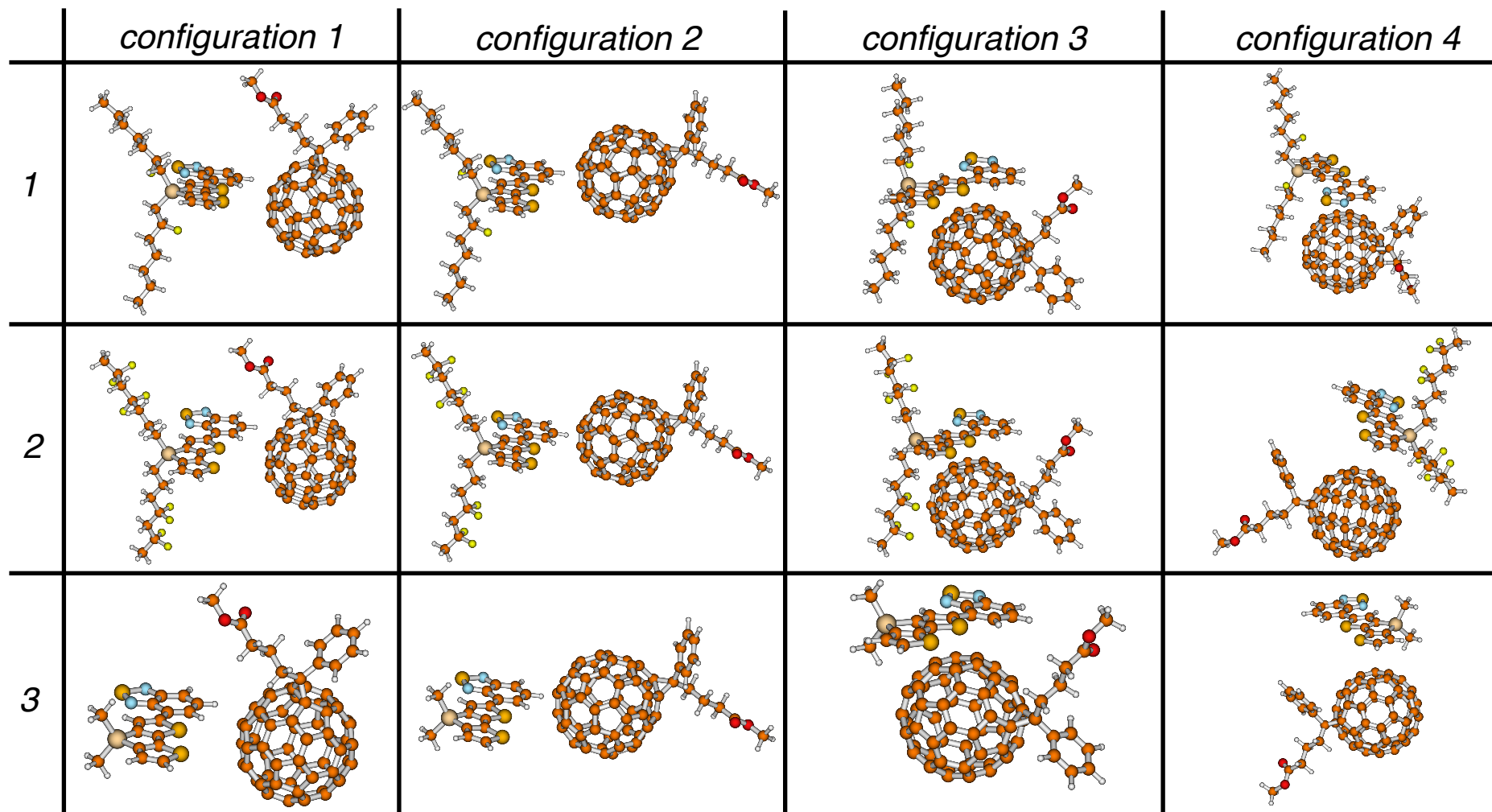
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Bulk effect of polymer side-chains

Winning team:



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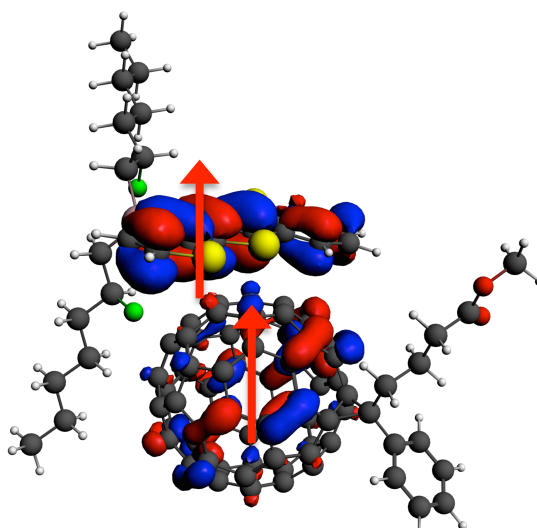
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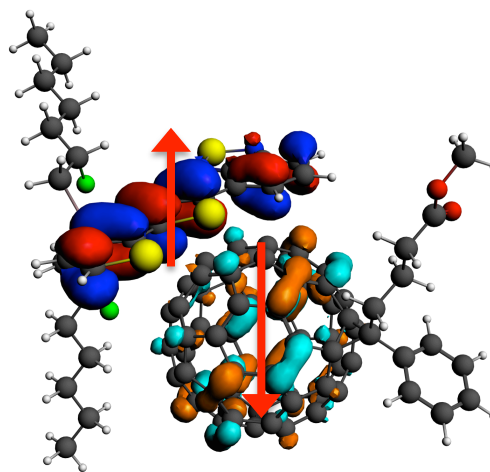
Theoretical Chemistry

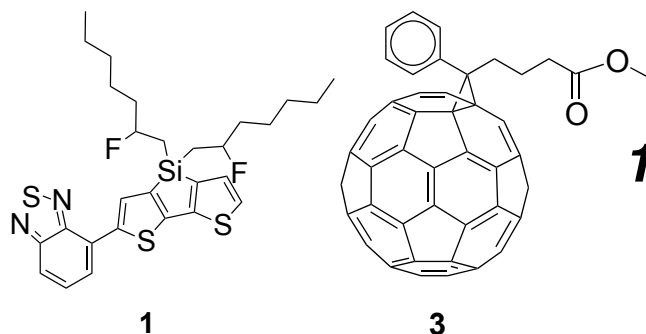
Lowest triplet excited state: energy and geometry

