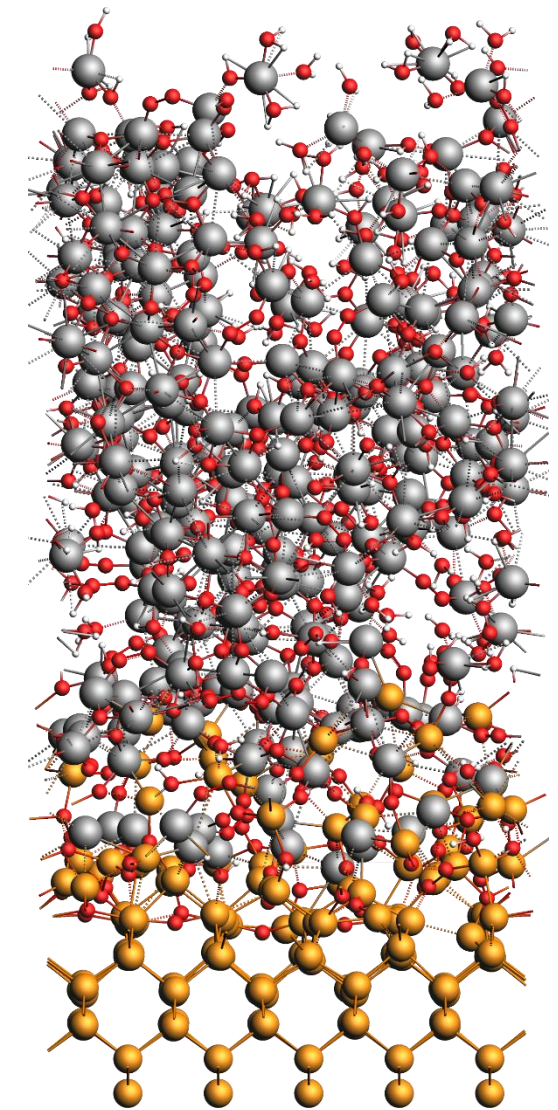
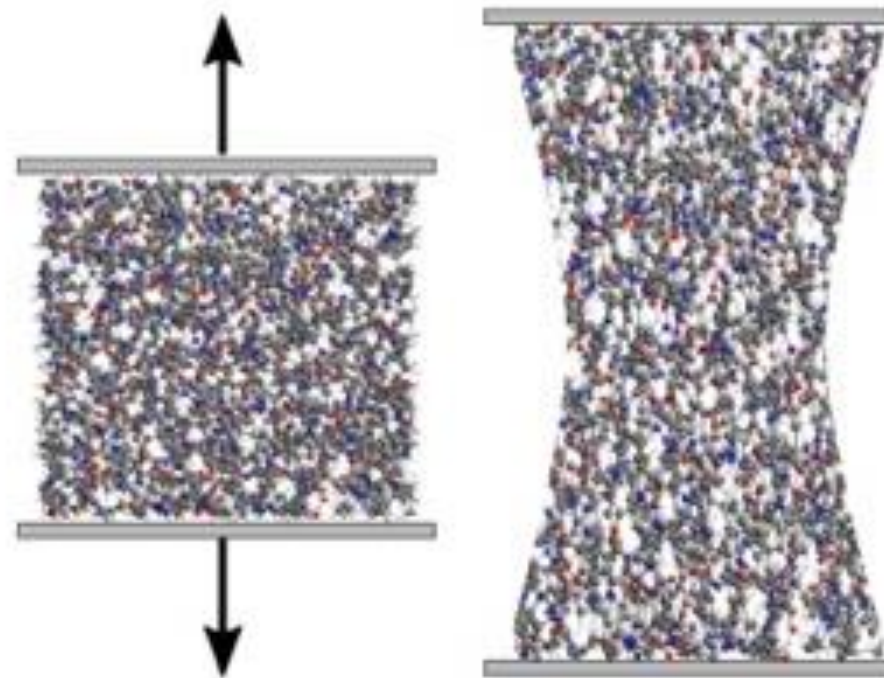
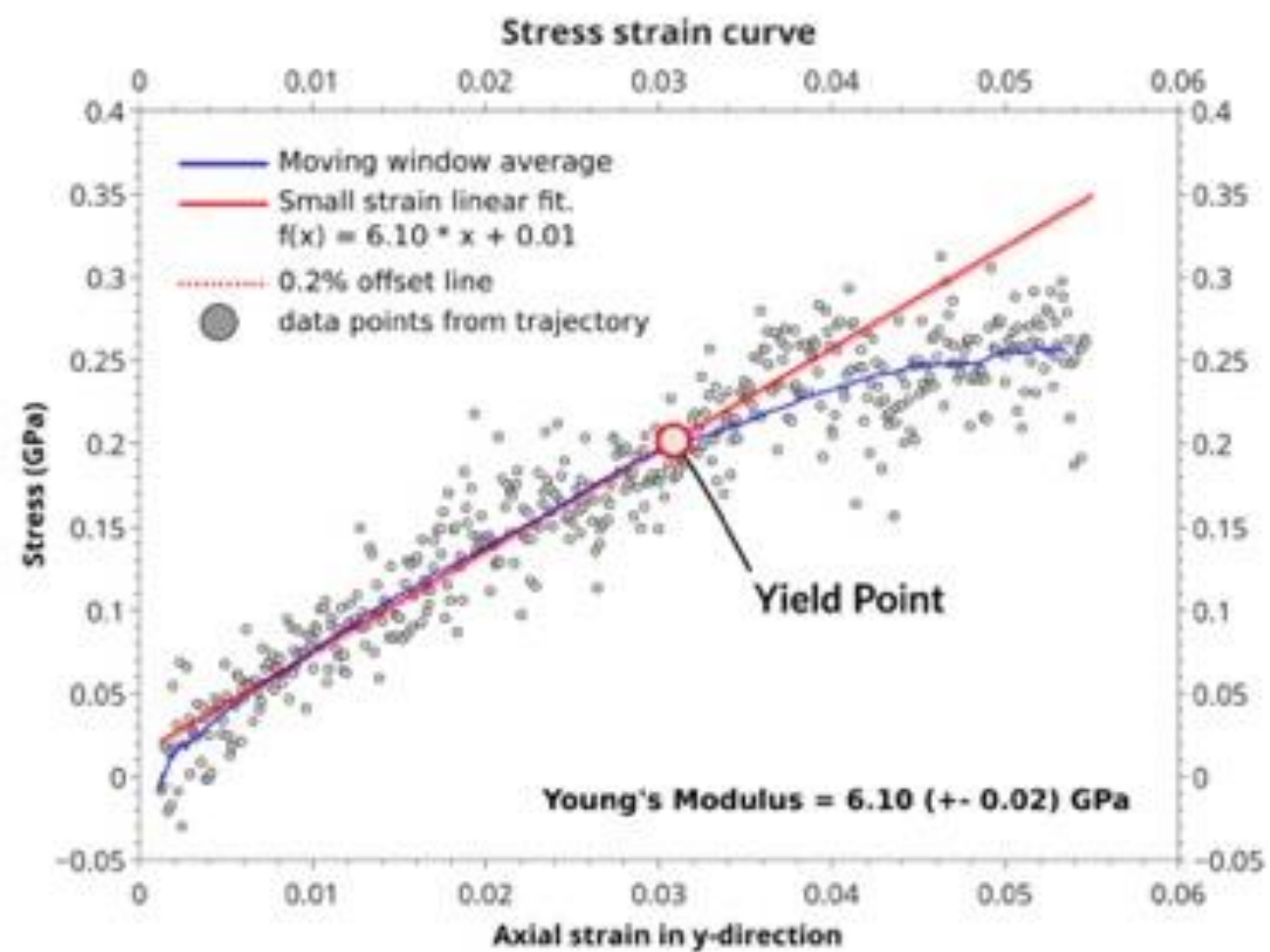


# Amsterdam Modeling Suite

## Atomistic & Multiscale Simulations for Chemistry & Materials

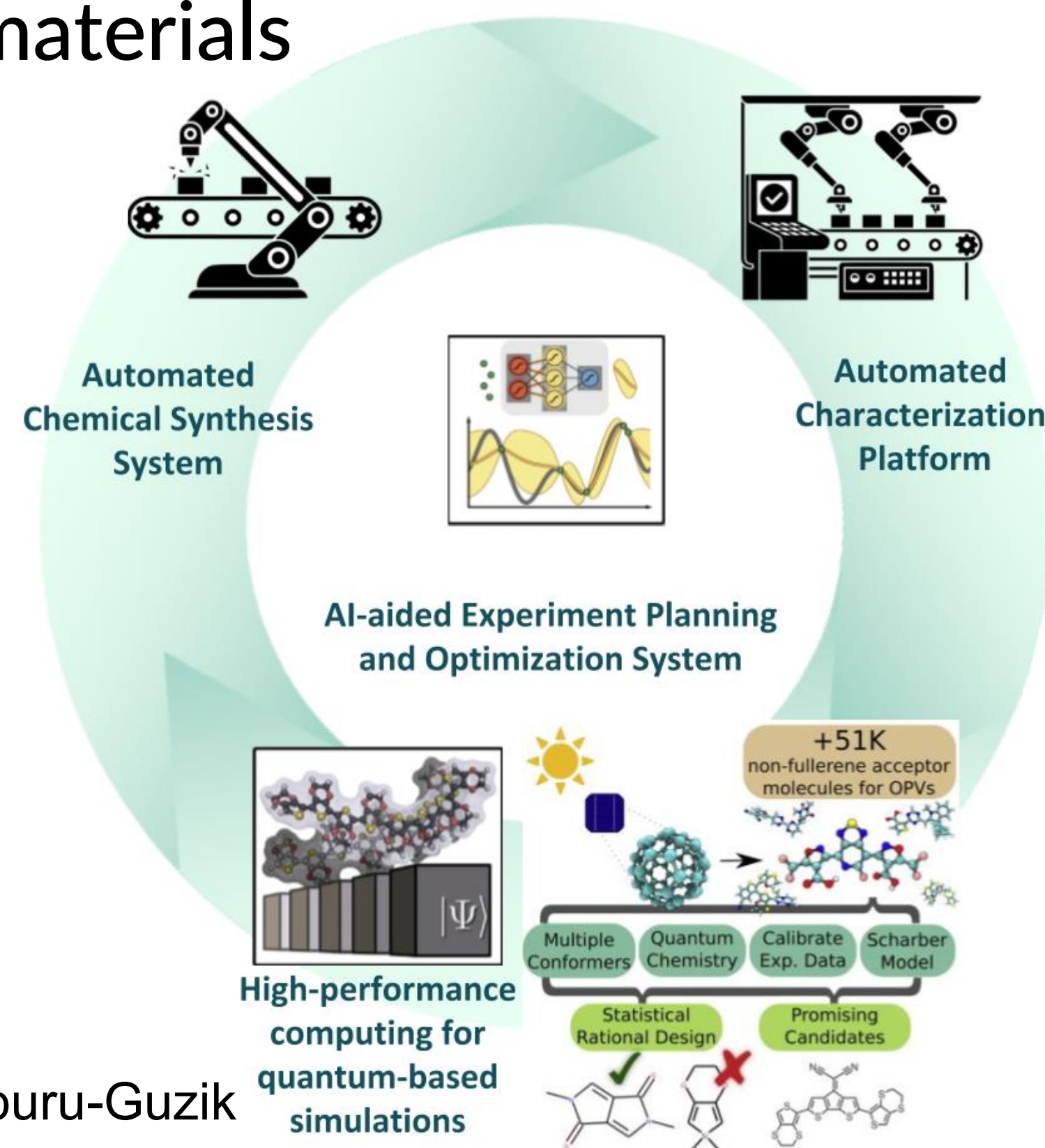


# New materials discovery too slow

- 8-19 years to develop materials solutions in new markets
- 80-85% R&D programs fail
- >50% R&D spending only incremental improvement
- **Catch 22:** slow discovery  $\Leftrightarrow$  few new materials

<https://www.mckinsey.com/industries/chemicals/our-insights/chemical-innovation-an-investment-for-the-ages>

Simulations -> predict new materials  
Robots -> make new materials  
AI -> improve simulations and DOE

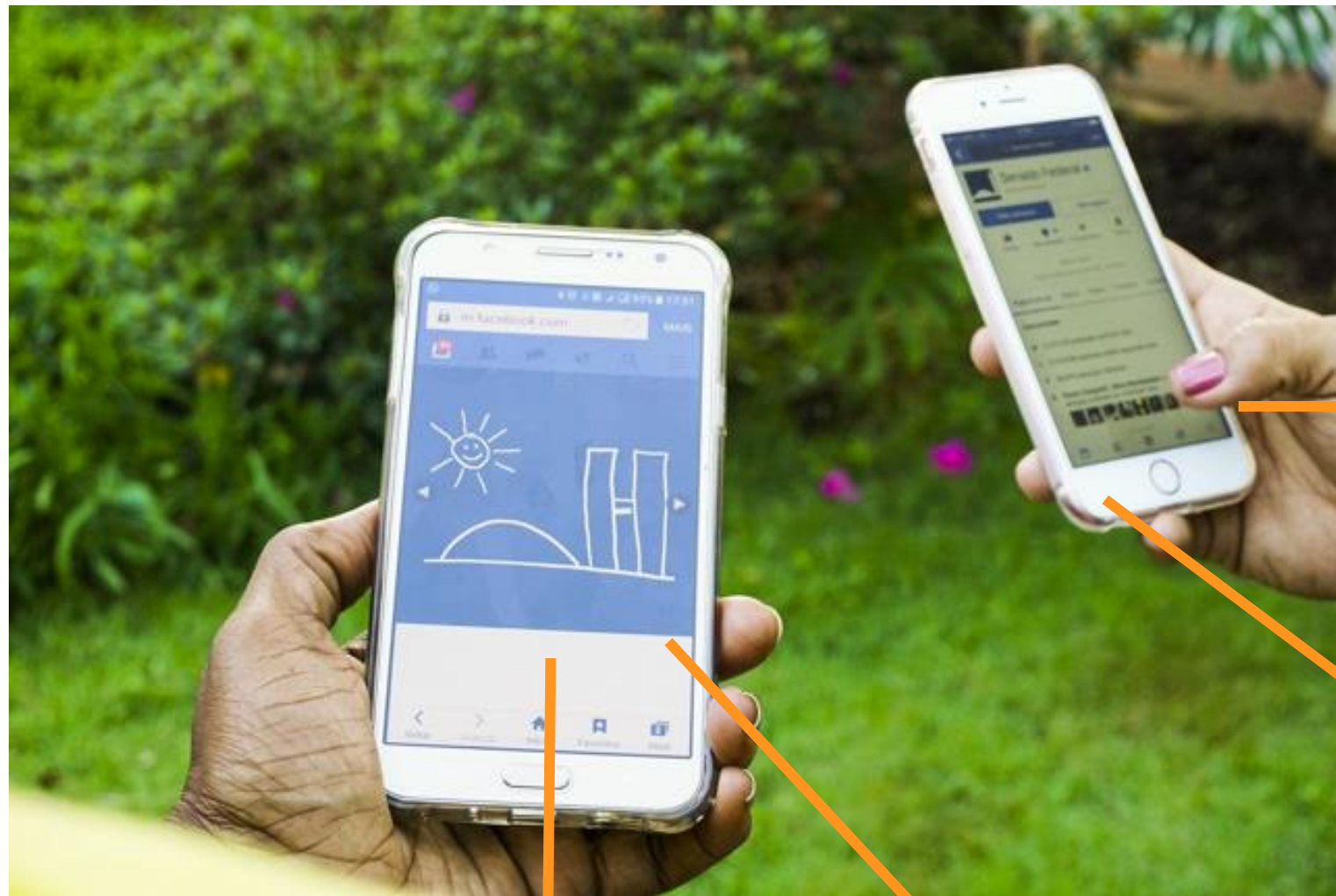


the Matter Lab - Aspuru-Guzik

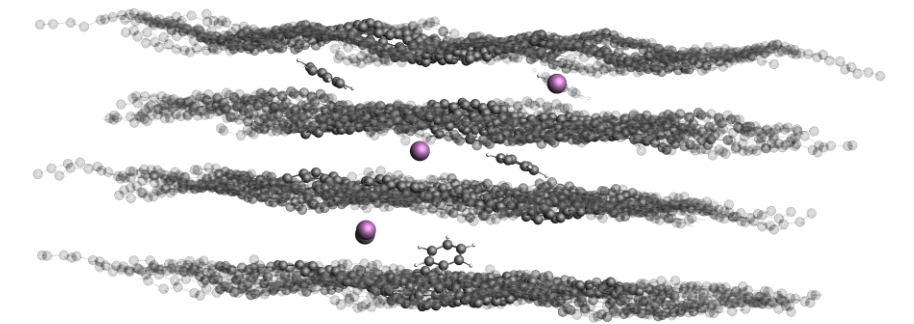
# Bottom up Property Prediction

## Atomistic modeling

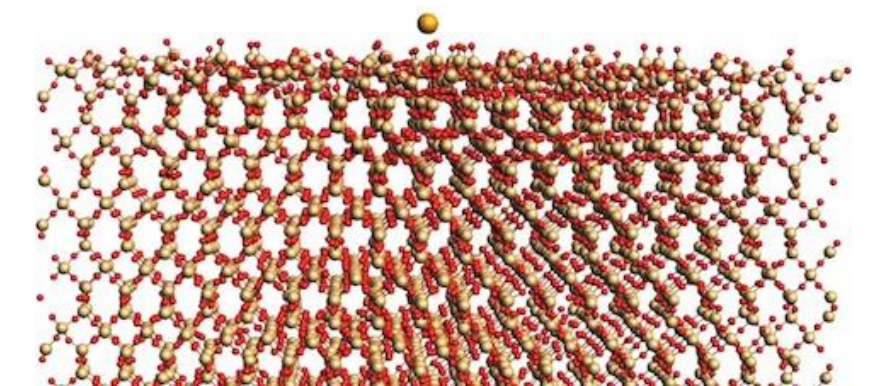
- Decreases search space
- Develops understanding  
Input for meso & macro



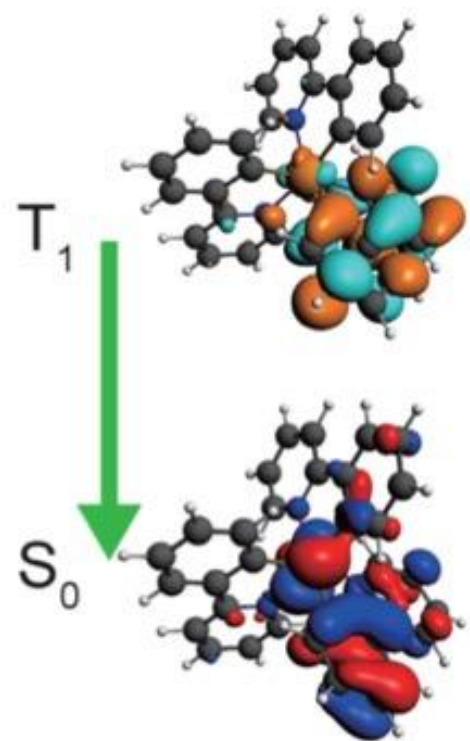
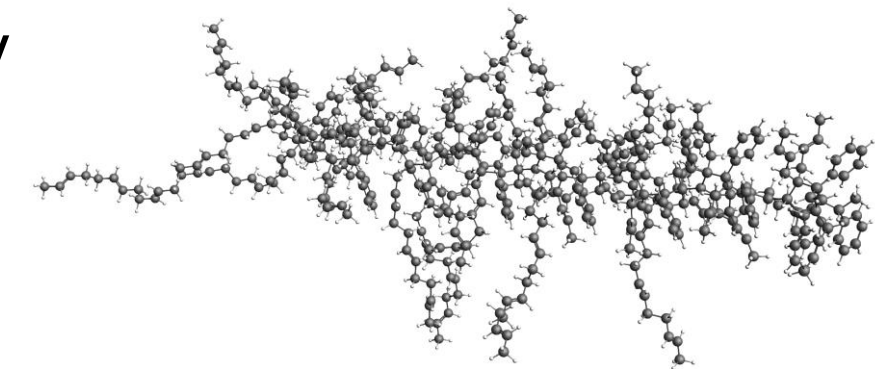
Batteries:  
fast recharge,  
high capacity



(m)CPU: high  
capacity, I-V profiles



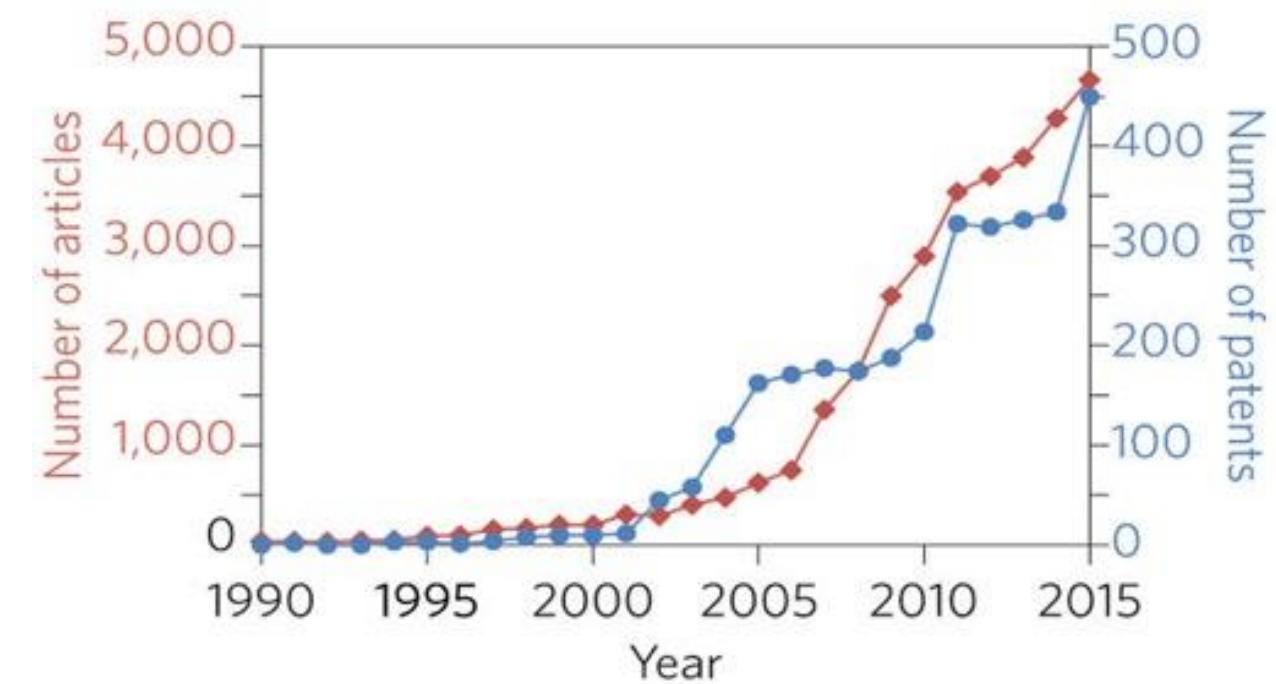
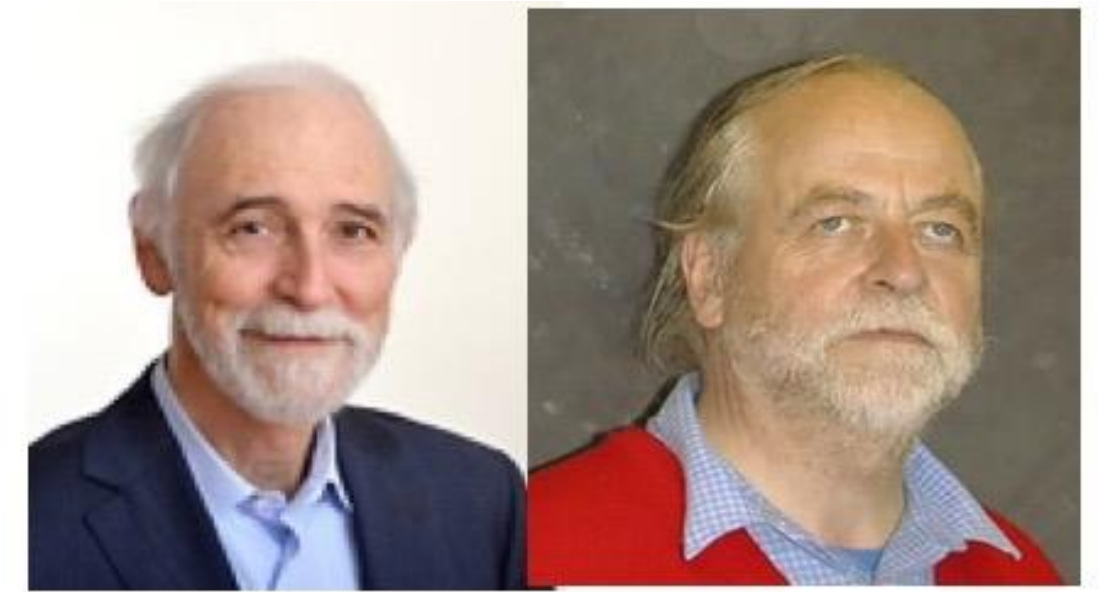
Glass, coating: optical  
properties, conductivity



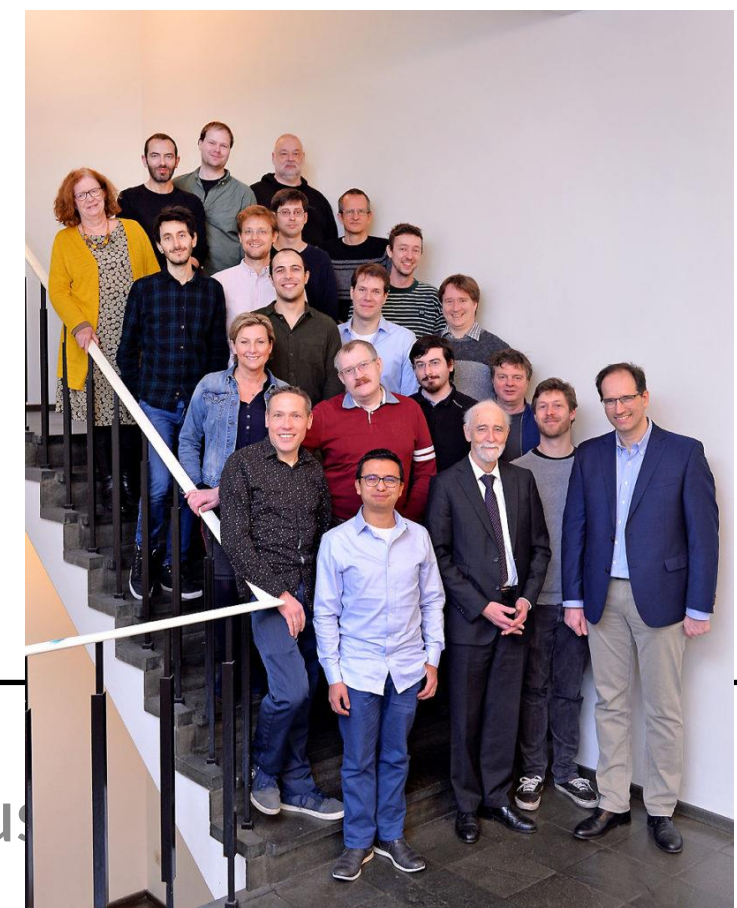
OLEDs, QLEDs: color,  
lifetime & efficiency

# History: Software for Chemistry & Materials

- 1970s ADF: first DFT code to understand chemistry
  - Baerends@VU & Ziegler@Calgary(+)
- 1980s: Mitsui, Shell, Akzo, Unilever:
  - Training in Amsterdam -> optimize catalysts
- 1995 SCM: Spin-off company
  - Continuous support & Development
- 2010s: DFTB, ReaxFF, COSMO-RS (Albemarle, DSM)
- 2019: Multi-scale: ReaxPro (BASF, Dow, Shell, JM)
- 2023: ~30 people (19 senior PhD's)
  - Many collaborations nonprofit & industry
  - SCM: development, debug, port, optimize, & support

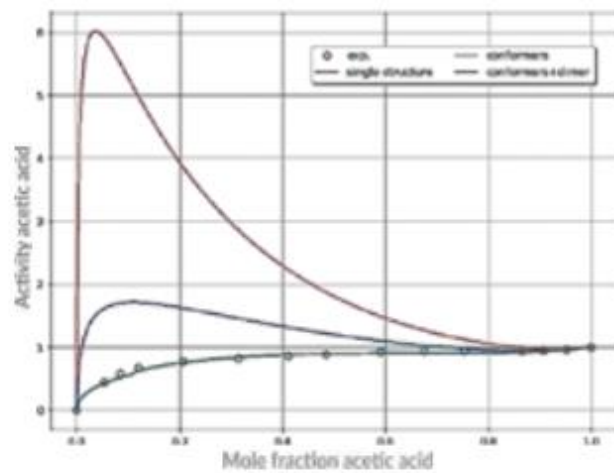


articles & patents in materials science with "density functional theory", Nat. Mat. 4619



# Amsterdam Modeling Suite

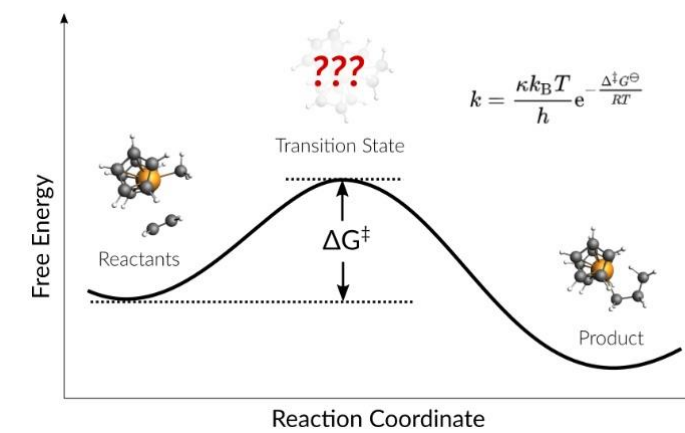
Continuum



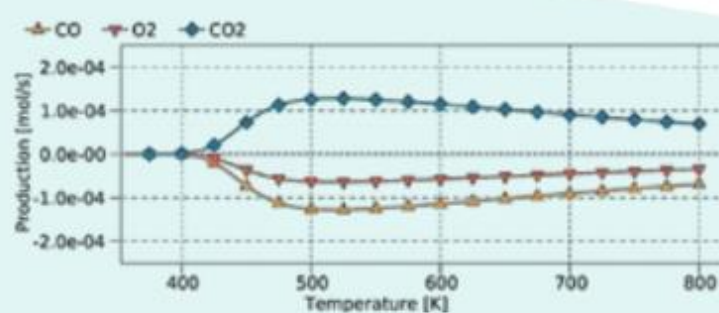
## Fluid Thermodynamics

COSMO-RS  
COSMO-SAC  
UNIFAC

AMS Driver: PES exploration,  
MD, MC, TS, IR, phonons, ...