$$E(T_1)_{\text{relaxed}} = 1.9 \text{ eV}$$



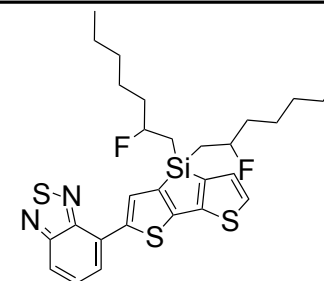
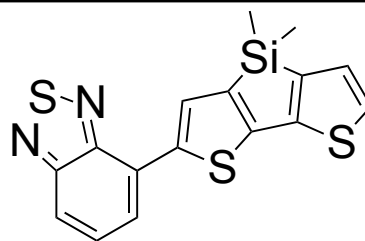
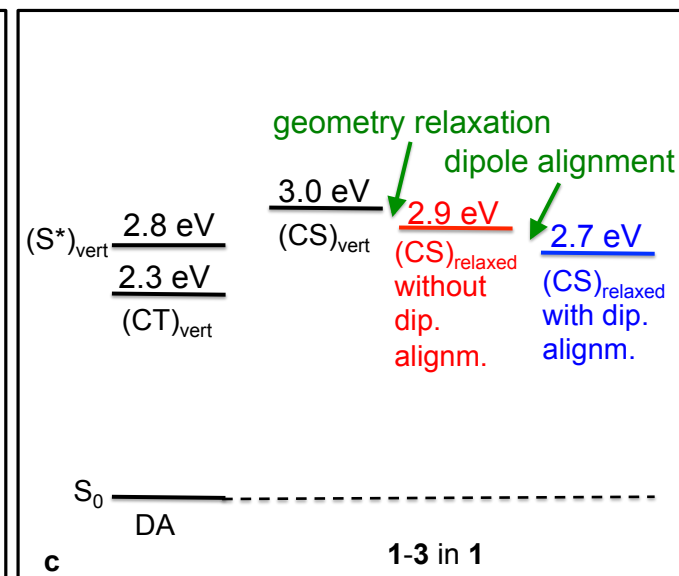
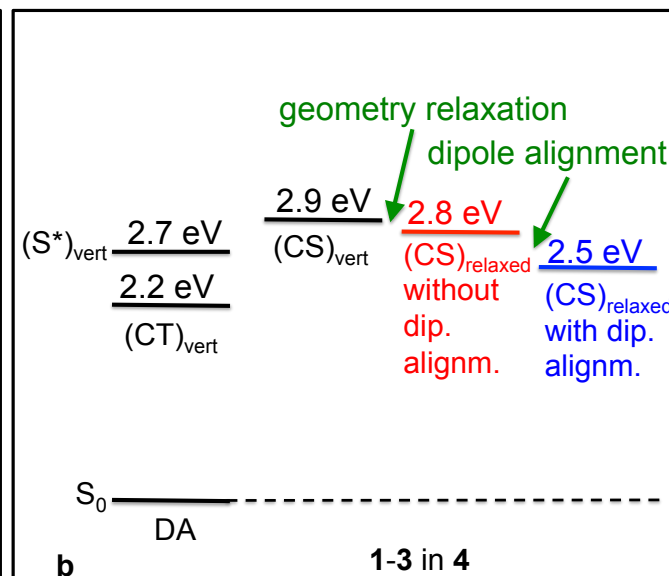
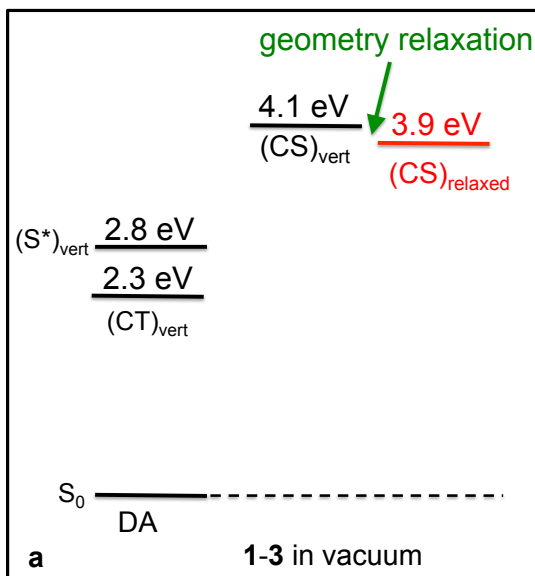
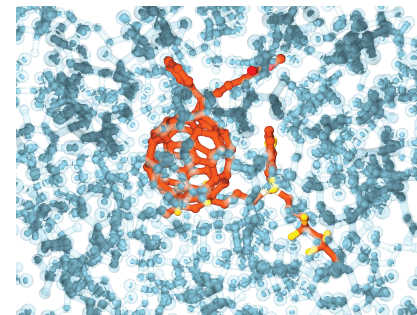
$$E(CT_1)_{T1} = E(CT_1)_{\text{relaxed}} = 2.1 \text{ eV}$$





Results

1-3 in vacuum, in 4 and in 1

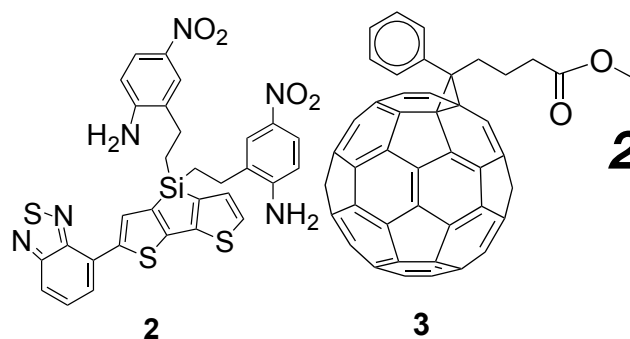


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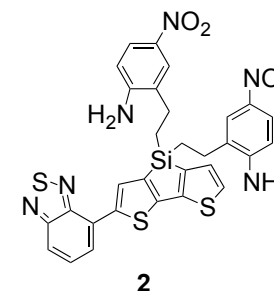
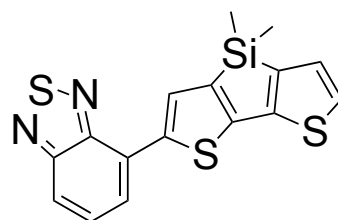
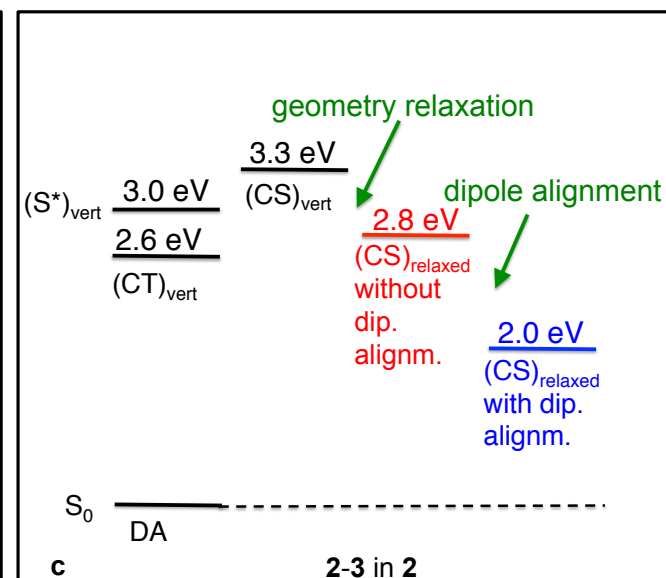
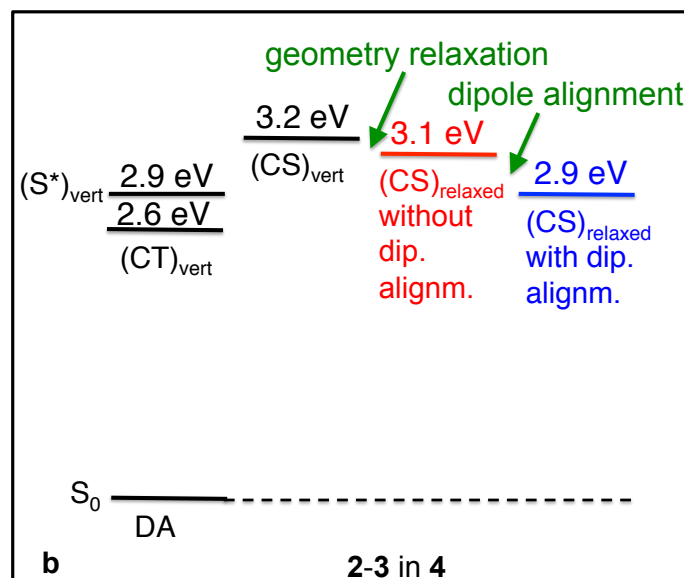
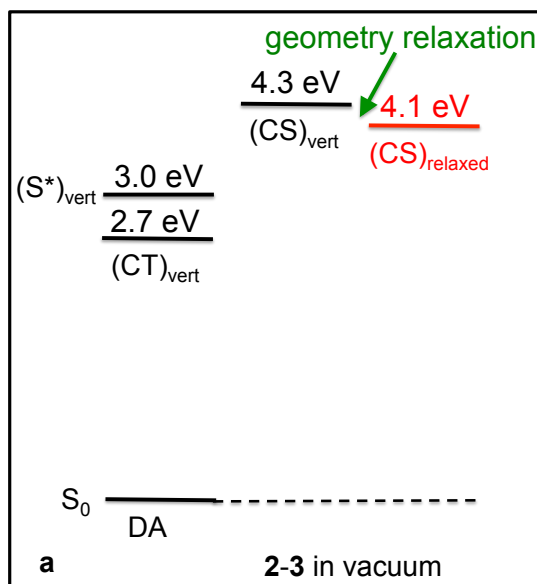
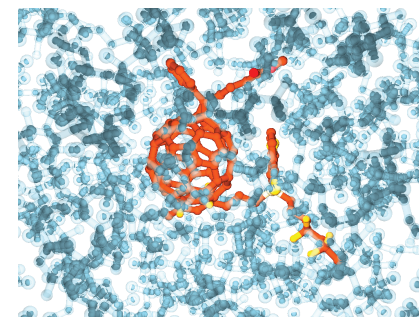
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Results

2-3 in vacuum, in 4 and in 2



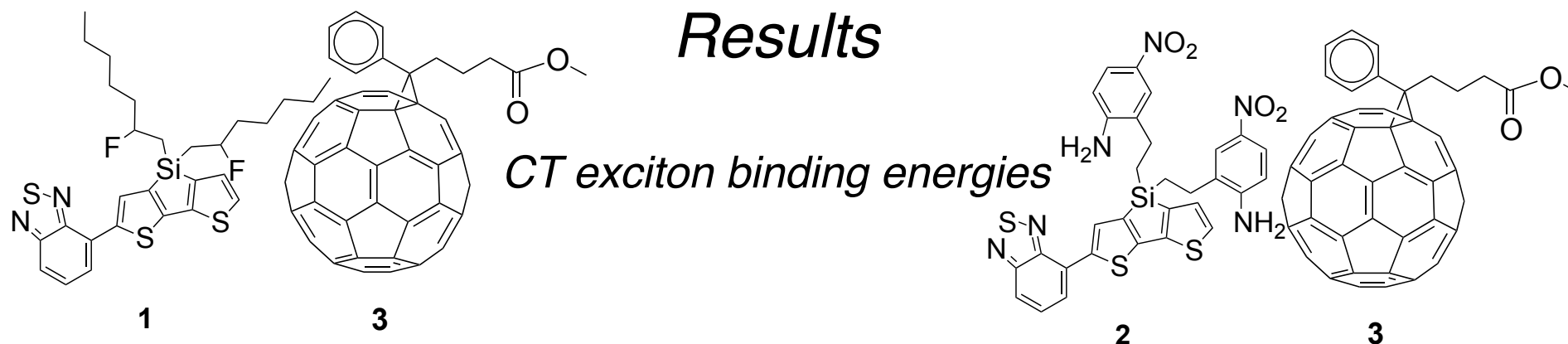
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Results

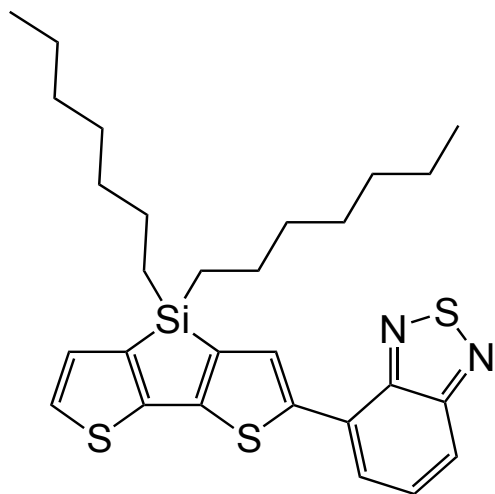


	1-3 in vac.	1-3 in 4	1-3 in 1
E_b no dip. alignm. (eV)	1.6	0.6	0.6
E_b with dip. alignm. (eV)	1.6	0.3	0.4
	2-3 in vac.	2-3 in 4	2-3 in 2
E_b no dip. alignm. (eV)	1.4	0.5	0.2
E_b with dip. alignm. (eV)	1.4	0.3	-0.6

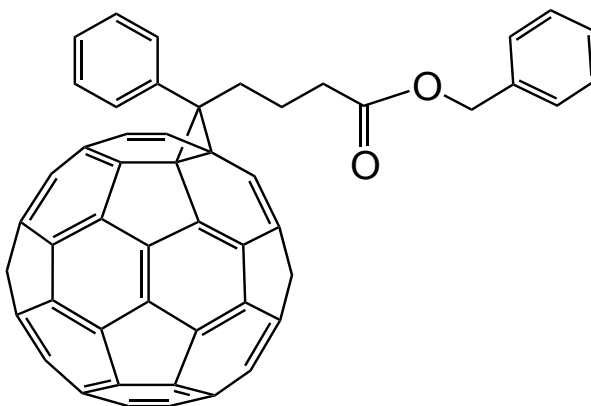


PCBM environment: Model systems

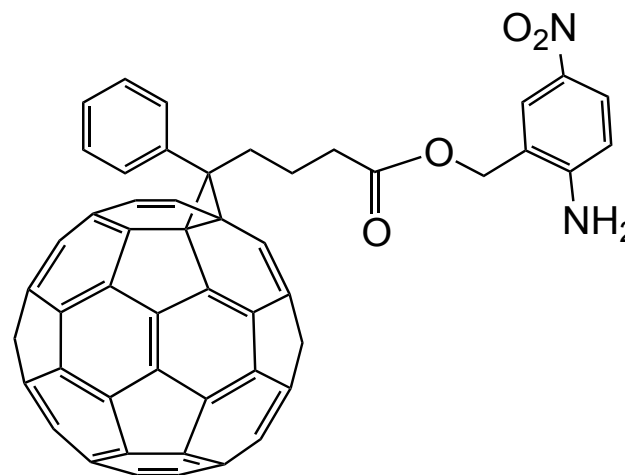
One donor-acceptor co-monomer (**1**) and one PCBM molecule (**2** or **3**),
embedded in only PCBM molecules (**2** or **3**)



1



2

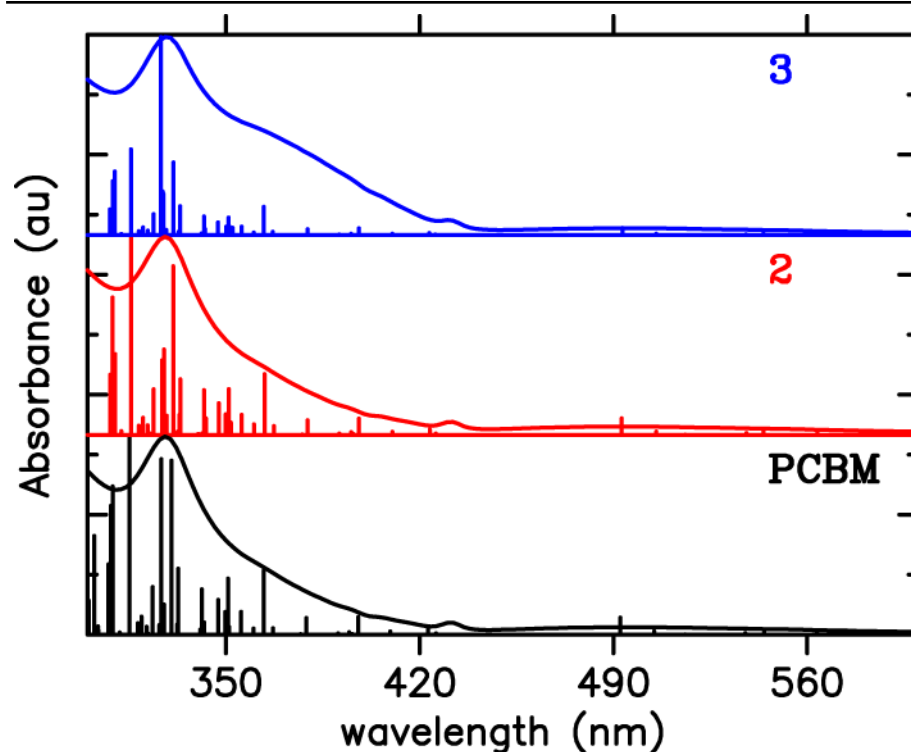


3



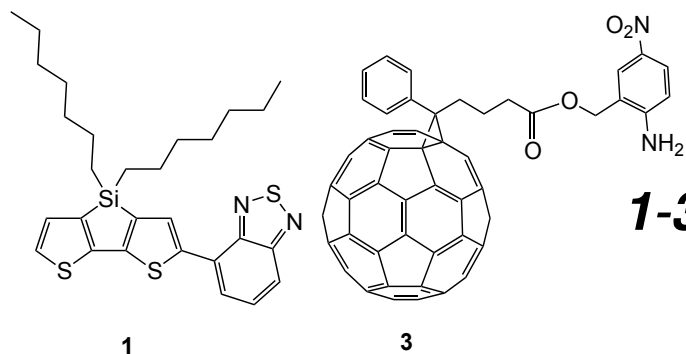
Electronic properties

	$E_{1/2-1, \text{red}}$ (V)	$E_{1/2-2, \text{red}}$ (V)	ϵ_{HOMO} (eV)	ϵ_{LUMO} (eV)	α/e^- (au)	μ (Deb)
PCBM	-1.089	-1.482	-7.92	-3.88		
2	-1.095	-1.489	-7.90	-3.86	1.56	2.35
3	-1.097	-1.487	-7.87	-3.83	1.53	5.97