Mesoscale

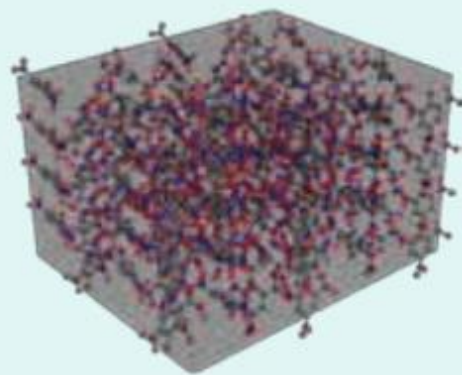


## Kinetics

Kinetic Monte Carlo  
Microkinetics

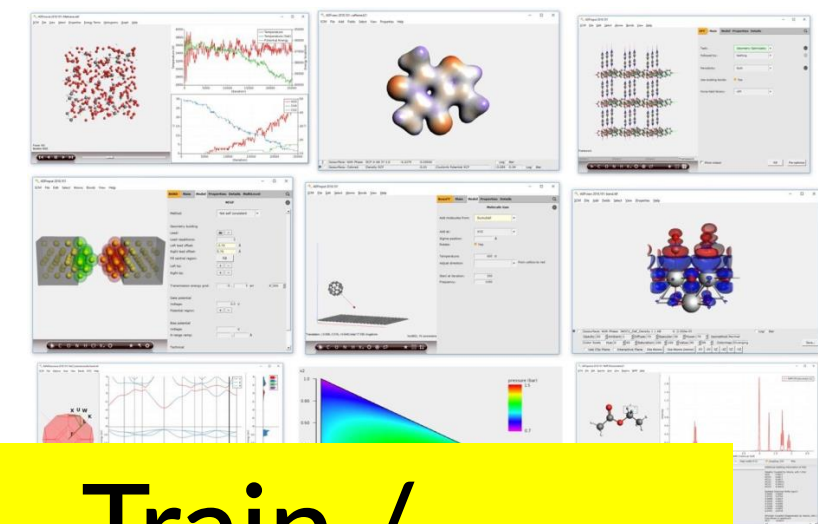
Integrated GUI, remote jobs

Materials

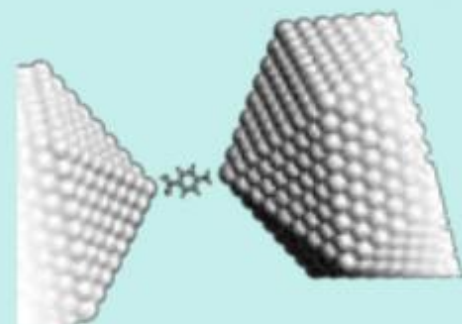


## Force Fields

ReaxFF, GFN-FF  
Machine Learning Potentials  
Apple & P



Nano

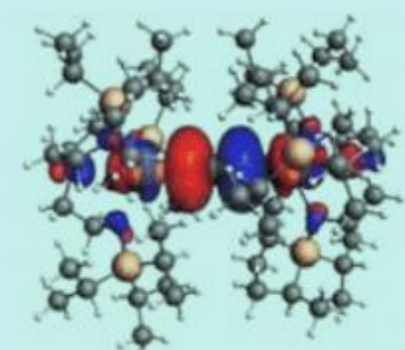


## QM/MM

FDE, Hybrid Engine

Train /  
parametrize

Atomistic



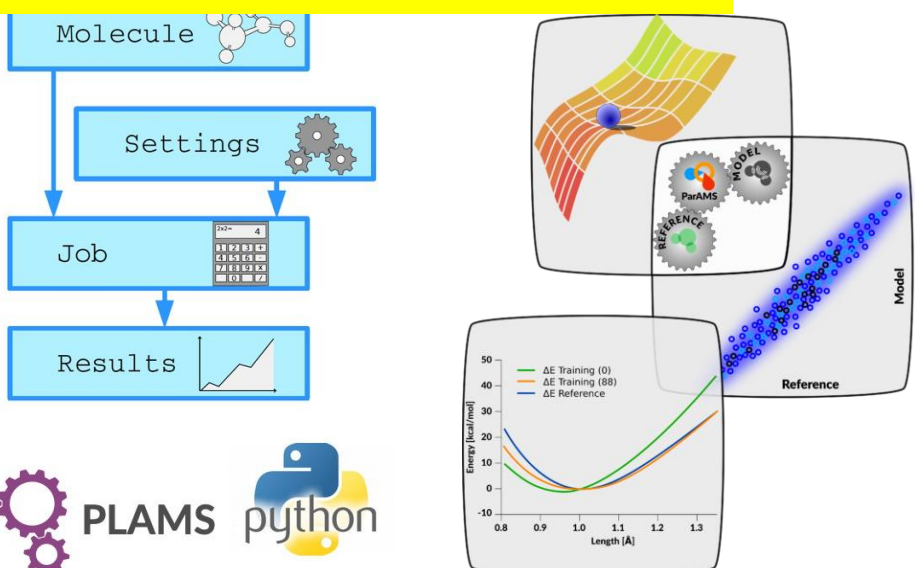
## Tight binding

GFN-xTB, DFTB

Py

## Periodic DFT

BAND, Quantum Espresso



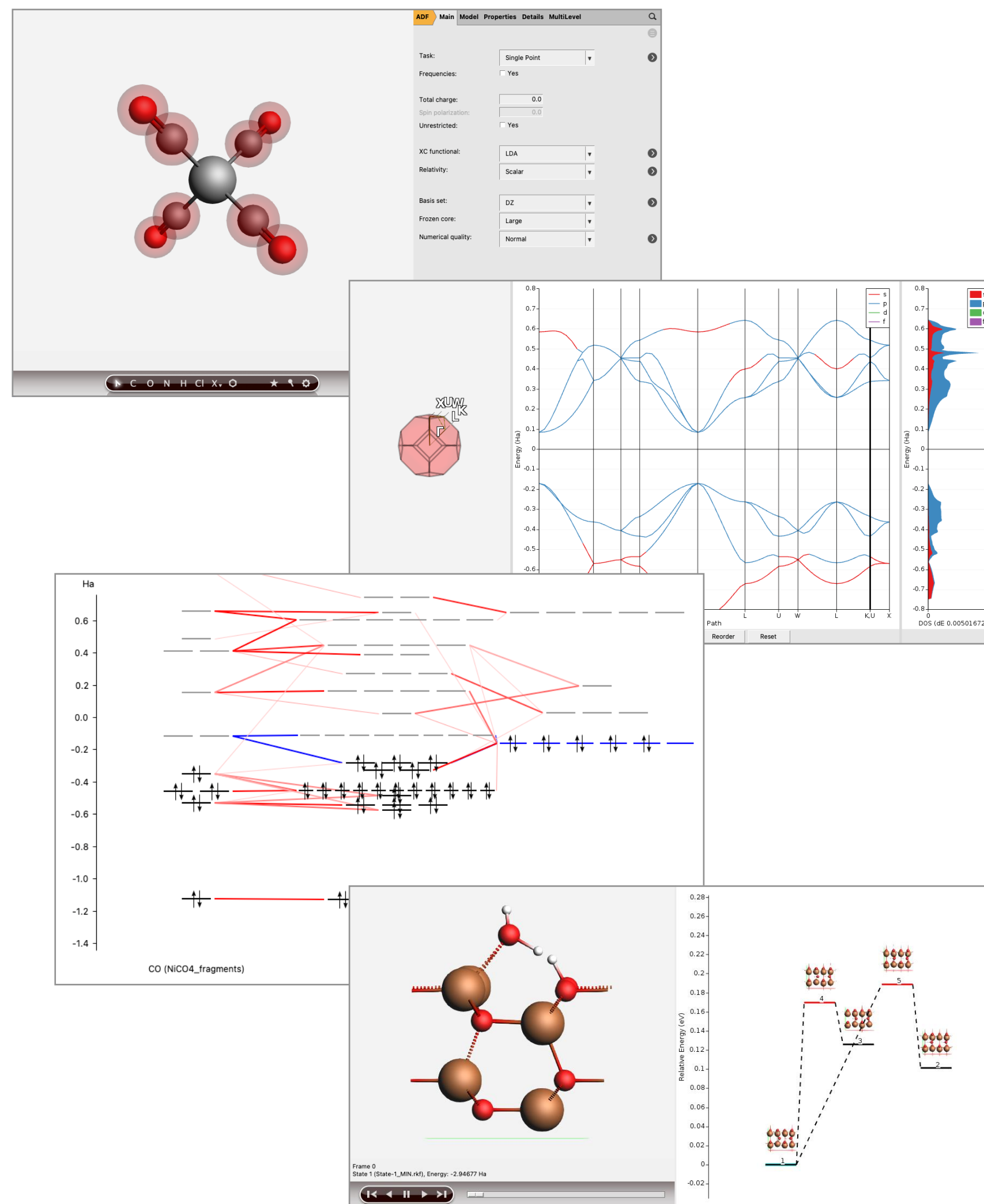
## Molecular DFT

ADF

# The graphical user interface (GUI)

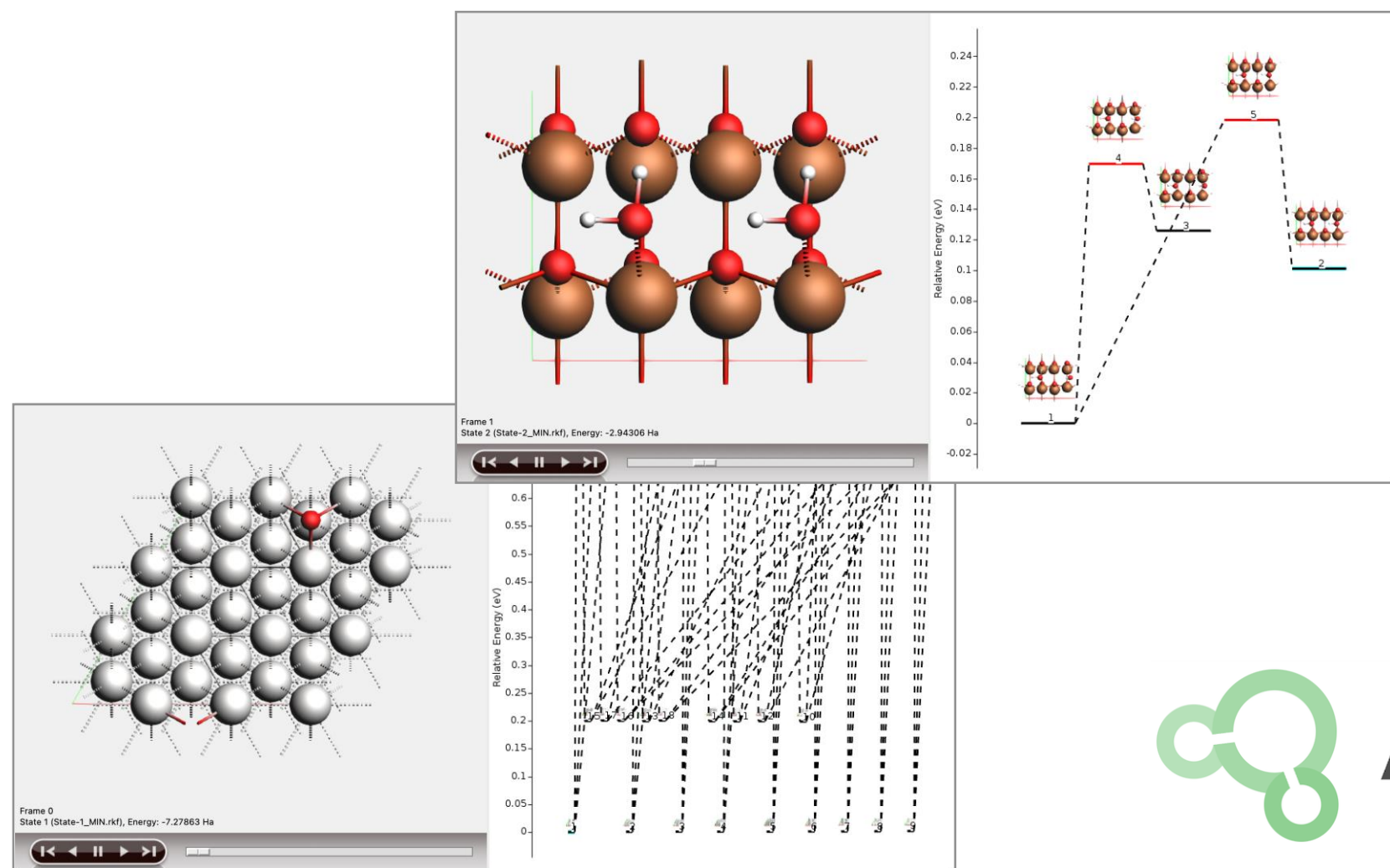
## Setup & analyze calculations

- AMSjobs
  - manage jobs, locally or remotely
  - extract summaries
- AMSinput
  - build molecules, periodic systems, polymers, etc.
  - import structures from many formats
- AMSview, AMSlevels, AMSspectra, etc.
  - analyze results
  - visualize structures, transition states, orbital densities, DOS, band structure, spectra, etc.



# AMS driver: MD with 'anything'

A unified driver to explore the Potential Energy Surface (PES)








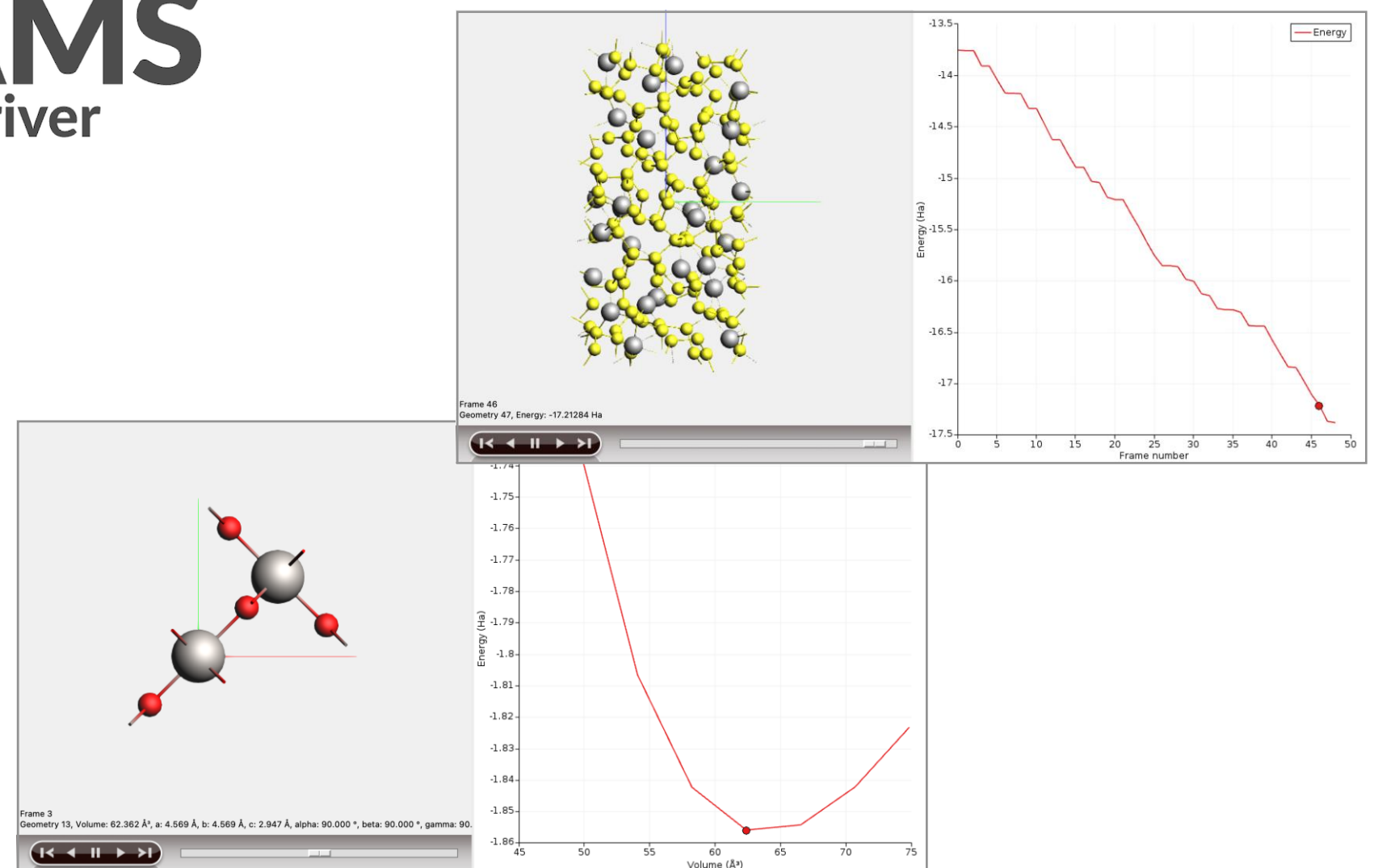
- Tasks
- Molecular dynamics
  - Frequencies & phonons
  - Stress & elastic tensors
  - Scan coordinates & constraints
  - Monte Carlo, etc.



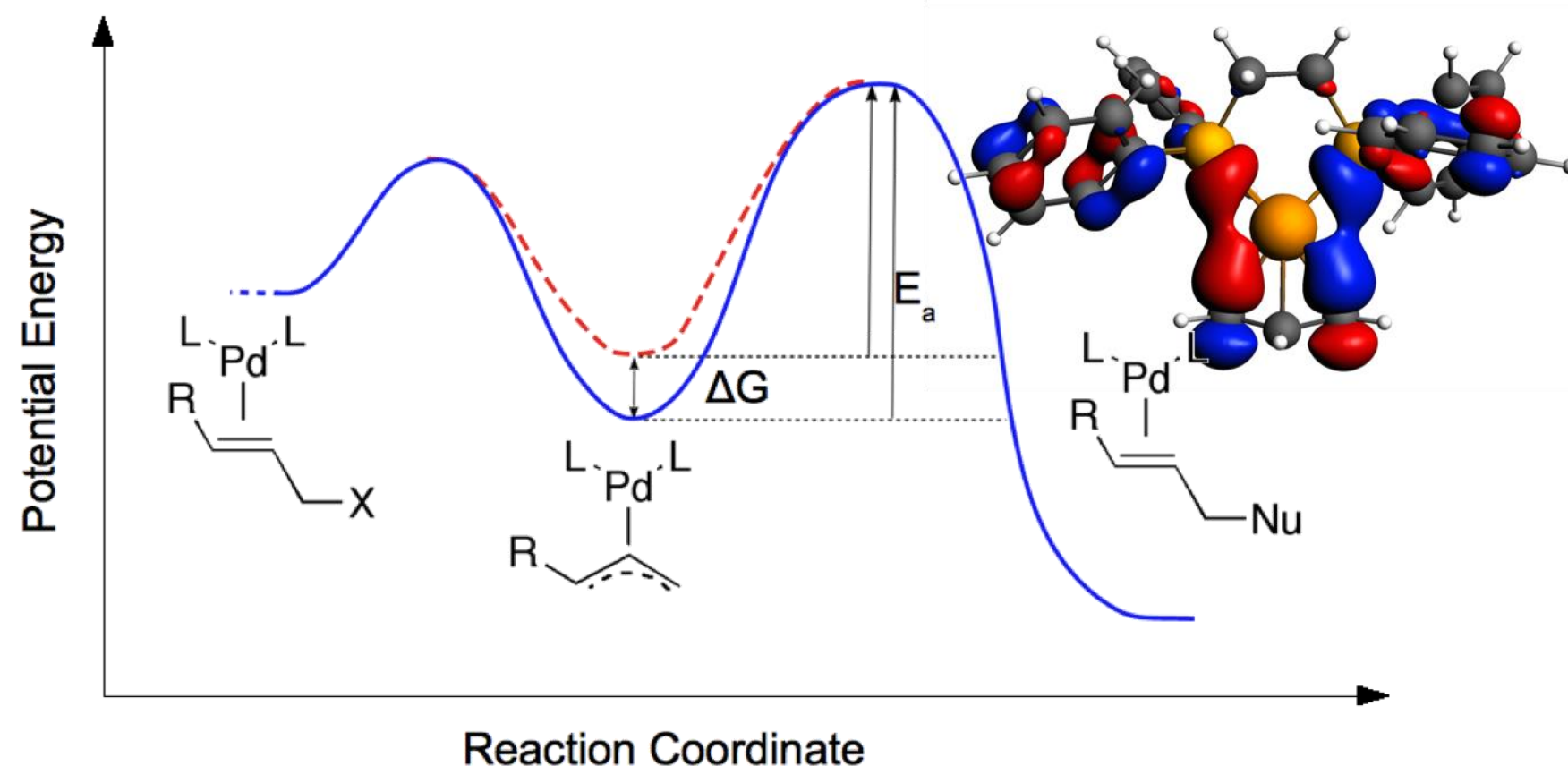
 **AMS**  
driver

Engines

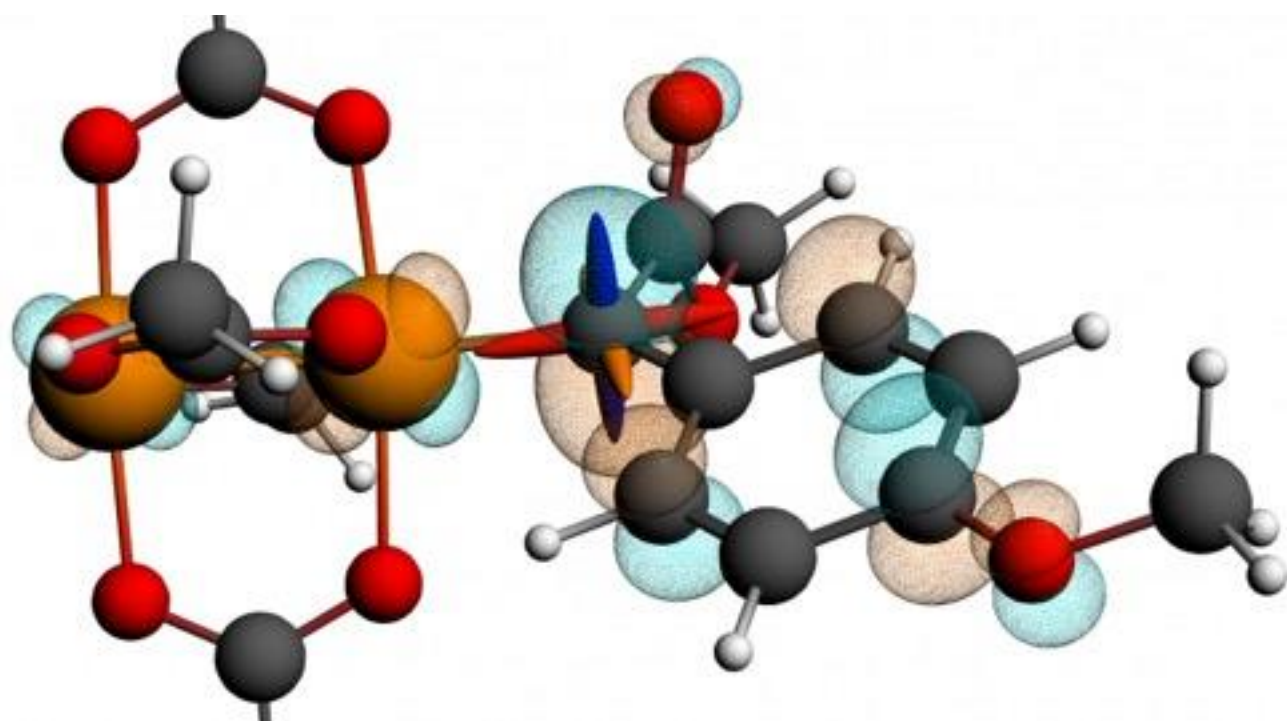
-  ADF
-  BAND
-  DFTB
-  ReaxFF
-  MLP



# ADF: Molecular DFT



Bonding analysis: Understanding Catalyst-Substrate Interactions [Nature Chem. 2, 417 \(2010\)](#)



NMR calculations locate <sup>13</sup>C di-Rh carbene catalyst intermediate, [Science, 342, 351 \(2013\)](#)

## Strong & unique points

- All-electron Slaters, **all elements**
- ZORA & X2C scalar relativistic
- **Spin-orbit coupling**
- Fast GW, BSE, double hybrids

## Spectroscopy

- EPR, NMR, IR, Raman, UV/VIS, XANES,

## Bonding analysis

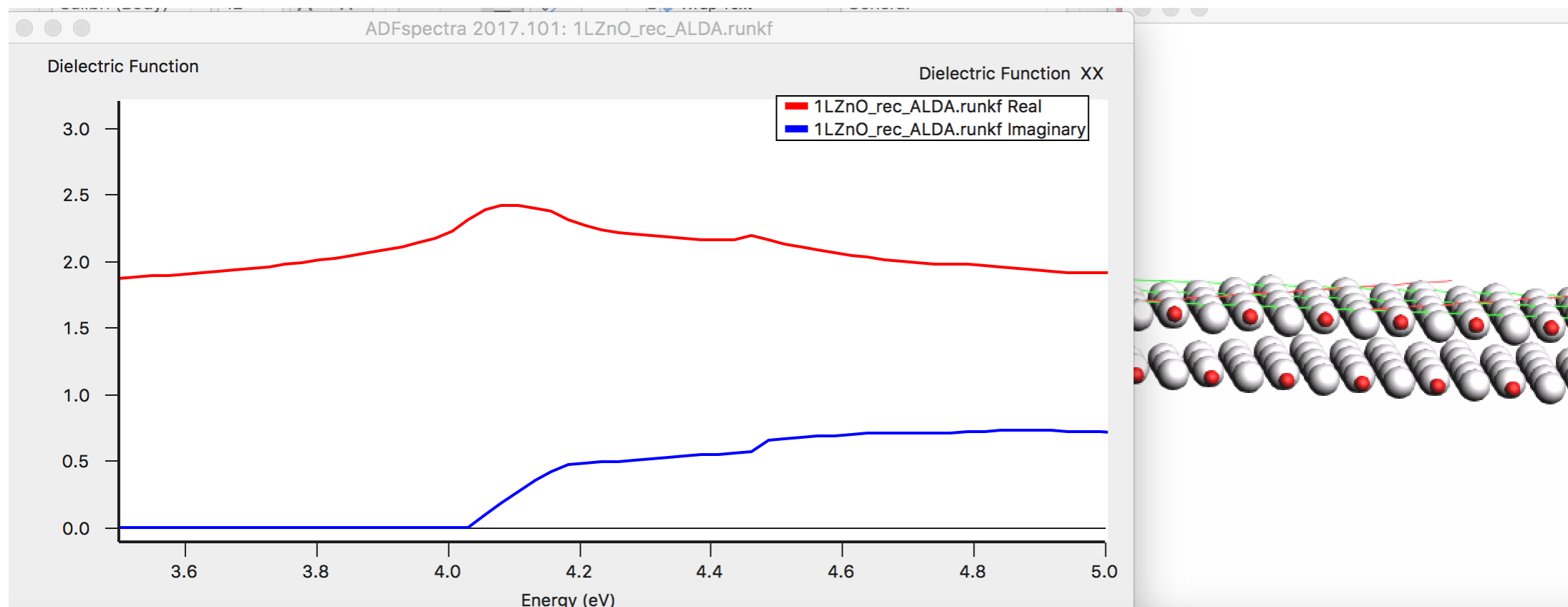
- ETS-NOCV, QTAIM, MO diagrams, NCI
- NTOs, CT descriptors
- TDM between excited states
- Transfer integrals

## Environments

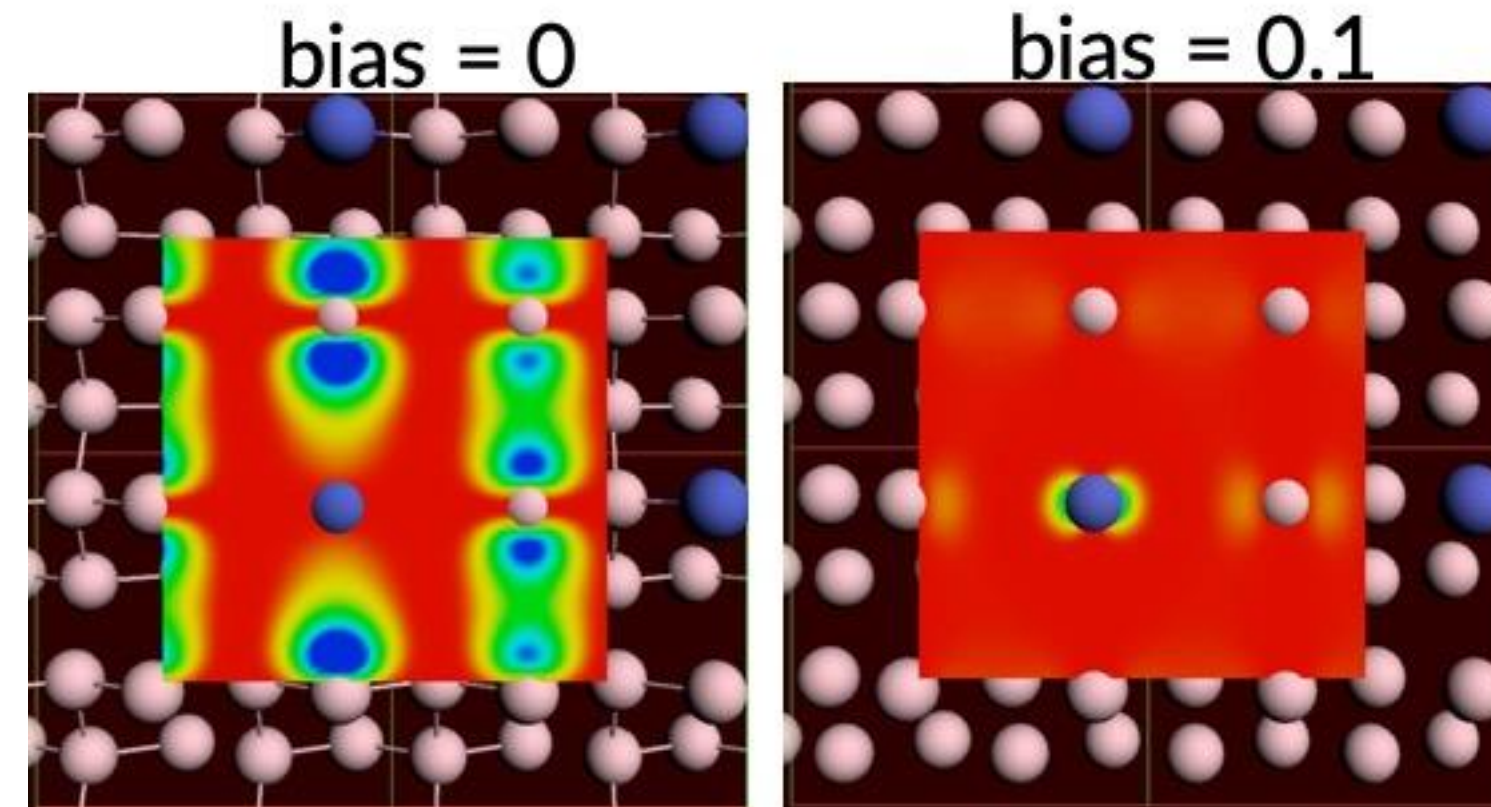
- COSMO, SM12, 3D-RISM
- Subsystem DFT (FDE)
- DRF, SCRF, DIM/QM, QM/FQ, multi-layer



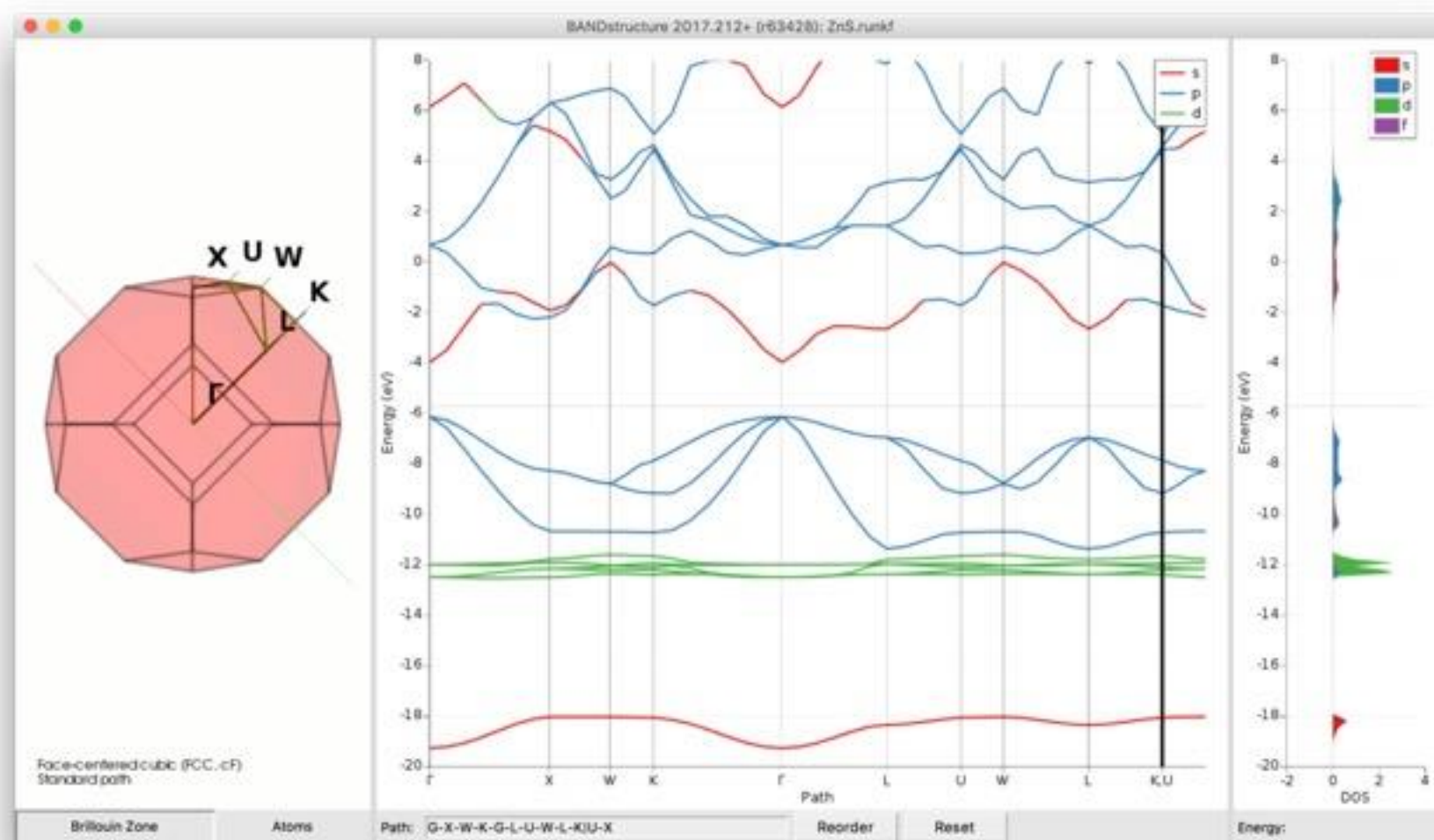
# BAND Periodic DFT with AOs



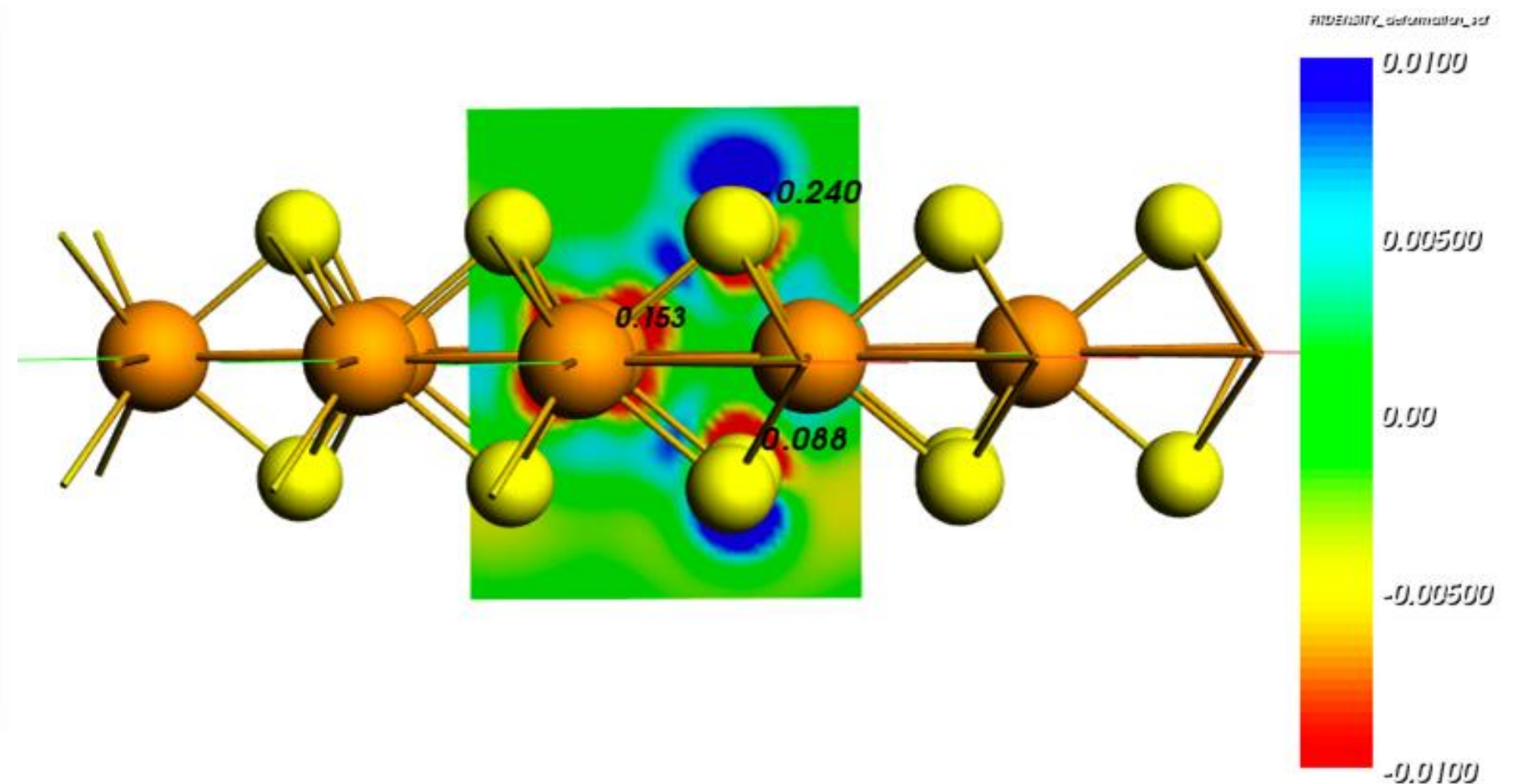
Dielectric function ML ZnO (2D TDCDFT)



LDOS: STM PtGe(100)



Band structure, pDOS, fat bands ZnS

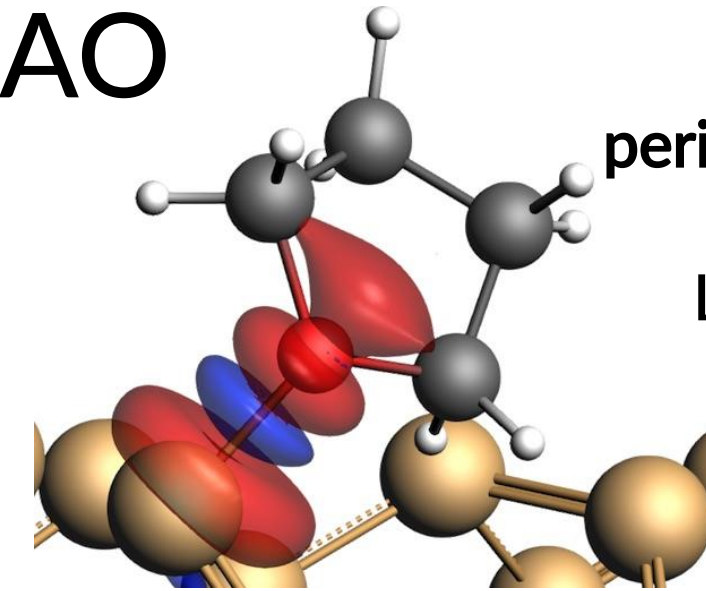


Polarizing MoS<sub>2</sub> with an electric field

# BAND + Plane Wave codes (QE)

- Atom centered basis functions, STO + NAO

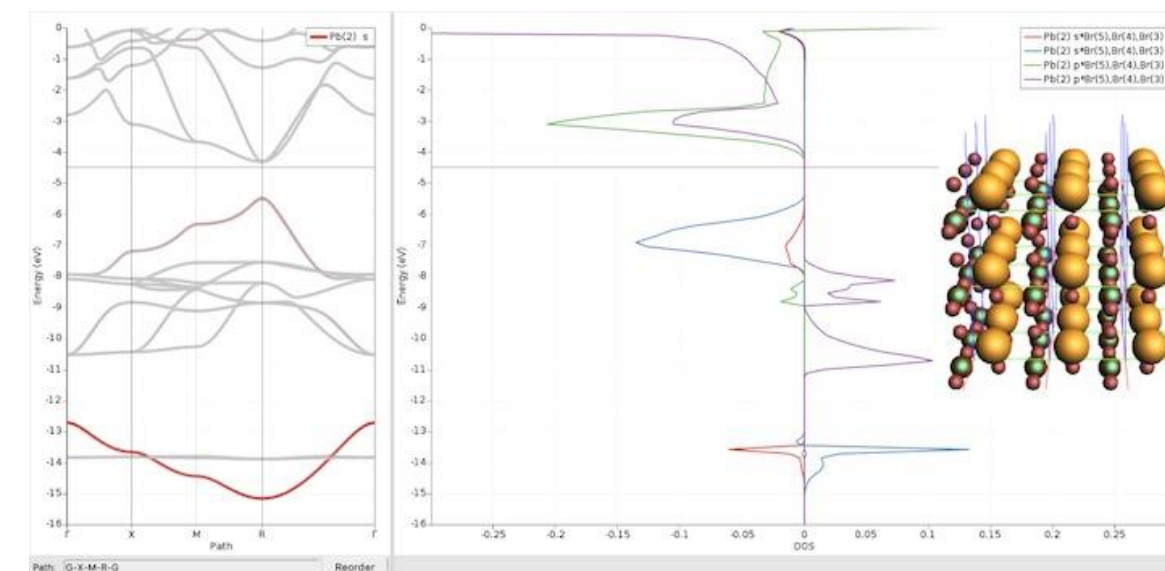
- Compare cluster with periodic
- No pseudopotentials, basis sets for all elements
- Core spectroscopy (core holes)
- Easy orbital analysis: pDOS, COOP, EDA
- Dielectric function: polarization functional, TDCDFT
- xc: TASK, SCAN, MN15-L, HSE06, GLLB-sc, D3(BJ), DFT-1/2, r2SCAN-D4
- Self-consistent spin-orbit coupling
- Self-consistent NEGF (biases, spin)



periodic energy decomposition analysis ([tutorial](#))  
L. Pecher and R. Tonner  
[WIREs CMS, \(2018\)](#)

- True 2D surfaces, 1D polymers

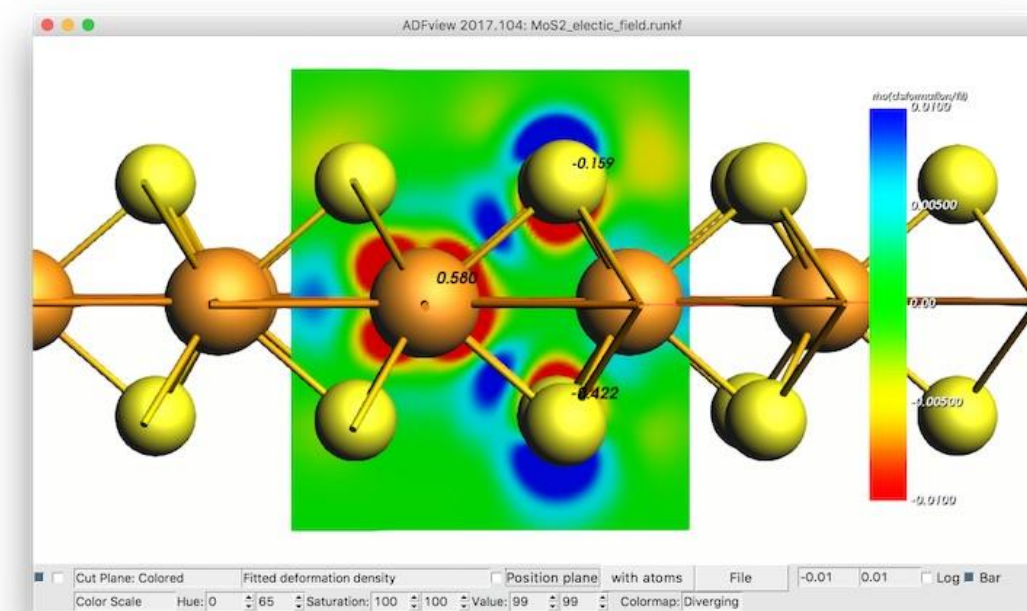
- Solvation: COSMO, SM12
- Polarization in z-direction
- 1D, 2D electronics (homogeneous E field)
- Nanotubes



COOP in perovskites ([tutorial](#))  
Goesten & Hoffmann  
[JACS \(2018\)](#)

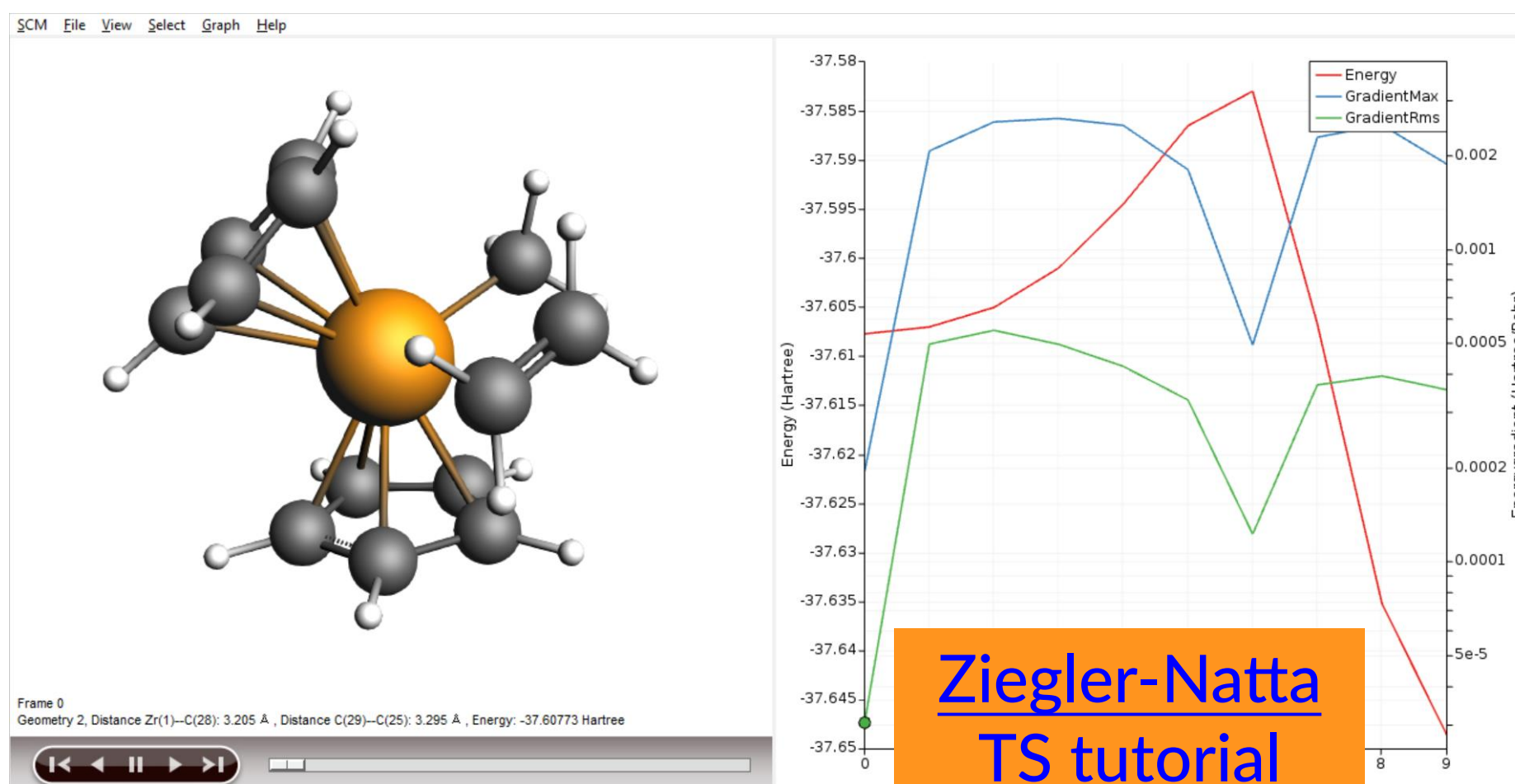
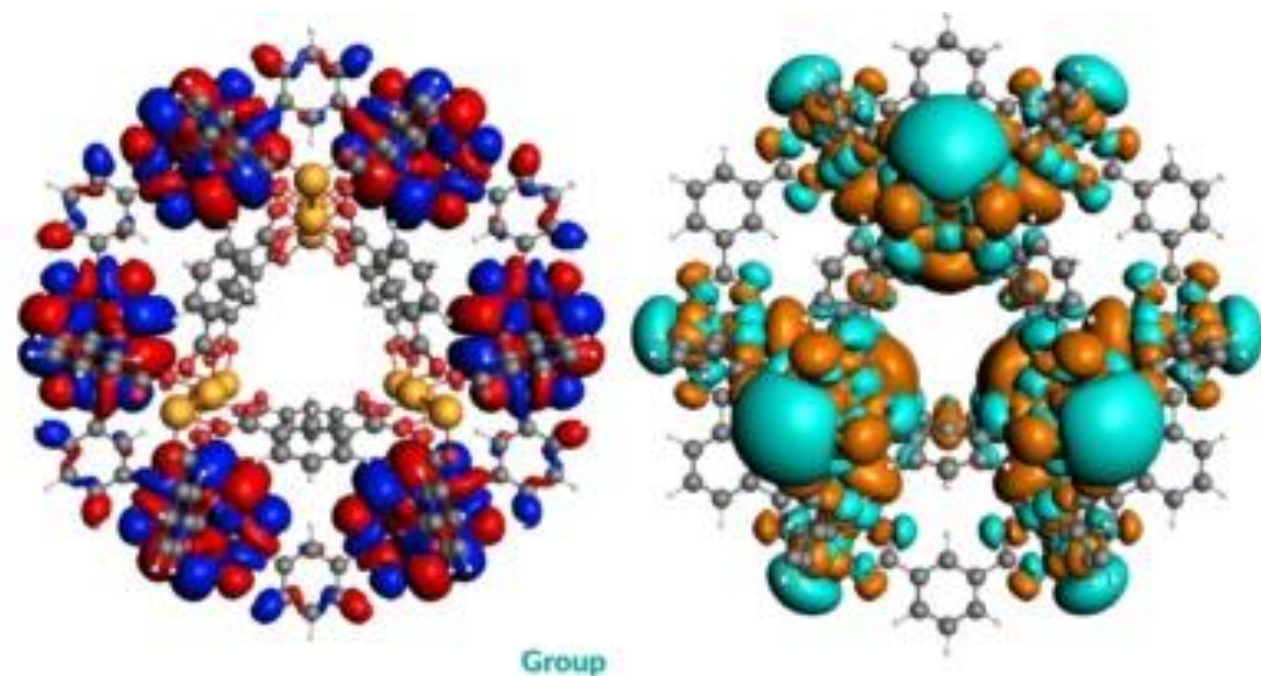
- Integrated GUI + Python I/F

- Easy set up & analysis
- Switch: ADF, DFTB, BAND, QE (VASP)
- Parametrize DFTB, ReaxFF, MLP



Polarizing 2D semiconductor ([tutorial](#))  
N. Zibouche et al.  
[PCCP \(2014\)](#)

# DFTB: 'fast DFT' for molecules & periodic



## Approximated DFT

- Nearest neighbor & minimal basis
- Tabulated elec & rep. parameters:
  - Grimme GFN-xTB (Z = 1-86)
    - GBSA solvation
  - QuasiNaNo & DFTB.org

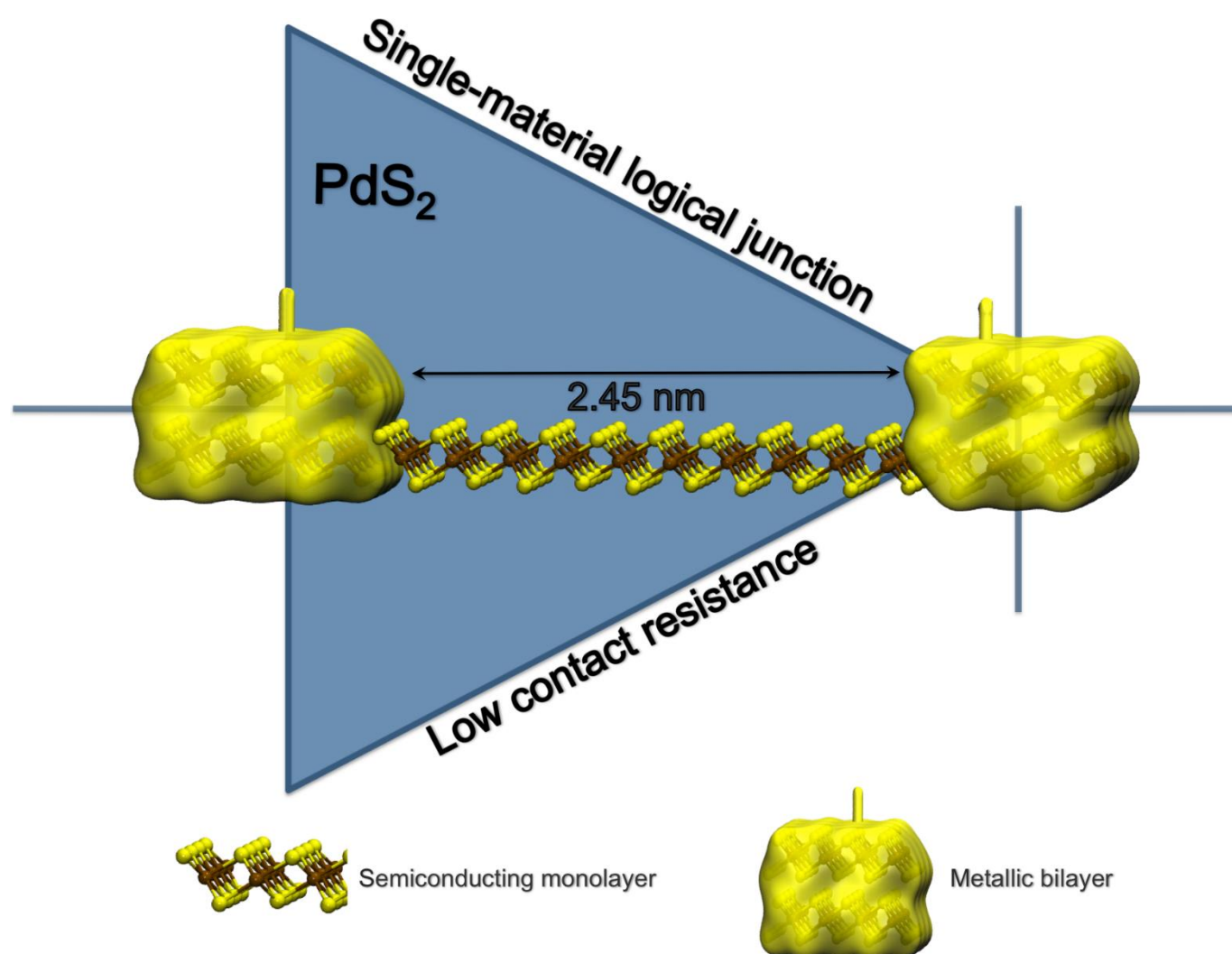
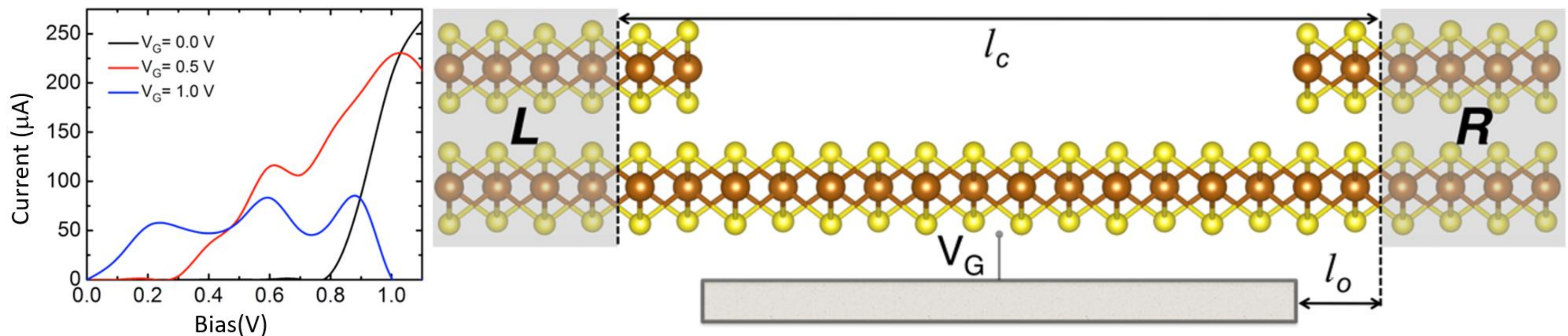
## Capabilities & Features

- UV/VIS for molecules (fast!)
- MOs, Band structures, DOS
- Molecules, 1D polymers, bulk
- NEGF: transmission

## DFTB & MOPAC + AMS driver

- Geometries, frequencies, phonons
- Stress tensors (optimize under p)
- Advanced MD, PES scans
- GCMC, molecule gun (deposition)

# Single 2D semiconductor material diode



2D PdS<sub>2</sub> is

- semiconducting as ML
- semimetallic as BL

## DFTB-NEGF

- channel lengths of about 2.5 nm
- I-V characteristics tunable by gate voltage

M.Ghorbani-Asl, A. Kuc, P. Miró, and T. Heine, *A Single-Material Logical Junction Based on 2D Crystal PdS<sub>2</sub>*, [Adv. Mater. \(2016\)](#)

# ReaxFF – reactive molecular dynamics

- Single atom type, bond-order based
- Charge equilibration
- Continuous F & E

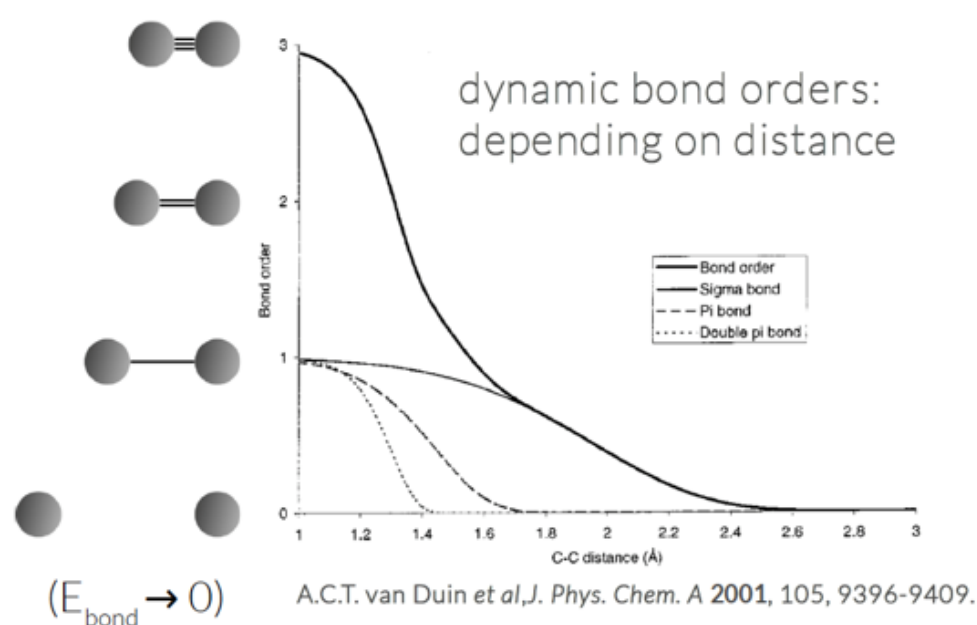
$$E_{\text{system}} = E_{\text{bond}} + E_{\text{lp}} + E_{\text{over}} + E_{\text{under}} + E_{\text{val}} + E_{\text{pen}} + E_{\text{coa}} + E_{\text{C}_2} + E_{\text{tors}} + E_{\text{conj}} + E_{\text{H-bond}} + E_{\text{vdWaals}} + E_{\text{Coulomb}}$$

$$BO'_{ij}(r_{ij}) = \exp \left[ p_{bo,1} \cdot \left( \frac{r_{ij}}{r_0} \right)^{p_{bo,2}} \right] + \exp \left[ p_{bo,3} \cdot \left( \frac{r_{ij}}{r_{0,\pi}} \right)^{p_{bo,4}} \right] + \exp \left[ p_{bo,5} \cdot \left( \frac{r_{ij}}{r_{0,\pi\pi}} \right)^{p_{bo,6}} \right]$$

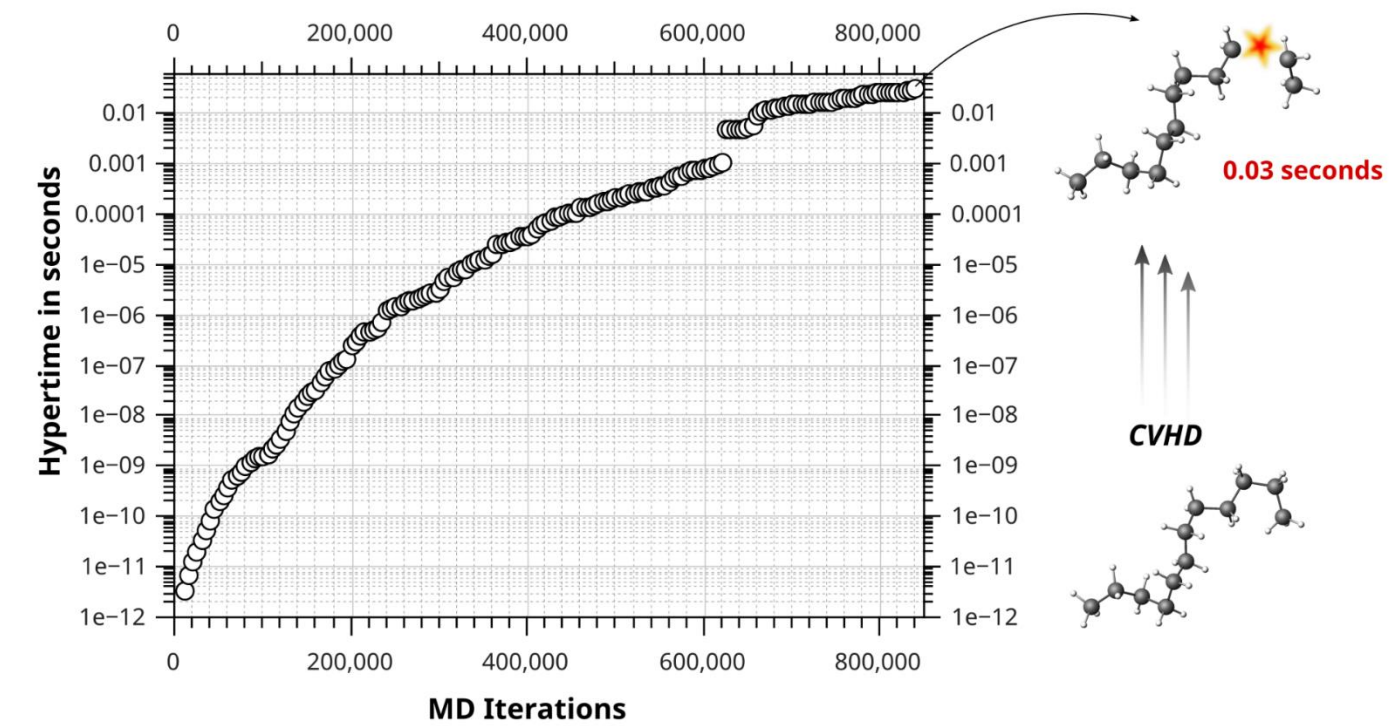
In: distance between atoms,  $r_{ij}$   
 Out: 1, 2, 1.42, etc...