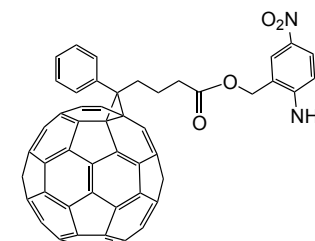
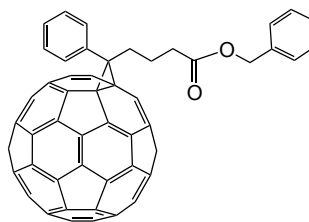
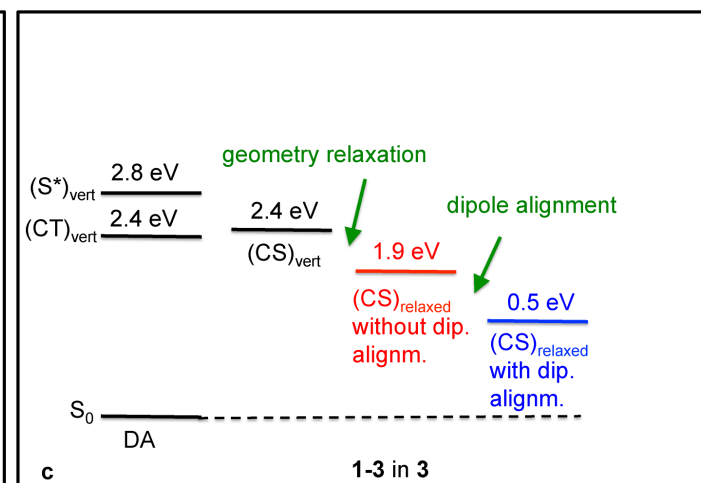
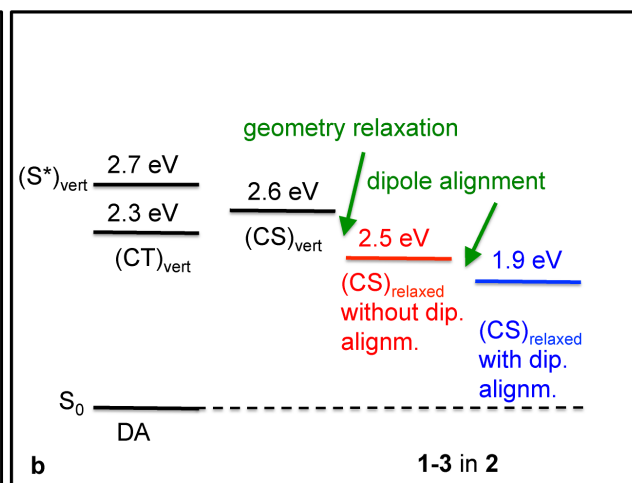
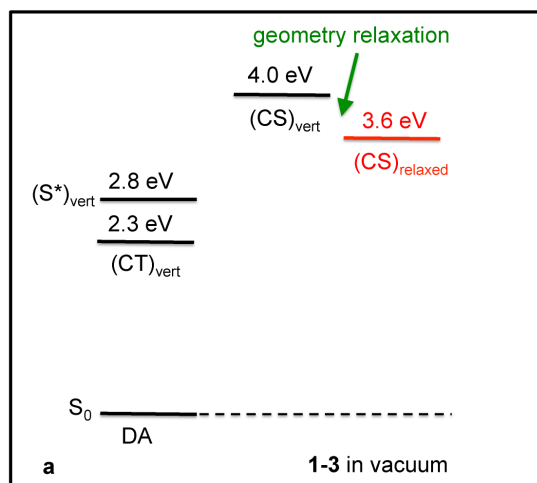
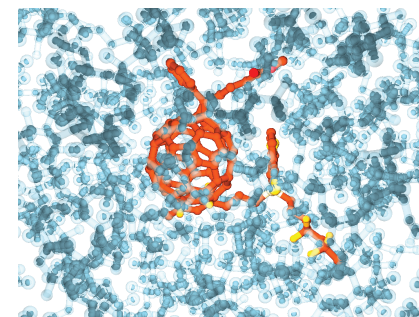
- All compounds have similar electronic properties, except for the dipole moment





Results

1-3 in vacuum, in 2 and in 3



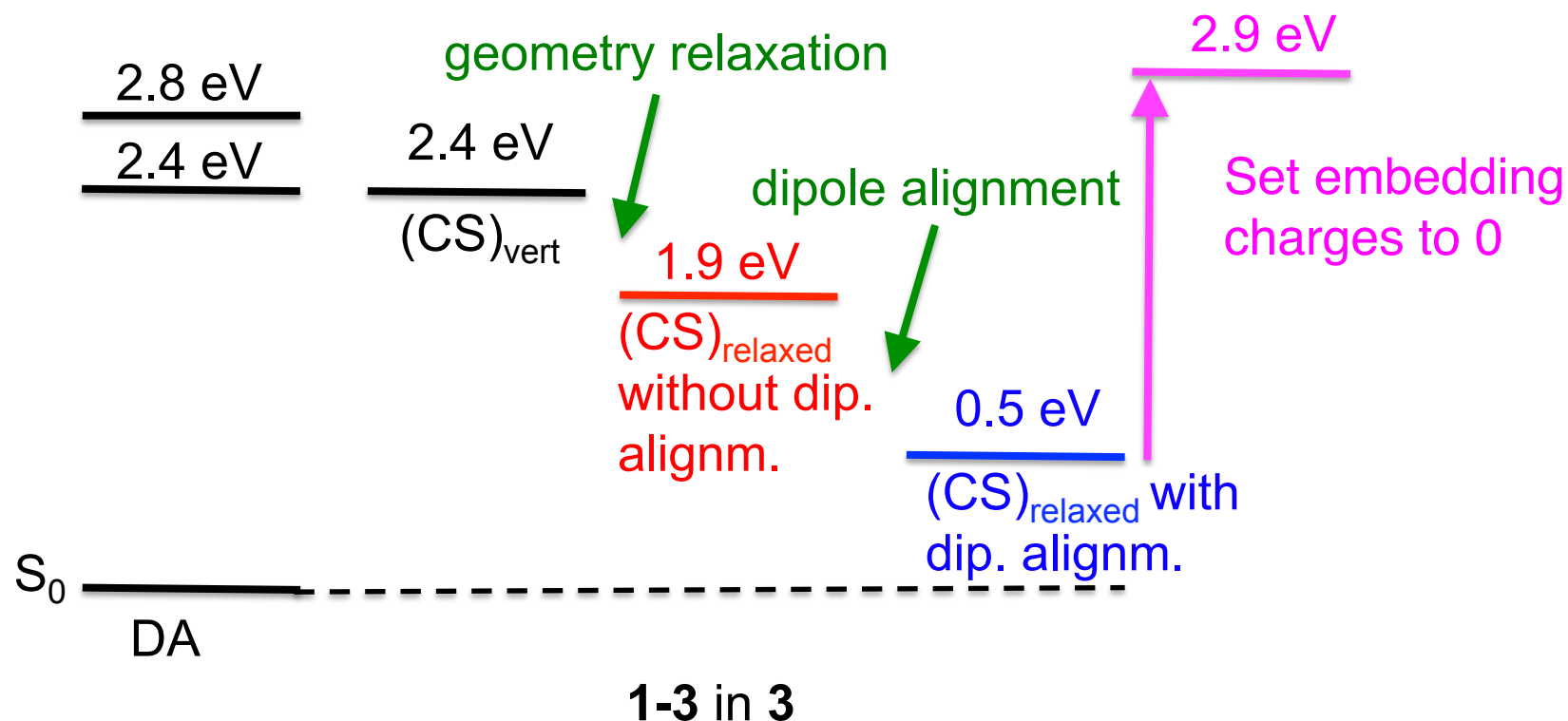
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Effect of dipoles?



- If embedding dipoles are set to 0, CS goes up again



What is still missing?

- A lot!
 - Geometry optimisation within the environment
 - Geometry relaxation of CT states
 - Entropy contributions
 - Energy transfer rates
 - Couplings between excited states
 - Role of vibrations?
 - Role of delocalisation?
- Currently we are working on several of these aspects in ADF



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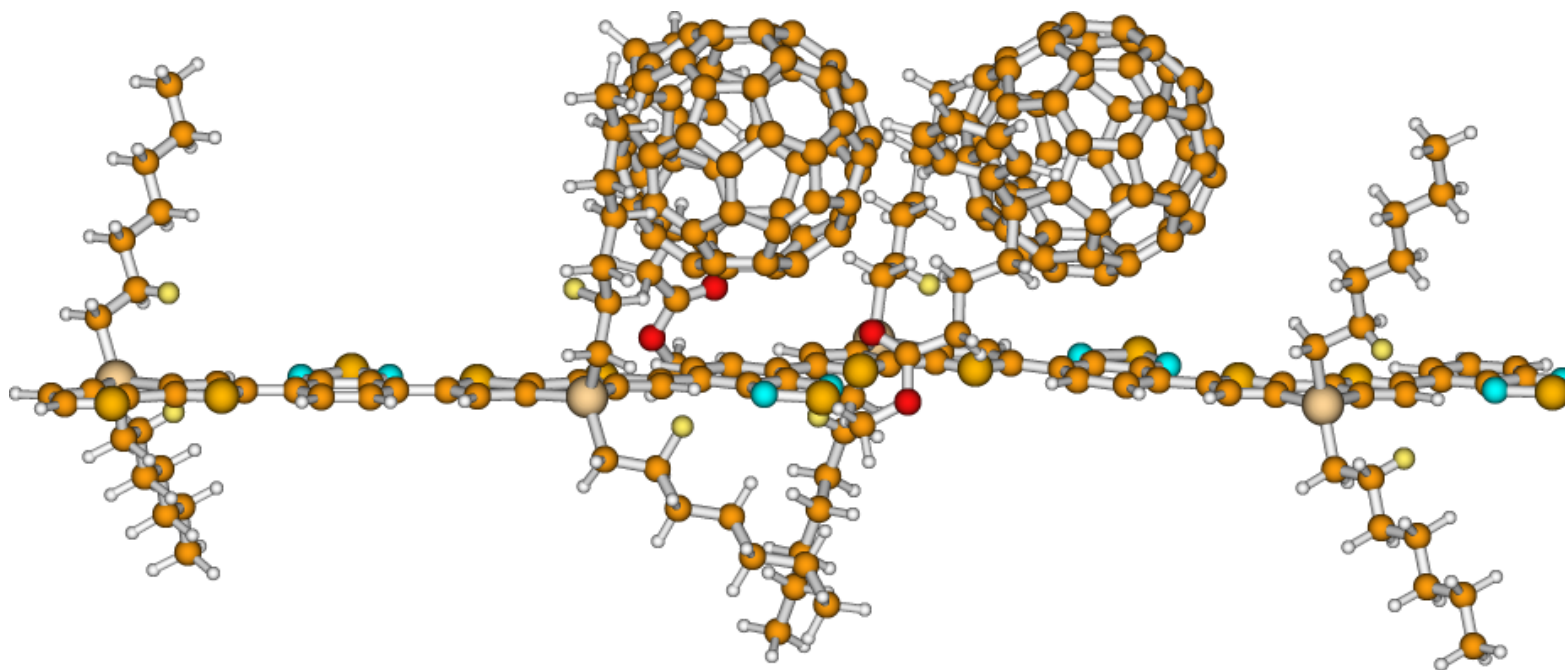
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Future directions

- Improve the theoretical methods
 - CC2 (NWChem) in combination with leading supercomputers
- Extend the system (10000 - 30000 bfs) to get a more realistic model of the donor-acceptor interface (DFT – ADF)



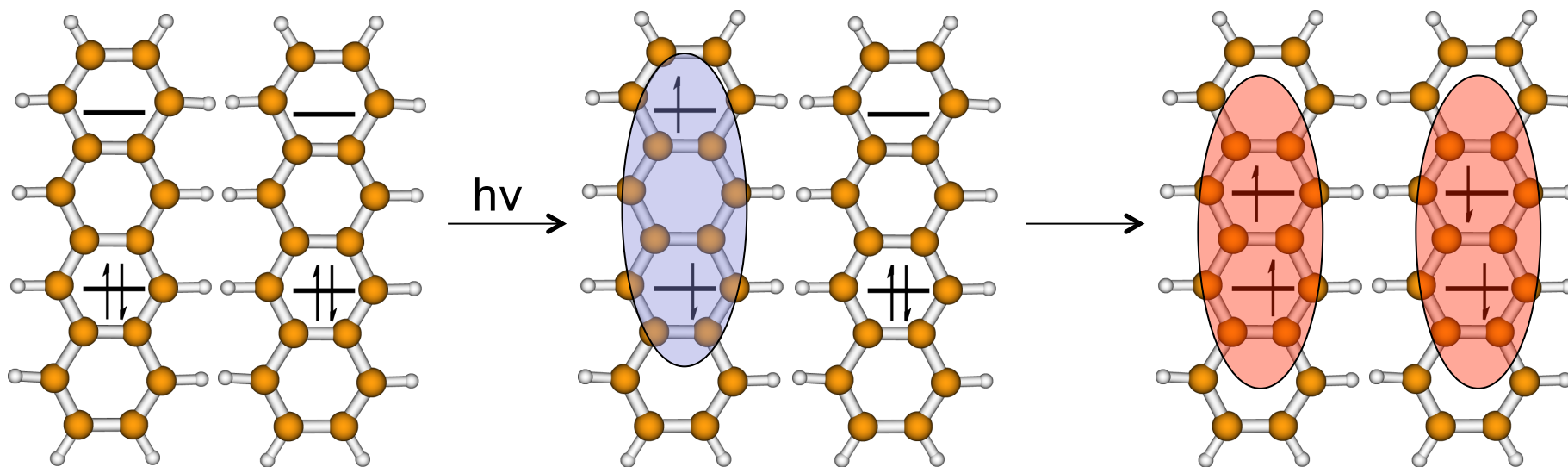
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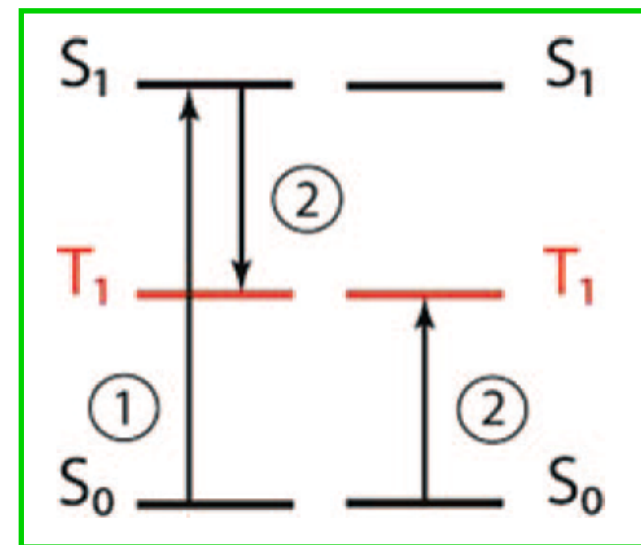
Theoretical Chemistry

Other aspects that may improve OPV: Singlet fission



- Spin allowed
- Radiationless process

M. B. Smith, J. Michl, *Chem. Rev.* **110** (2010), 6891



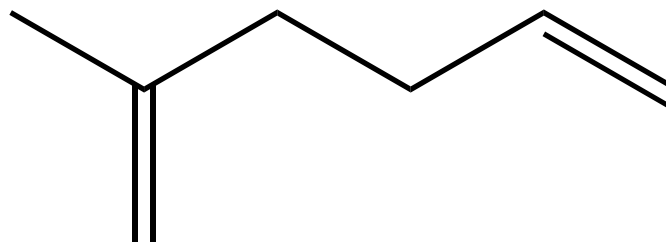
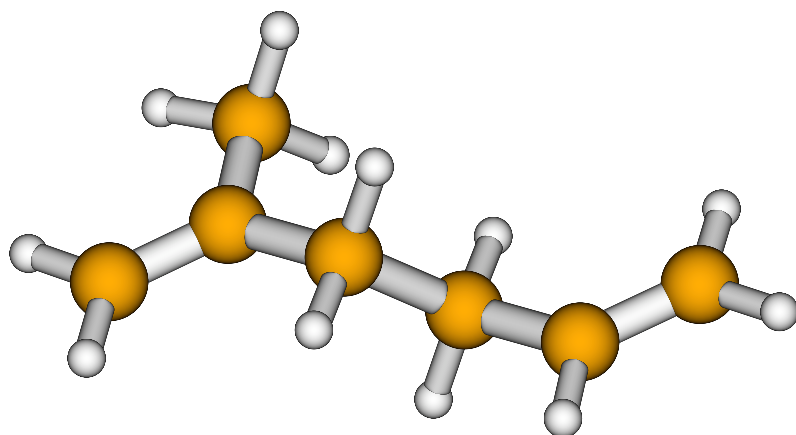
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Model calculations



- Two weakly coupled chromophores



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Demands on Computational Techniques

^1TT state is a singlet state with four open-shells:

Multi determinant wavefunction

- MCSCF
 - CAS/RASPT2, MRCI
- Non orthogonal CI
 - compact wavefunctions
 - clear chemical/physical interpretation

- Geometries/vibrational frequencies:

Can we use ^5TT instead of ^1TT ?