Parameters = 16  
 $p_{bo,1}, p_{bo,2}, p_{bo,3}, p_{bo,4}, p_{bo,5}, p_{bo,6}, r_0, r_{0,\pi}, r_{0,\pi\pi}$   
 $val_1, val_2, \lambda_1, \lambda_2, \lambda_3, \lambda_4, \lambda_5$

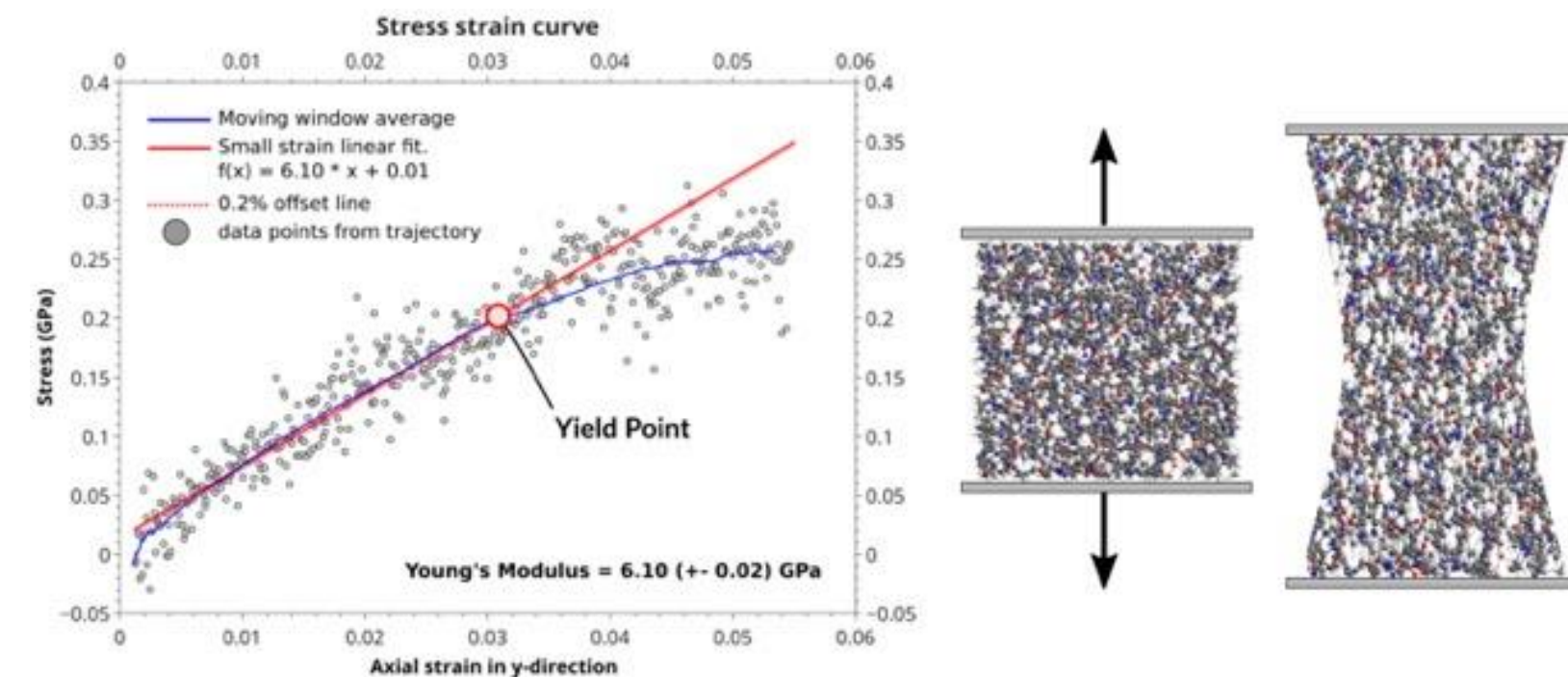
Correction terms  $f_1, f_2, f_3$ :  $BO_{ij}(r_{ij}) = BO'_{ij}(r_{ij}) \cdot f_1(BO'_{ij}) \cdot f_2(BO'_{ij}) \cdot f_3(BO'_{ij})$



Review NPJ Comp. Materials (2016):  
[The ReaxFF reactive force field: development, applications and future directions](#)



[Combustion, pyrolysis](#)  
 Chem. Sci., 2016, 7, 5280

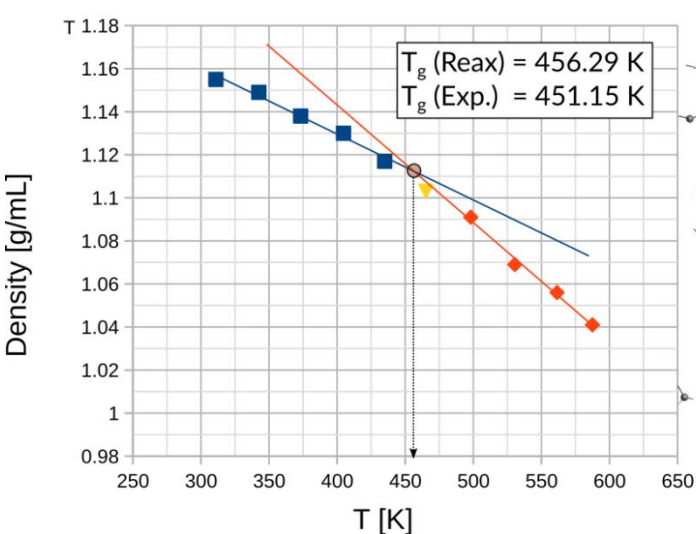
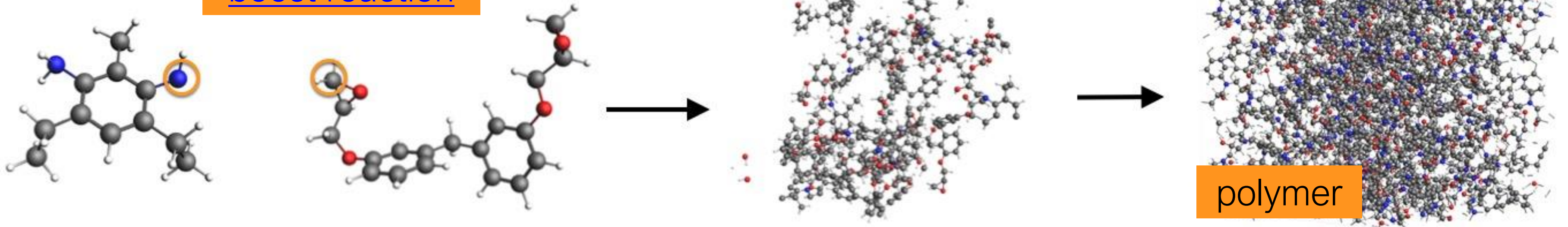


[Polymer formation & Tg, CTE, yield point](#)  
 J. Polym. Sci. B 2018, 56, 255

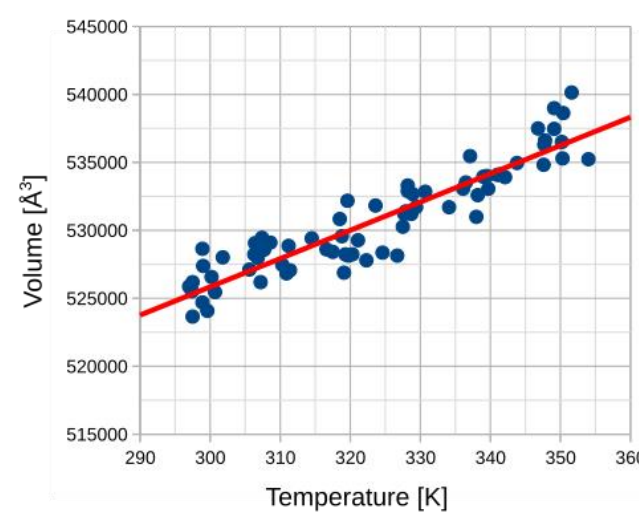
# Epoxy polymers: structure & properties

- Understand & predict how cross-linking effects mechanical properties
  - Exp = slow: minutes to hours to reach ~80% cross-linking
  - ReaxFF: simulate few ns => accelerate kinetics to get highly xlinked structures

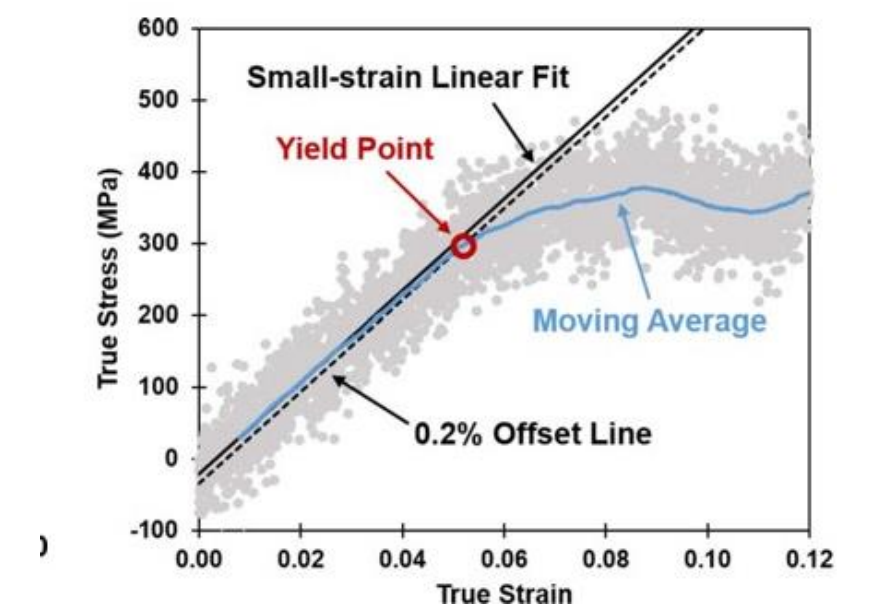
epoxy + amine  
boost reaction



density(T) using npT  
glass transition  $T_g$

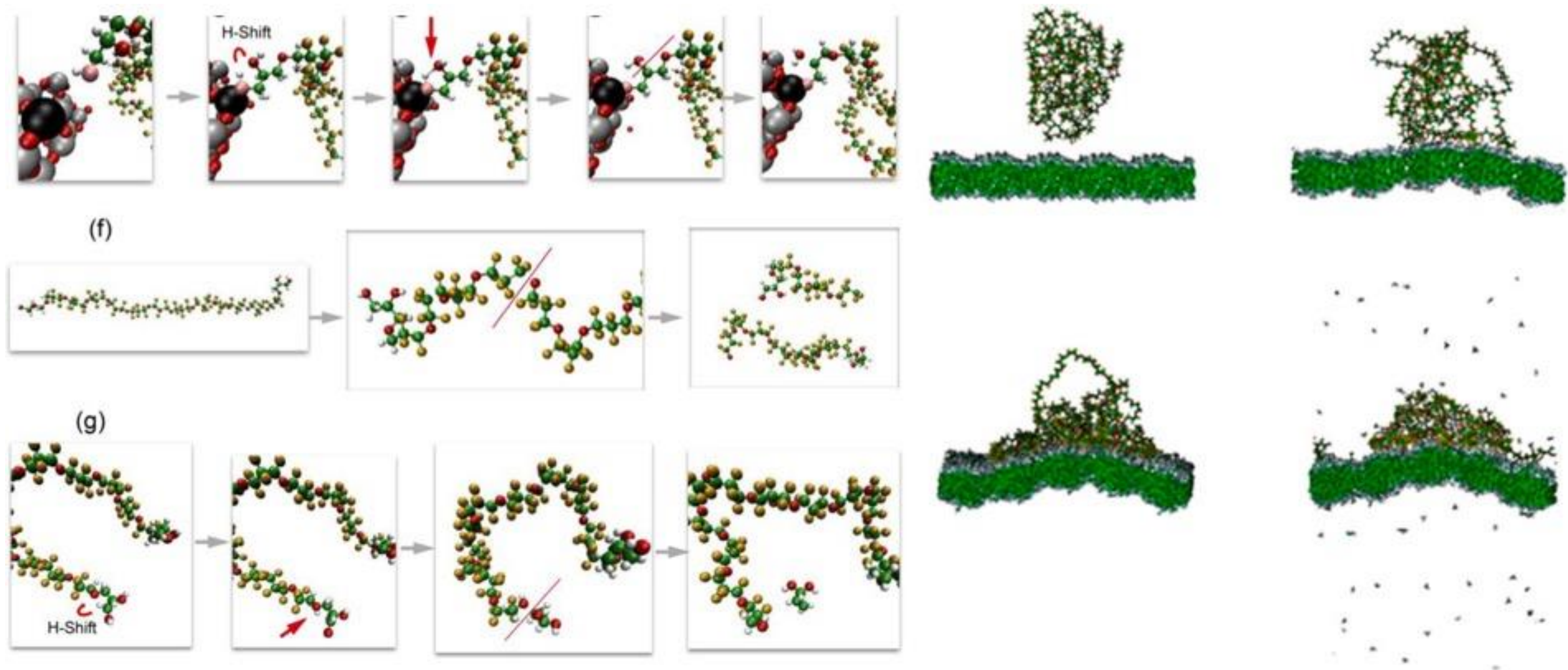


volume(T) npT  
CTE



Stress-strain:  
yield point, Poisson ratio

# ReaxFF simulations: degradation



Degradation mechanisms perfluoropolyether lubricant on SiO<sub>2</sub>, Fe<sub>2</sub>O<sub>3</sub> nanoparticles & DLC, effect of oxygen and water. Work by van Duin (Penn State/RxFFconsulting) with **Western Digital**

[J. Phys. Chem. C, 120, 27433 \(2016\)](#)

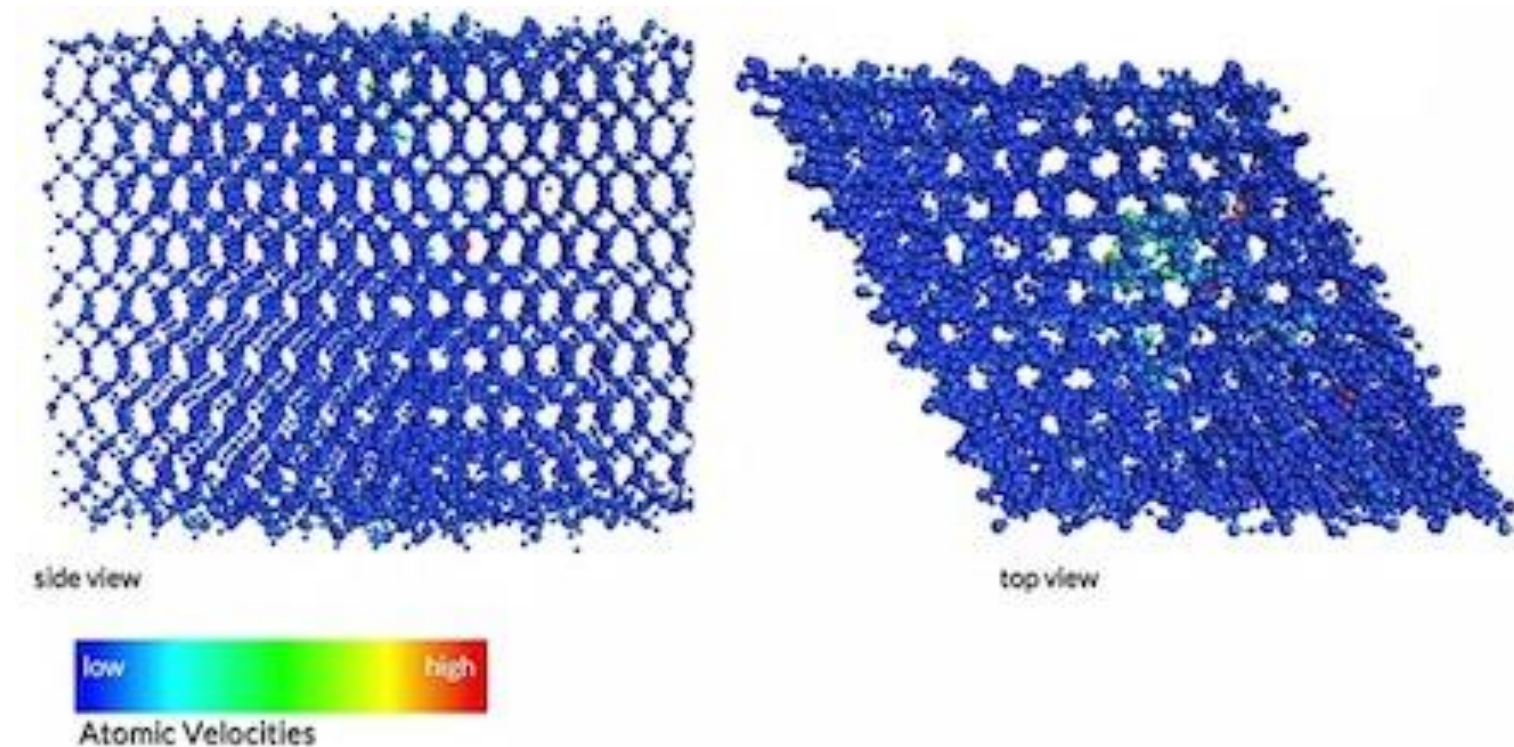
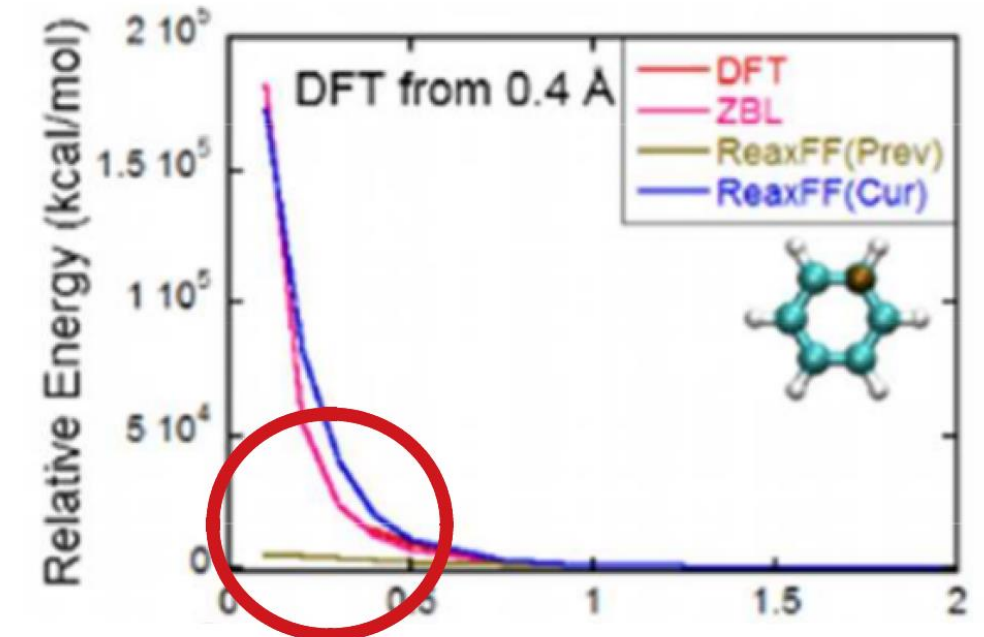
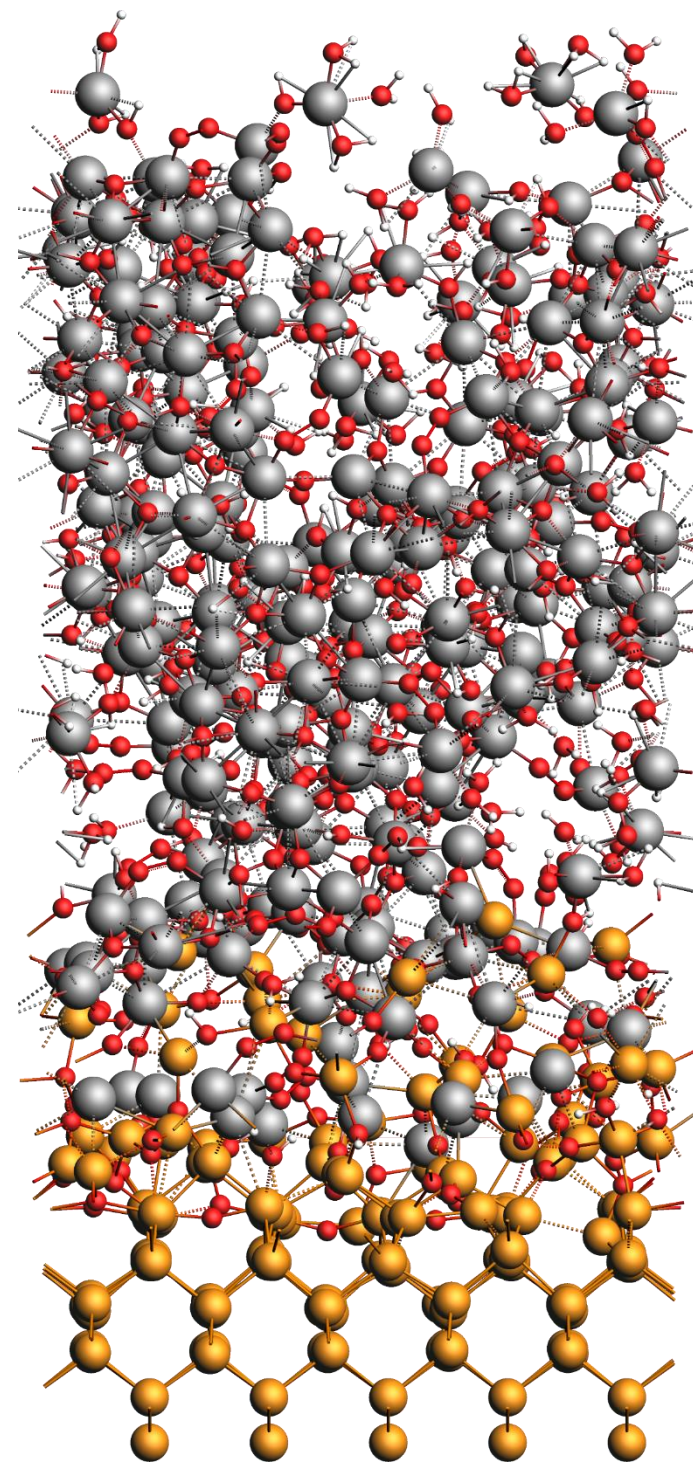
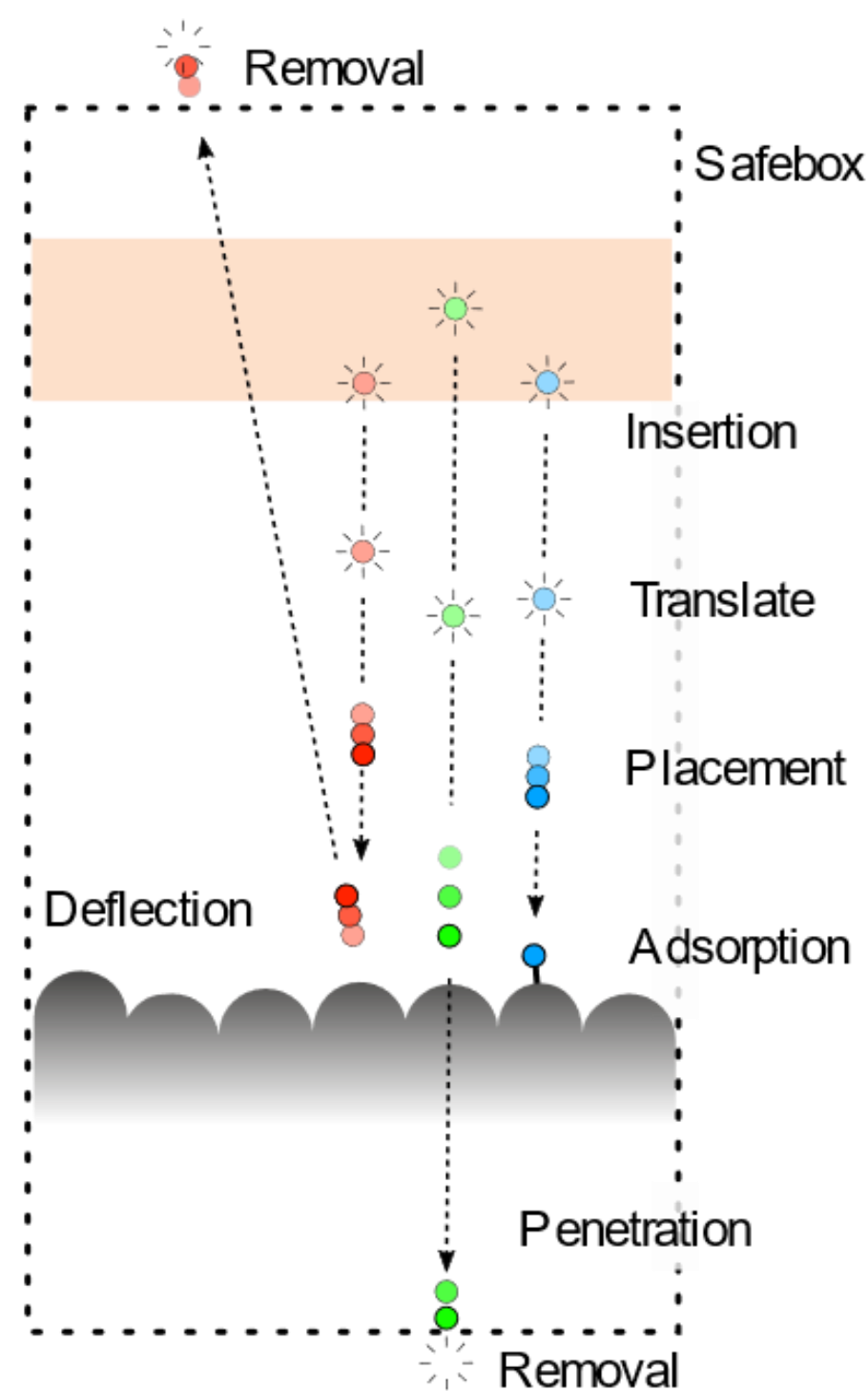
[J. Phys. Chem. C, 122, 2684 \(2018\)](#)

# Sputtering, etching, CVD, ALD, PVD

Shoot molecule/atomic targets at surface

- High E: sputtering, (plasma/photo?)etching
- Low E: CVD, ALD, PVD

- Need good force fields:  
ReaxFF refit short distances
- Sputtering yields (E, angle)  
([movie](#): Ar / SiO<sub>2</sub>)

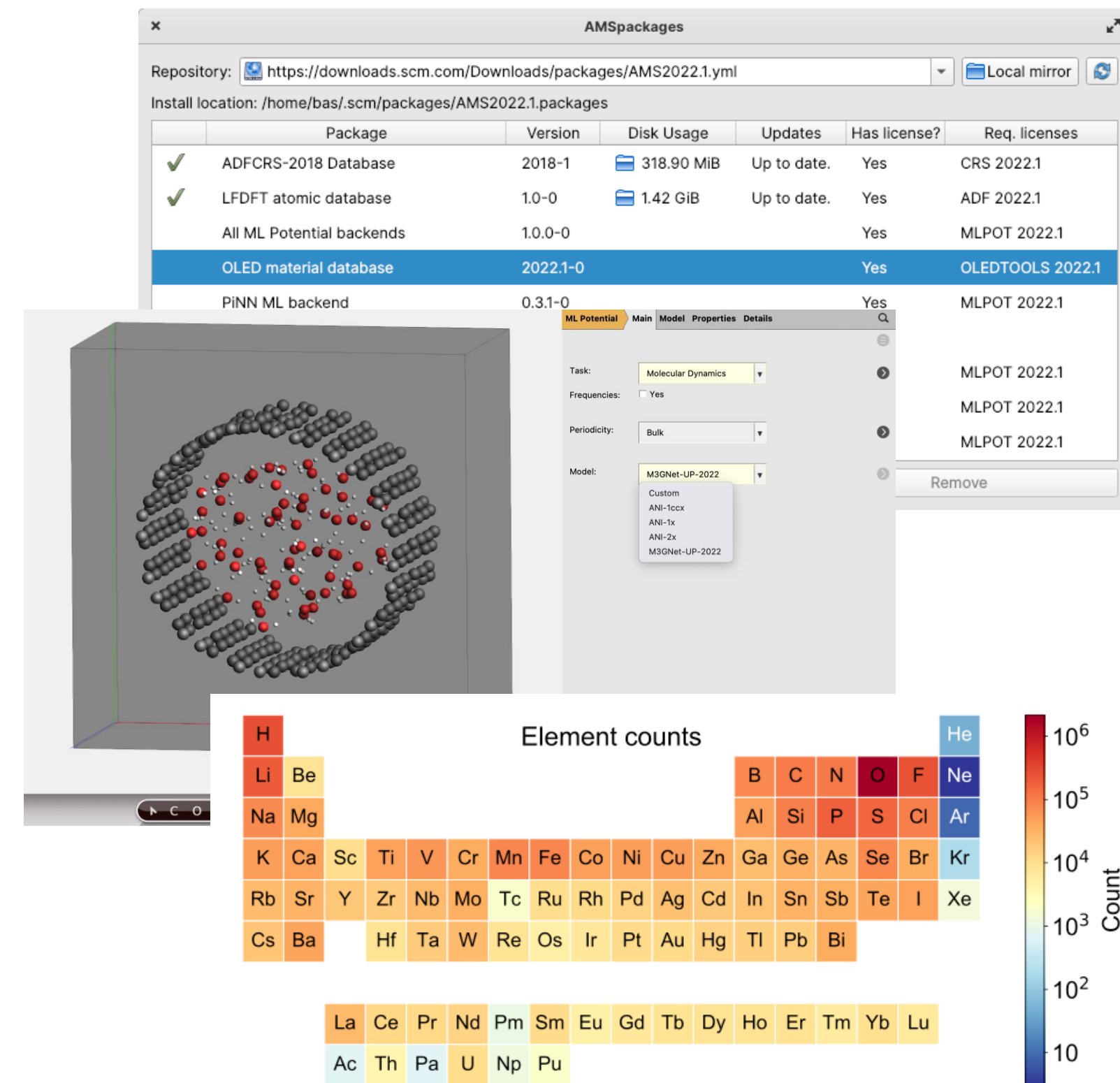


[Tutorial CVD: Alumina / Ge\(001\)](#)



# Machine Learning Potentials

- Automatically install popular ML Backends
  - Pre-parametrized
    - i) ANI-1x and 2x (H, C, N, O, F, S, Cl)
    - ii) M3GNet (“Universal”)
  - Backends, via ASE
    - iii) NEquIP, FLARE (on-the-fly)
    - iv) sGDML
    - v) SchNet
    - vi) CHGNet
- Use MLP with all the tasks in the AMS driver
  - PES scans, reaction discovery, conformers, IR, phonons, MD, MC etc.
  - Hybrid (multi-layer): combine with other methods
- CUDA-enabled PyTorch and Tensorflow

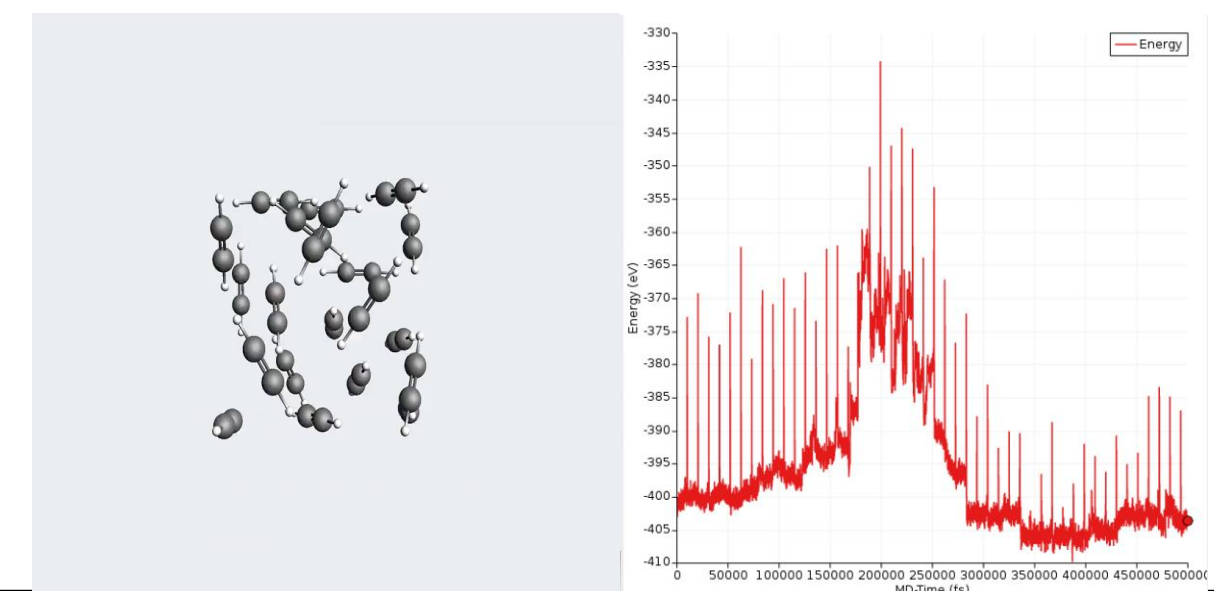
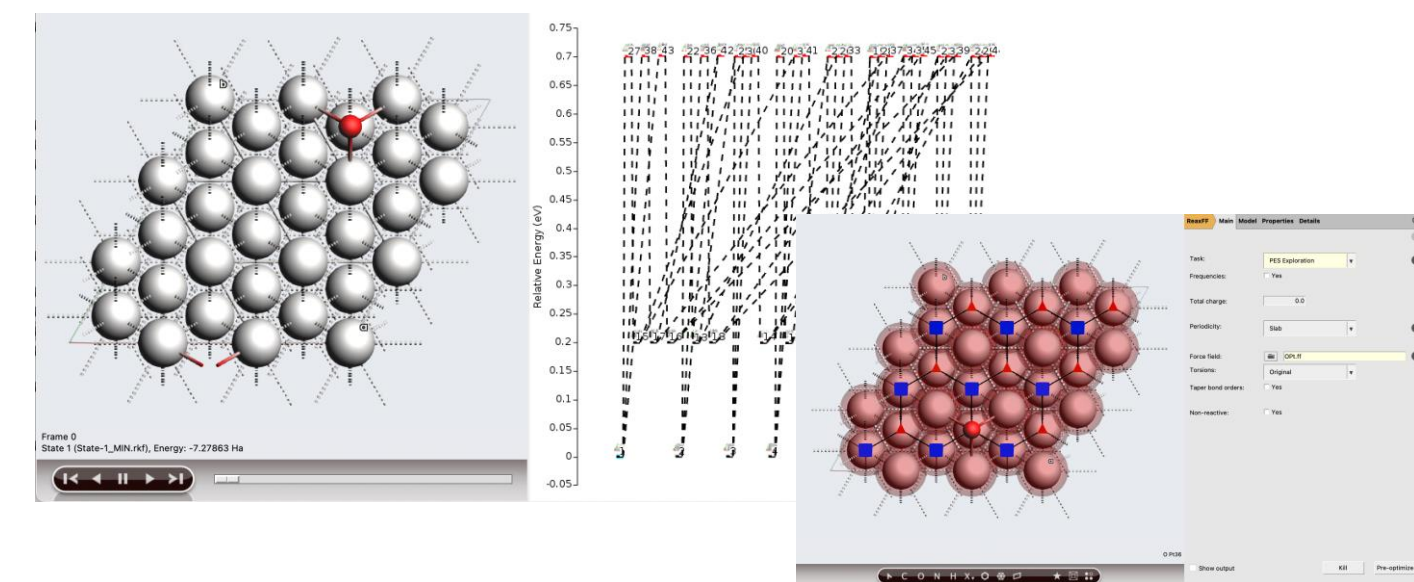
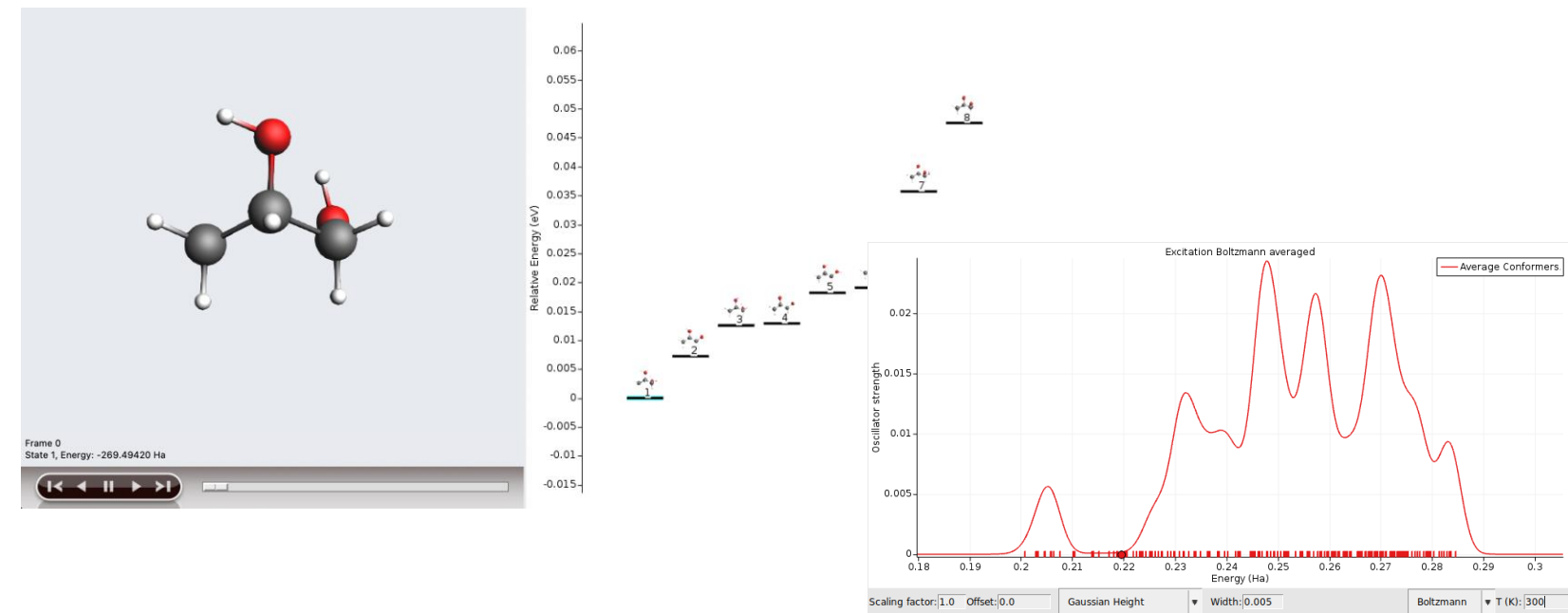