DFT would then be ok



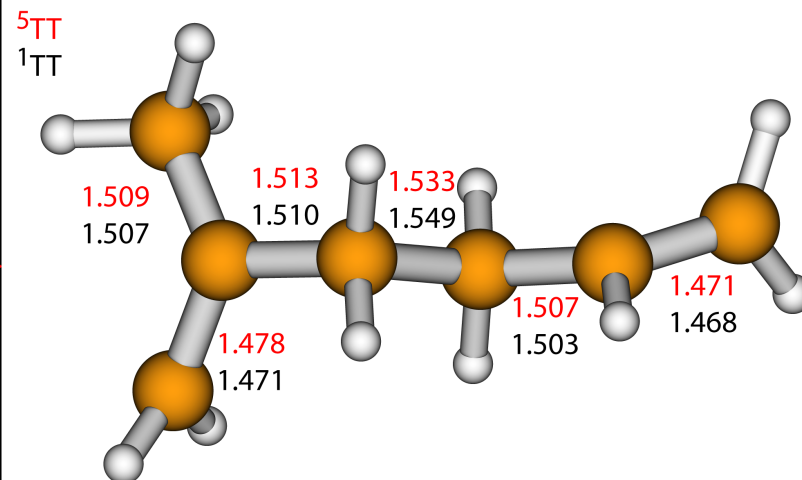
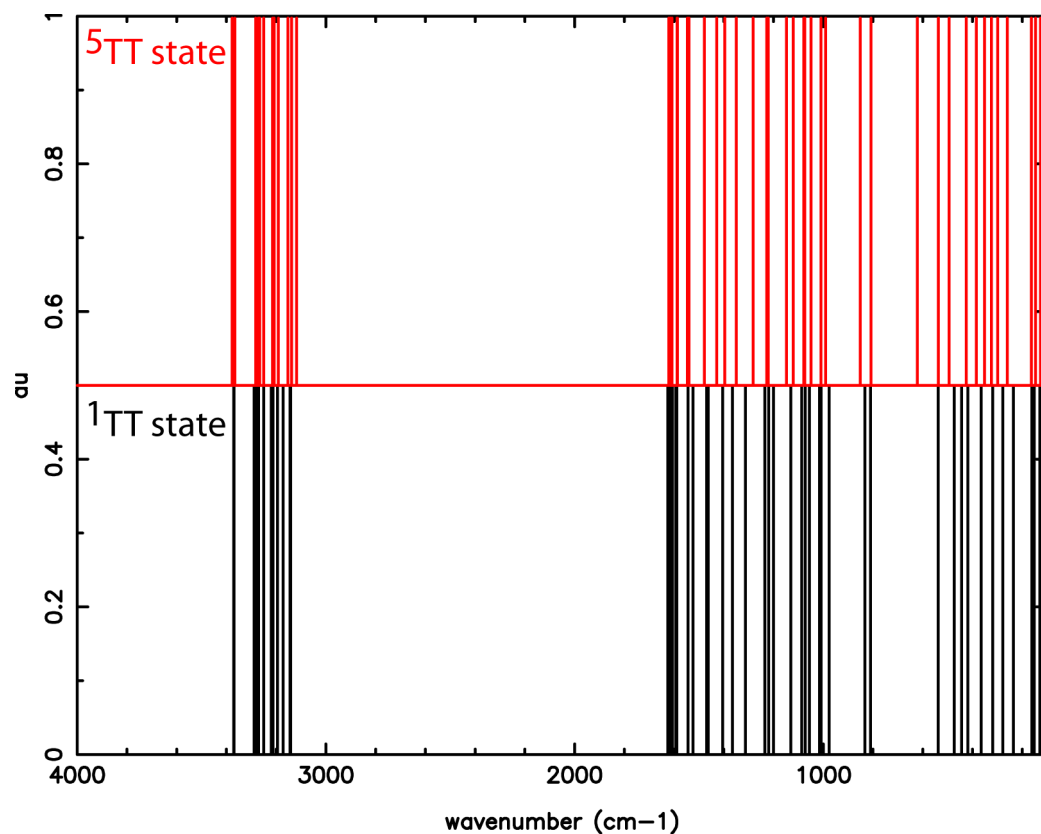
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Geometry and frequencies (CASSCF)



5TT is a sufficiently good representation of 1TT



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Electron transfer rate

- Fermi Golden rule in diabatic representation (Marcus theory)

$$k_{ET} \propto \left| \langle \Psi_f | H | \Psi_i \rangle \right|^2$$

- Electronic coupling between diabatic states
- Adiabatic representation: Non-adiabatic couplings (Landau-Zener model)
 - Potential energy surfaces and conical intersections/crossings

P.F. Barbara, T.J. Meyer, M.A. Ratner, *J. Phys. Chem.* **100** (1996), 13148
F. Bernardi, M. Olivucci, and M.A. Robb, *Chem. Soc. Rev.* **25** (1996), 321



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MCSCF-CASPT2-MRCI

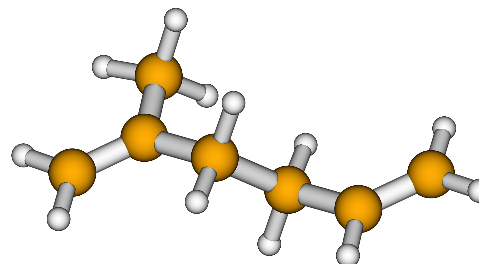
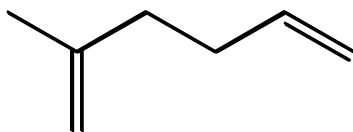
$$\Psi = \Psi_0 + \sum_{i,a} c_i^a \Psi_i^a + \sum_{ij,ab} c_{ij}^{ab} \Psi_{ij}^{ab} + \dots$$

- Balanced description of all relevant states
- Accurate
- Non-adiabatic coupling elements $\langle \Psi_i | \frac{\partial}{\partial Q} | \Psi_f \rangle$
- Expensive or impossible for large systems
- Chemical interpretation is not trivial

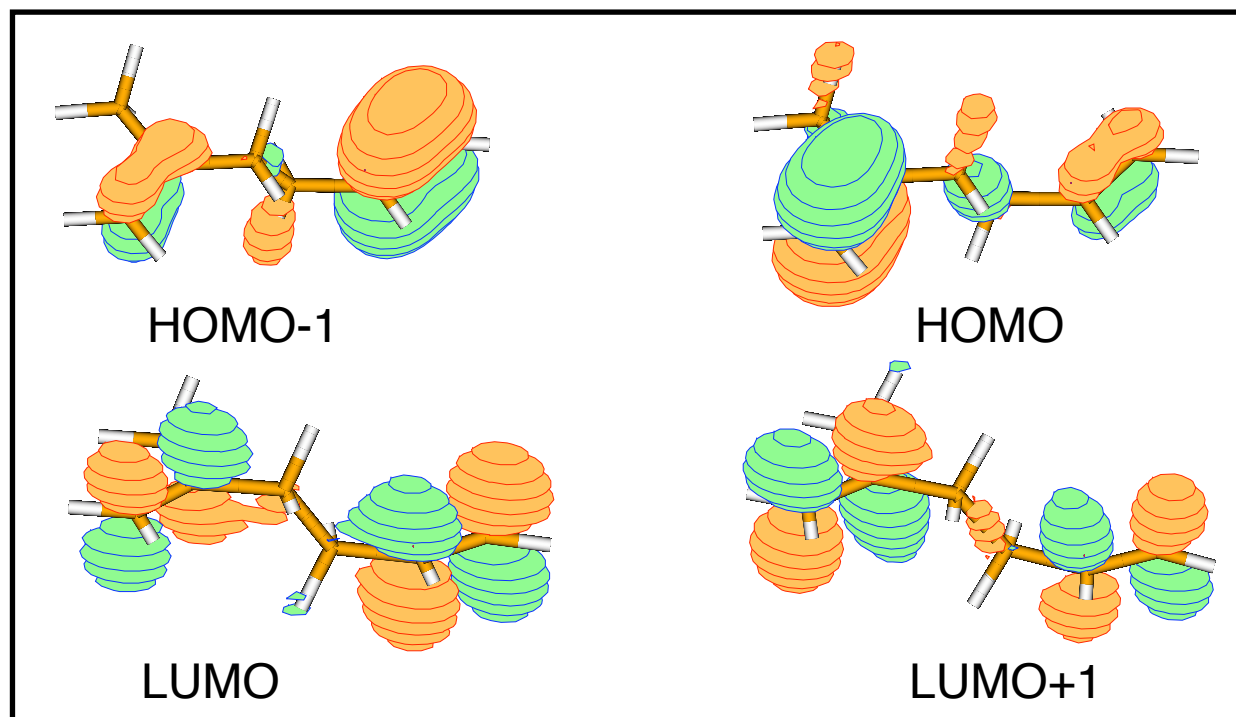


2-methyl-1,5-hexadiene

Molecule:



active
space



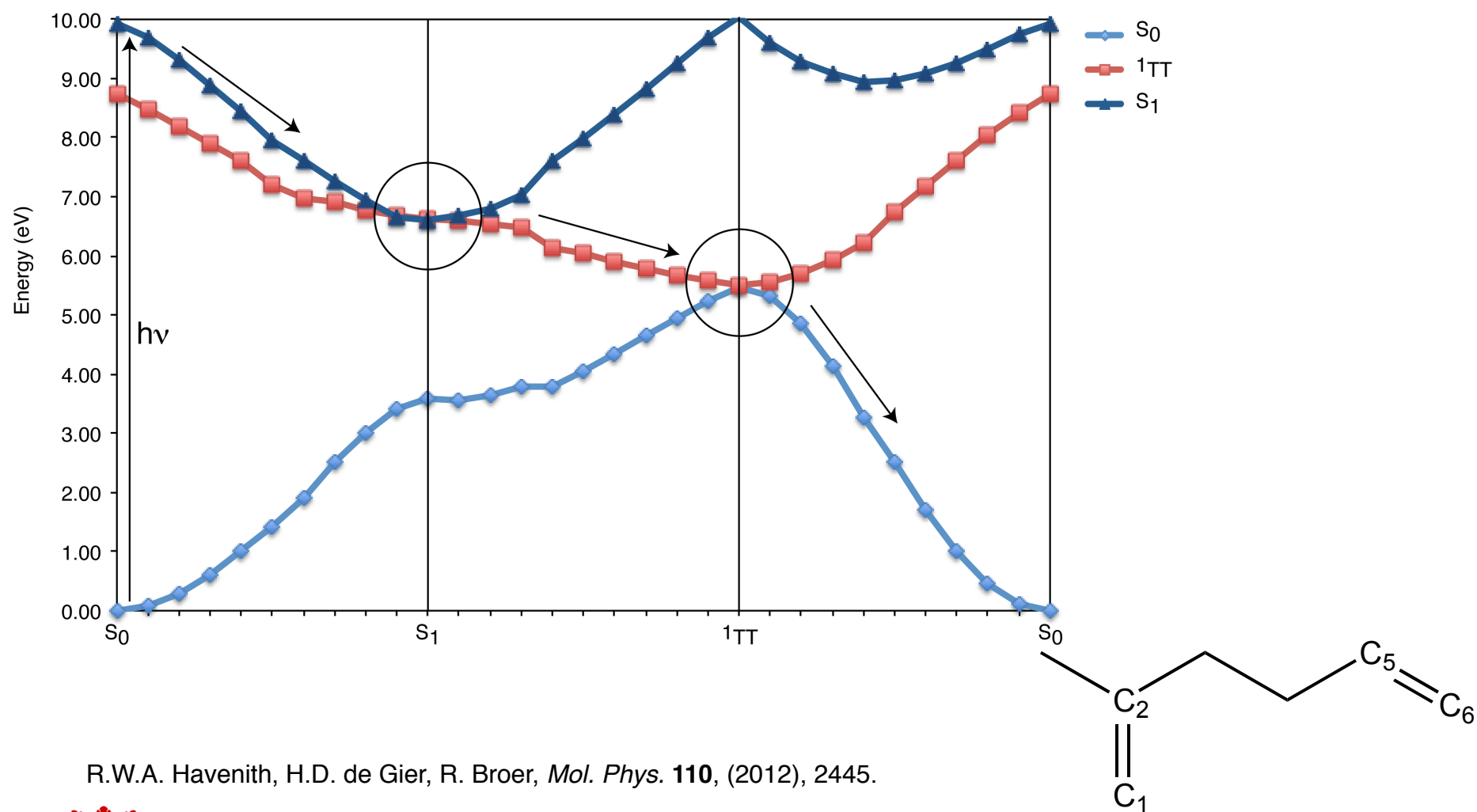
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Adiabatic potential energy surfaces



R.W.A. Havenith, H.D. de Gier, R. Broer, *Mol. Phys.* **110**, (2012), 2445.



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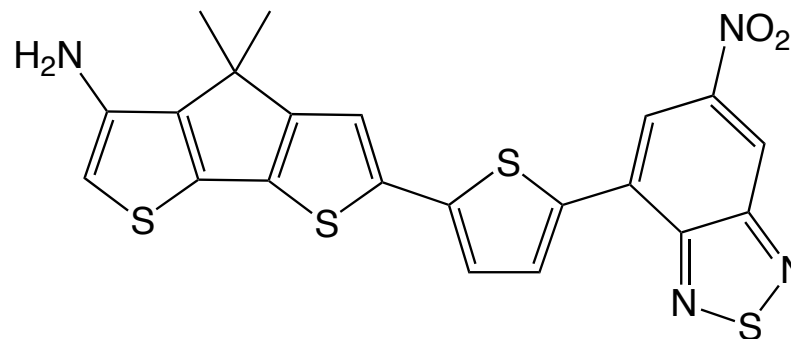
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Theoretical Chemistry

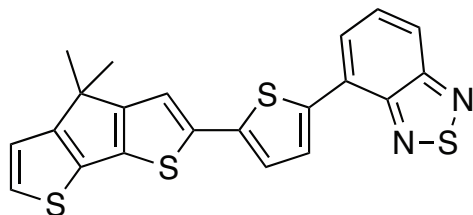
Promoting charge separation

- Modify electronic structure of the chromophore
 - Introduction of dipoles: push-pull substitution

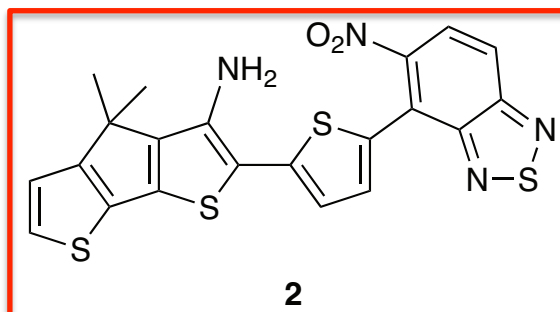


Functionalisation of polymer

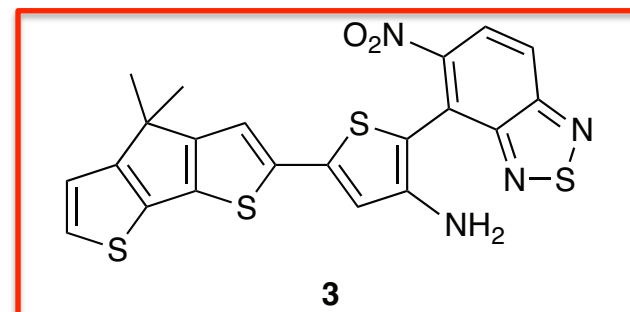
- Addition of push-pull groups to lower the exciton binding energy:
 - Mesomeric: **linear-** (2, 3) *versus* **cross-**conjugation (4, 5)
 - **Inductive** (6)



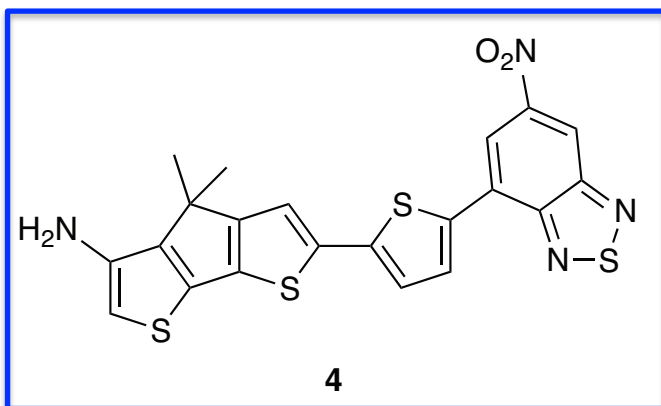
1



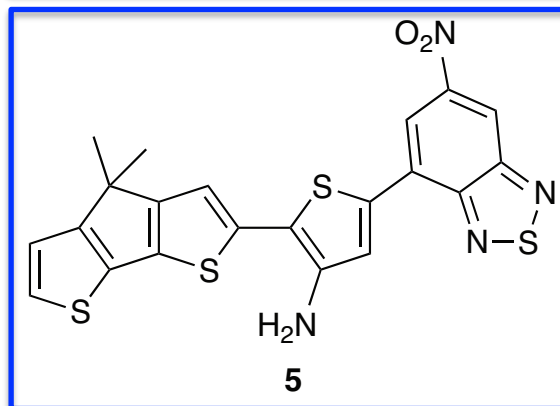
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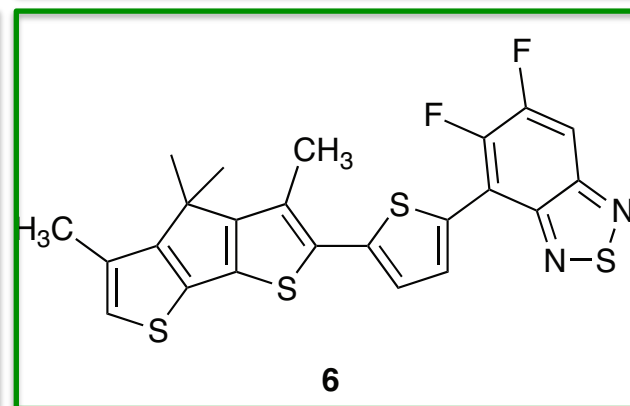
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4