- i) O. Isayev et al. *Chem. Sci.*, 2017, 8, 3192–3203
- ii) C. Chen, S.P. Ong., *Nature Comp. Sci.* 2, 718–728 (2022)
- iii) S. Batzner et al. *Nature Comm.* 13: 2453 (2022)
- iv) S. Chmiela et al. *Comp. Phys. Commun.* 240 (2019) 38-45
- v) K. T. Schütt et al., *J. Chem. Theory Comput.* 15 (2019) 448-455

# MLPotentials for chemistry

## From organic chemistry to catalysis

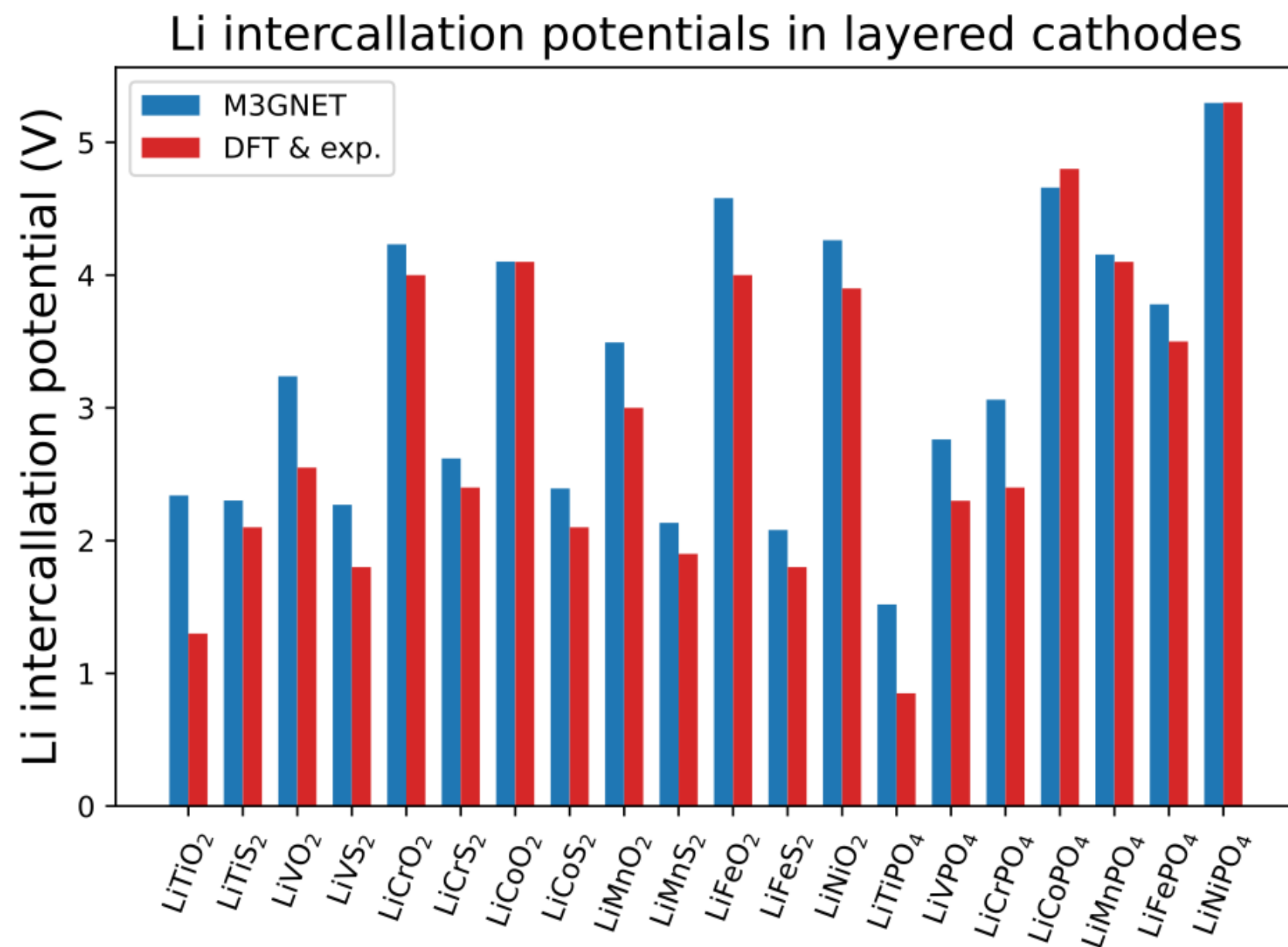
- Explore conformers with ANI-1x/2x
  - CREST & RDKit conformer generation
  - Re-score with DFT
  - Boltzman averaged properties
- Screen adsorption sites with PESexploration
  - Refined (or fully computed) with MLP
  - OCP (Open Catalyst Project) via ASE engine <https://opencatalystproject.org/>
- Discover chemical reactions with ANI-1xnr (2024)
  - Use the nano reactor to accelerate reactions
  - Analyze all chemical reactions with ChemTrayZer



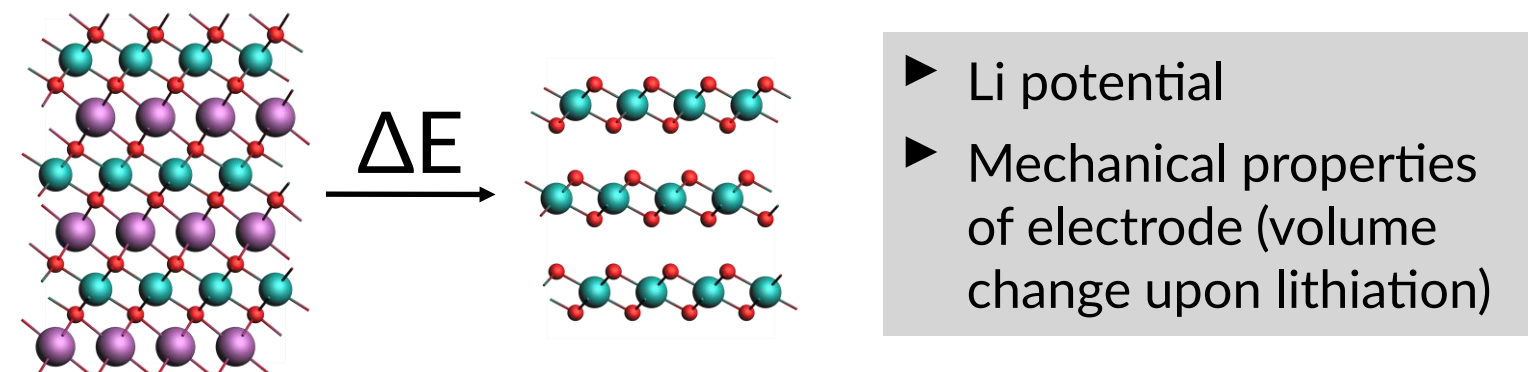
# MLPotentials for (battery) materials

## M3GNet, trained to Materials Project 1000x faster than DFT

C. Chen, S.P. Ong., *Nature Comp. Sci.* 2, 718–728 (2022)

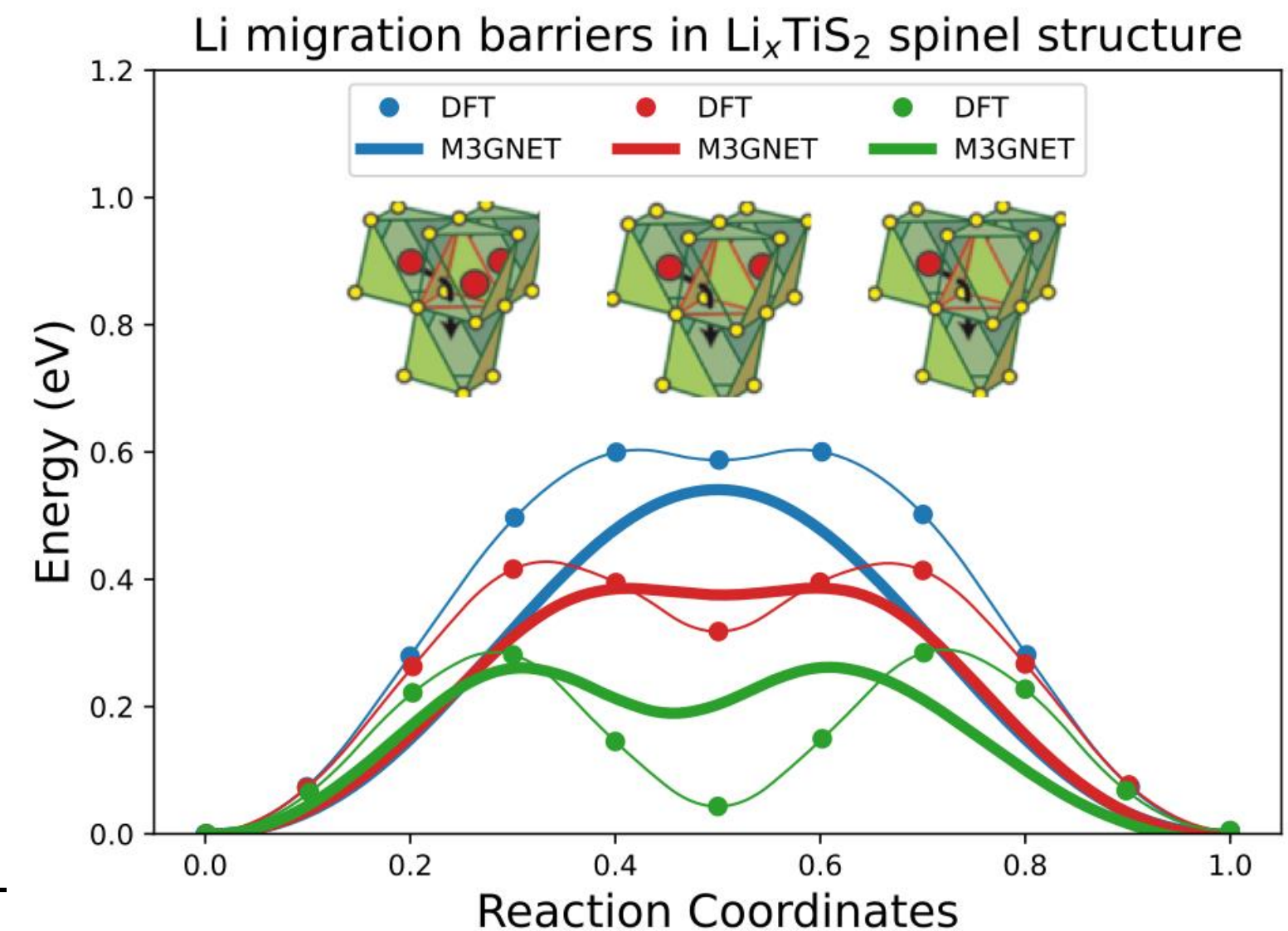
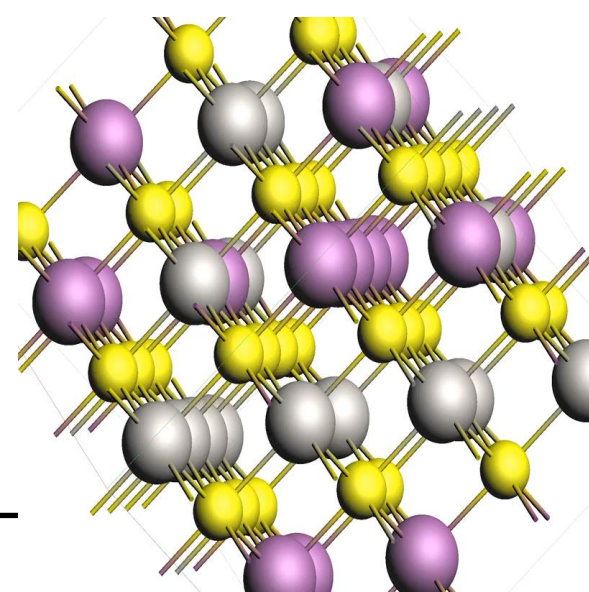


- Li intercalation potentials accurately predicted with DFT (~100 atoms)
- M3GNet reproduces DFT really well



- Li diffusion through NEB or PES scan
- M3GNet can accelerate by 1000x
- (Dis)charge curves: GCMC

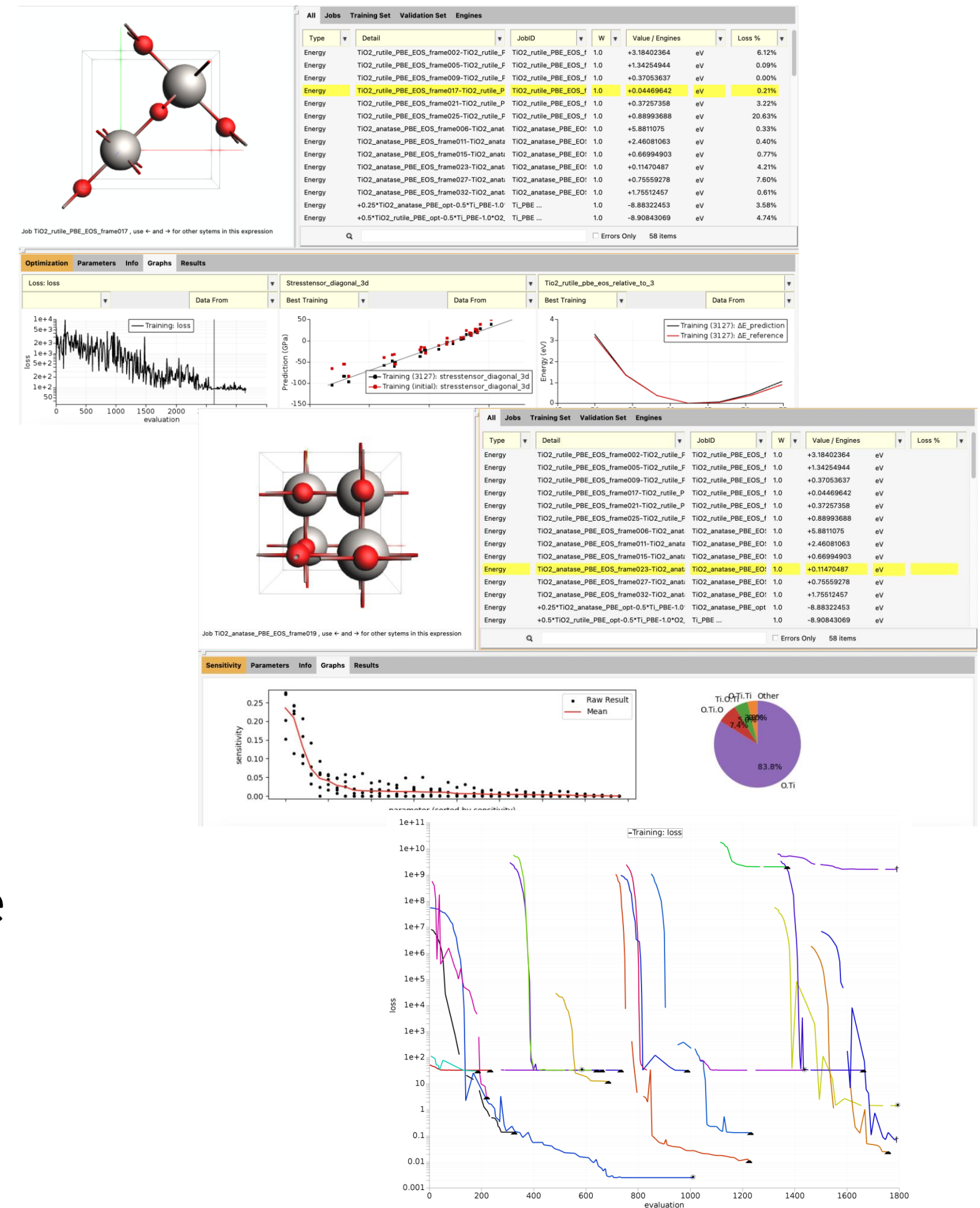
- ▶ Activation energy
- ▶ Diffusion (kinetics)



# Training FF with ParAMS

## ParAMS making parameters optimization easier

- Import, build and visualize training data
- Use data from AMS, VASP, QE, and experiments
- Tune parameters to describe energies, forces, geometries, stress tensor, charges, bandgaps, etc.
- Use validation sets to prevent overfitting
- Submit multiple optimization at the same time to explore parameters space
- Explore the sensitivity of the parameters to select the most important subset to optimize
- Train DFTB, ReaxFF, Force field, and ... MLP



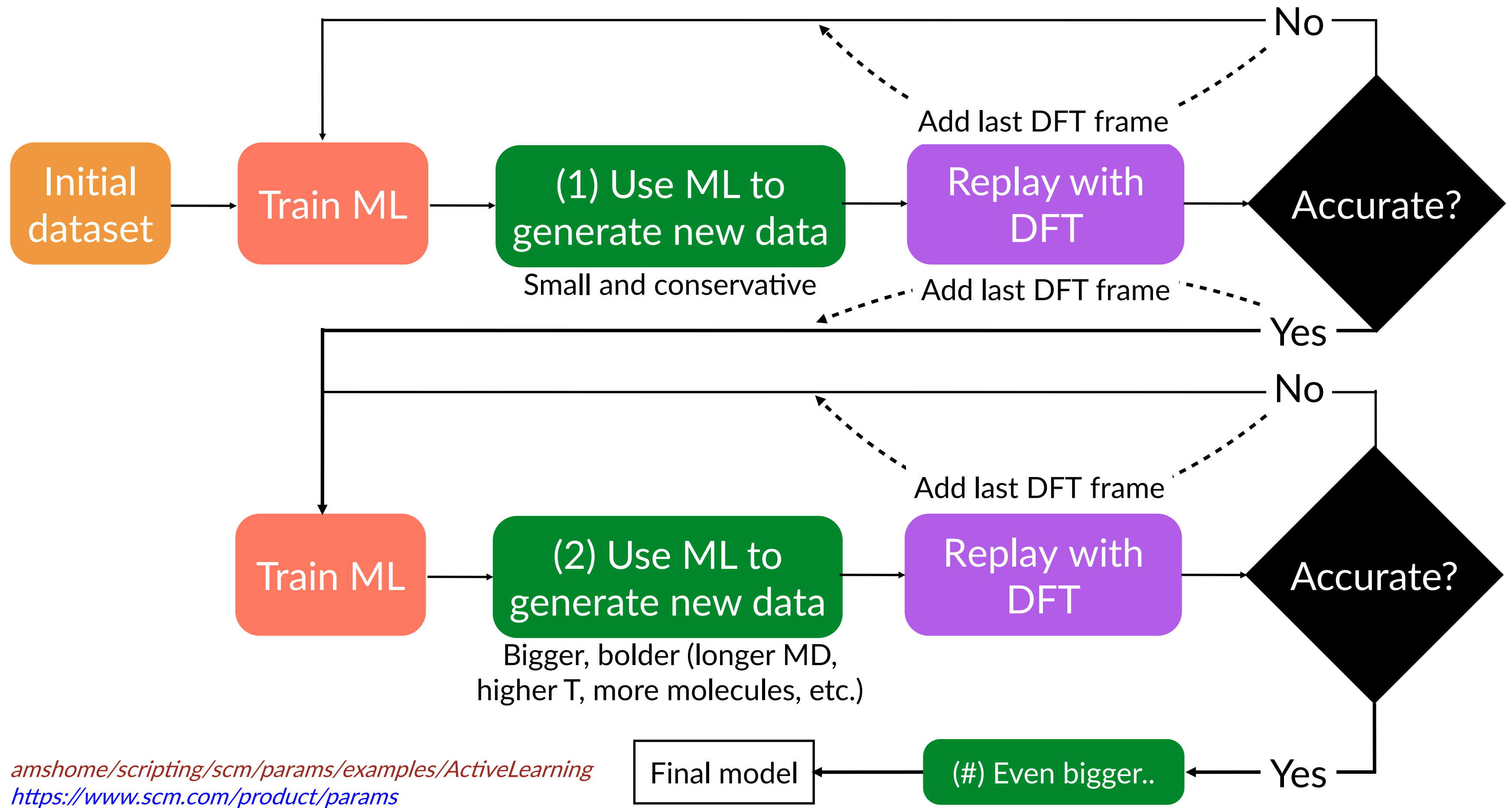
*J. Chem. Inf. Model.* 2021, 61, 8, 3737–3743

*J. Chem. Theory Comput.* 2023, 19, 9, 2557–2573

<https://www.scm.com/product/params>

# Training MLP with ParAMS

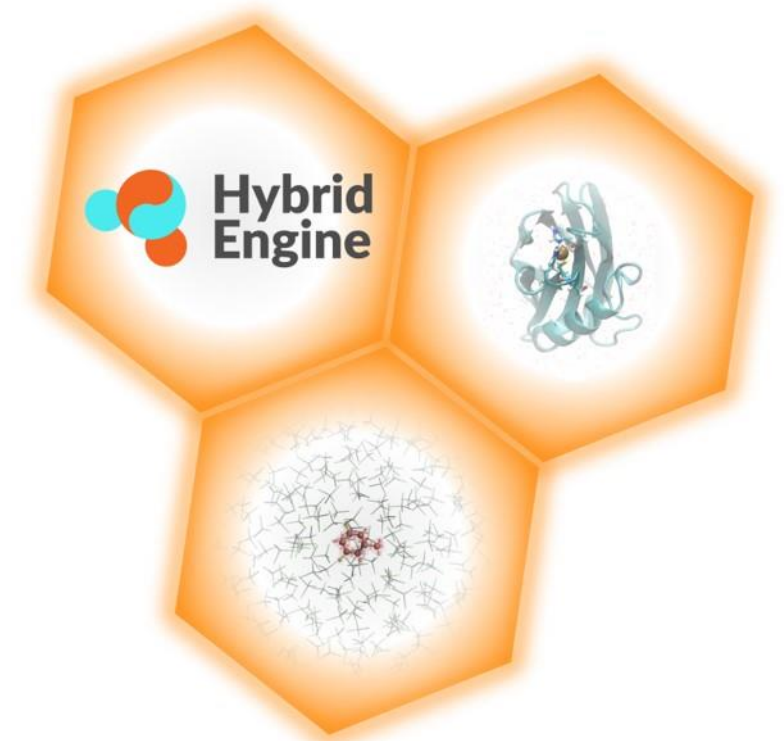
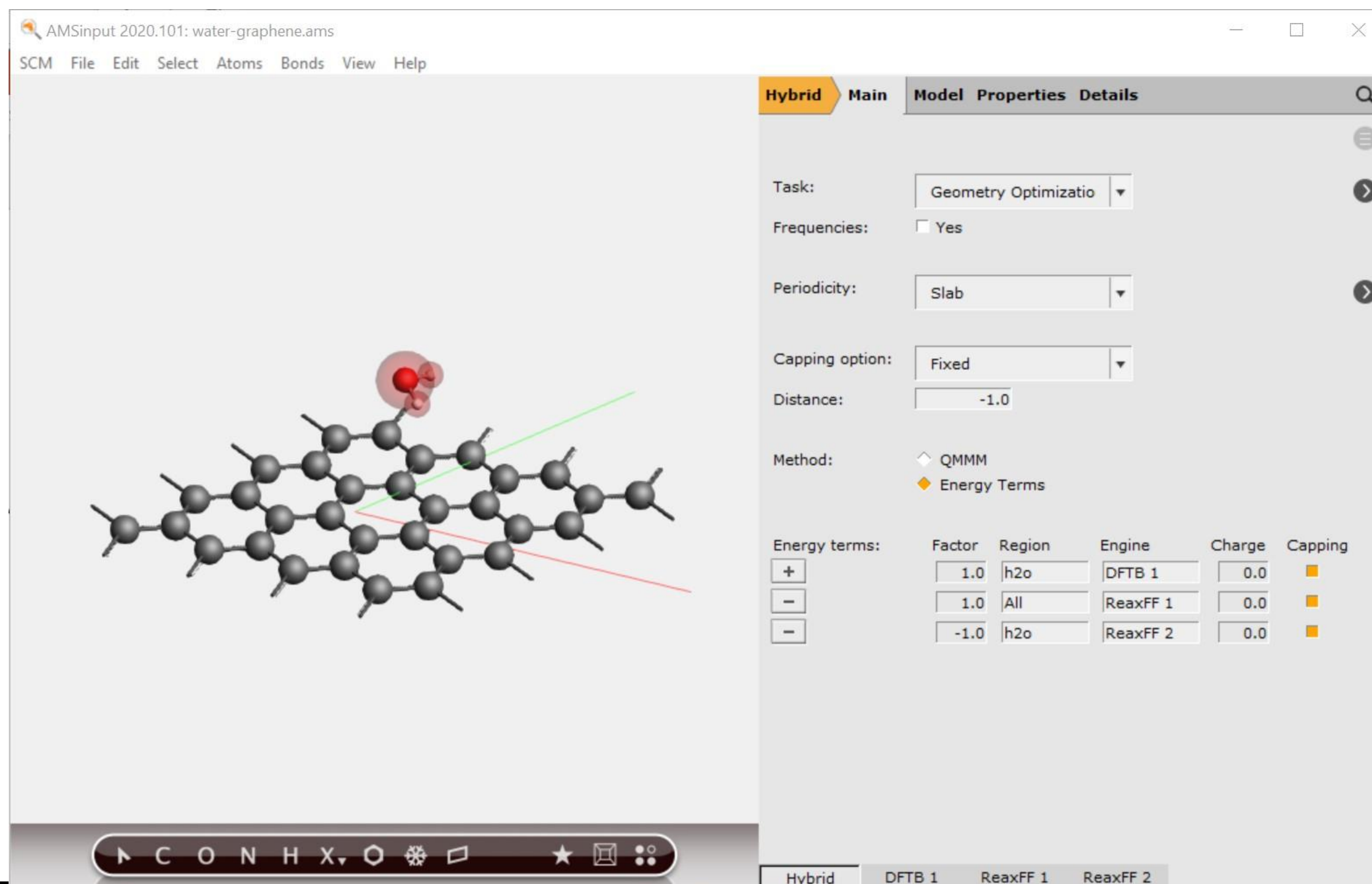
## Active learning workflow



[amshome/scripting/scm/params/examples/ActiveLearning](https://www.scm.com/product/params)  
<https://www.scm.com/product/params>

# Hybrid Engine

- Multi-layer (subtractive, QUILD, ONIOM)
  - combine any periodicity, number of layers, and QM or FF methods
- 2-layer: (additive) QM/MM
  - any periodicity
  - QM: ADF, DFTB, BAND, MM: Force Field engine



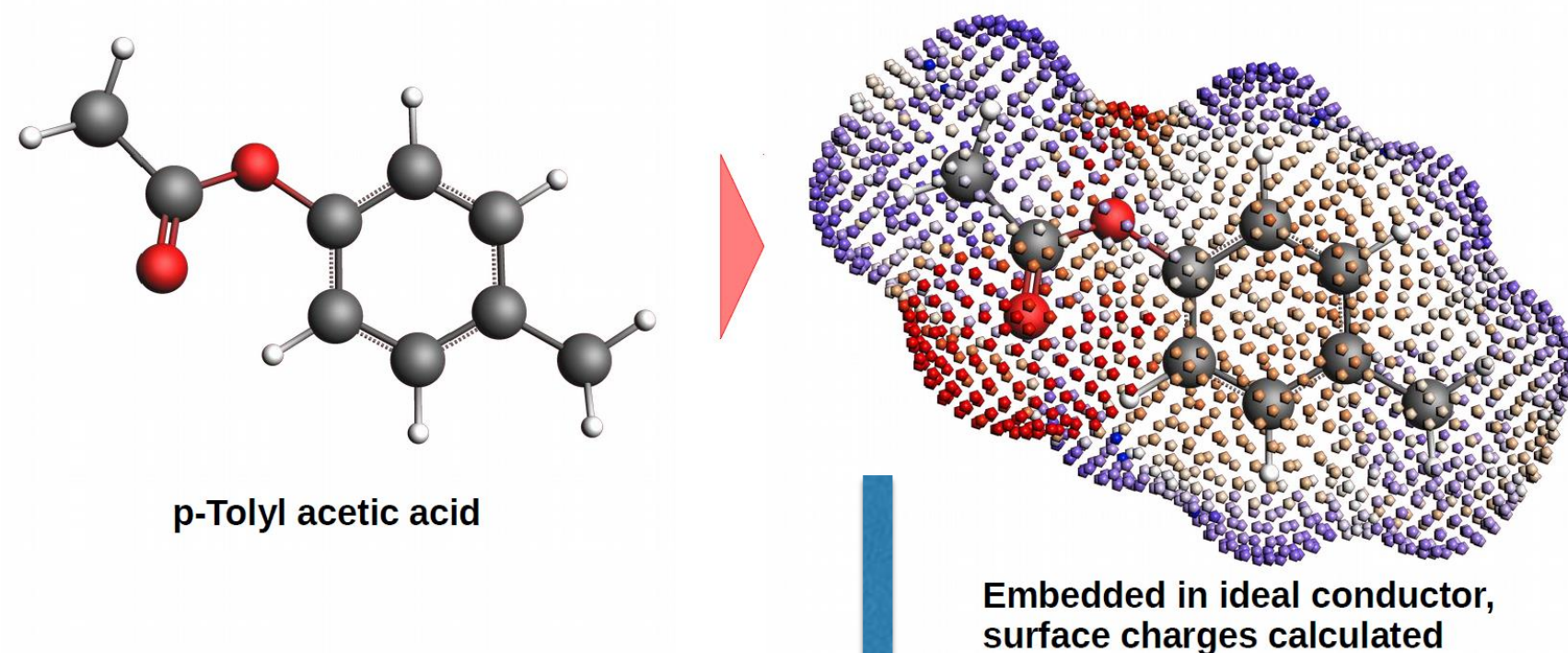
[Demo video](#)

# COSMO-RS/SAC: thermodynamic properties of fluids

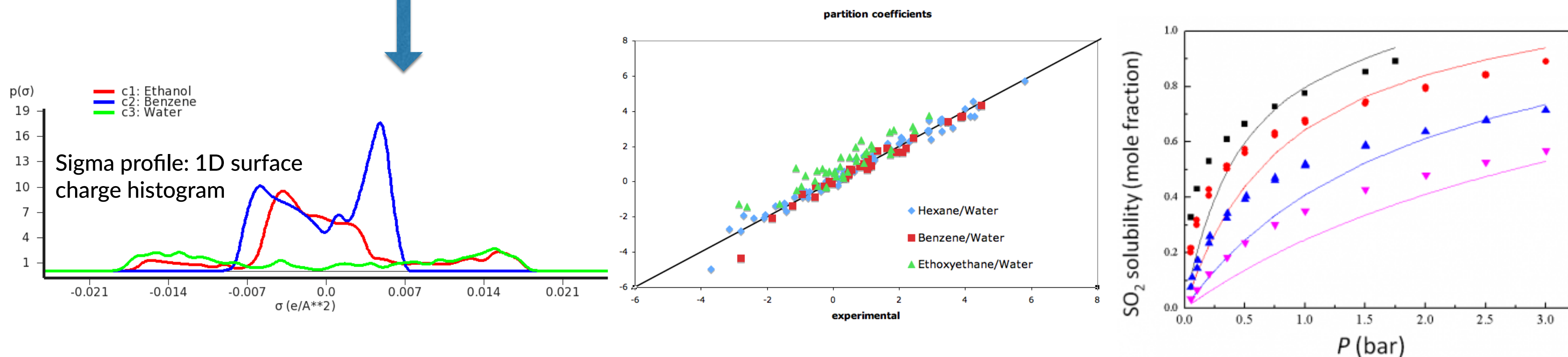
## Quantum Chemistry & QSPR for quick property predictions

COntinuum Solvation MOdel + RS (Klamt), SAC (Sandler)

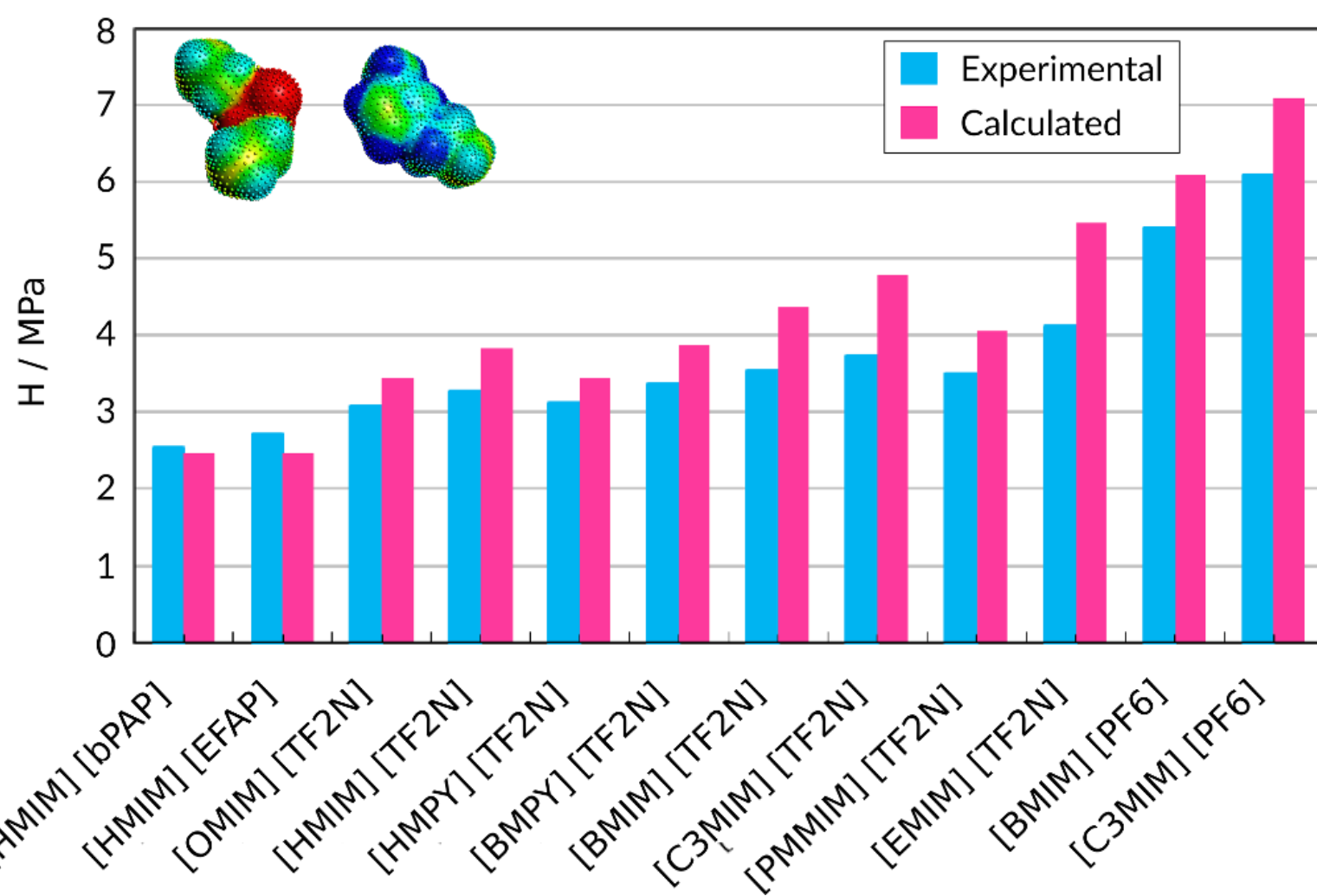
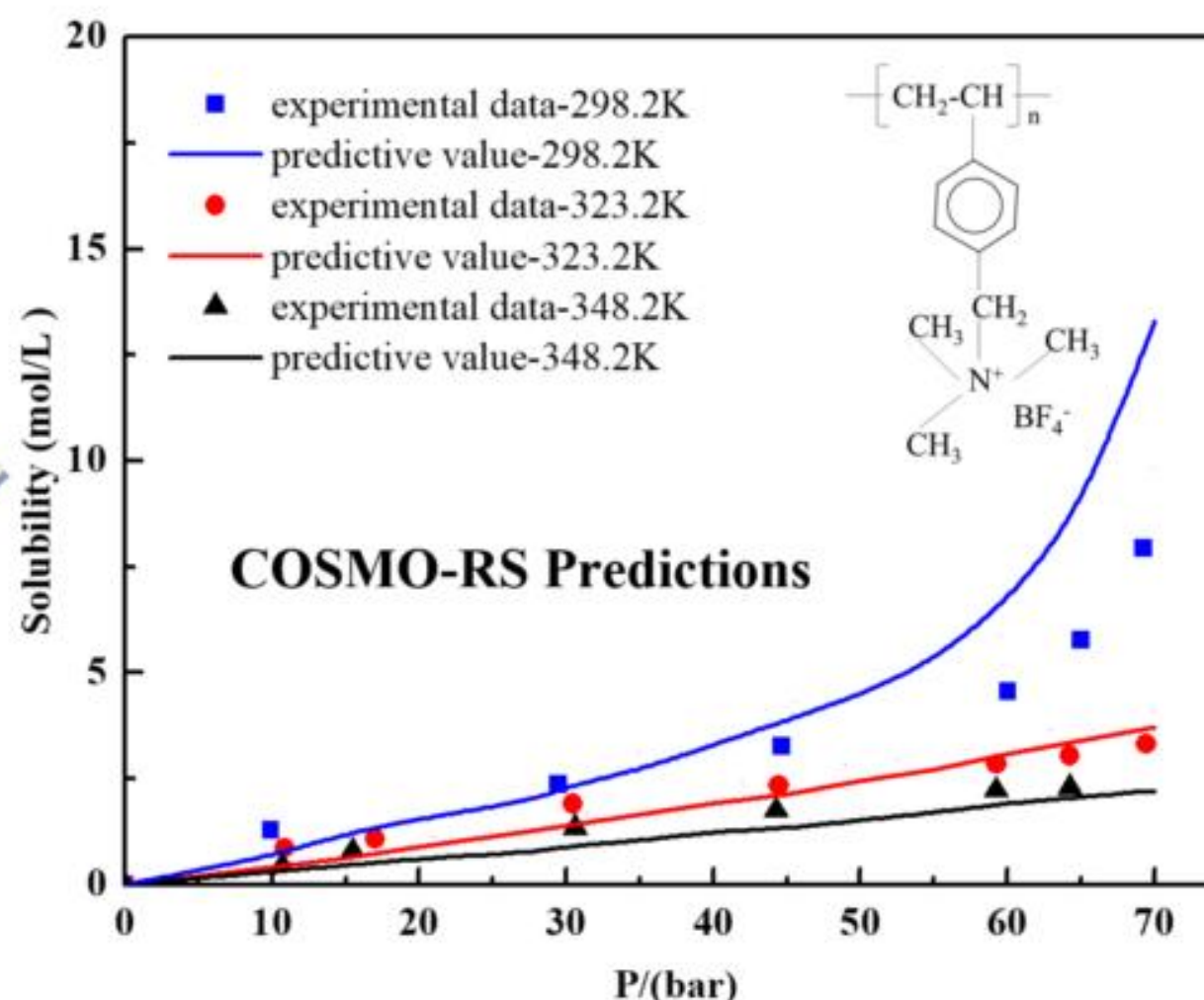
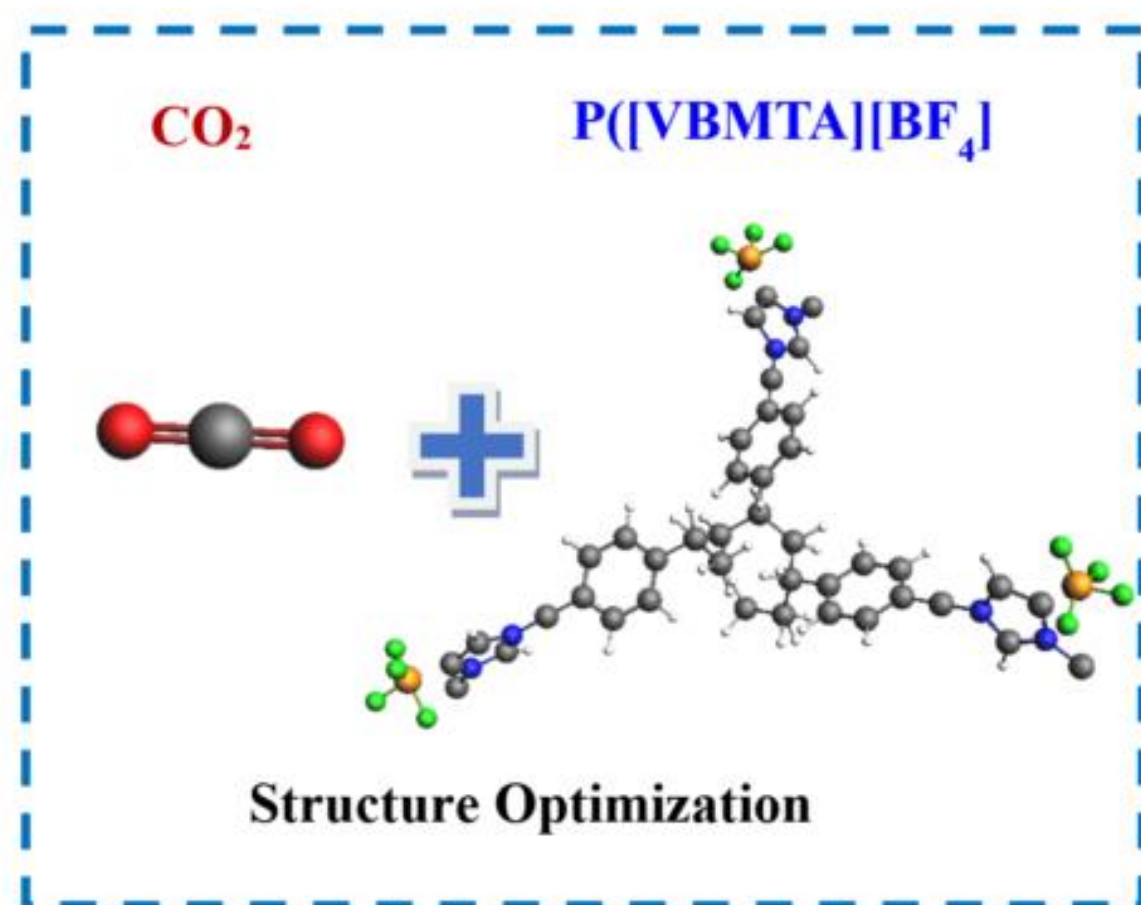
chemical potential => activity coefficients => instantaneous properties



- Solvation & excess energies, pKa
- Solubilities, LLE, VLE, boiling points
- Optimize mixtures: solubility, LLE
- Polymers: Flory-Huggins X
- QSPR & fast sigma: using SMILES



# COSMO-RS: solubility in (polymeric) ionic liquids



- COSMO-RS predicts beyond parametrization (opposed to UNIFAC)
- Works well for SO<sub>2</sub> solubilities
- Improvements for CO<sub>2</sub> (2018)
- Combined with polymer terms (2021)

Lei group: [Chem. Rev., 114, 1289–1326 \(2014\)](#),  
[Green Energy & Environment \(2018\)](#),  
[Green Energy & Environment \(2021\)](#)



# Enjoy the workshop!



## Amsterdam Modeling Suite

Making Computational  
Chemistry Work  
For You

**Try for free**

[scm.com/trial](https://scm.com/trial)

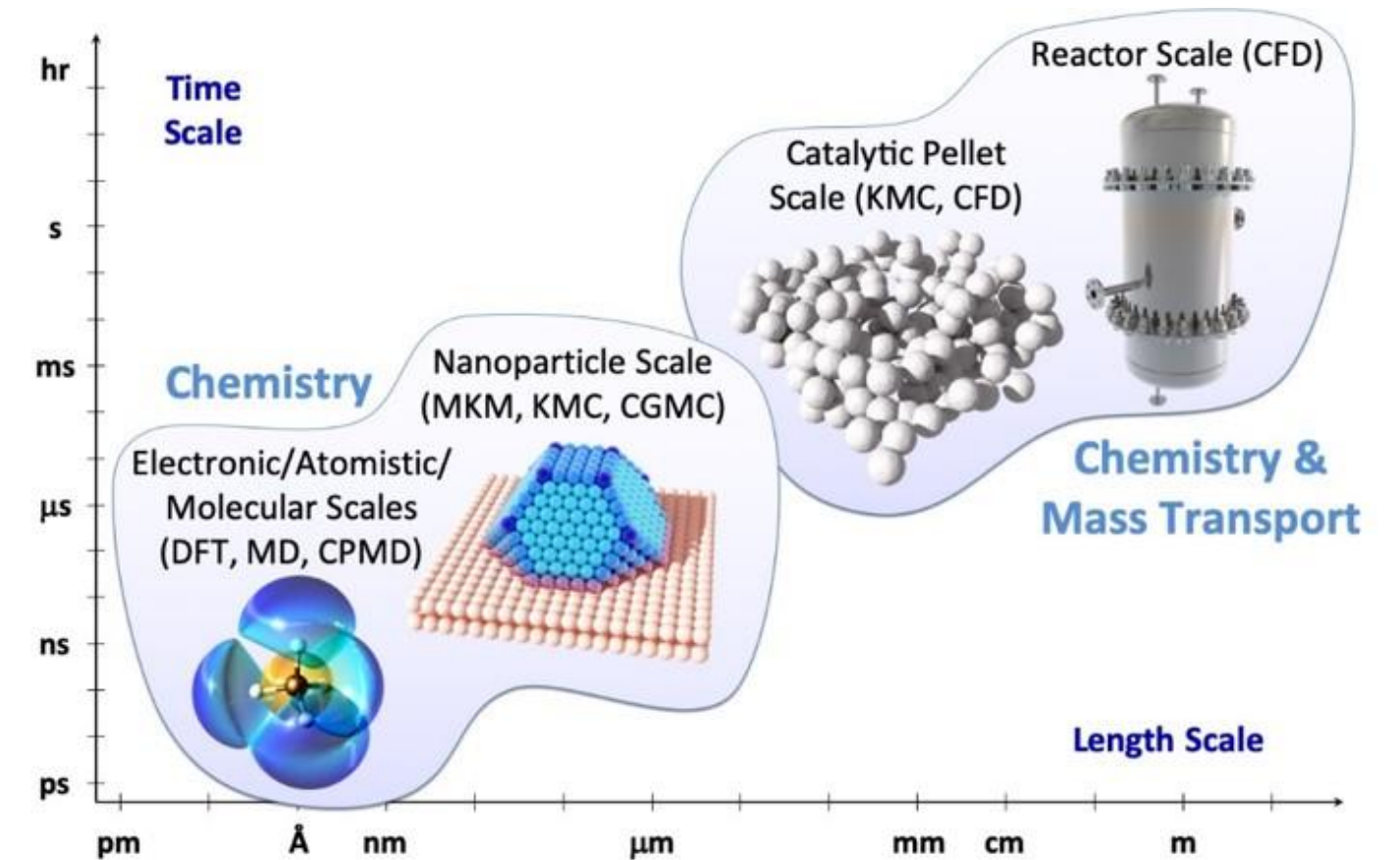
We love to hear about your R&D challenges!

[goumans@scm.com](mailto:goumans@scm.com) | [www.scm.com](http://www.scm.com)

生菜科技有限公司 | [www.sunrise.tw](http://www.sunrise.tw)

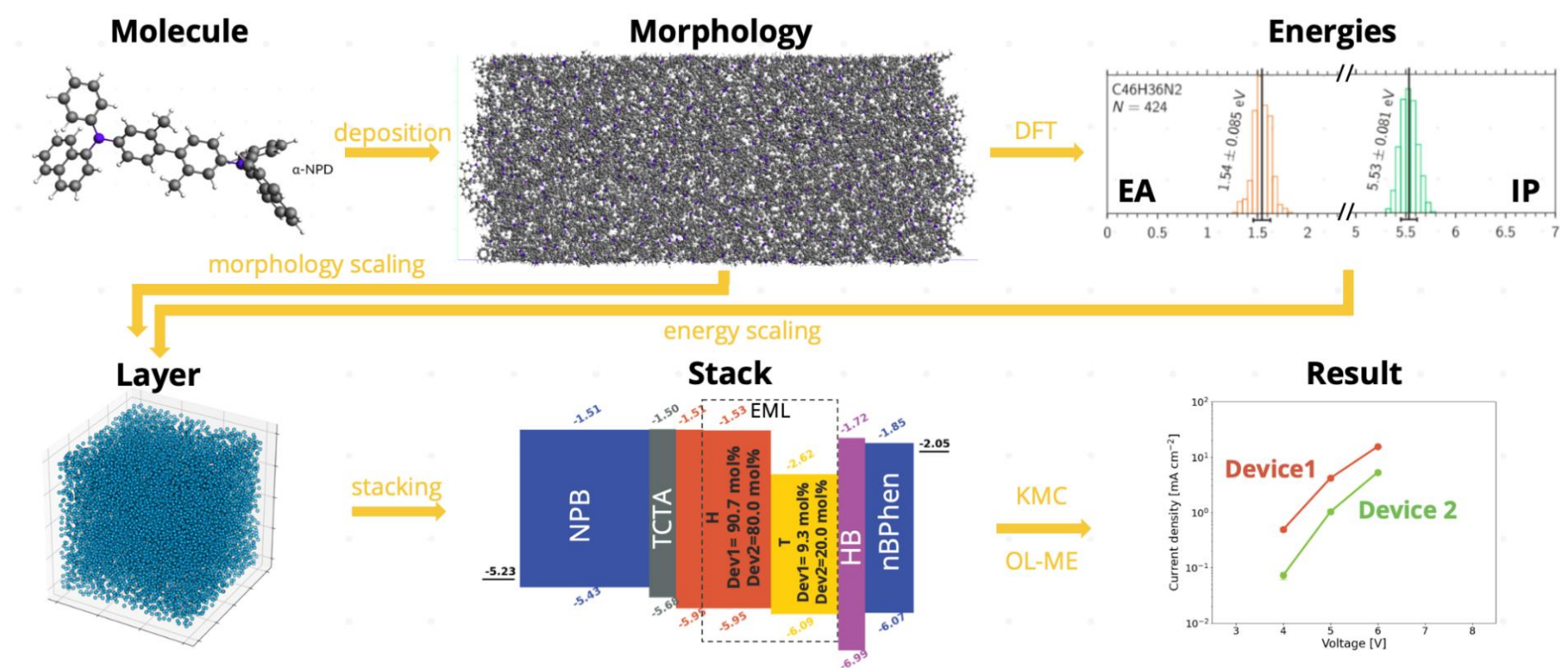
# Summary: Amsterdam Modeling Suite

- Couple atomistic, meso & macro
- Integrated User Interfaces
  - Graphical
  - Python layer
  - Connections to third party

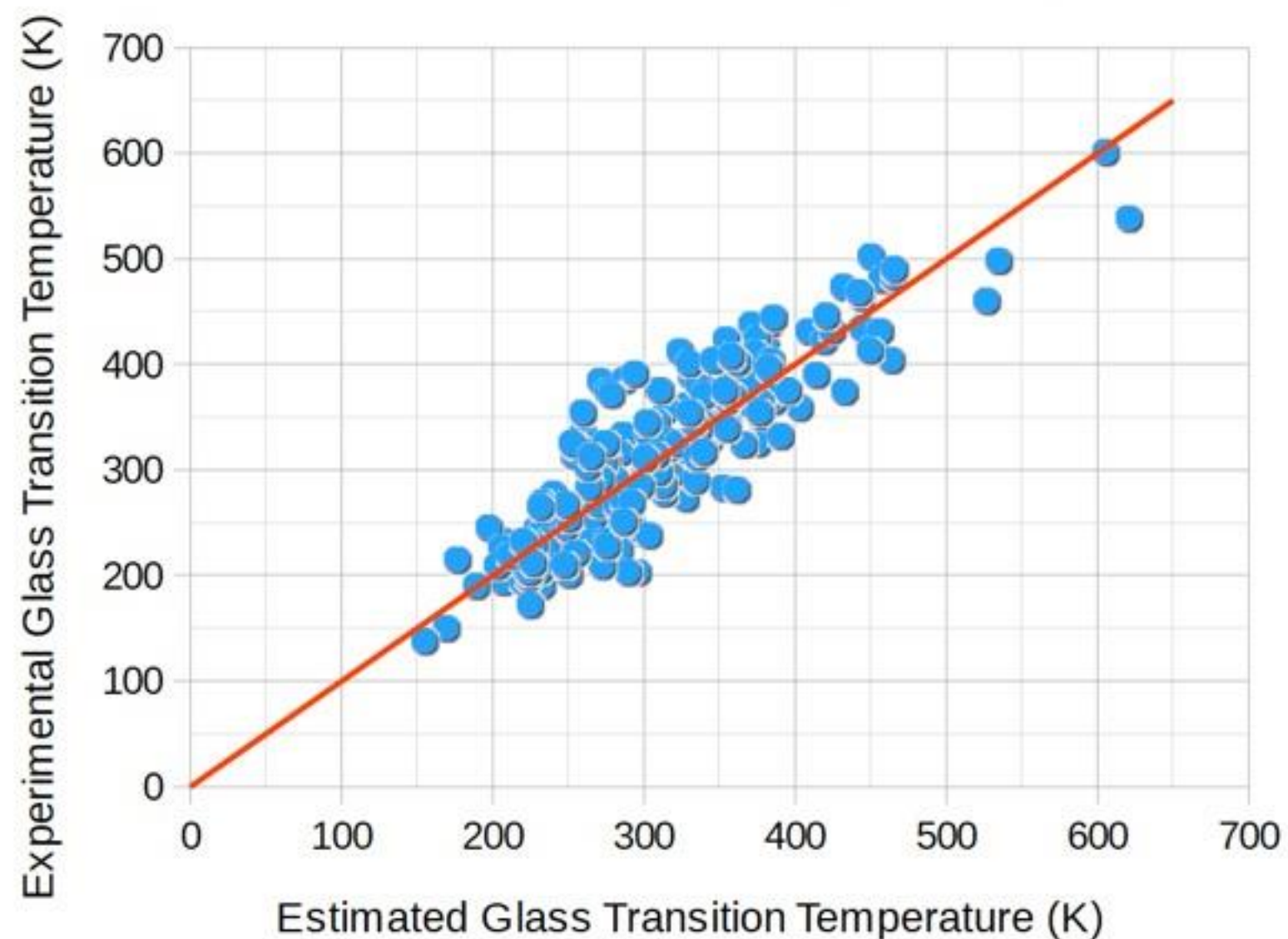
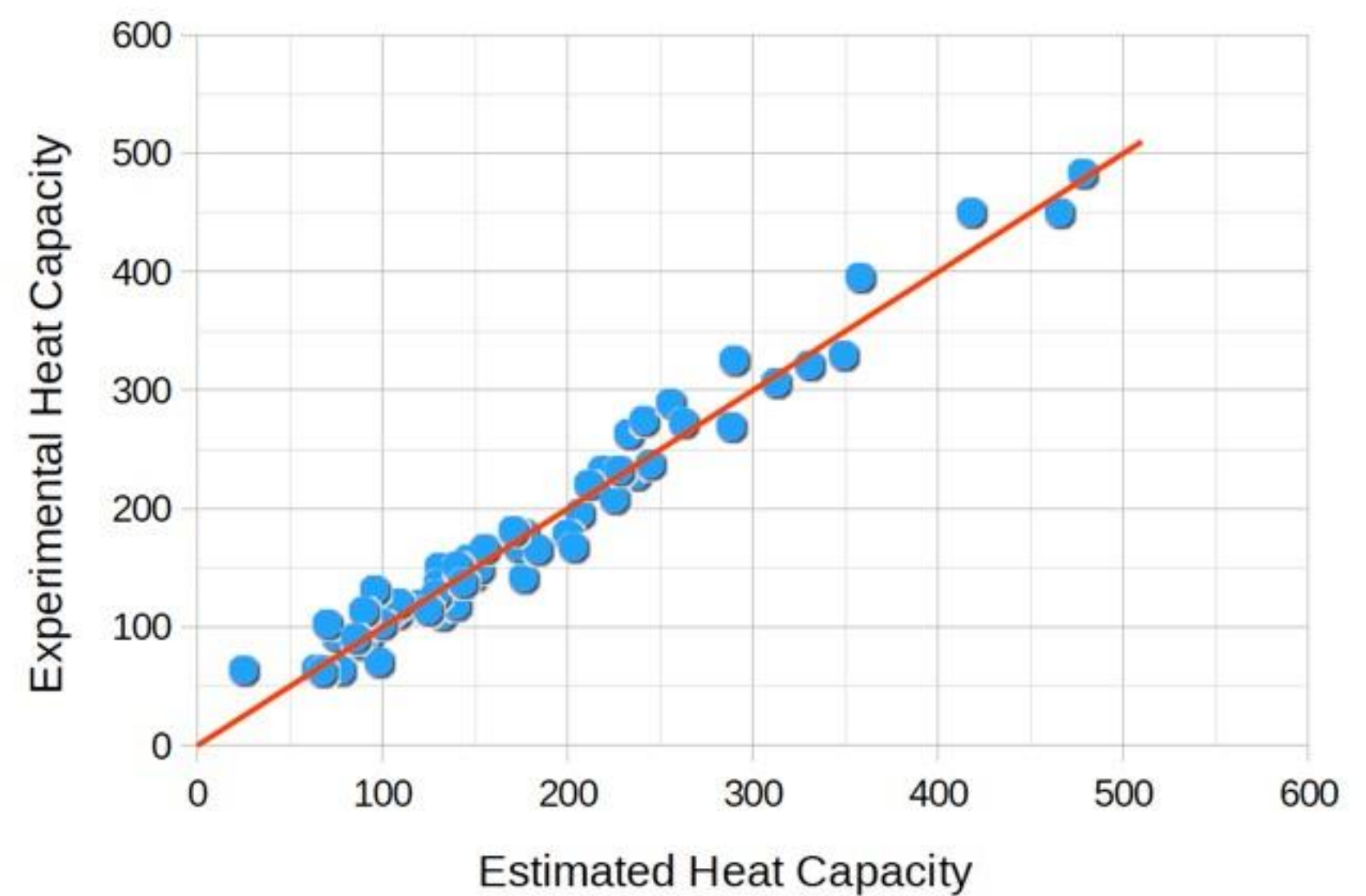
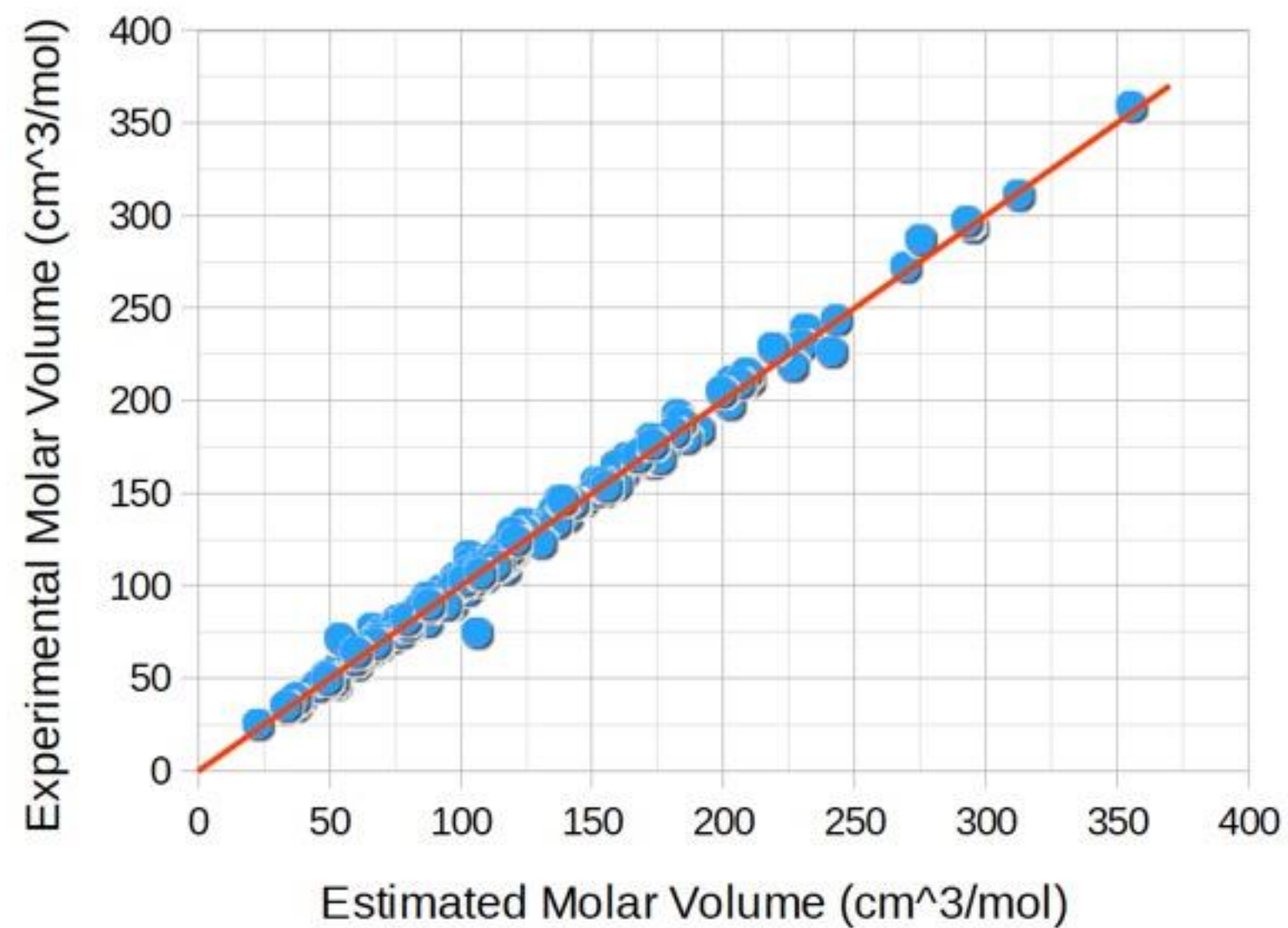


- Applications

- Catalysis
- Organic electronics
- Batteries
- Polymers



# Polymers: QSPR with sigma-moments



## Basic statistics

$r^2$ :

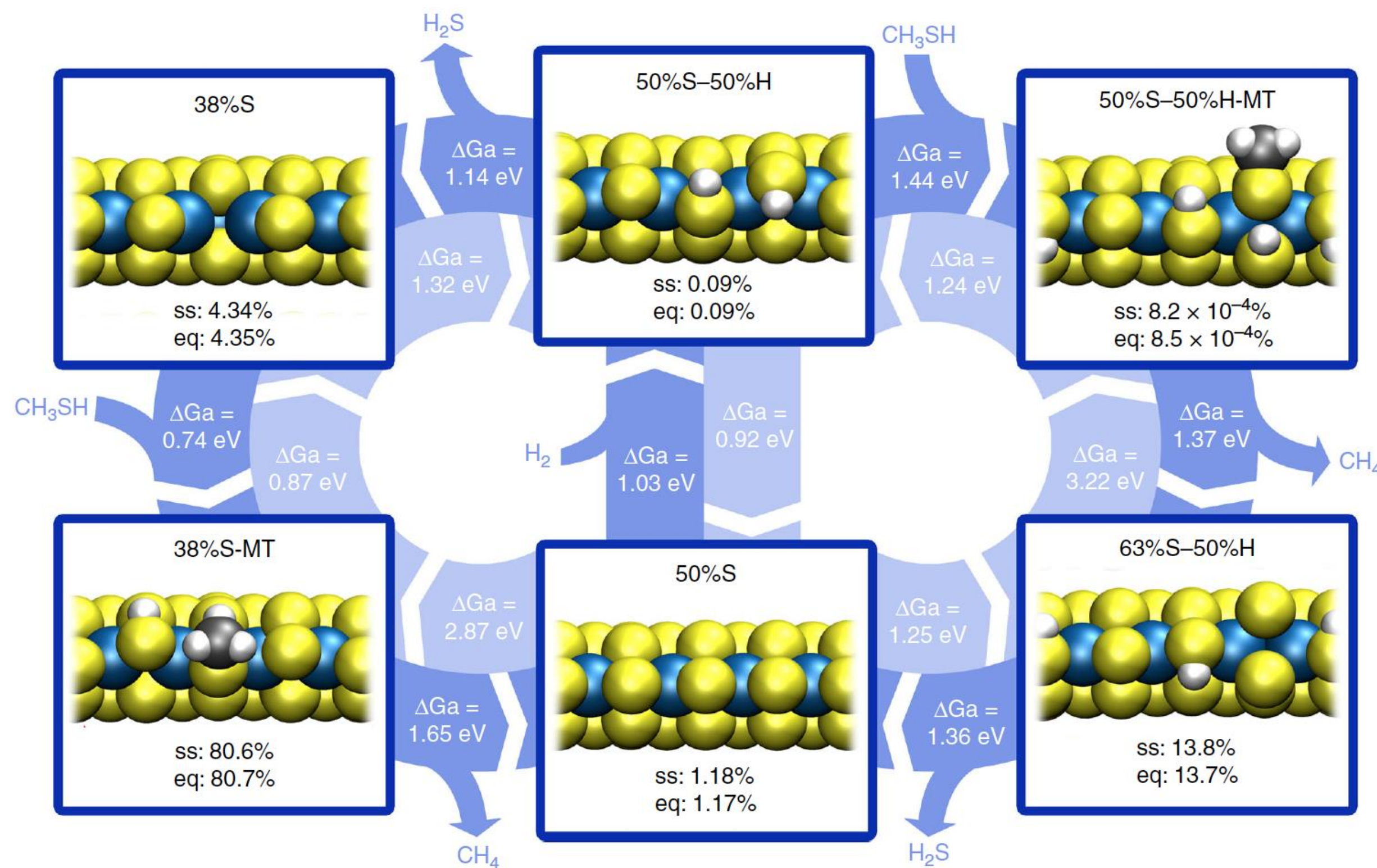
- Molar volume: > 0.99
- Heat Capacity: 0.96
- Glass Transition Temp. : 0.81

Average absolute error:

- Molar volume: 3.1 cm<sup>3</sup>/mol
- Heat Capacity: 13.8 J/(mol K)
- Glass Transition Temp. : 45 K