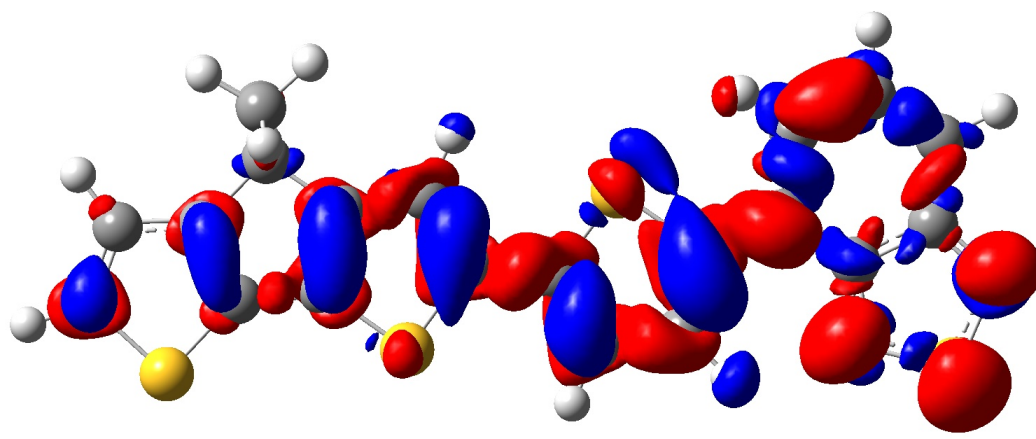
5



6



Enhancement of charge transfer



0.19

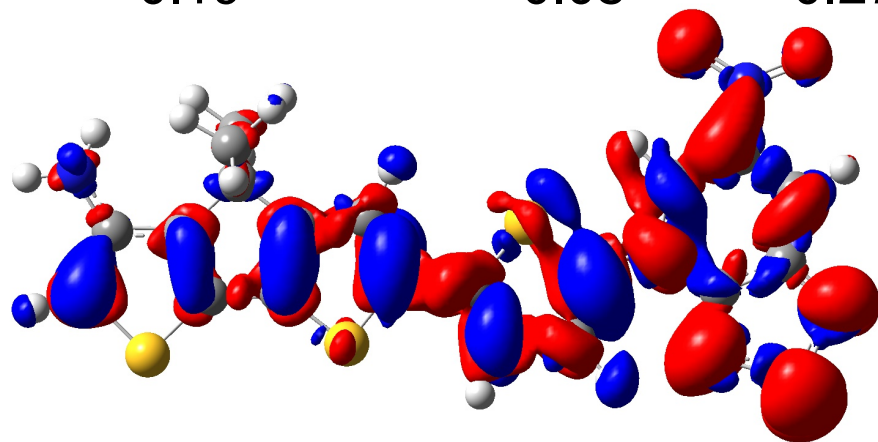
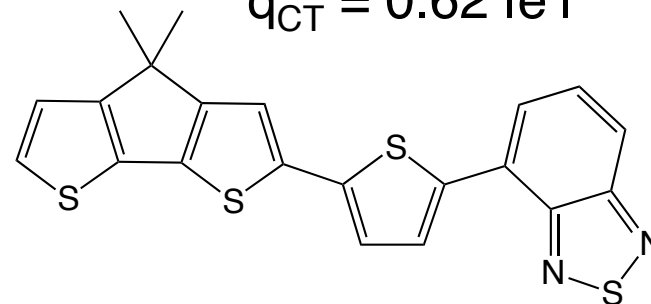
0.08

-0.27

1

$$D_{CT} = 4.02 \text{ \AA}$$

$$q_{CT} = 0.62 \text{ } |e^-|$$



0.30

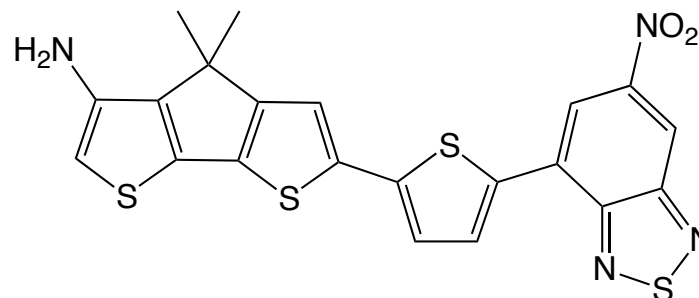
0.10

-0.40

4

$$D_{CT} = 4.53 \text{ \AA}$$

$$q_{CT} = 0.82 \text{ } |e^-|$$



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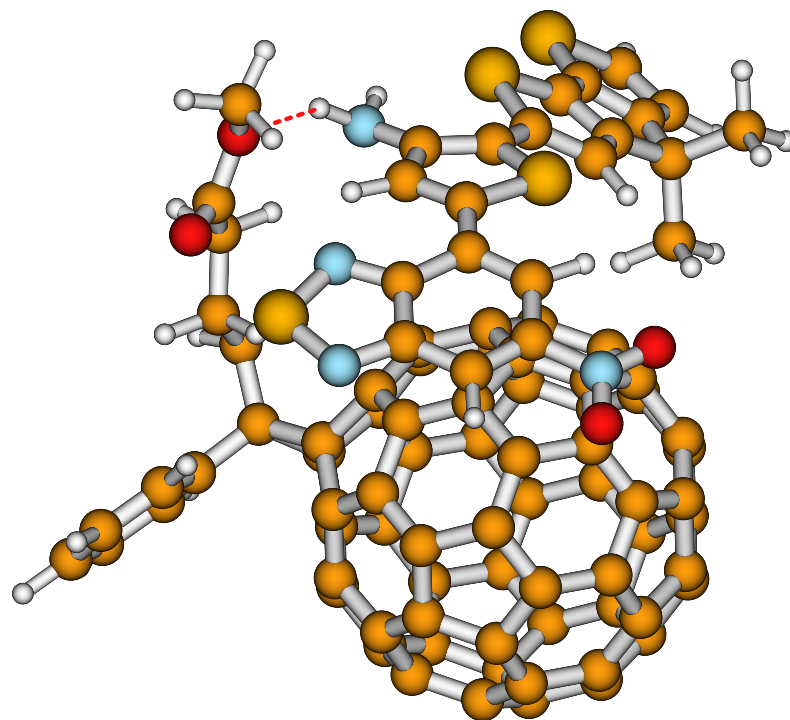
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Theoretical Chemistry

Exciton binding energies

Exciton binding energies in eV

	μ (Deb)	$(E_b^{\text{exc}})_{\text{polymer}}$	$(E_b^{\text{exc}})_{\text{PCBM}}$
1	3.02	2.70	1.97
2	2.39	2.61	2.01
3	2.79	2.79	2.07
4	4.04	2.46	1.95
5	4.52	2.42	1.93
6	2.35	2.73	1.97

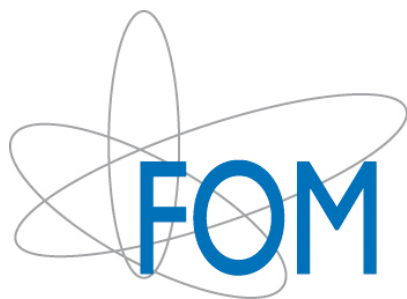


- Marked differences between linear and cross conjugation disappear and all CT exciton binding energies are similar



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Theoretical Chemistry



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