# Hydrodesulfurization MoS<sub>2</sub>: exp + calc

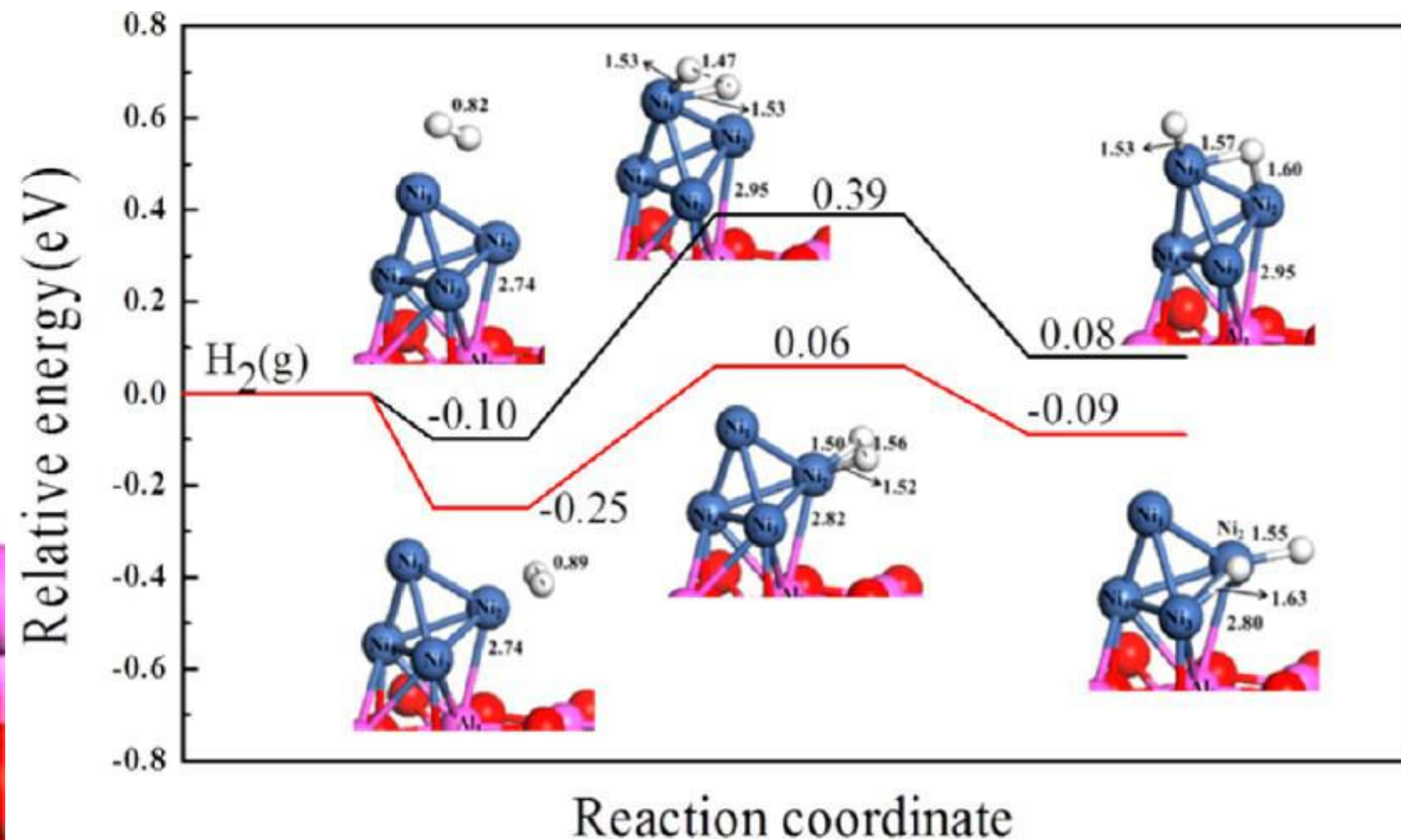
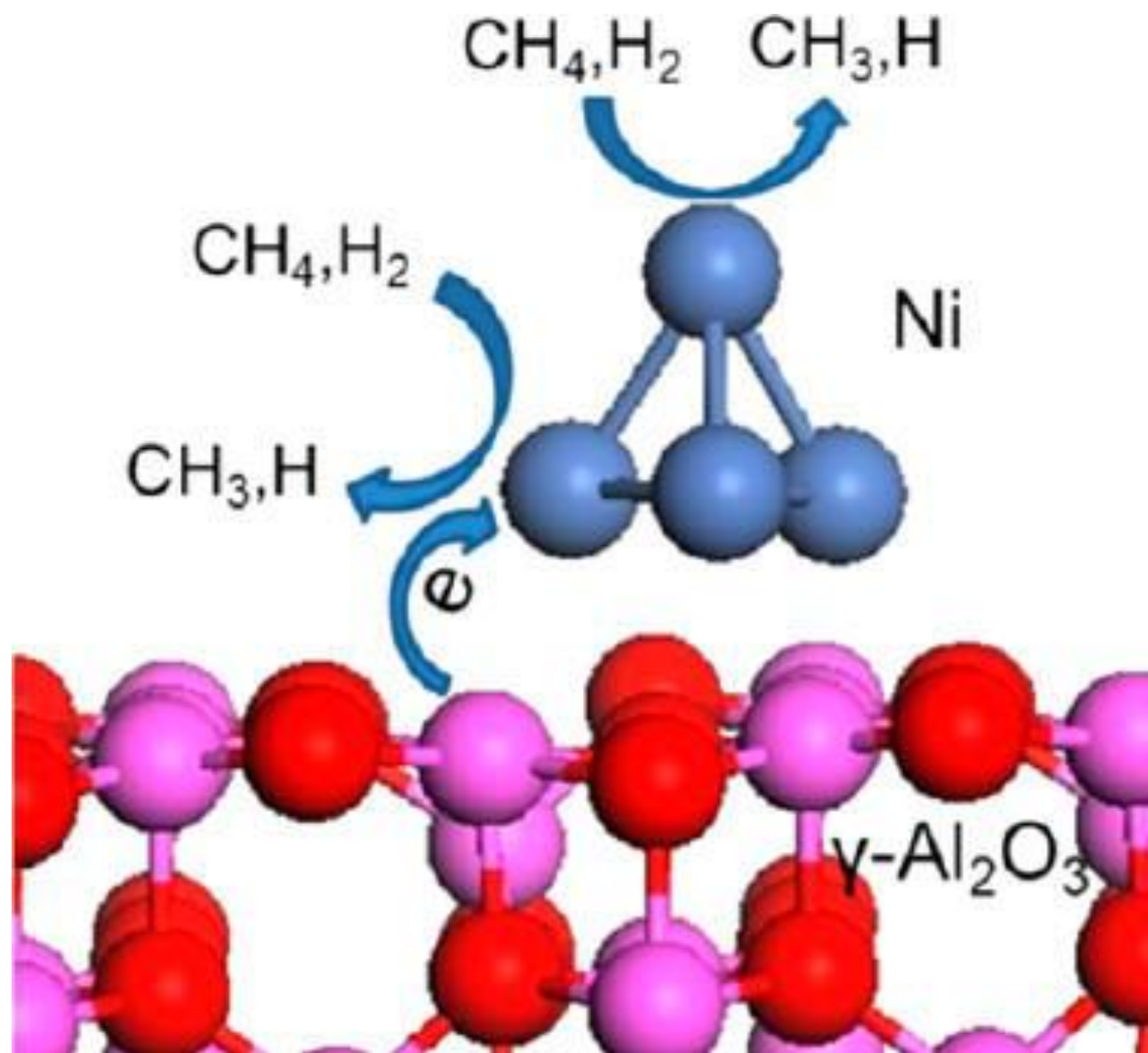
- STM with high p & T (different H<sub>2</sub>/H<sub>2</sub>S pressure)
- BAND calc. 2D=> phase diagram of edge -> compare with exp
- Reaction pathways (Albemarle: partial Hessian in BAND very useful!)



Nature Communications **10**, 2546 (2019)

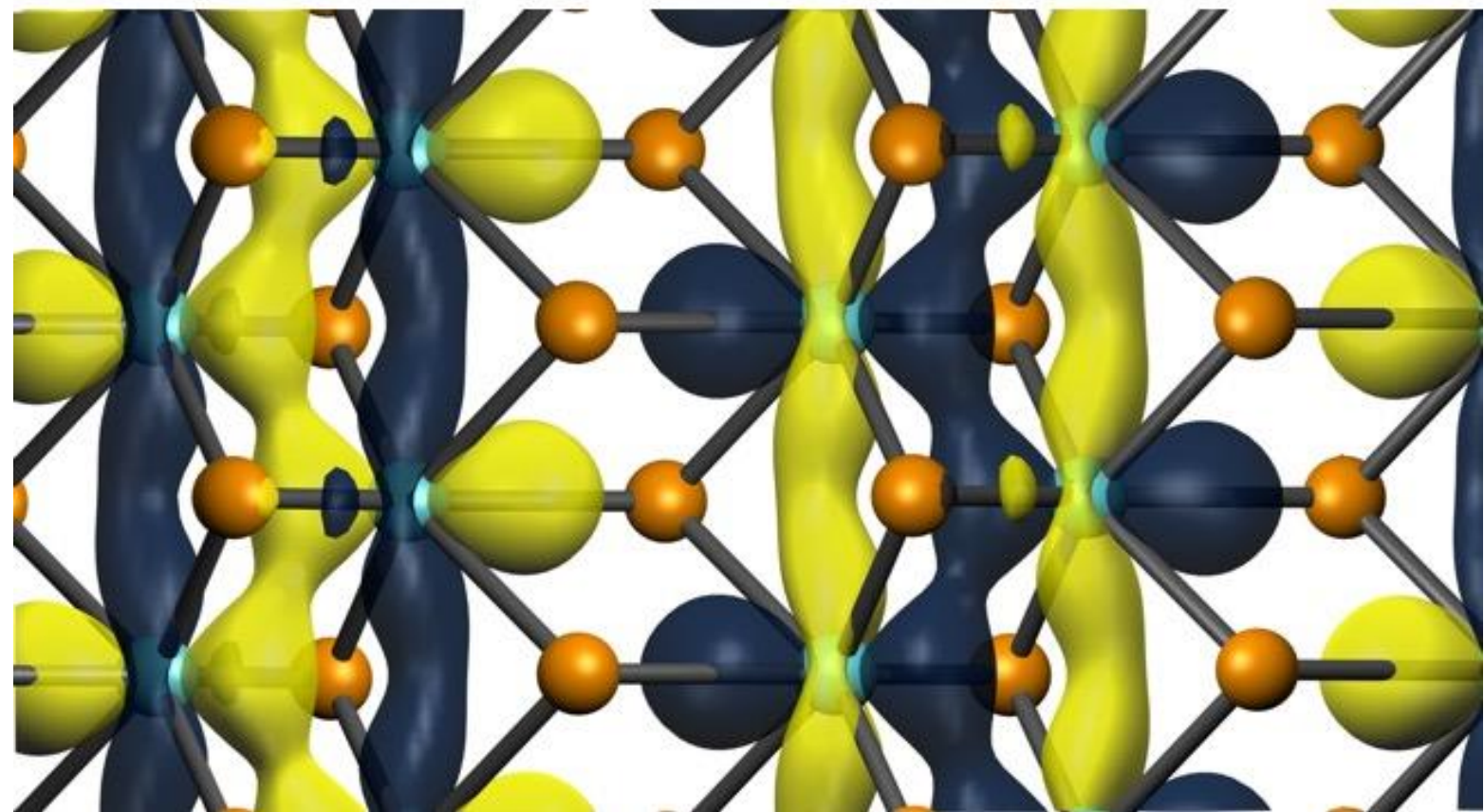
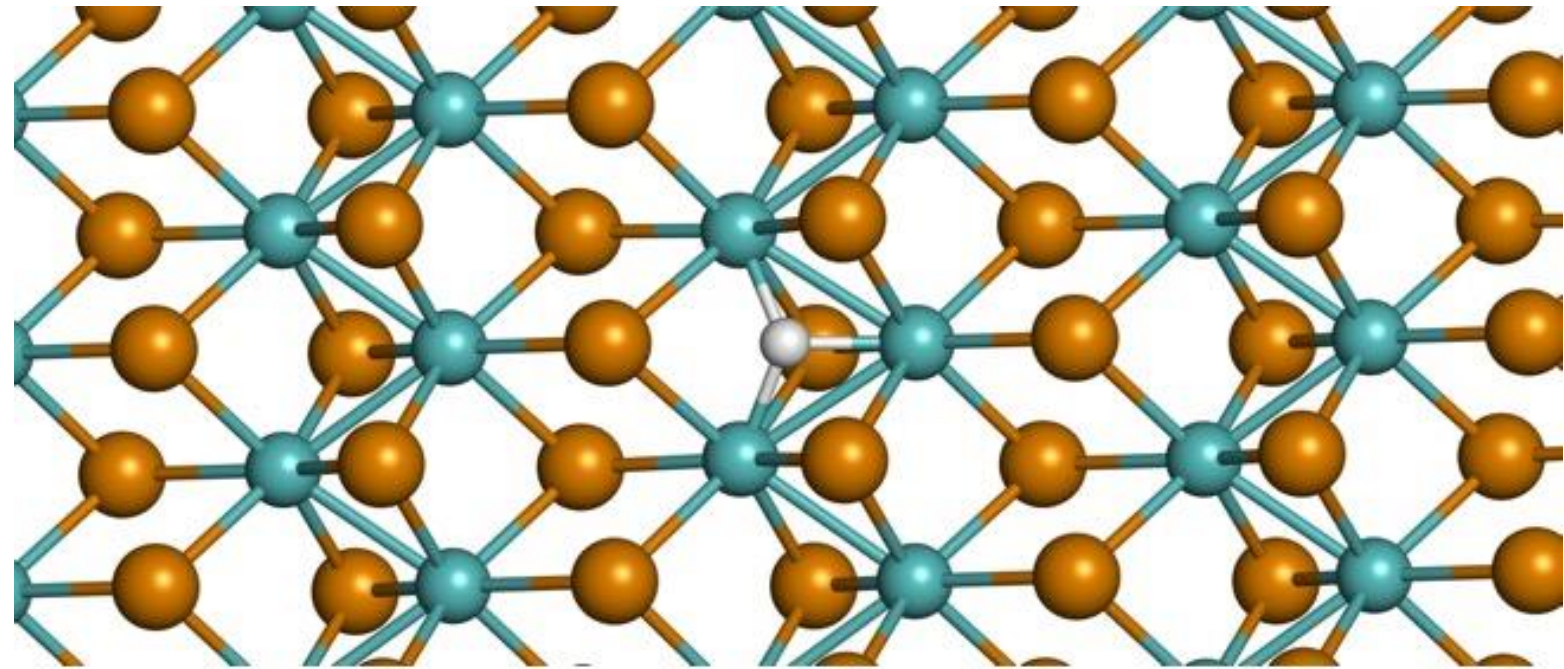
# CH<sub>4</sub> and H<sub>2</sub> dissociation on Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>

- Dissociation at TMO – NP interface preferred
- Aluminum acts as electron donor
- Need 2D to polarize surface



Li, Croiset, Ricardez-Sandoval, *J. Phys. Chem. C* **2013**, *117*, 16907

# Electrochemical activation $\text{MoTe}_2$ for $\text{H}_2$ evolution



- $1\text{T}'\text{-MoTe}_2$  catalyst improved at bias
- 2D surfaces + COSMO solvation
- H adsorbs on Te at cathodic bias

The rapid electrochemical activation of  $\text{MoTe}_2$  for the hydrogen evolution reaction [Nature Comm. 2019](#)

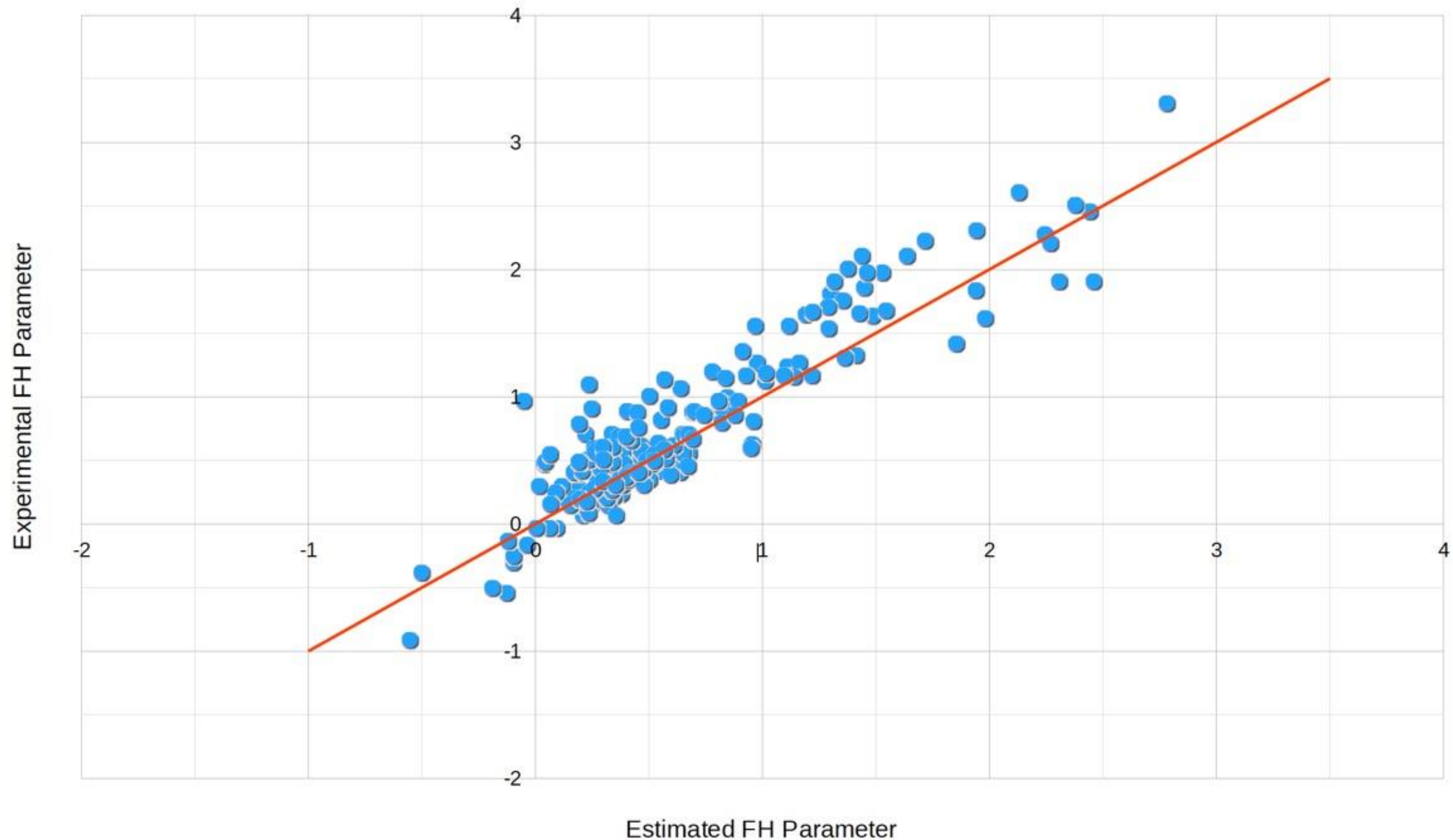
# Flory-Huggins from COSMO-RS

$N_{p/s}$  – Number of moles of polymer / solvent

$\phi_{p/s}$  – Volume fraction of polymer / solvent

$\chi_{ps}$  – Flory – Huggins Parameter

$$\frac{\Delta G_{mix}}{RT} = N_p \ln(\phi_p) + N_s \ln(\phi_s) + N_s \phi_p \chi_{ps}$$

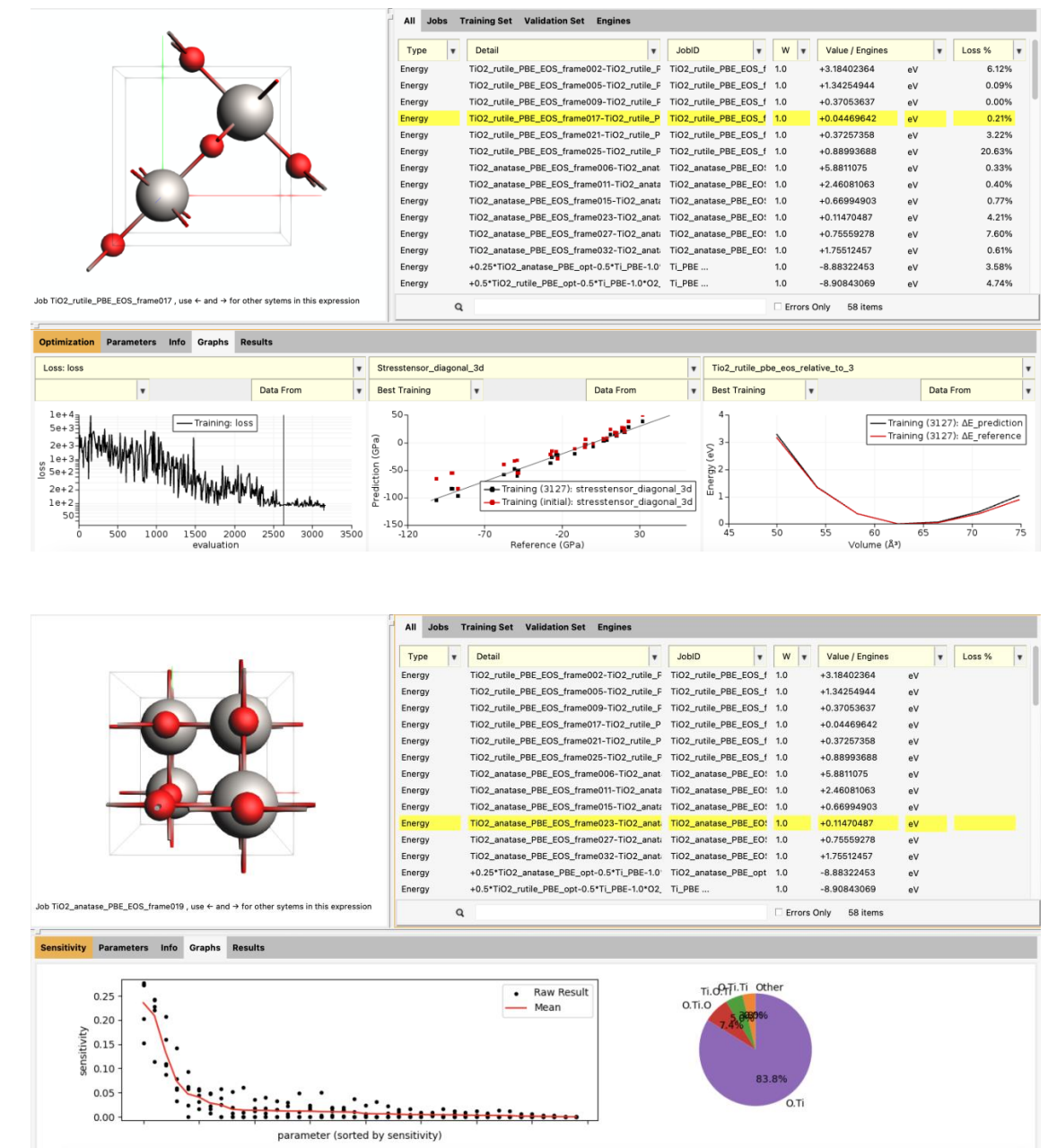


# Training FF with ParAMS

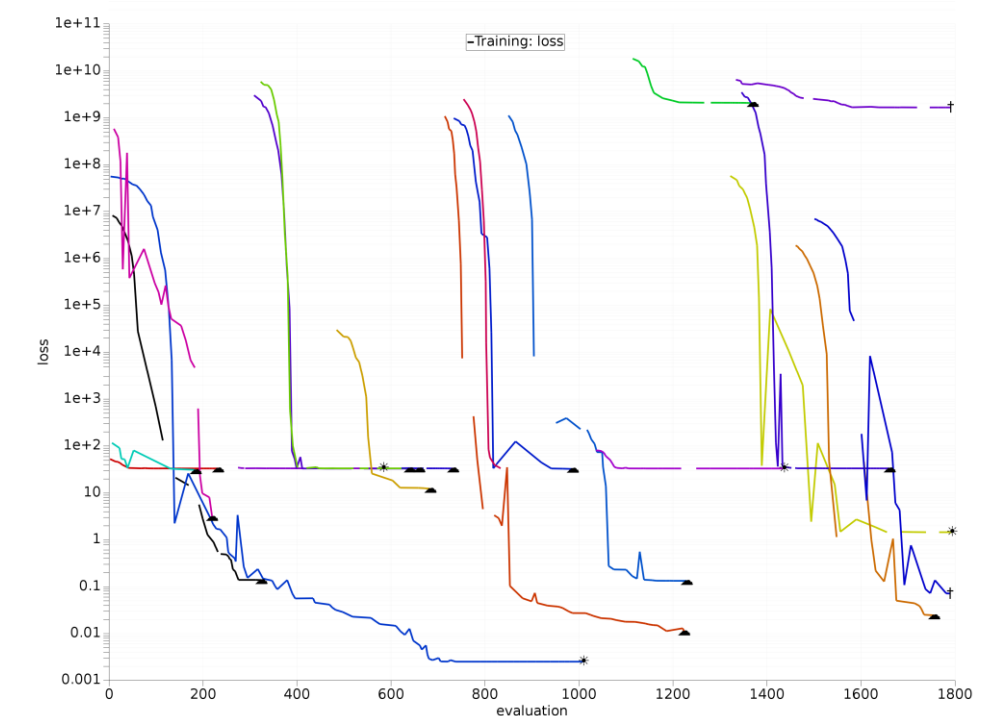
## Making parameters optimization simple

What can you do with ParAMS?

- Import, build and visualize training data
- Use data from AMS, VASP, QE, and experiments
- Tune parameters to describe energies, forces geometries, stress tensor, charges, bandgaps, etc.
- Use validation sets to prevent overfitting
- Submit multiple optimization at the same time to explore parameters space
- Explore the sensitivity of the parameters to select the most important subset to optimize
- Train DFTB, ReaxFF, Force field, and ... MLP



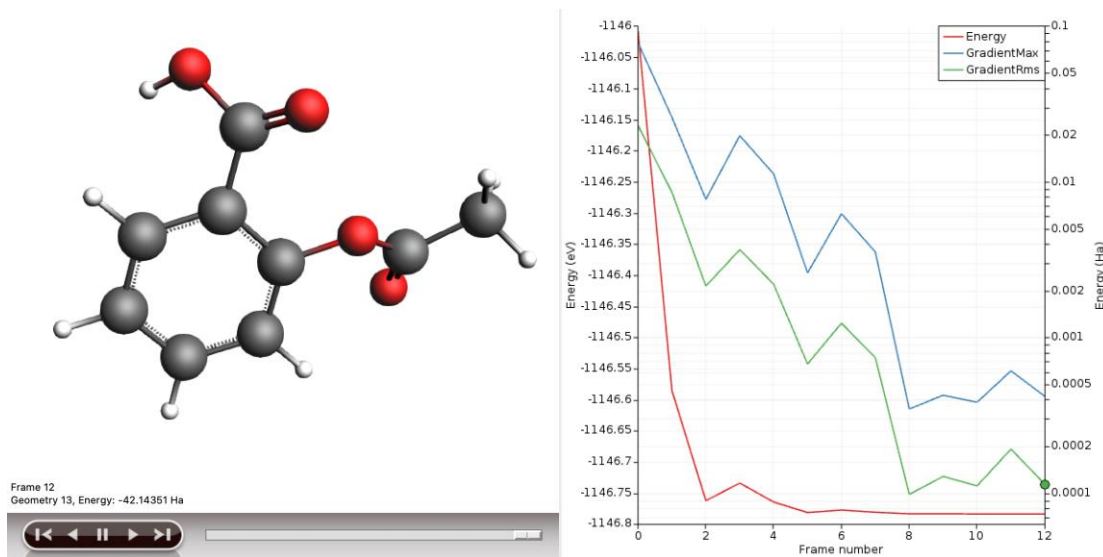
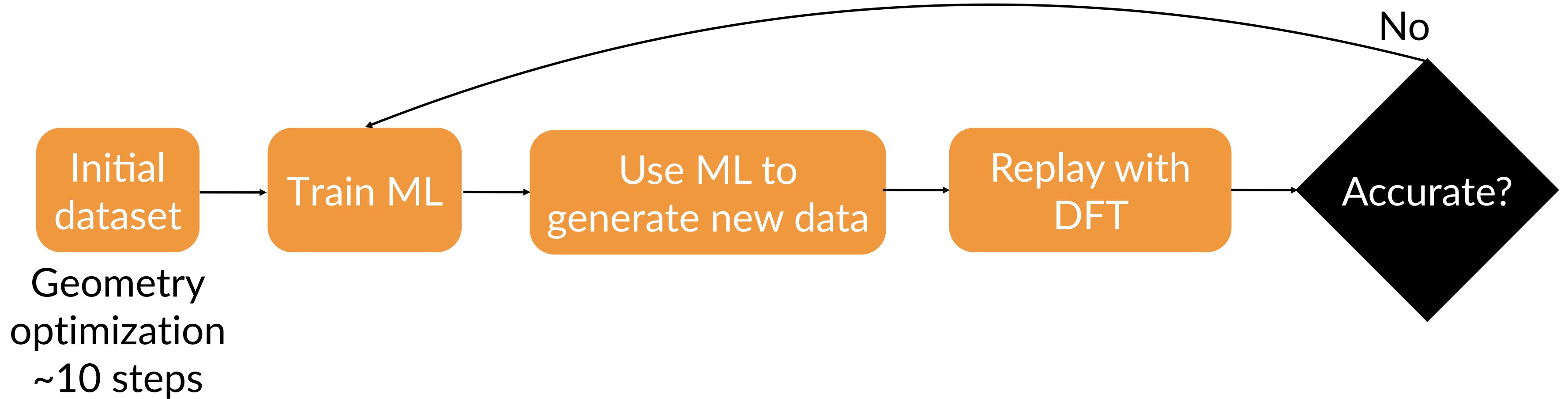
*J. Chem. Inf. Model.* 2021, 61, 8, 3737–3743  
*J. Chem. Theory Comput.* 2023, 19, 9, 2557–2573  
<https://www.scm.com/product/params>





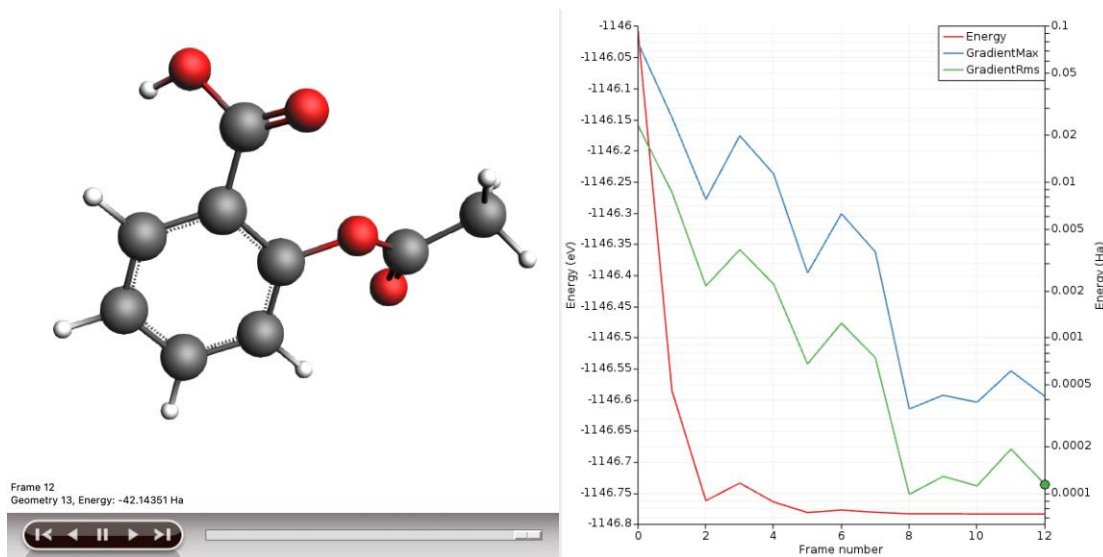
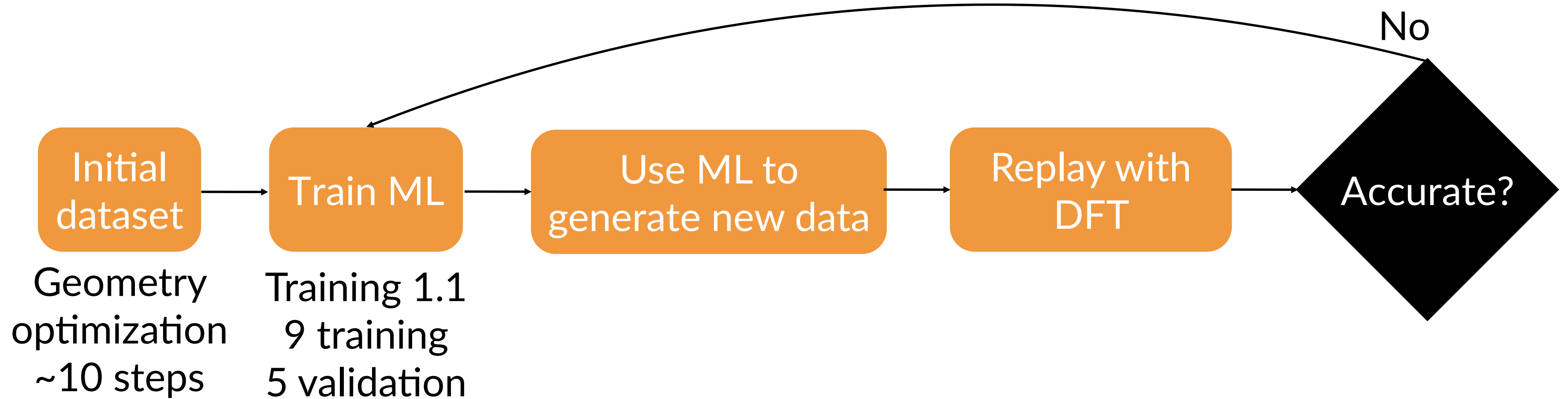
# Training MLP with ParAMS

## Active learning workflow



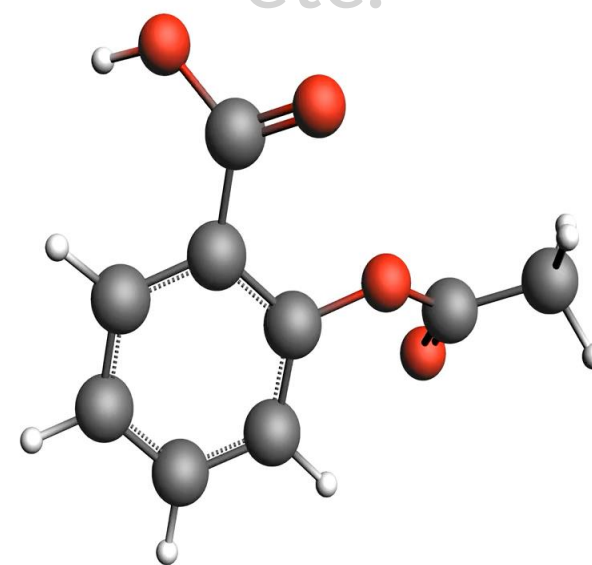
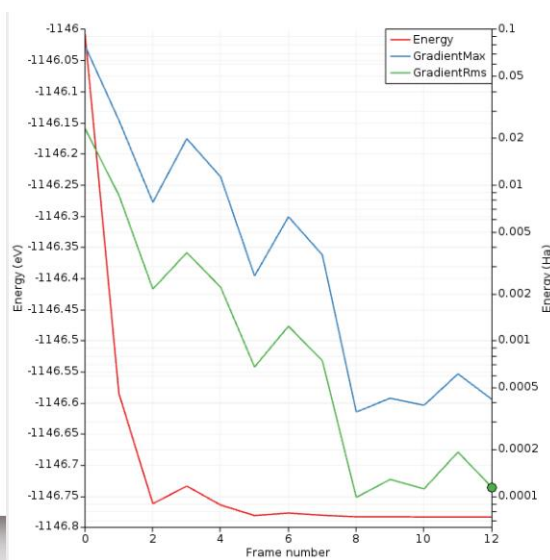
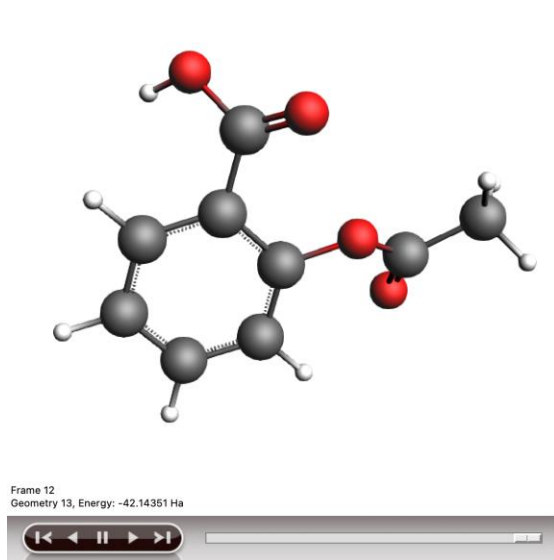
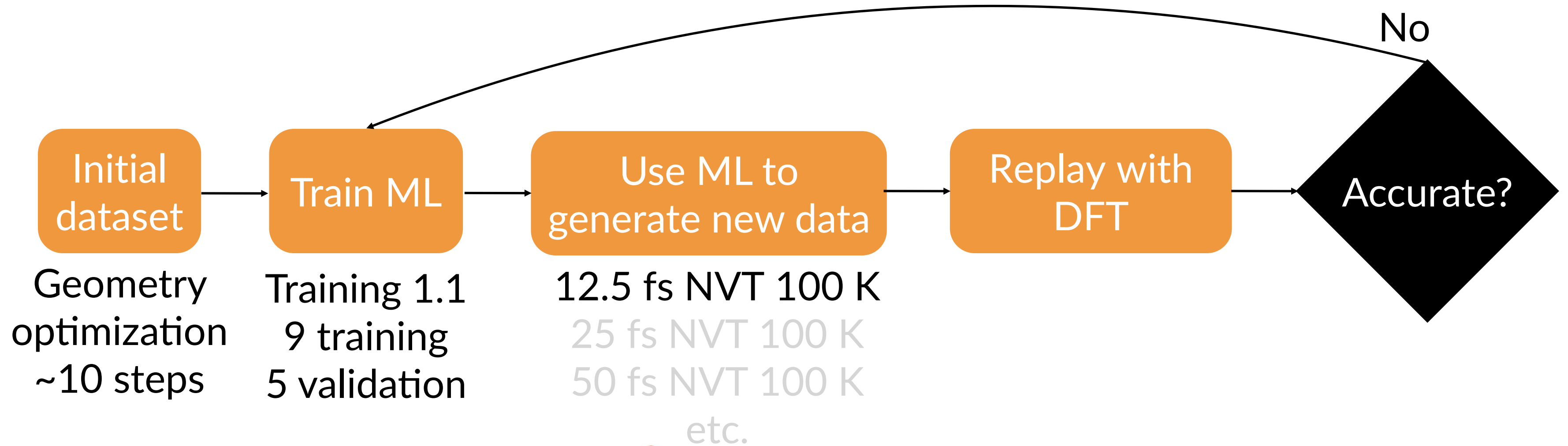
# Training MLP with ParAMS

## Active learning workflow



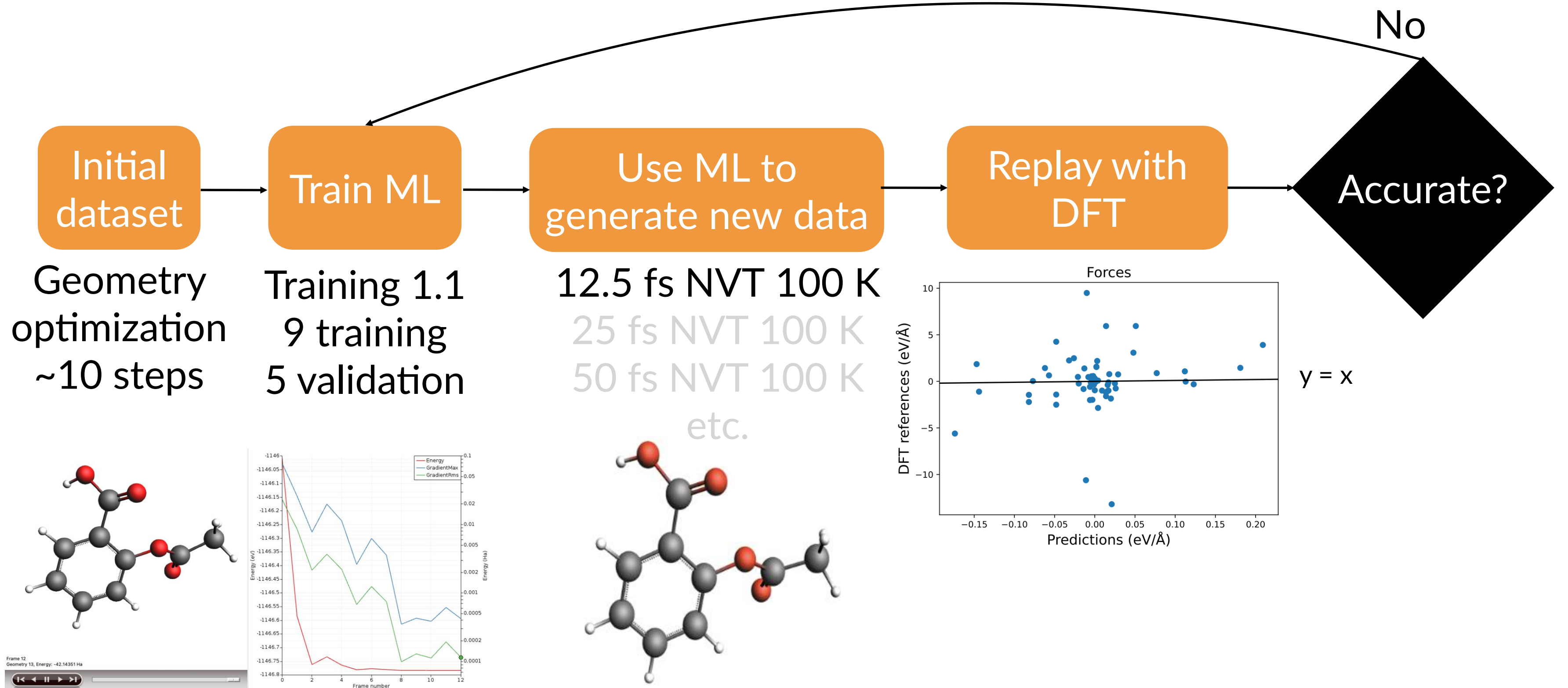
# Training MLP with ParAMS

## Active learning workflow



# Training MLP with ParAMS

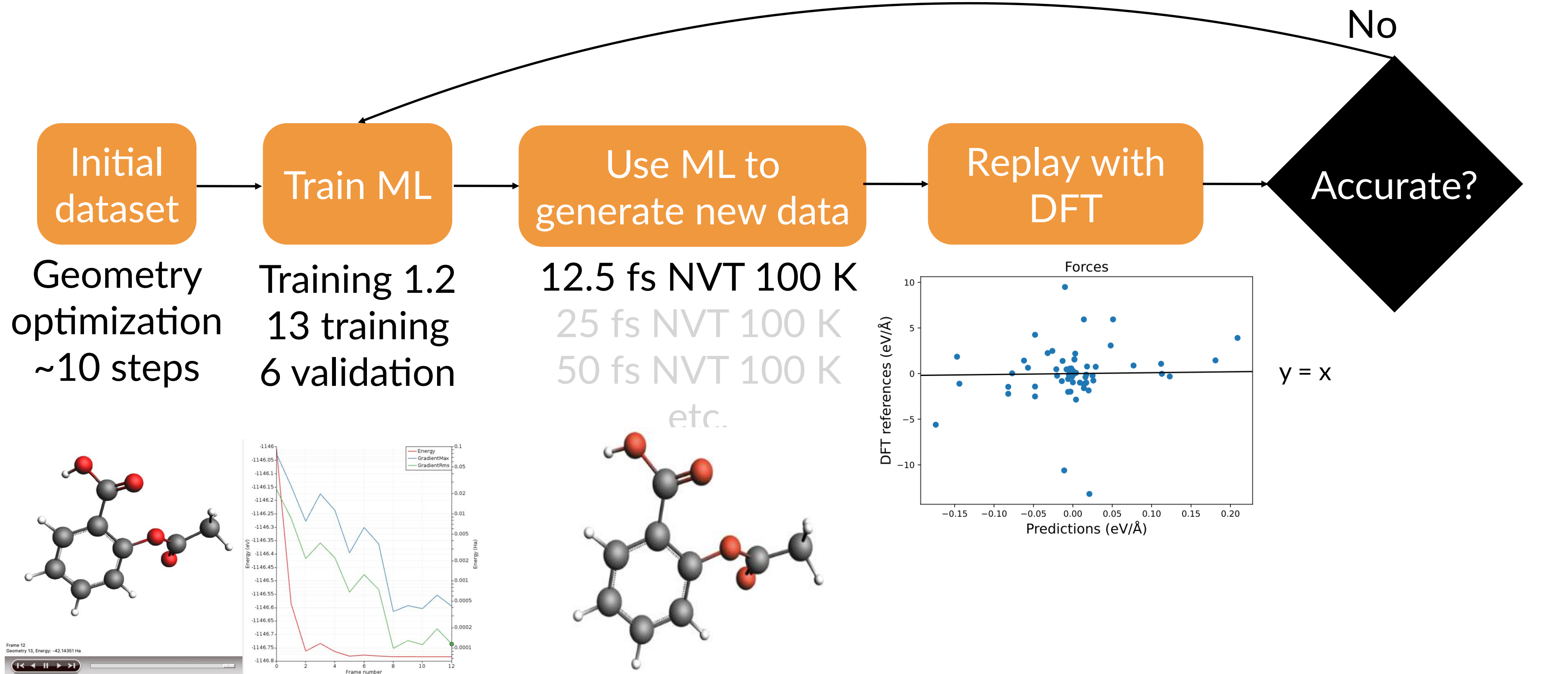
## Active learning workflow



$$\text{MAE} = 1.83 \text{ eV/\AA}$$

# Training MLP with ParAMS

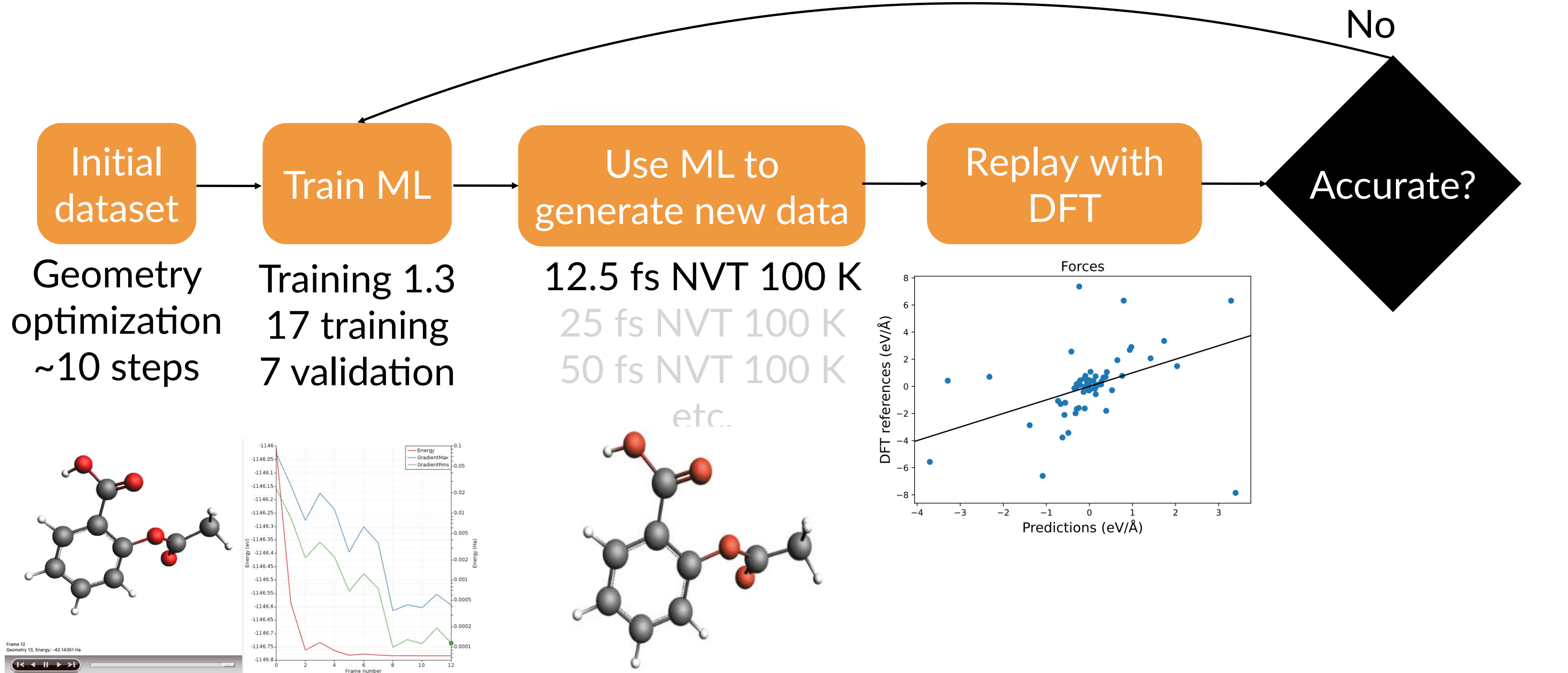
## Active learning workflow



$$\text{MAE} = 1.83 \text{ eV/\AA}$$

# Training MLP with ParAMS

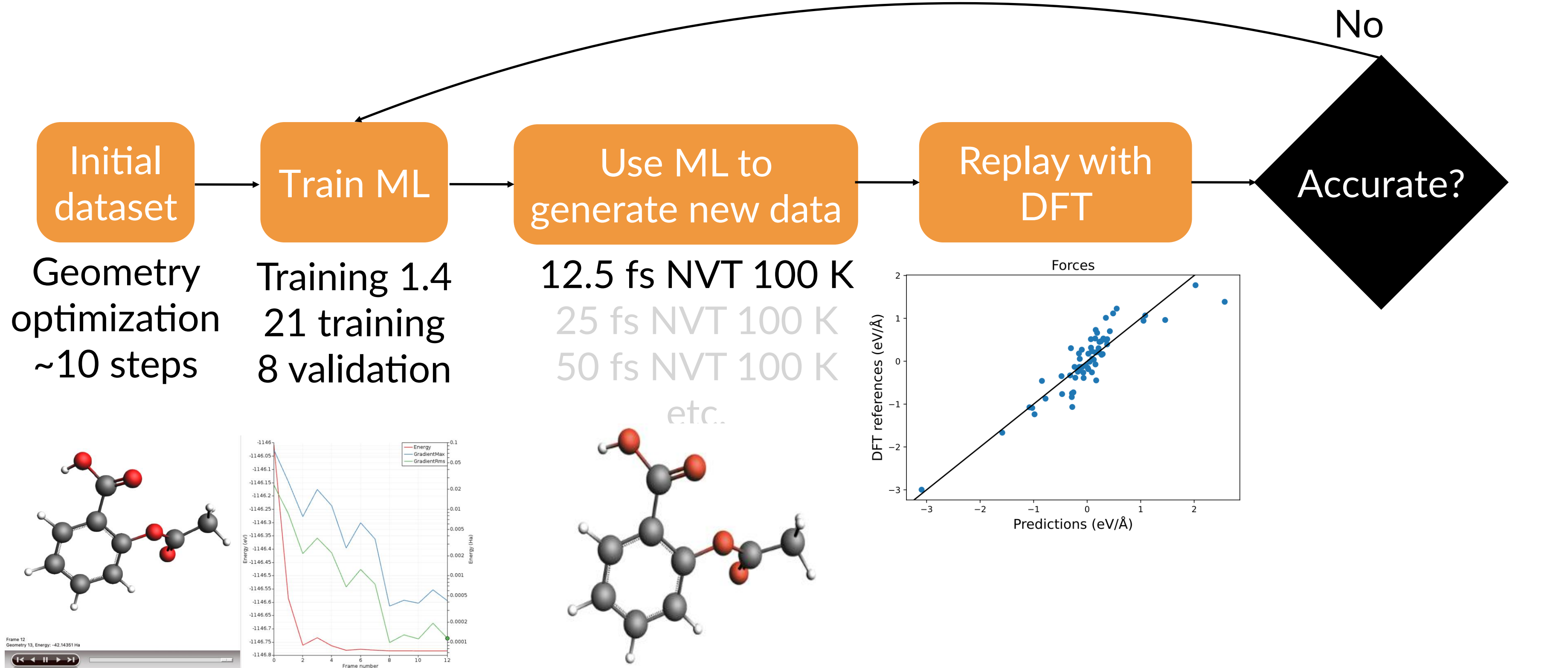
## Active learning workflow



$$\text{MAE} = 1.32 \text{ eV/\AA}$$

# Training MLP with ParAMS

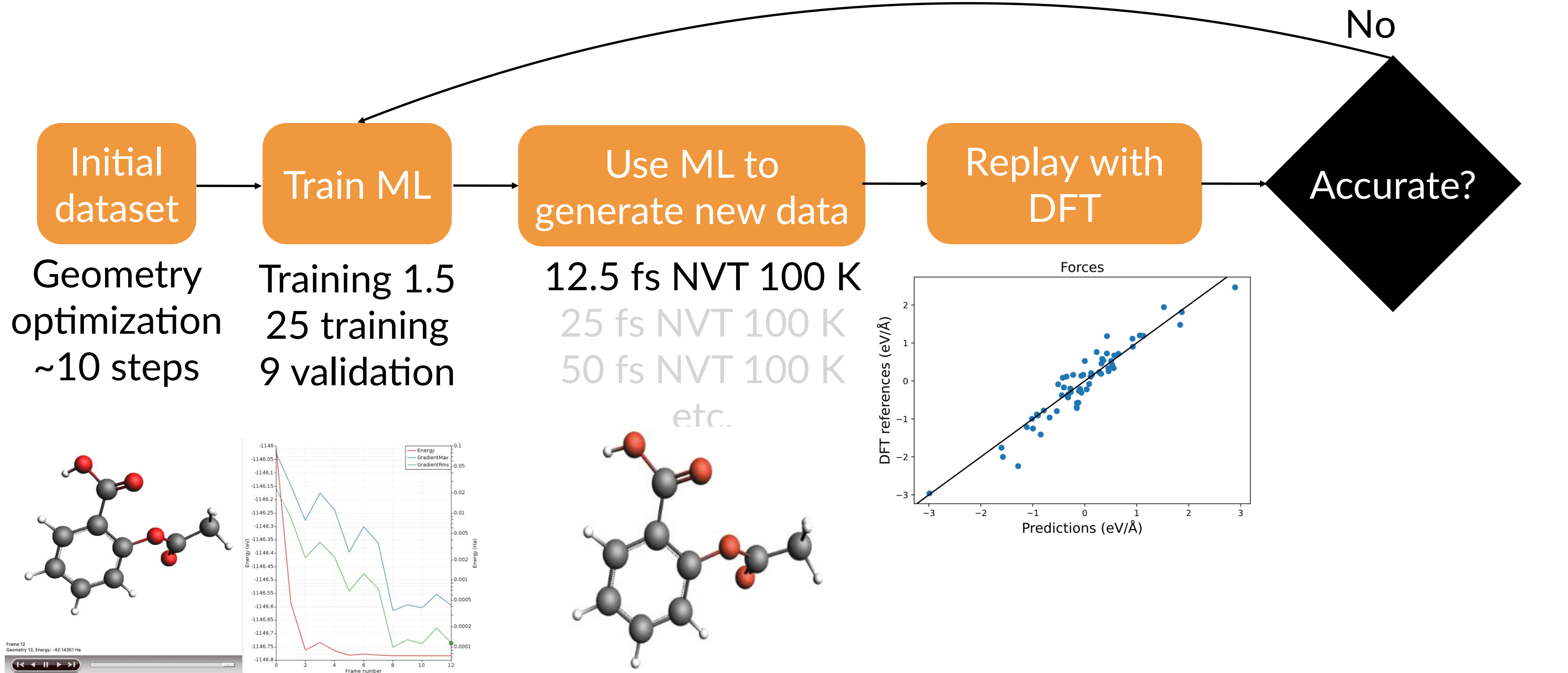
## Active learning workflow



$$\text{MAE} = 0.26 \text{ eV/\AA}$$

# Training MLP with ParAMS

## Active learning workflow

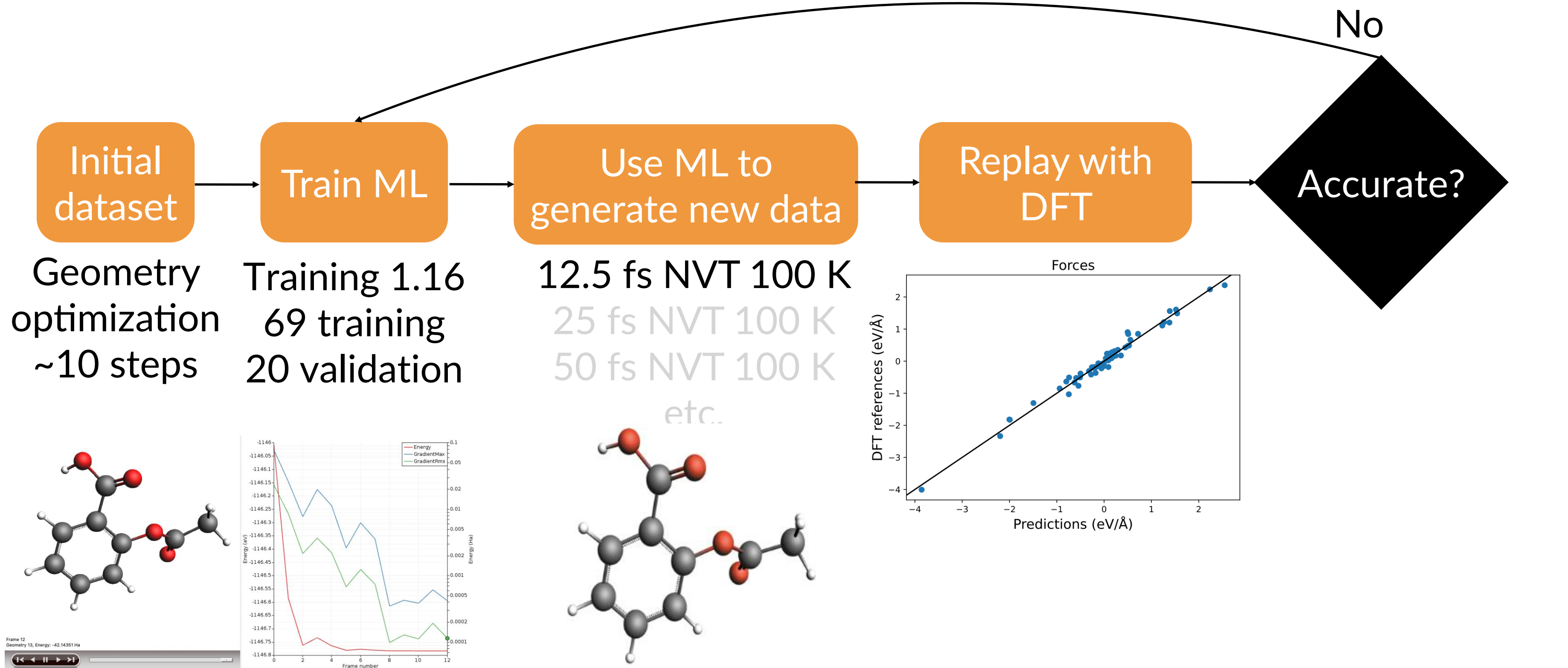


$$\text{MAE} = 0.25 \text{ eV/\AA}$$



# Training MLP with ParAMS

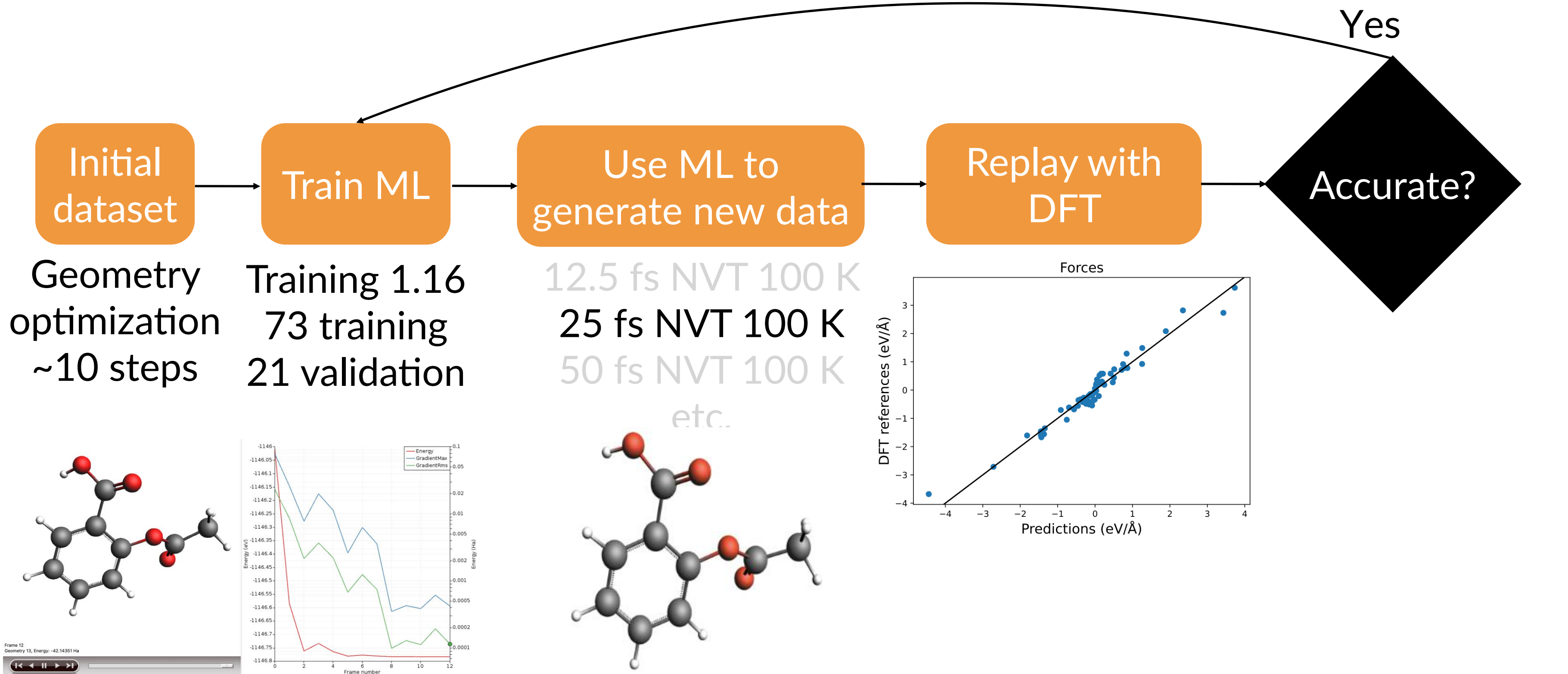
## Active learning workflow



$$\text{MAE} = 0.10 \text{ eV/\AA}$$

# Training MLP with ParAMS

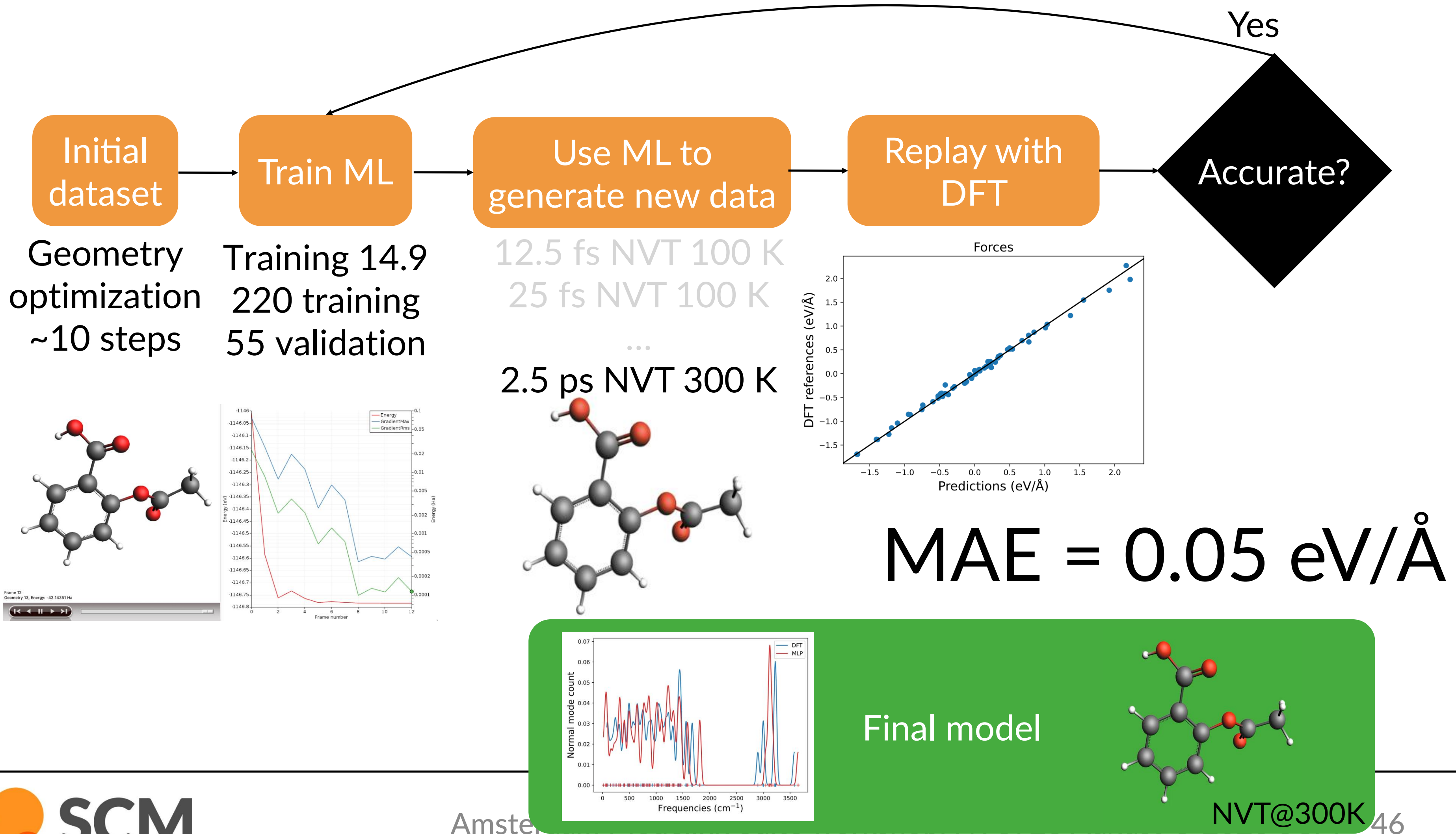
## Active learning workflow



$$\text{MAE} = 0.18 \text{ eV/\AA}$$

# Training MLP with ParAMS

## Active learning workflow



# Training MLP with ParAMS

## Summary

- Compute the initial training set of reference calculations
- Select the ML model to optimize (nequip, flare)
- Define the series of test tasks (default/custom)
- Run an interactively follow the training procedure
  - MAE plot
  - Parity plot
  - Dynamics